

Chemoselectivity: Far to be a kid's game

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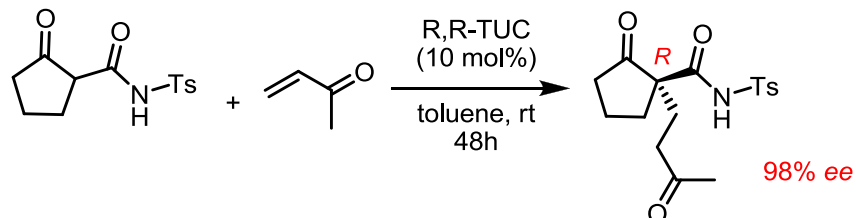
- Selectivities
- Importance of chemoselectivity
- Transformations with saturated Carbon-Heteroatom bonds
- Transformations with unsaturated Carbon-Heteroatom bonds
- Metal-promoted Carbon-Carbon bonds
- Conclusion

Selectivities

Stereoselectivity: The preferential formation of one stereoisomer over another.

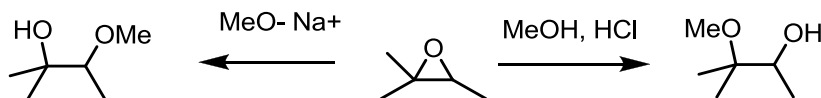
Enantio and diastereoselectivity

Q. : How it will react?



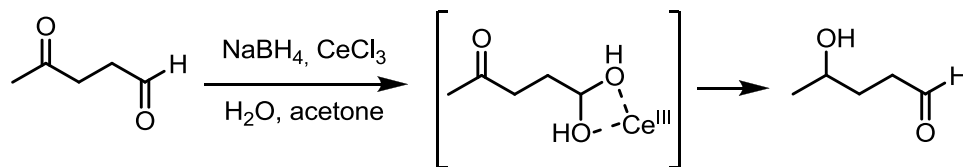
Regioselectivity: directional preference of the breaking or making of a chemical bond.

Q. : Where it will react?



Chemoselectivity: The preferential reaction of a chemical reagent with one of two or more different functional groups.

Q. : Which functional group will react?



Importance of Chemoselectivity

In Total Synthesis:

➔ Obstacle to complex molecules synthesis

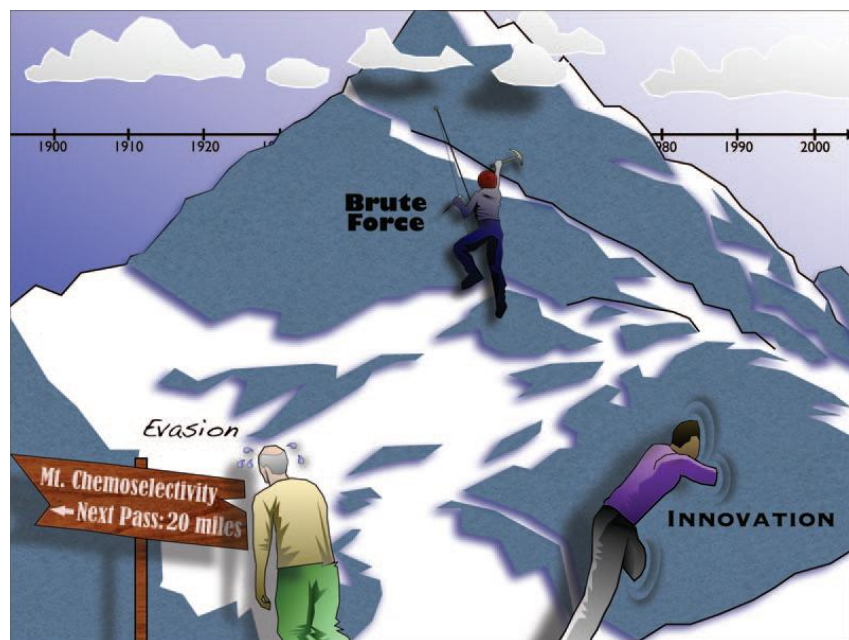
Nature vs Man

Tools to compare or predict reactivity:

Electronic/steric effect,
redox potential, pKa values,
hard/ soft acid/base

Advantage:

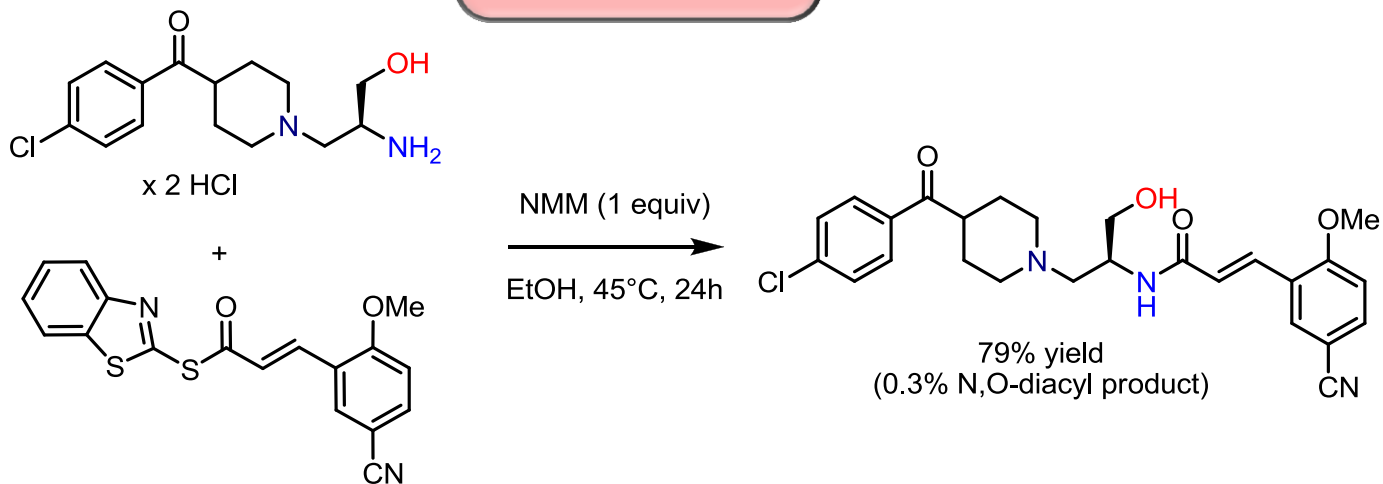
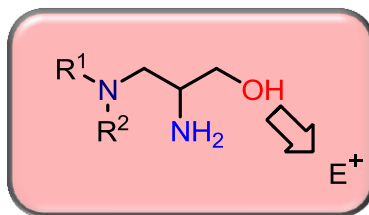
- Minimal reliance on protecting groups
- Atom and step economy



Saturated Carbon-Heteroatom

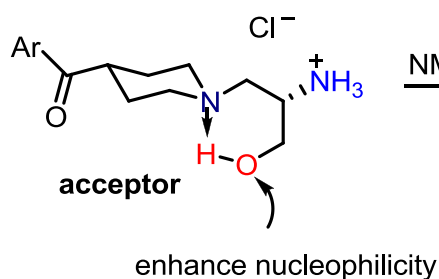
Amine vs Hydroxy groups

Proton-Mediated chemoselectivity



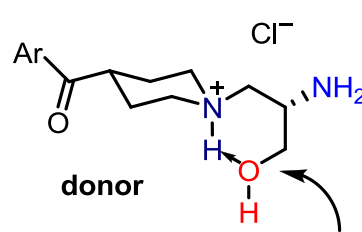
Mercapto benzothiazolyl thioester

Starting material



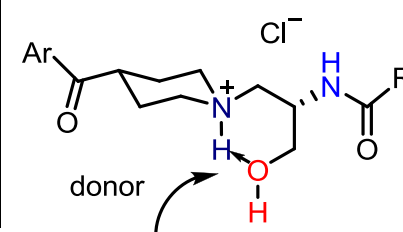
enhance nucleophilicity

NMM (1 equiv)



donor

Product

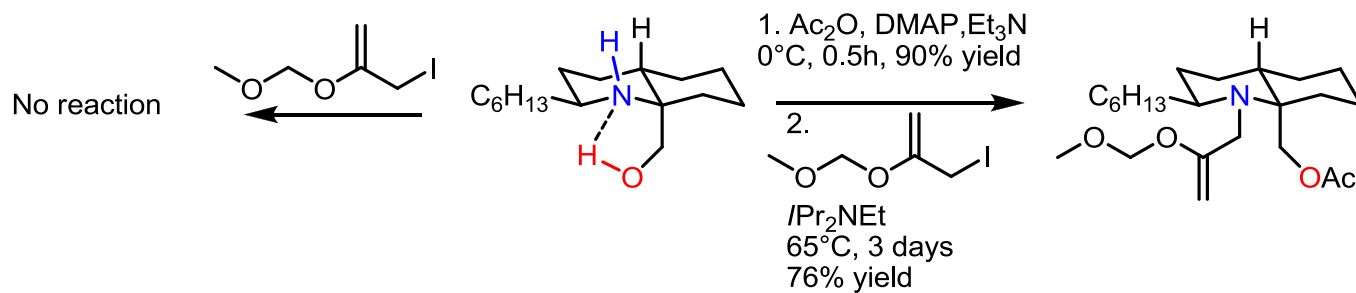
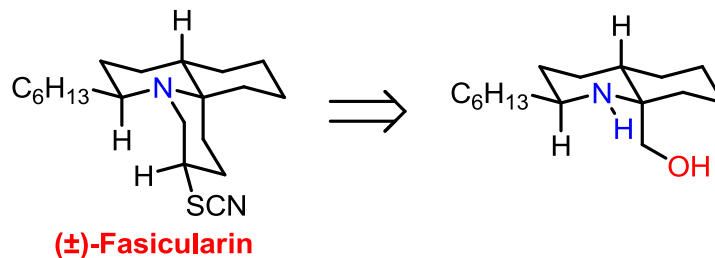
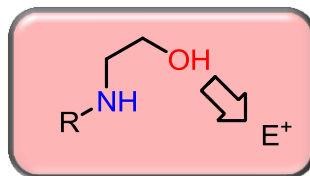


reduced nucleophilicity

Saturated Carbon-Heteroatom

Amine vs Hydroxy groups

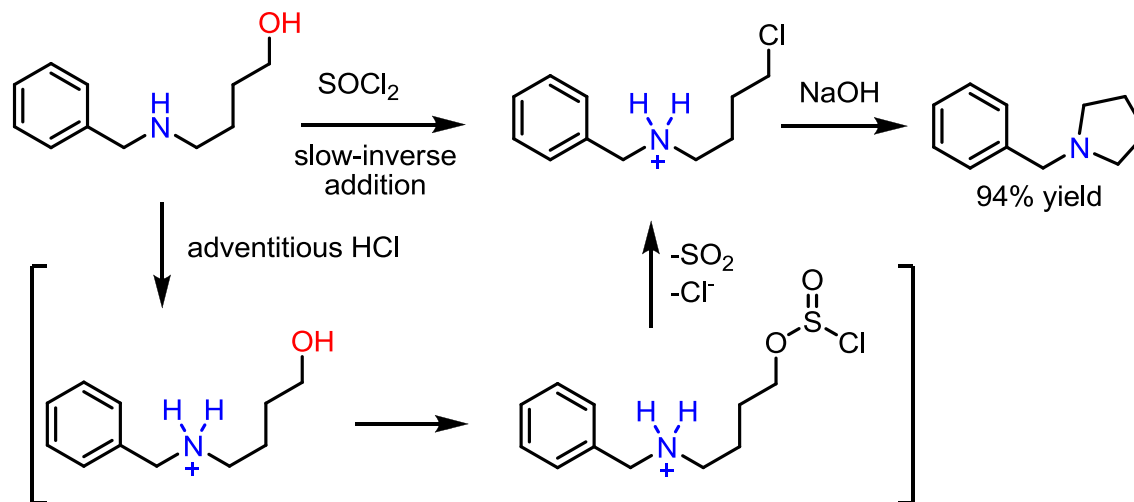
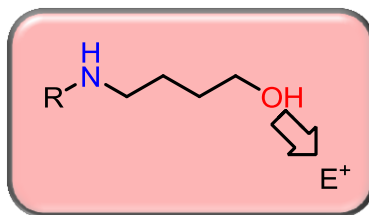
Proton-Mediated chemoselectivity



Saturated Carbon-Heteroatom

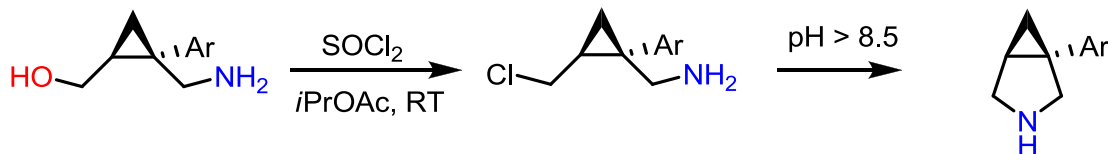
Amine vs Hydroxy groups

Proton-Mediated chemoselectivity



In situ protection of the amine by a proton.

Application:



(+)-Bicifadine, 65% yield (Ar = *p*-MeC₆H₄)
DOV2194, 57% yield (Ar = 3,4-Cl₂C₆H₃)

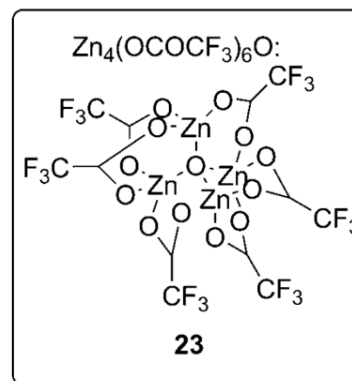
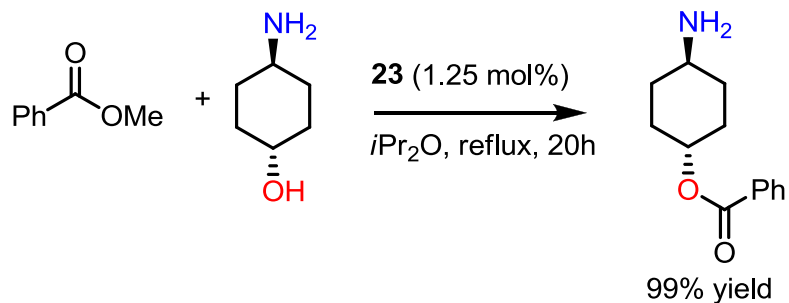
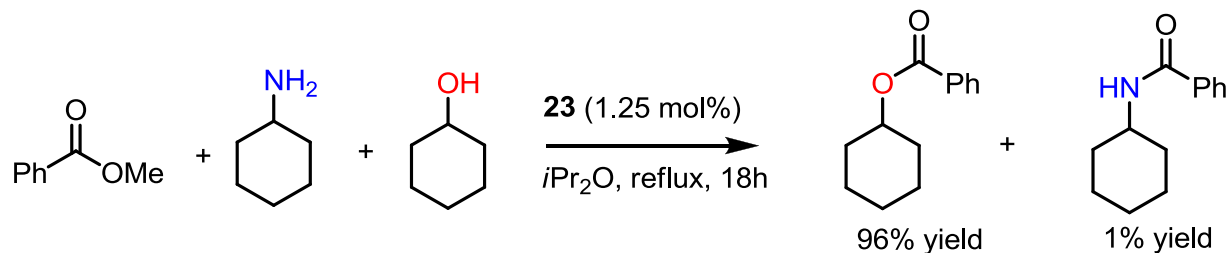
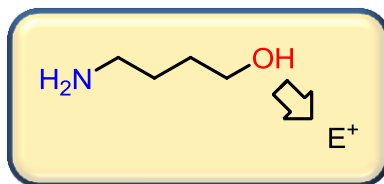
F. Xu, B. Simmons, R. A. Reamer, E. Corley, J. Murry, D. Tschaen, *J. Org. Chem.* **2008**, *73*, 312 – 315.

F. Xu, J. A. Murry, B. Simmons, E. Corley, K. Fitch, S. Karady, D. Tschaen, *Org. Lett.* **2006**, *8*, 3885 – 3888.

Saturated Carbon-Heteroatom

Amine vs Hydroxy groups

Metal-Mediated Chemoselectivity



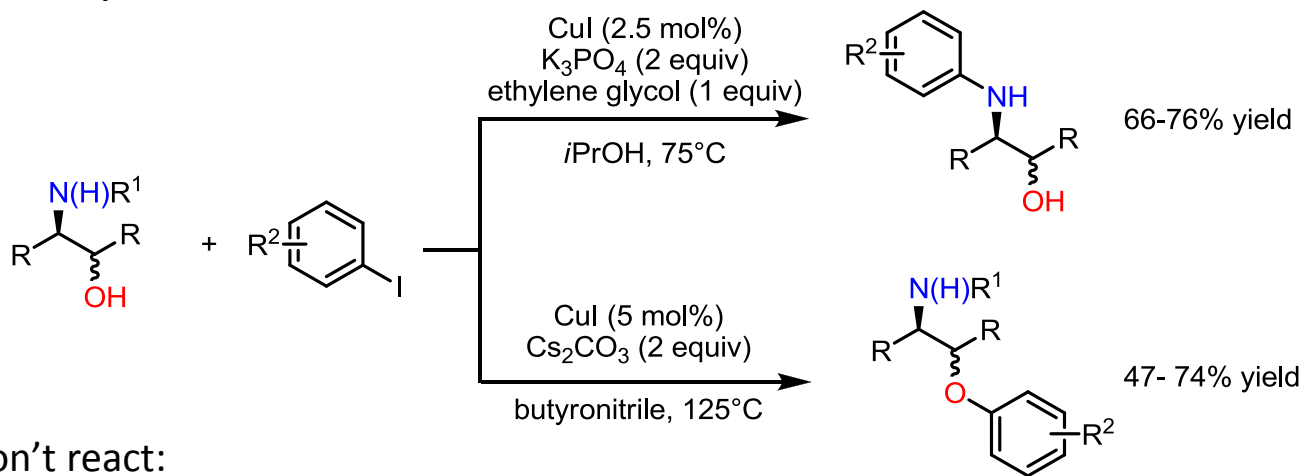
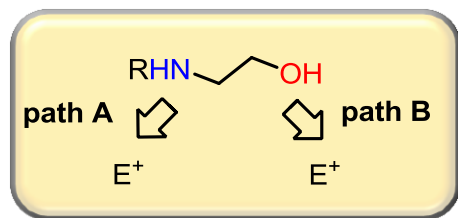
Chemoselectivity controlled by the high oxophilicity of the Zn cluster.

enzyme-like chemoselective reaction

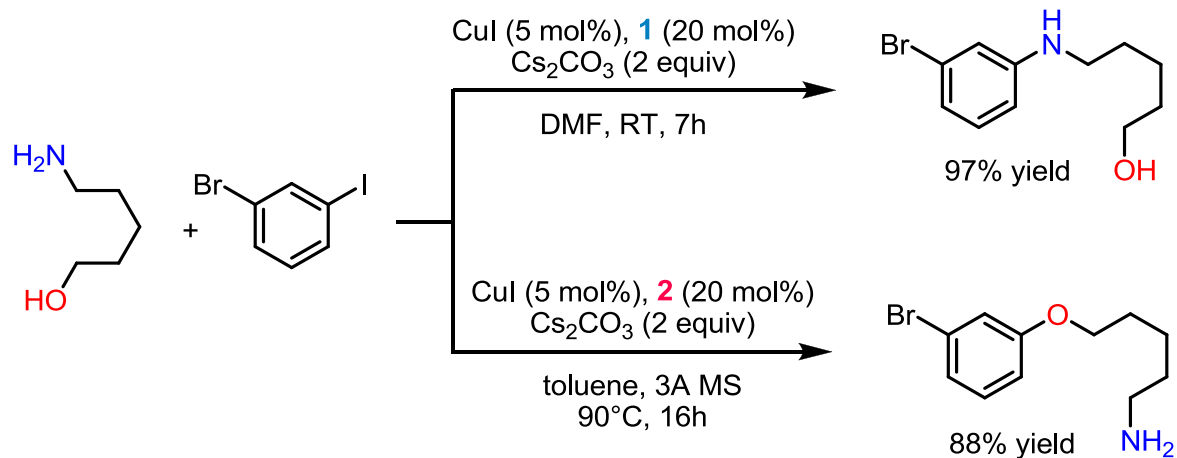
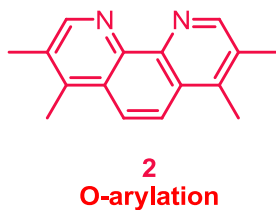
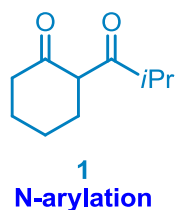
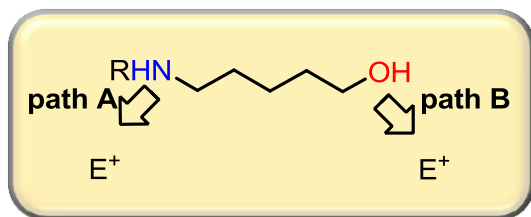
Saturated Carbon-Heteroatom

Amine vs Hydroxy groups

Metal-Mediated Chemoselectivity



Simple alcohols or amines don't react:
chelating ability of β -amino alcohols for reactivity



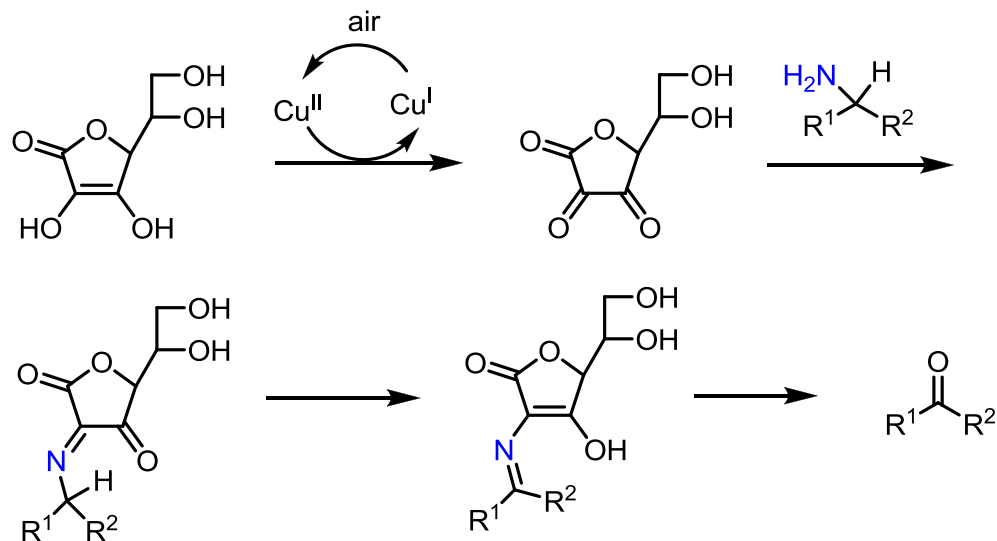
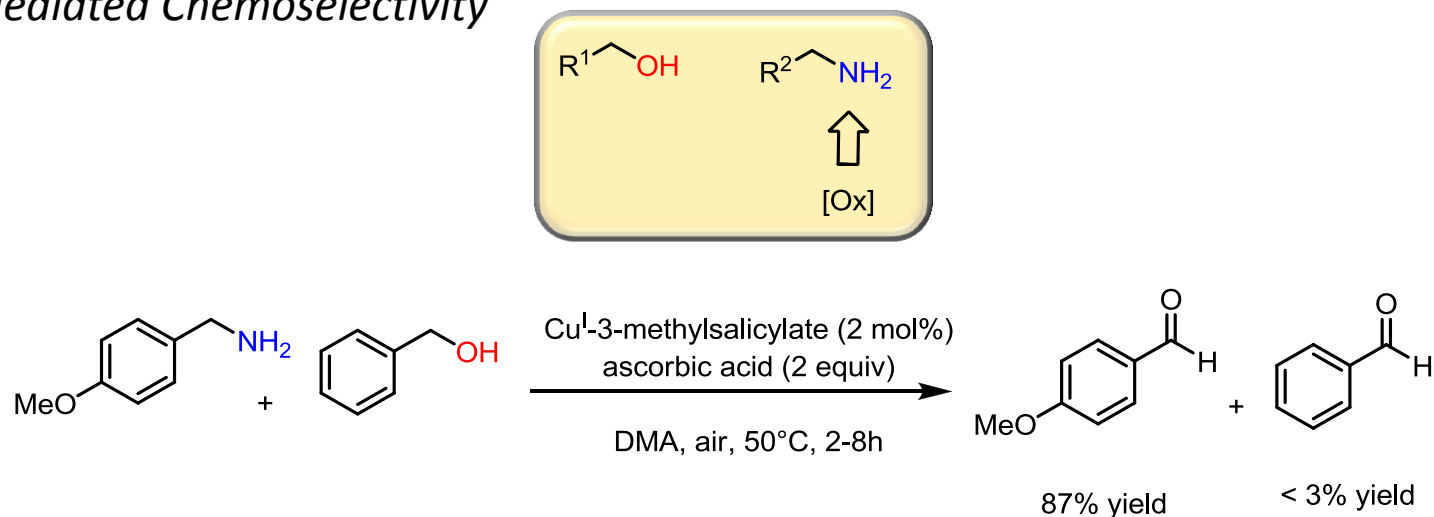
G. E. Job, S. L. Buchwald, *Org. Lett.* **2002**, 4, 3703 – 3706.

A. Shafir, P. A. Lichtor, S. L. Buchwald, *J. Am. Chem. Soc.* **2007**, 129, 3490 – 3491.

Saturated Carbon-Heteroatom

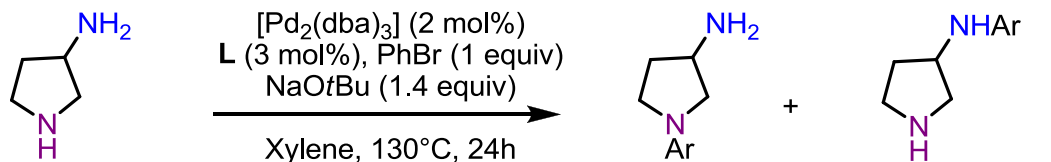
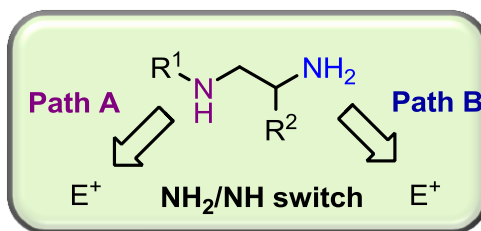
Amine vs Hydroxy groups

Metal-Mediated Chemoselectivity



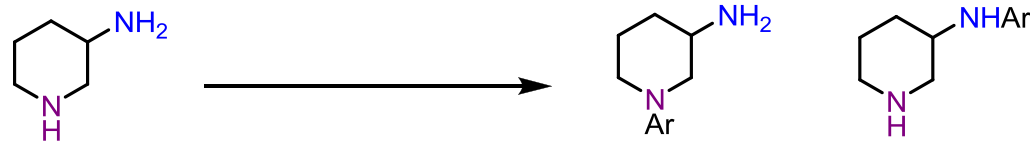
Saturated Carbon-Heteroatom Oligoamines

Metal-Mediated Chemoselectivity



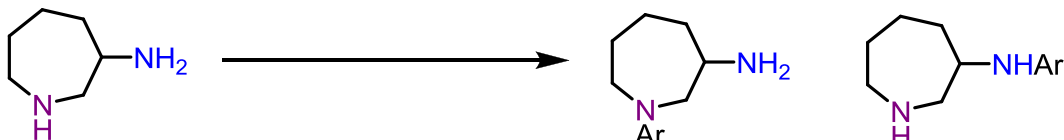
L1 60% yield
L2 85% yield

L1 0% yield
L2 0% yield



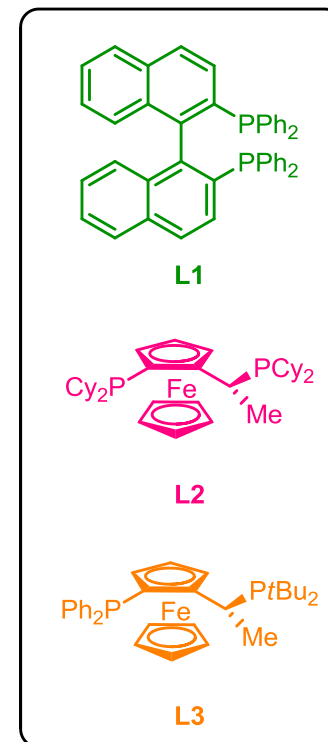
L1 43% yield
L3 16% yield

L1 4% yield
L3 33% yield



L1 18% yield
L3 8% yield

L1 39% yield
L3 32% yield



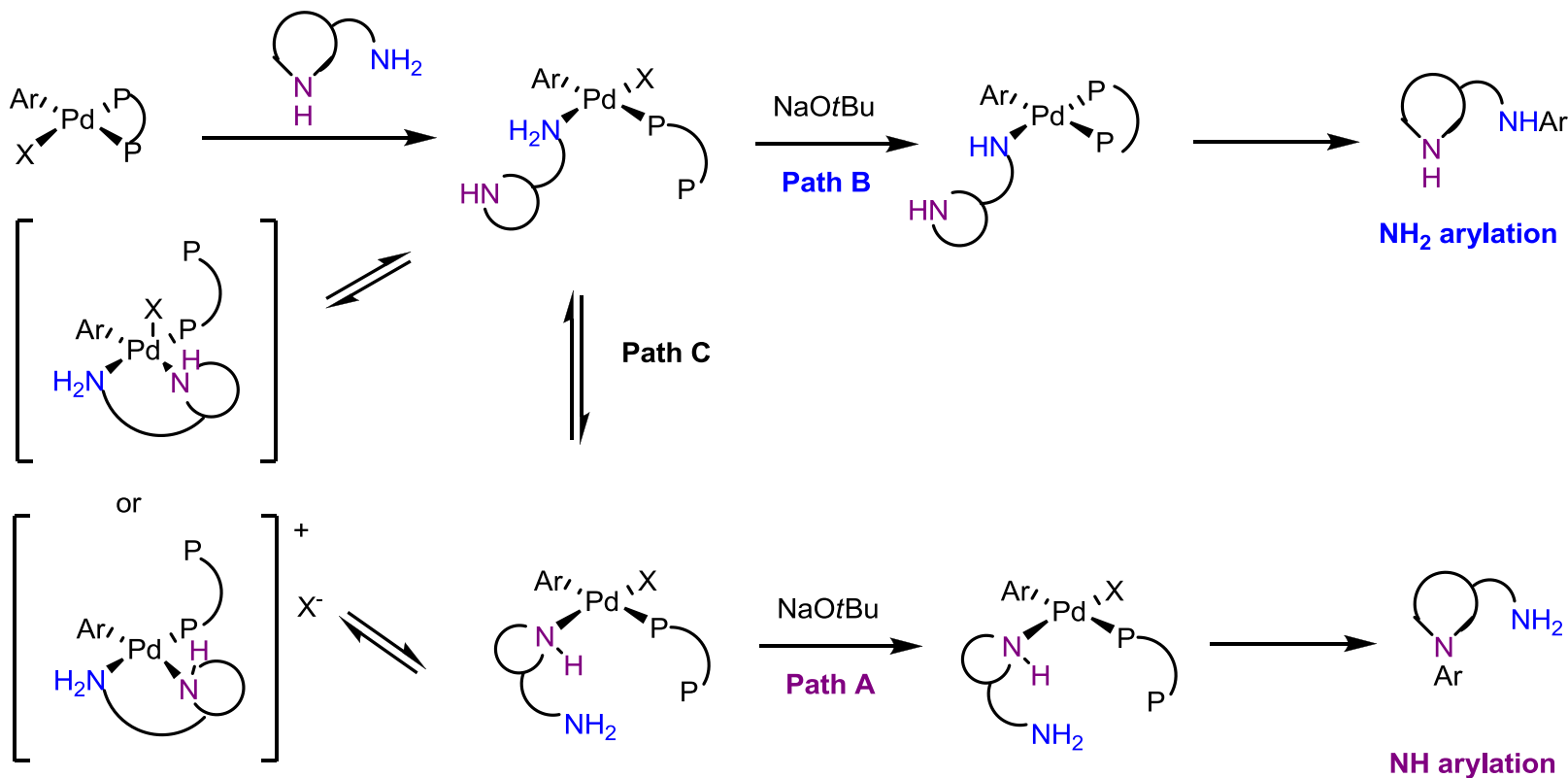
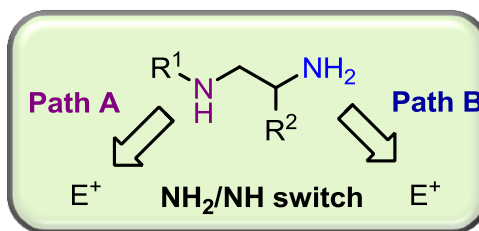
Chemoselectivity controlled by:

- Steric/electronic environment of **L**
- Ring size of diamine

Saturated Carbon-Heteroatom Oligoamines

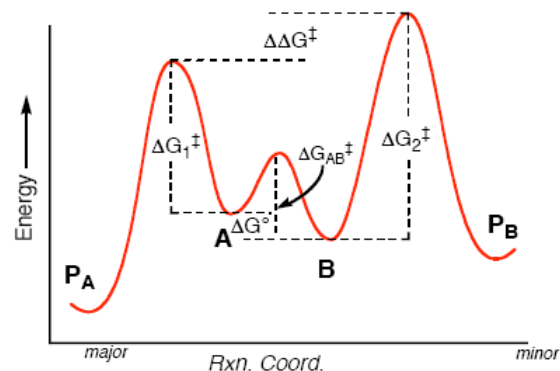
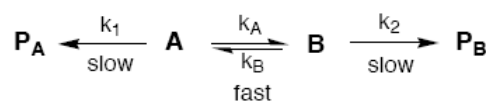
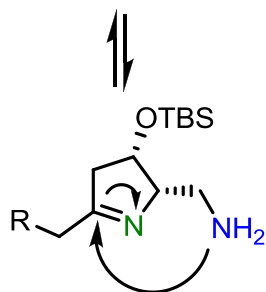
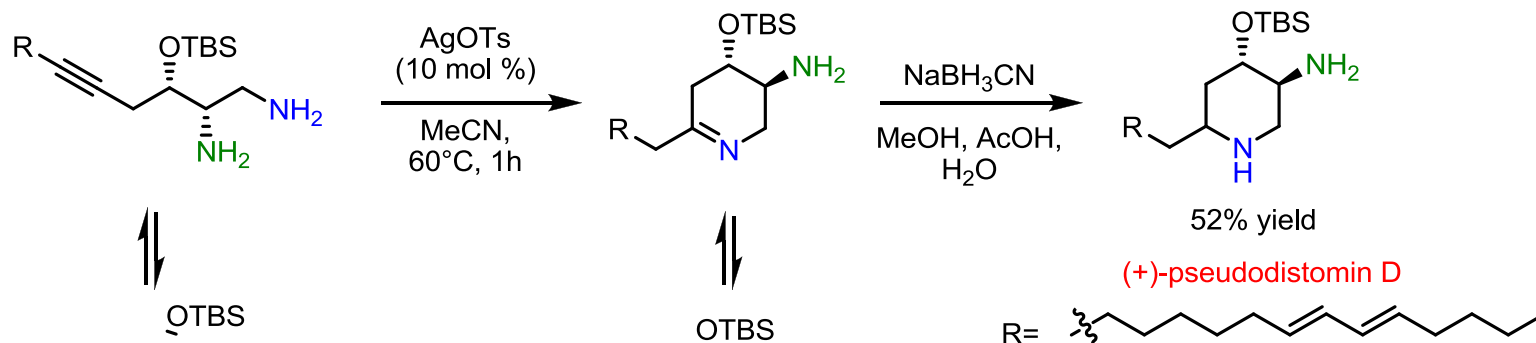
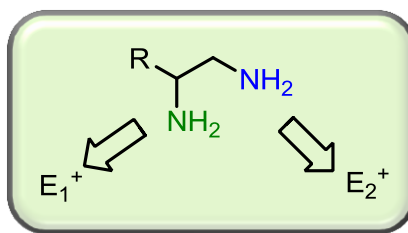
Metal-Mediated Chemoselectivity

Mechanism



Azanorbornyl conformations

Saturated Carbon-Heteroatom Oligoamines



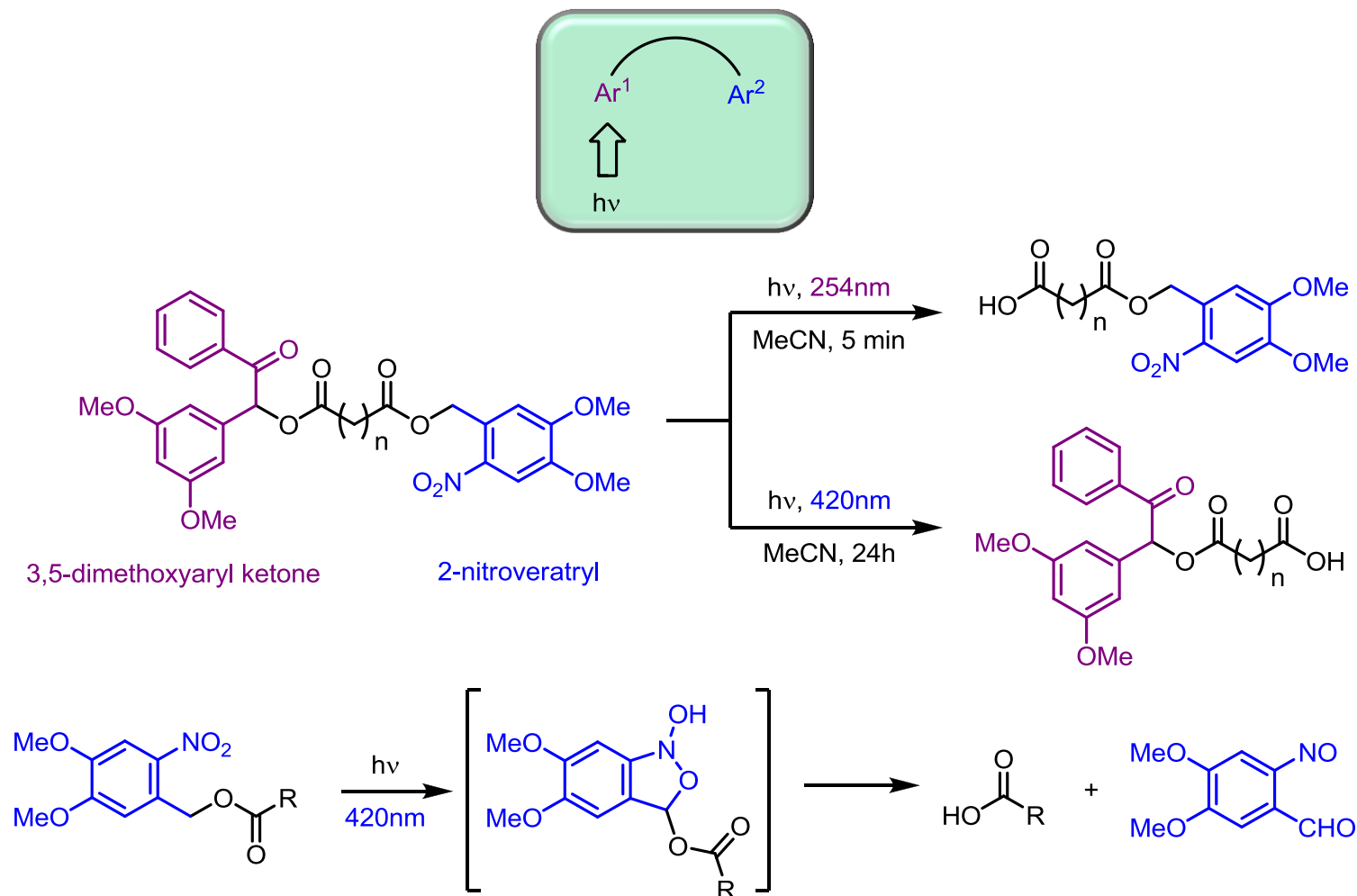
Selectivity based on **Curtin-Hammett principle**:

« The product composition P_A vs P_B is not solely dependent on relative proportions of the conformational isomers in the substrate;

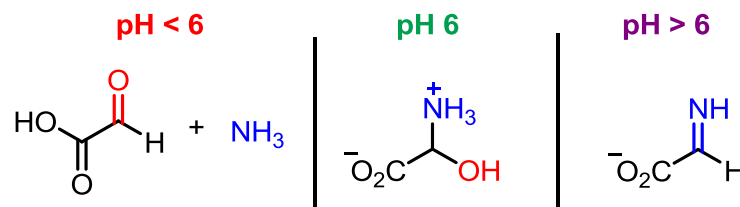
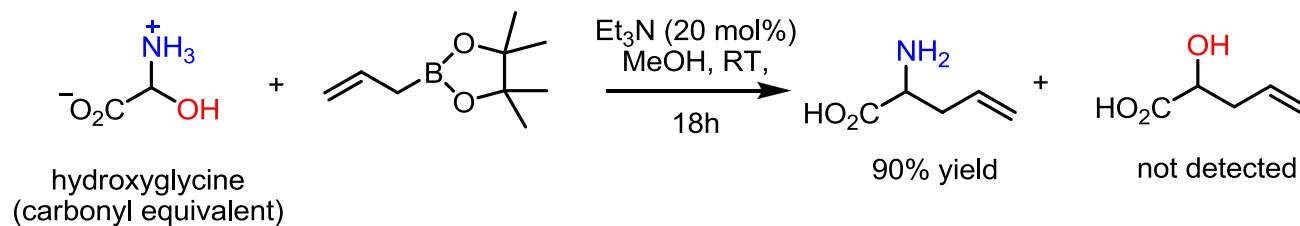
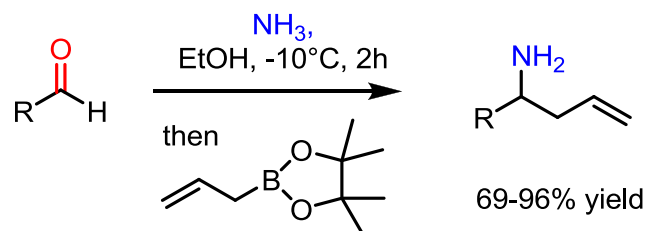
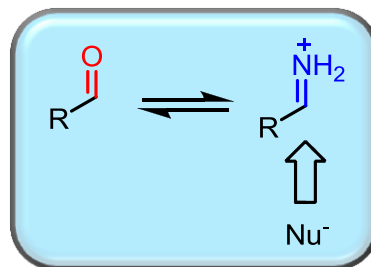
It is controlled by the difference in the free energy ($\Delta\Delta G^\ddagger$) of the respective transition states. »

Saturated Carbon-Heteroatom

Chromatic orthogonality



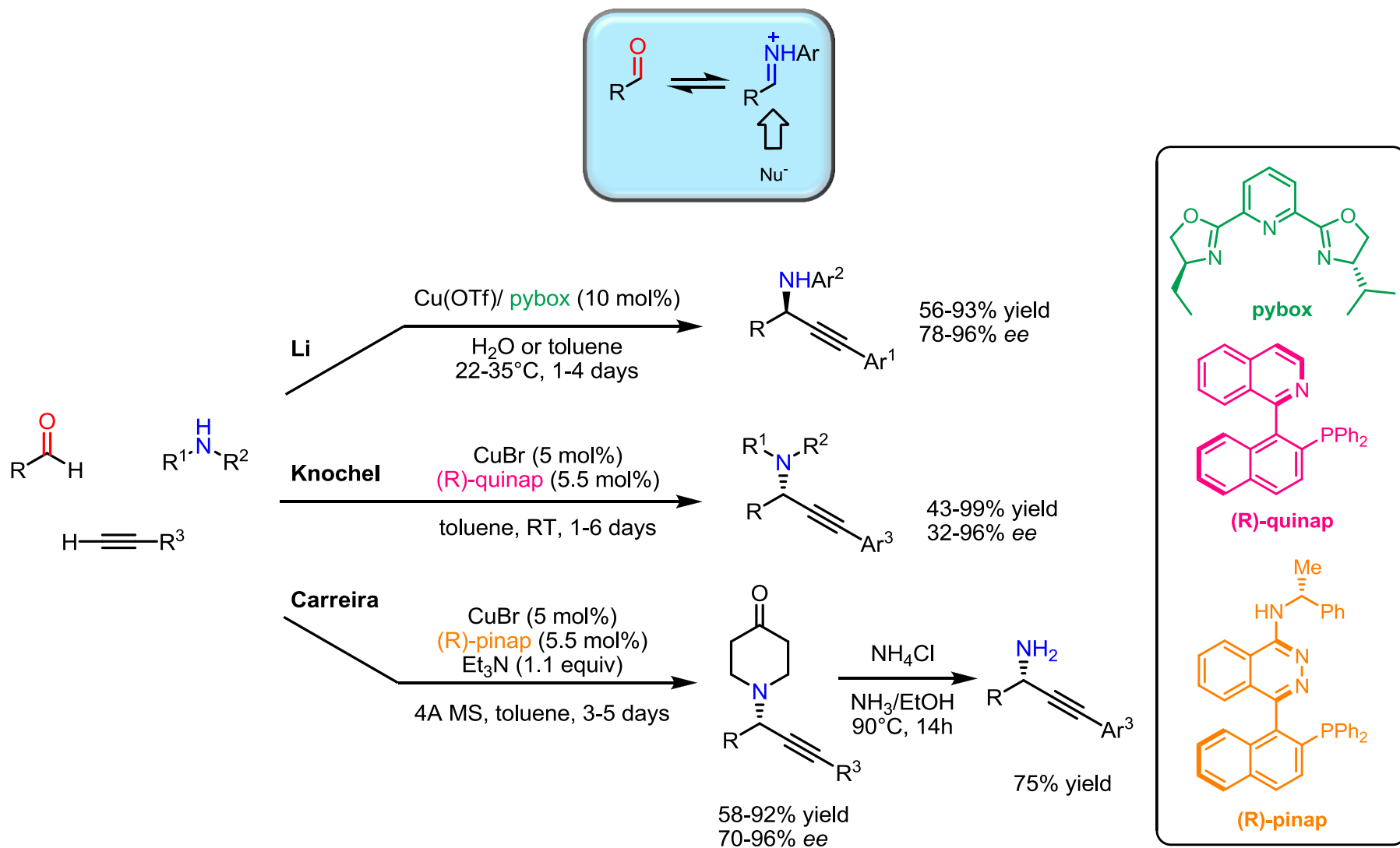
Unsaturated Carbon-Heteroatom Iminium electrophiles



M. Sugiura, K. Hirano, S. Kobayashi, *J. Am. Chem. Soc.* **2004**, *126*, 7182 – 7183.

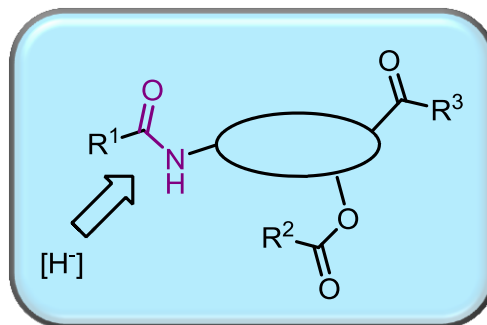
M. Sugiura, C. Mori, K. Hirano, S. Kobayashi, *Can. J. Chem.* **2005**, *83*, 937 – 942.

Unsaturated Carbon-Heteroatom Iminium electrophiles



C. Wei, C.-J. Li, *J. Am. Chem. Soc.* **2002**, *124*, 5638 – 5639.; N. Gommermann, C. Koradin, K. Polborn, P. Knochel, *Angew. Chem. Int. Ed.* **2003**, *42*, 5763 – 5766; P. Aschwanden, C. R. J. Stephenson, E. M. Carreira, *Org. Lett.* **2006**, *8*, 2437 – 2440.

Unsaturated Carbon-Heteroatom Iminium electrophiles



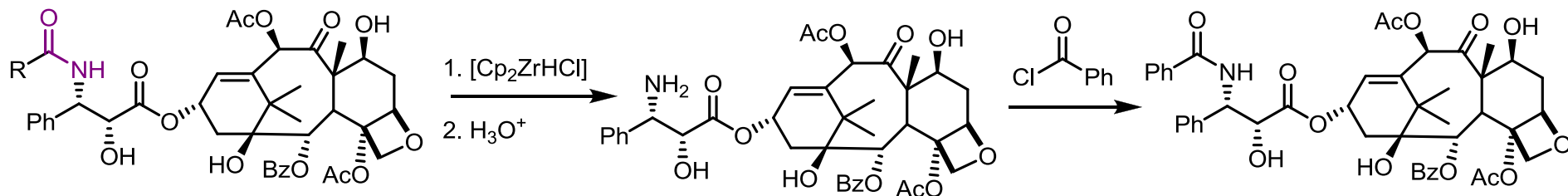
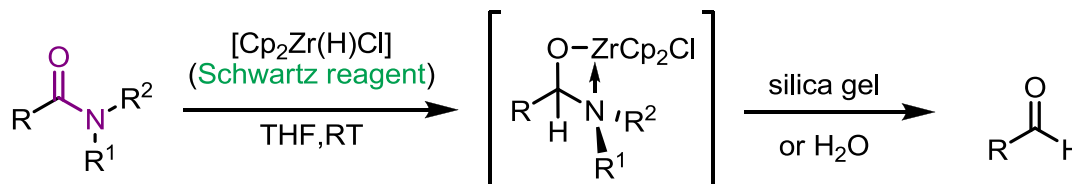
Compatible with:

esters, nitriles, nitro, carbamates,
alkenes, internal alkynes

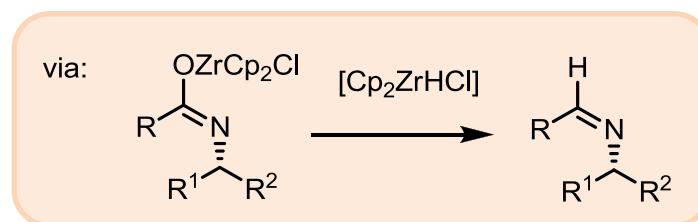


Not compatible with:

aromatic ketones,
terminal alkynes

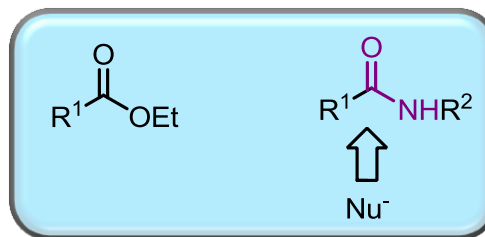


Taxol



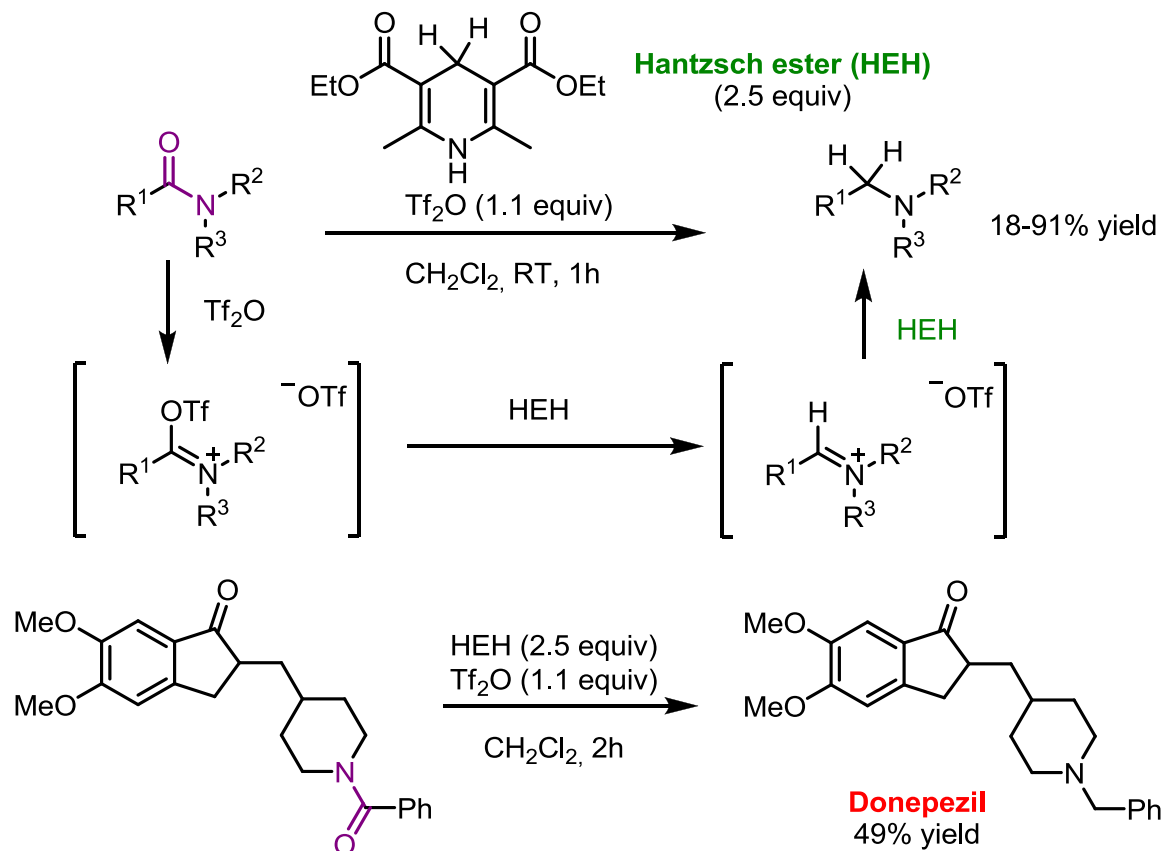
Unsaturated Carbon-Heteroatom

Iminium electrophiles



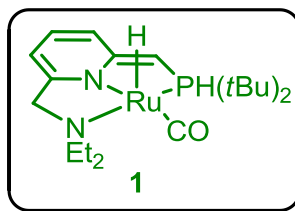
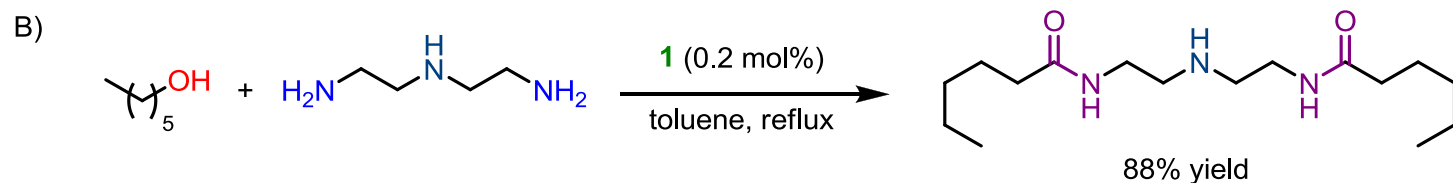
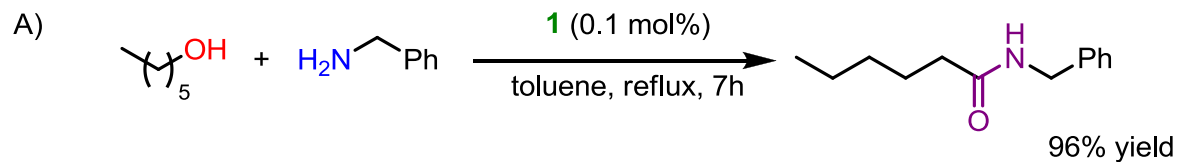
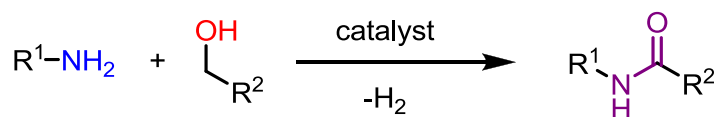
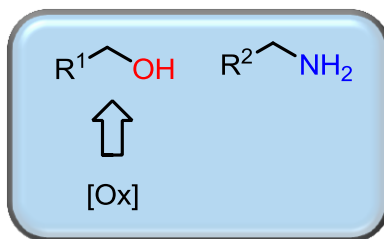
Compatible with:

ketones, esters, α,β -unsaturated esters, nitriles, epoxides, alkynes, ethers



Unsaturated Carbon-Heteroatom

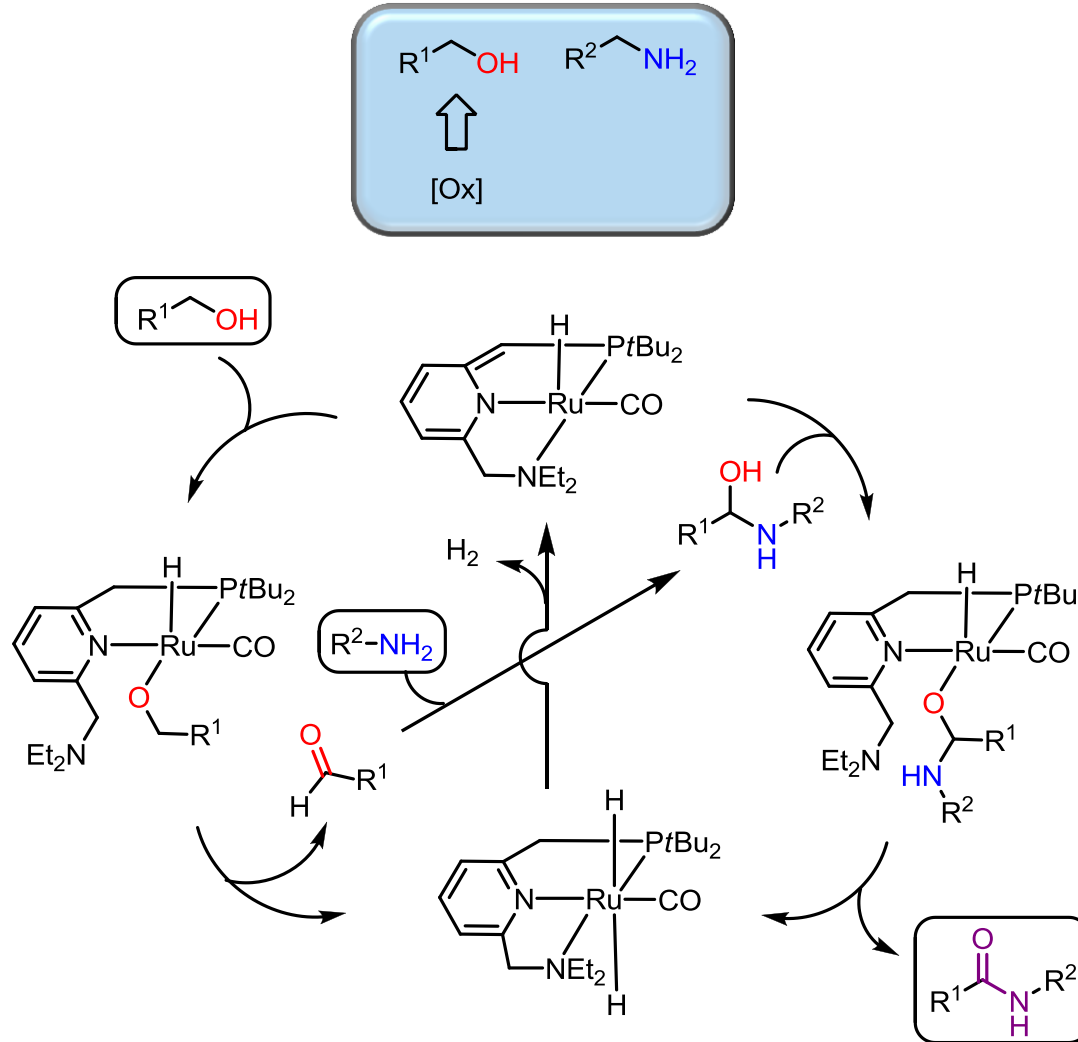
Redox processes



Unsaturated Carbon-Heteroatom

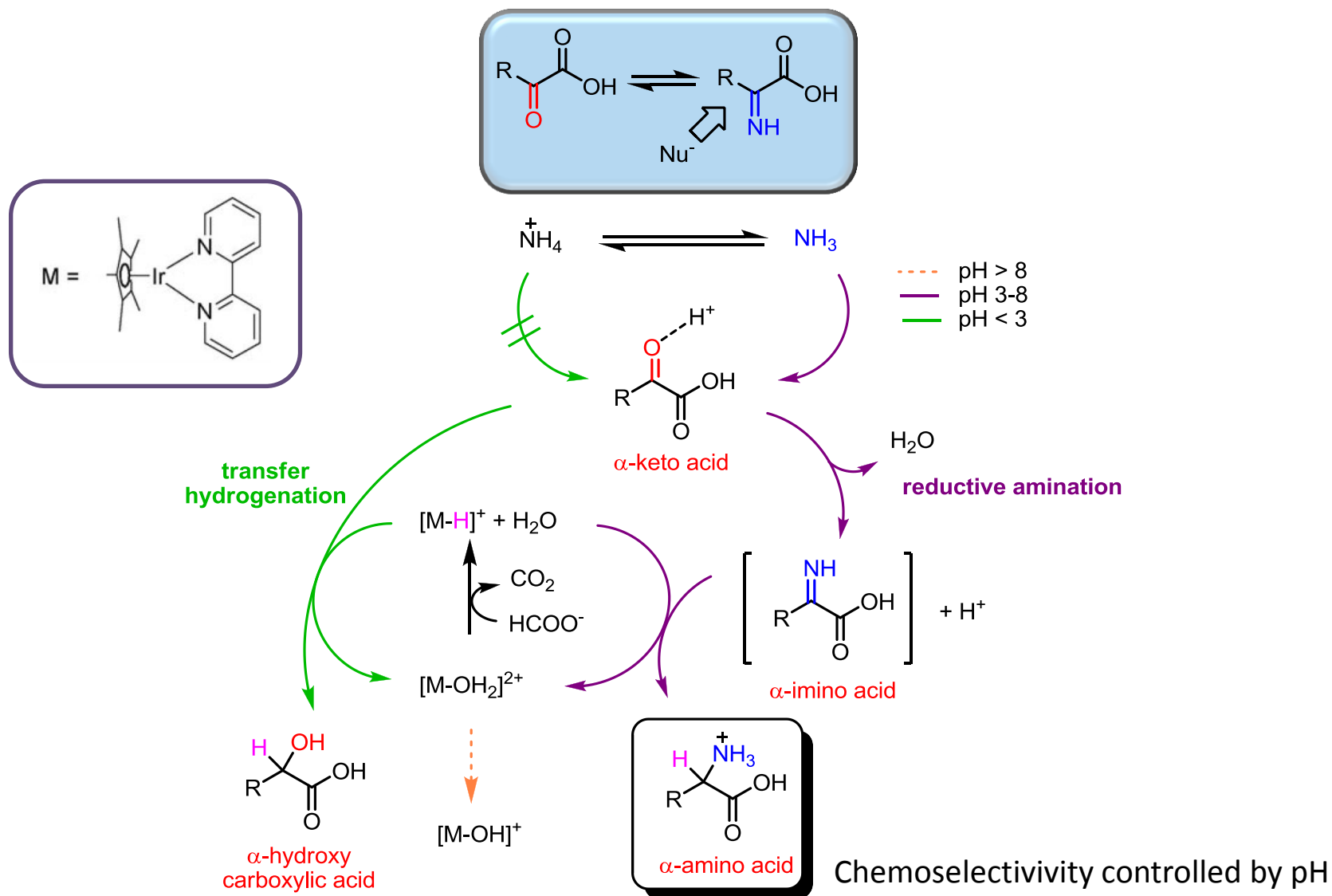
Redox processes

Mechanism:



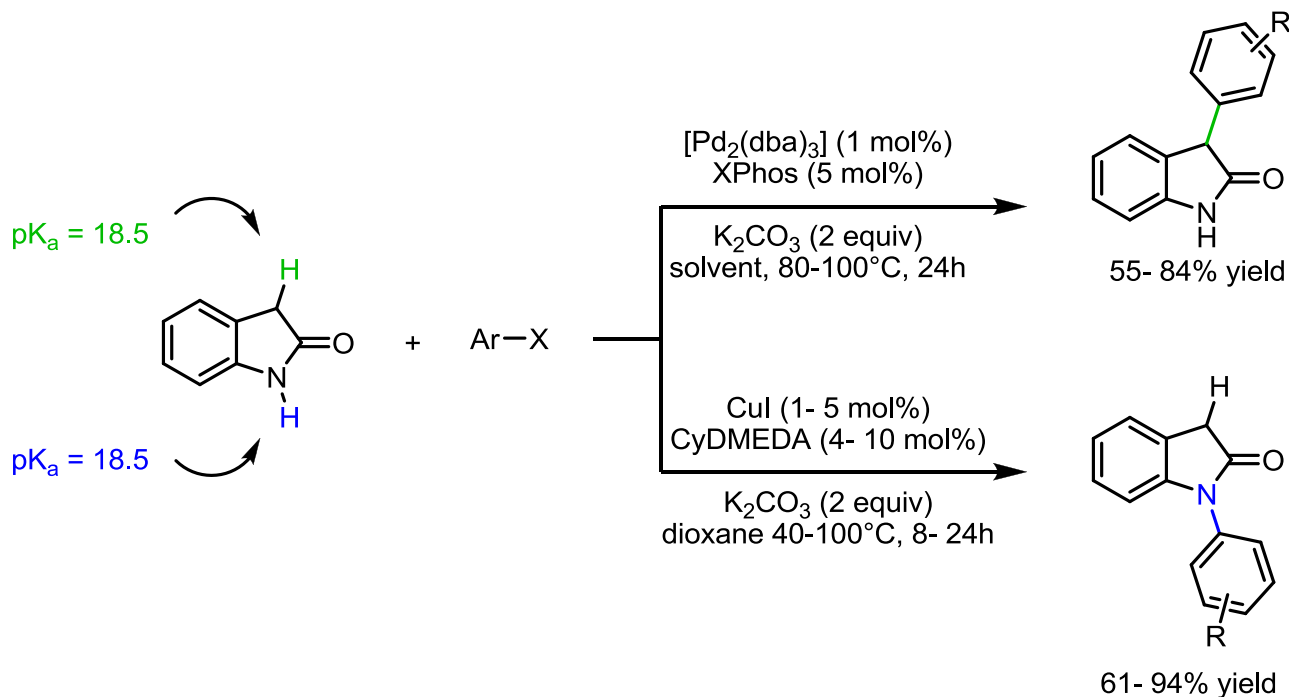
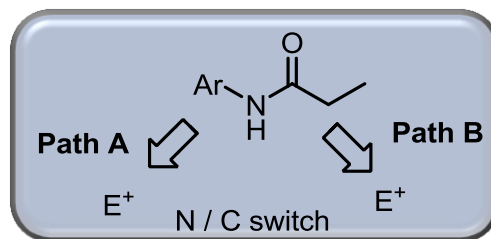
Unsaturated Carbon-Heteroatom

Redox processes



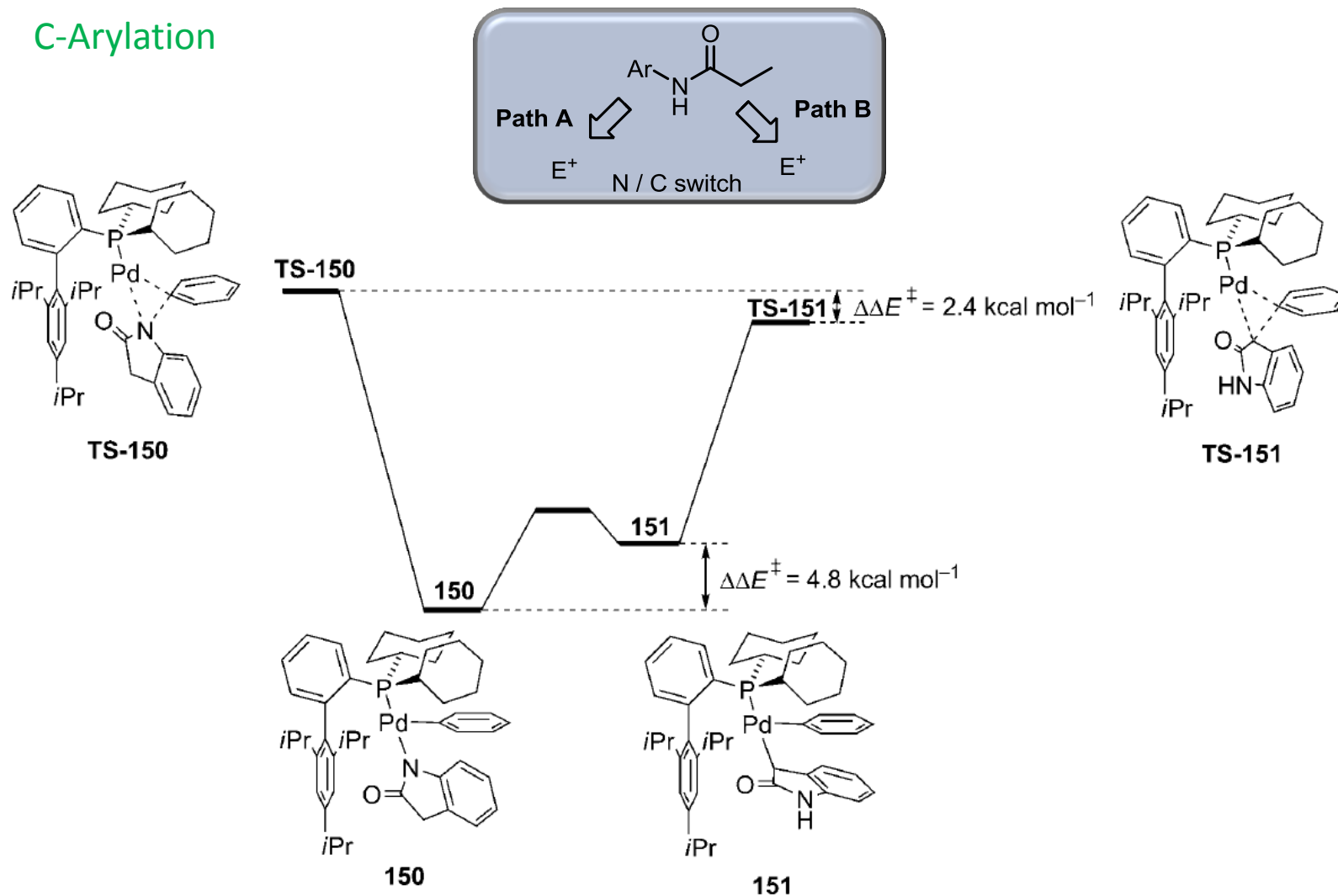
Unsaturated Carbon-Heteroatom

Carbon-Carbon vs Carbon-Heteroatom



Unsaturated Carbon-Heteroatom Carbon-Carbon vs Carbon-Heteroatom

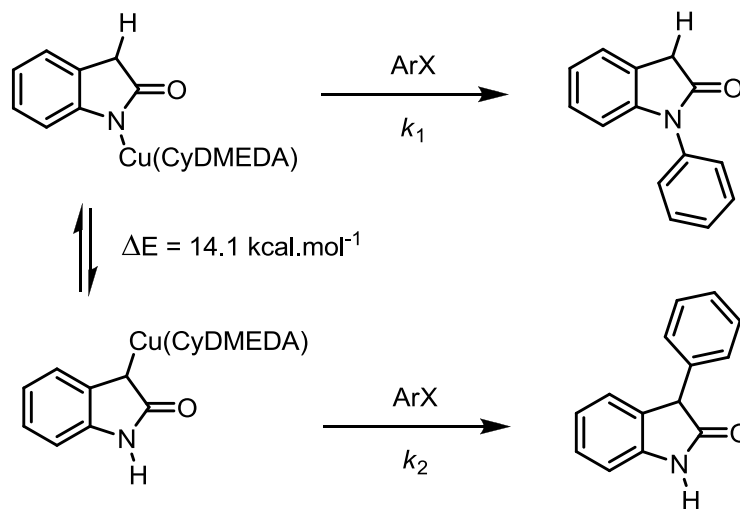
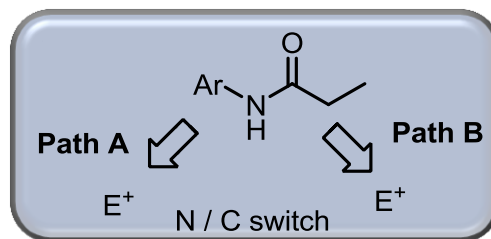
C-Arylation



Chemoselectivity controlled according to the Curtin-Hammett principle

Unsaturated Carbon-Heteroatom Carbon-Carbon vs Carbon-Heteroatom

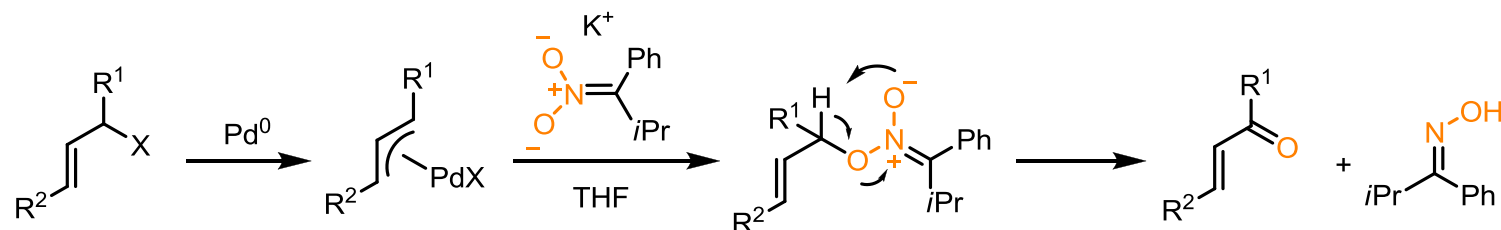
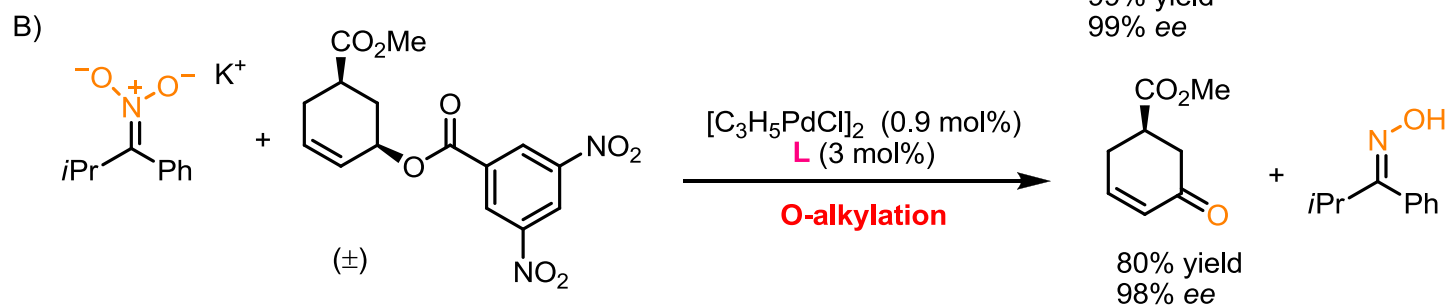
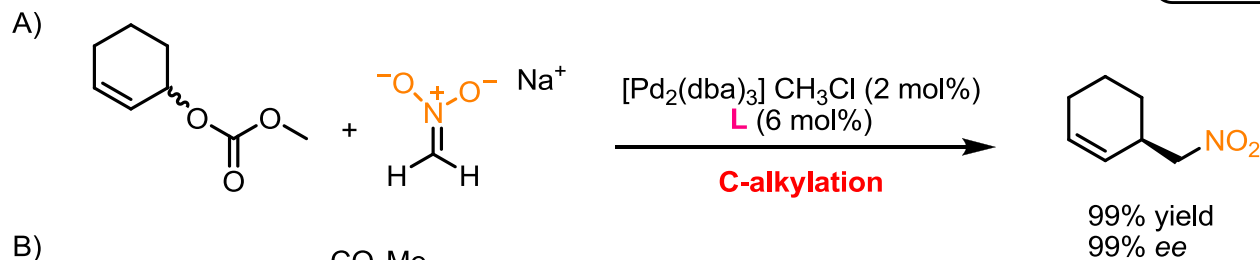
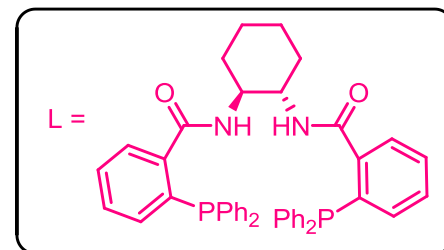
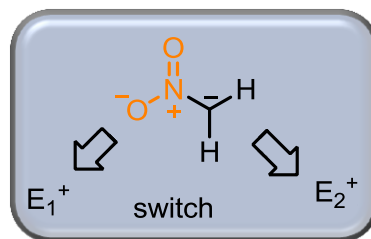
N-Arylation



2 possible explanations of selectivity:

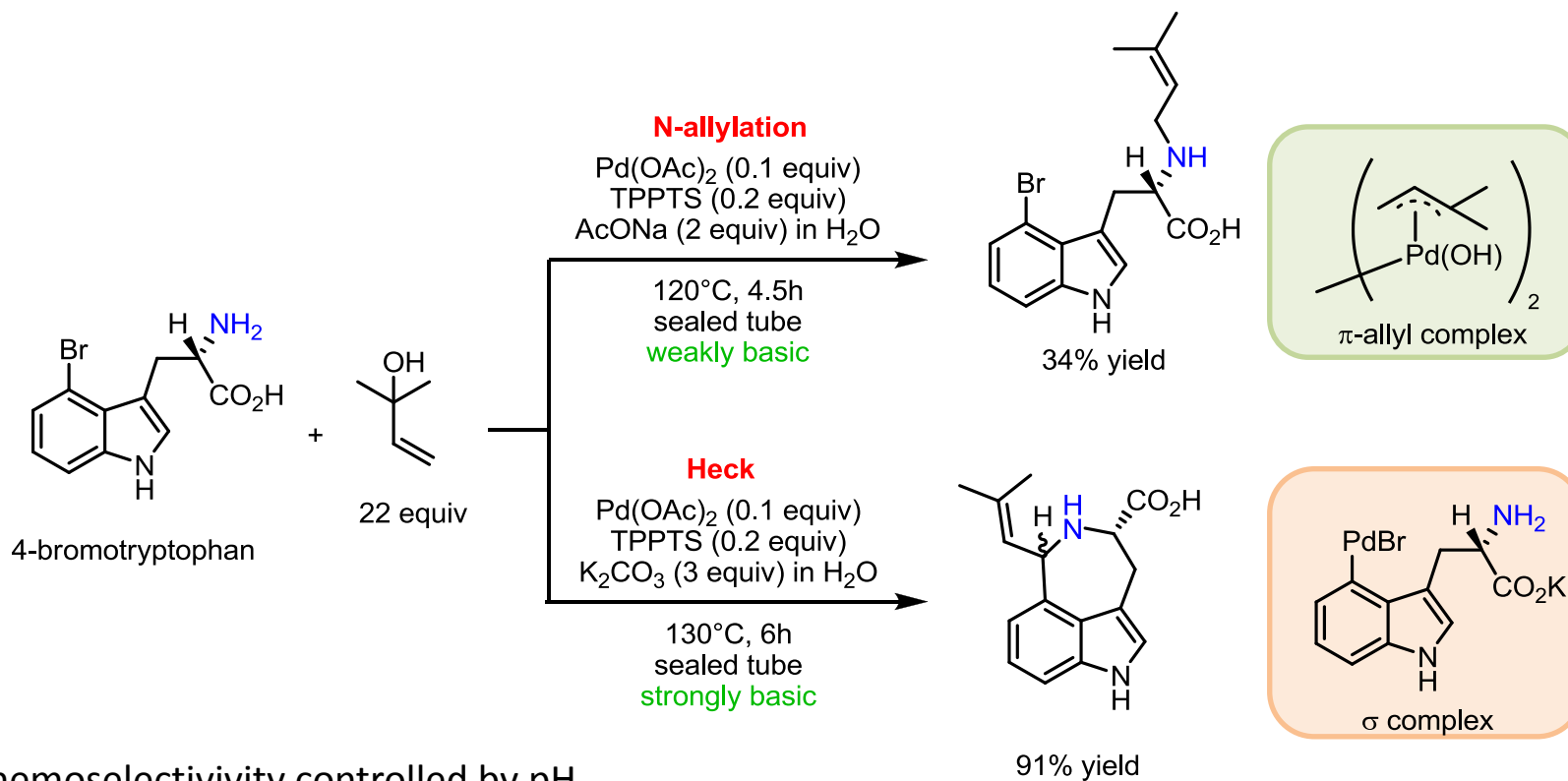
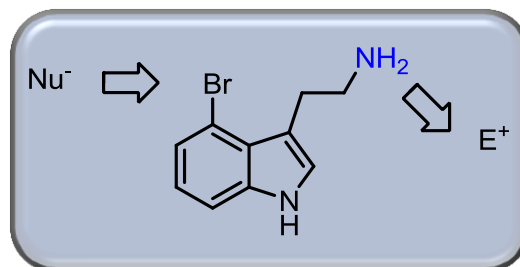
- Not interconversion
- $k_1 \gg k_2$

Unsaturated Carbon-Heteroatom Carbon-Carbon vs Carbon-Heteroatom



Chemoselectivity controlled by :
steric/ electronic properties of the nitronate

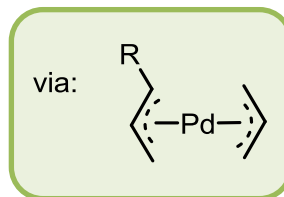
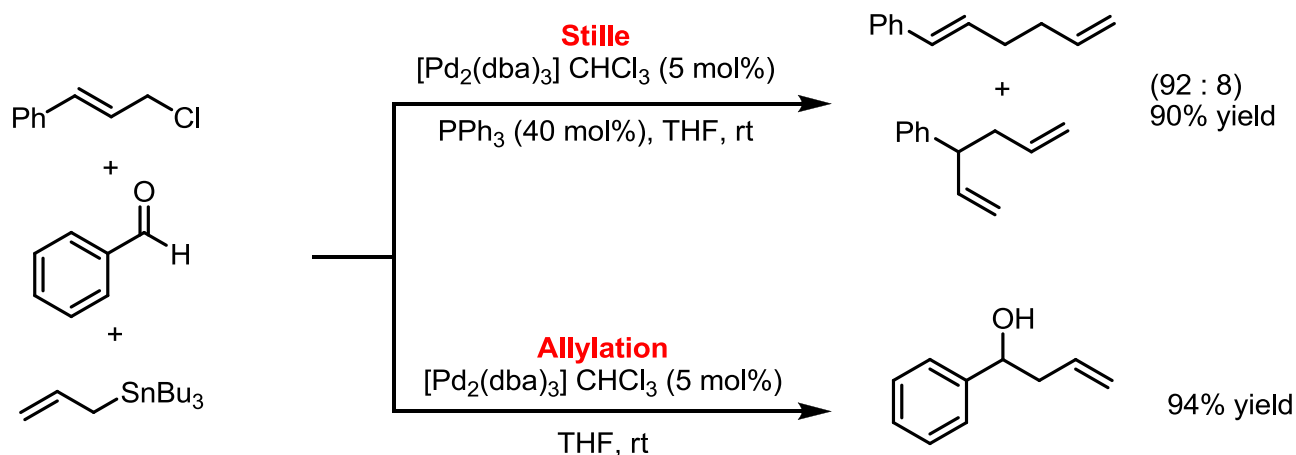
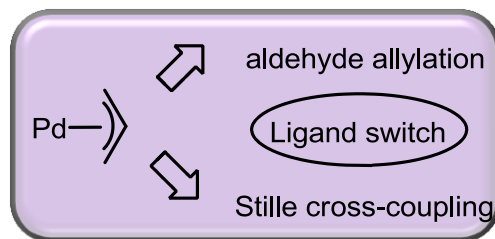
Unsaturated Carbon-Heteroatom Carbon-Carbon vs Carbon-Heteroatom



Chemoselectivity controlled by pH

Unsaturated Carbon-Heteroatom

Carbon-Carbon couplings

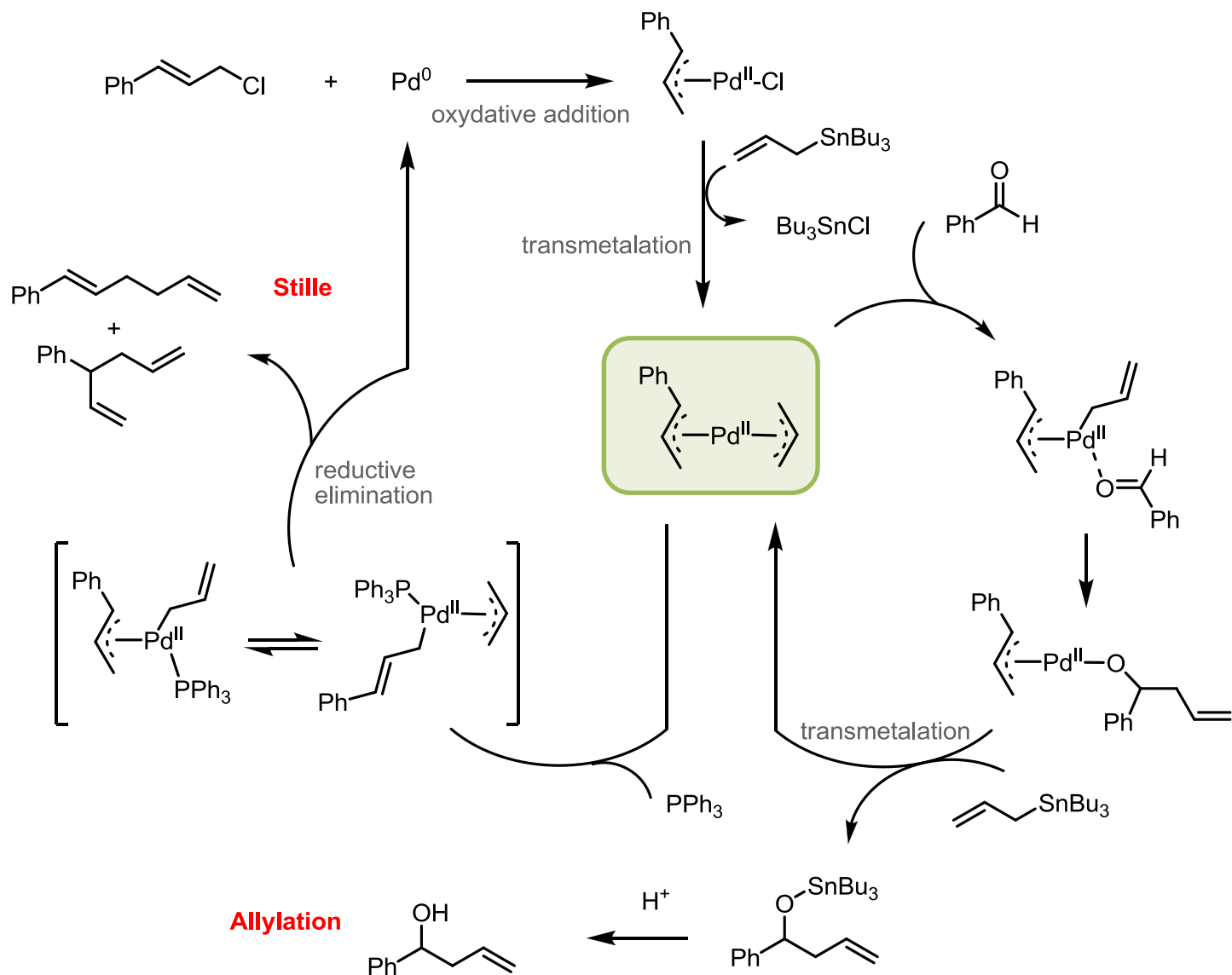


Chemoselectivity controlled by:
presence/absence of PPh₃

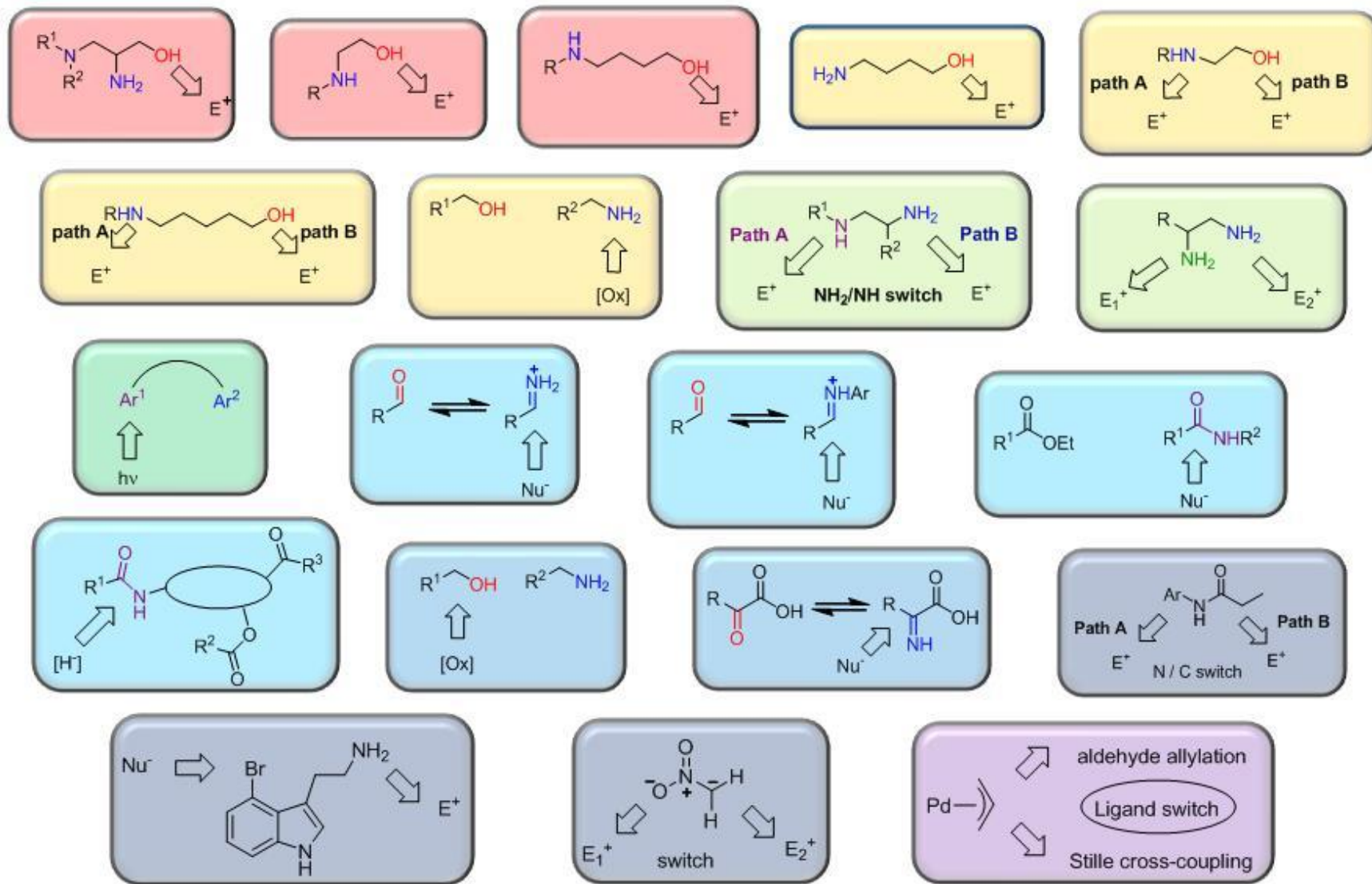
Unsaturated Carbon-Heteroatom

Carbon-Carbon couplings

Mechanism:



Conclusion



Conclusion

