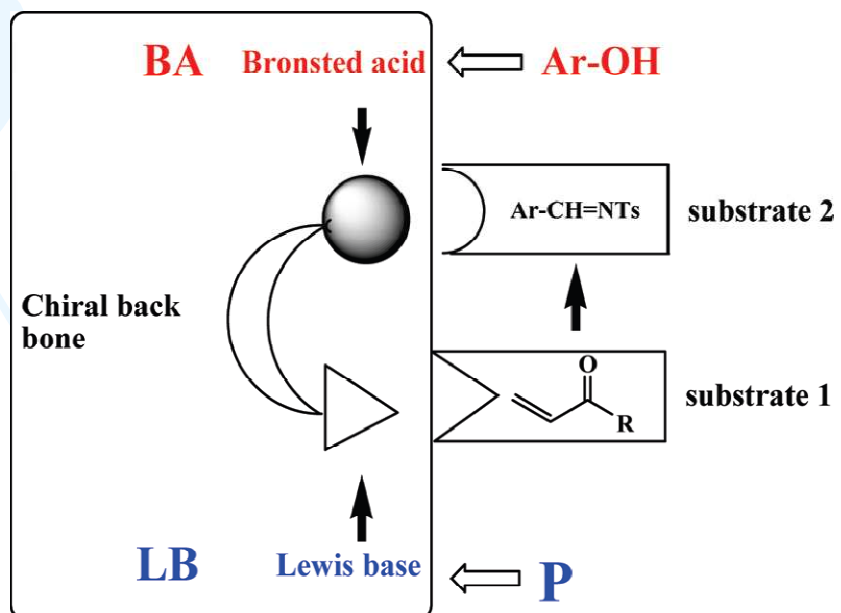


## Multifunctional Chiral Phosphine Organocatalysts in Catalytic Asymmetric Morita-Baylis-Hillman and Related Reaction



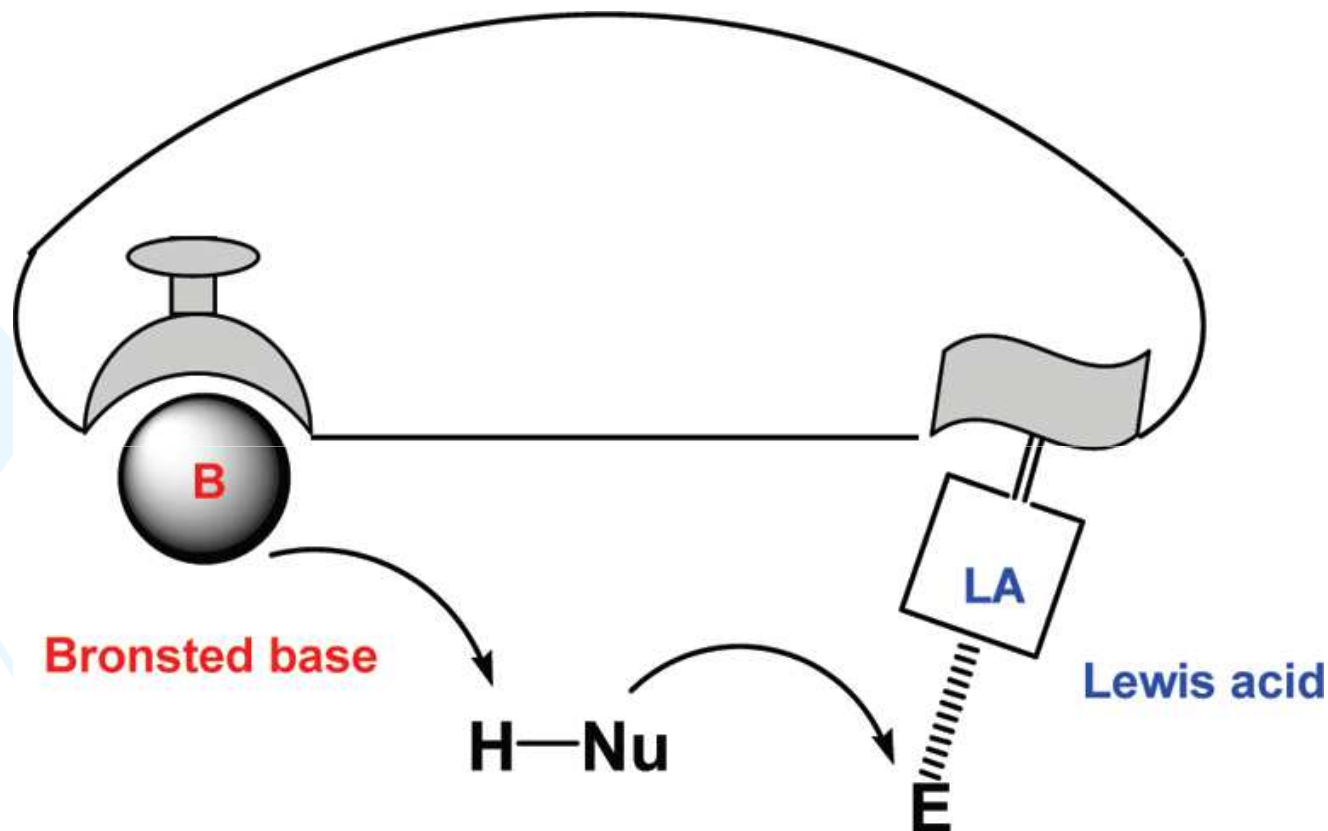
# Introduction

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- Catalytic asymmetric synthesis has received considerable attention over the past few decades.
- Becoming a highly dynamic area of chemical research with significant contributions to the field of organic synthesis.
- Artificial catalysts now provide highly economic access to many desirable compounds, but the general adaptability and reactivity of these platforms remain problematic, particularly in comparison to nature's catalysts, enzymes.
- The multifunctional organocatalysts described in this Account represent another positive step in the synthetic chemist's efforts to profitably mimic nature's catalytic platform, helping develop small-molecule catalysts with enzyme-like reactivities and selectivities

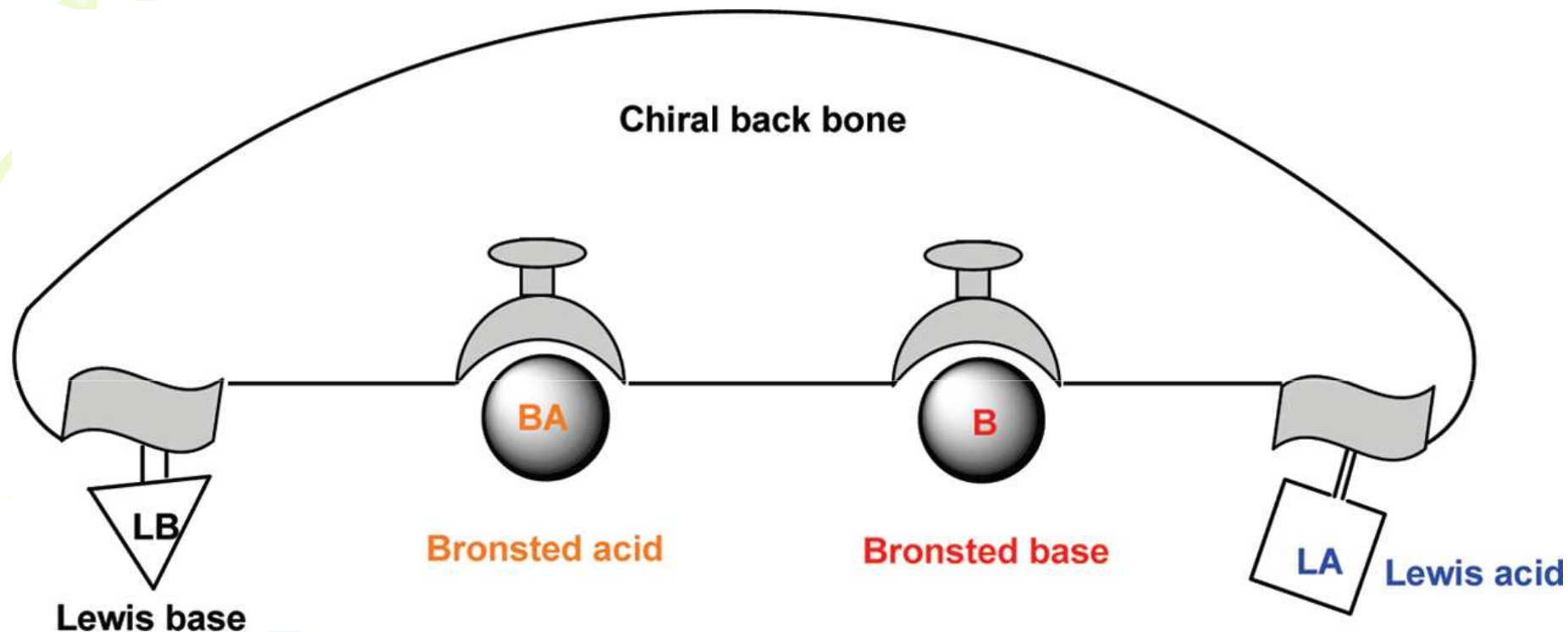
***These multifunctional chiral phosphines, which contain Lewis basic and Brønsted acidic sites within one molecule, provide good-to-excellent reactivities and stereoselectivities in the asymmetric aza-MBH reaction, the MBH reaction, and other related reactions***

## Introduction



Multifunctional catalysts employing the synergistic function of a Lewis acid and a Brønsted base: LA, Lewis acid; B, Brønsted base; E, electrophile; Nu-H, nucleophile.

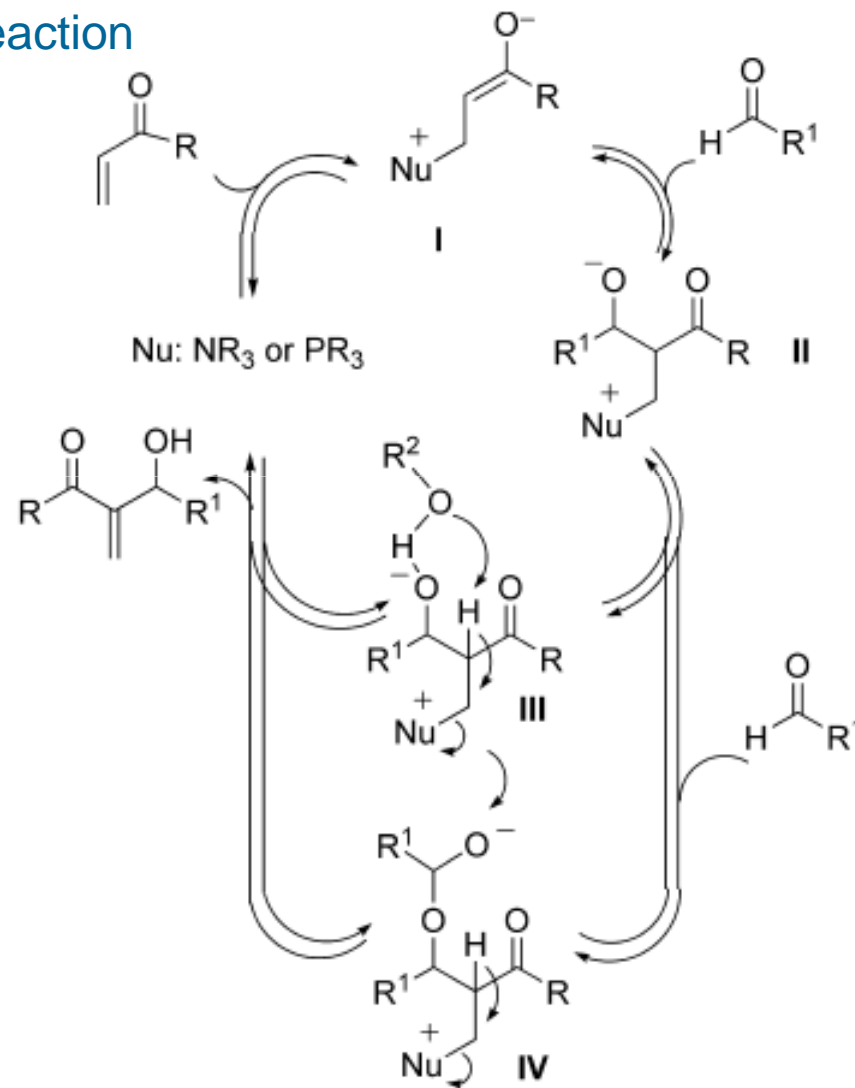
## Introduction



Ideal multifunctional chiral catalysts containing Lewis acid, Brønsted base, Brønsted acid, and Lewis base as active catalytic sites: LA, Lewis acid; B, Brønsted base; BA, Brønsted acid; LB, Lewis base.

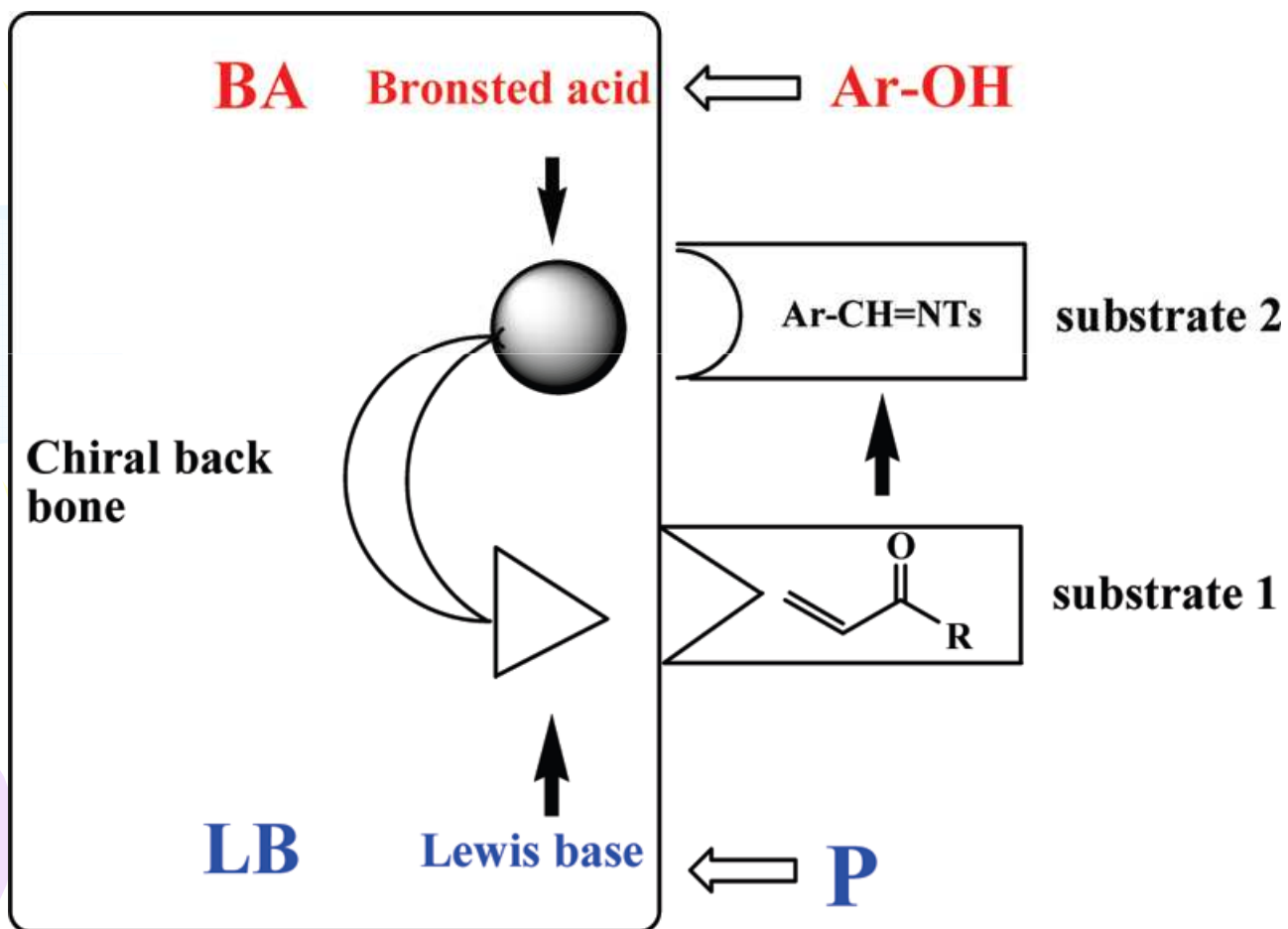
# Introduction

## The Proposed Mechanism of MBH Reaction



## Introduction

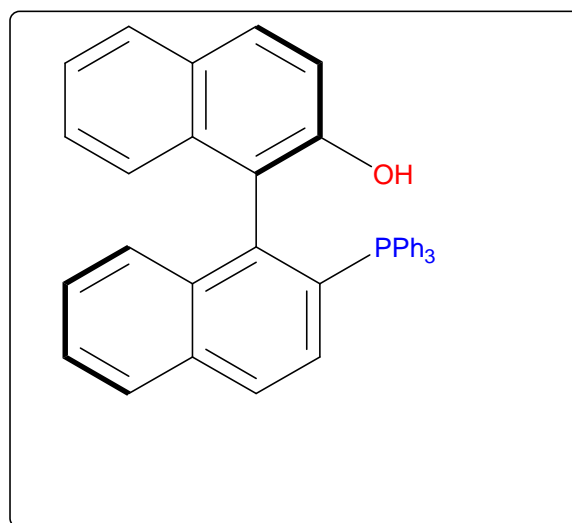
Multifunctional chiral phosphine Lewis base catalyst, LBBA bifunctional catalytic system:  
LB, Lewis base; BA, Brønsted acid.



## Multifunctional Chiral Phosphine Catalysts in Aza-Morita-Baylis-Hillman Reaction:

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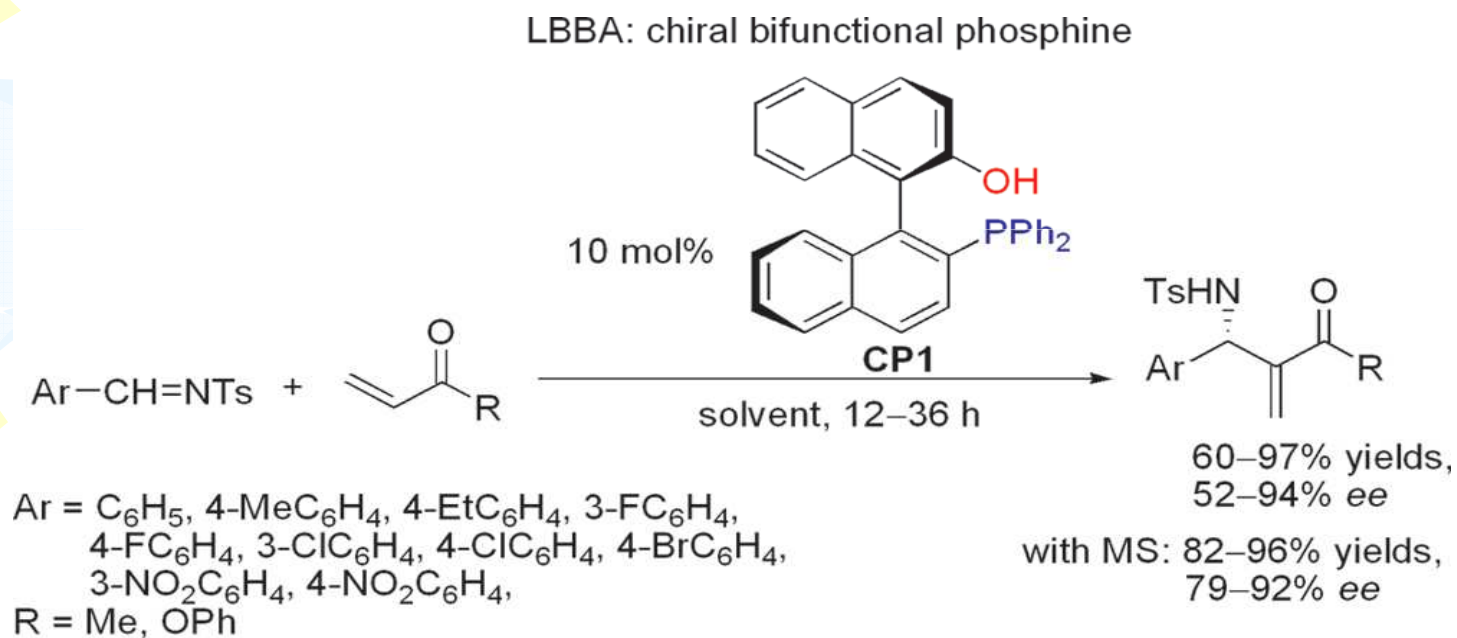
In 1993, Hayashi first synthesized the chiral phosphorus compound **CP1** as a chiral monodentate phosphine ligand:



CP1

## Multifunctional Chiral Phosphine Catalysts in Aza-Morita-Baylis-Hillman Reaction:

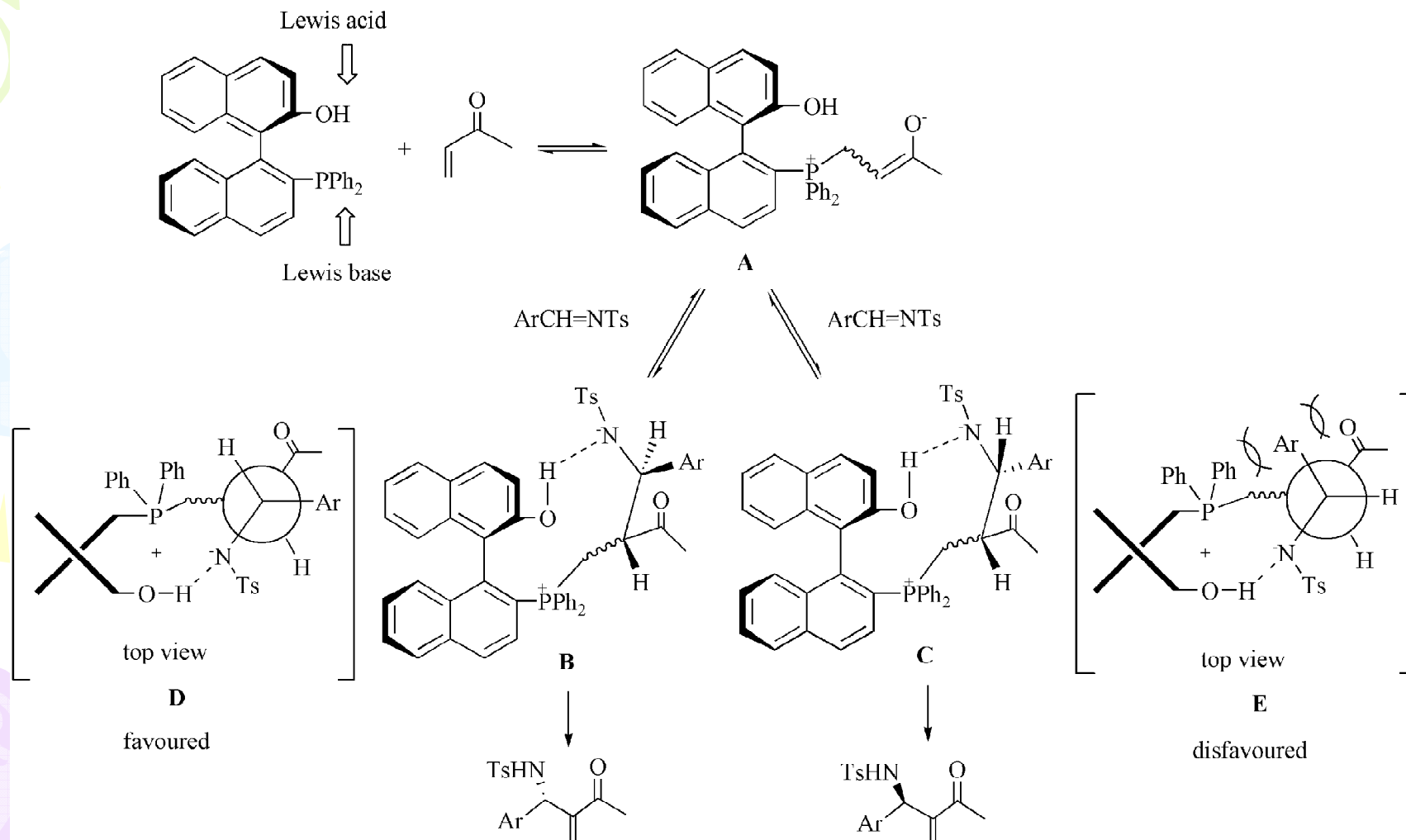
In 2002, Min Shi first demonstrated that this 1,1'-bi-2,2'-naphthol(BINOL)-derived chiral LBBA bifunctional phosphine **CP1** (LB) (PPh<sub>3</sub>, BA) (Ph-OH)





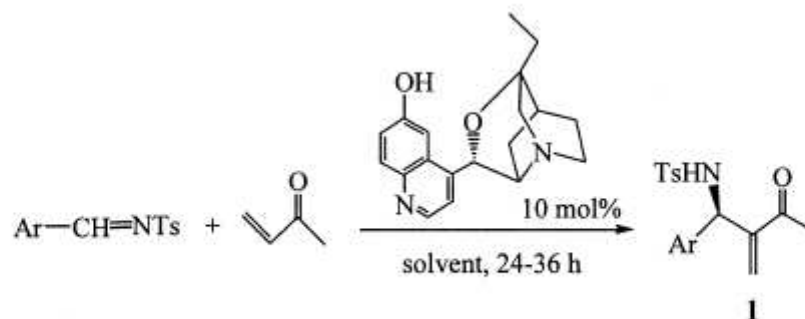
# Multifunctional Chiral Phosphine Catalysts in Aza-Morita-Baylis-Hillman Reaction:

## Mechanistic speculation on the chiral Lewis base CP1



## Multifunctional Chiral Phosphine Catalysts in Aza-Morita-Baylis-Hillman Reaction:

The asymmetric induction of CP1 catalyst is comparable to that of the quinidine derivatives

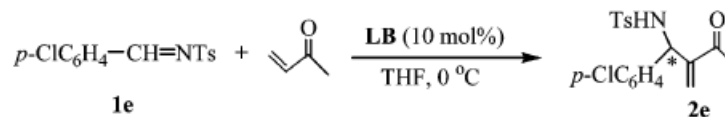
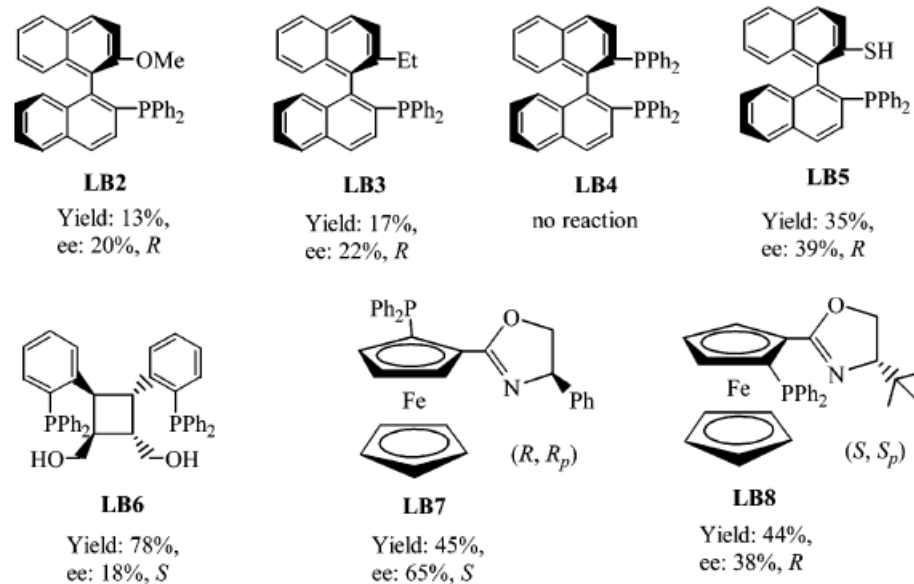


Entry	Ar	<b>1</b>	Solvent	<i>t</i> [h]	<i>T</i> [°C]	Yield of <b>1</b> [%] <sup>[a]</sup>	<i>ee</i> value [%]
1	<i>p</i> -EtC <sub>6</sub> H <sub>4</sub>	c	THF	36	20	30	62
2	<i>p</i> -EtC <sub>6</sub> H <sub>4</sub>	c	THF	24	-25	33	76
3	<i>p</i> -EtC <sub>6</sub> H <sub>4</sub>	c	MeCN	24	0	50	78
4	<i>p</i> -EtC <sub>6</sub> H <sub>4</sub>	c	MeCN	24	-20	64	86
5	<i>p</i> -EtC <sub>6</sub> H <sub>4</sub>	c	DMF	24	-20	55	93
6	<i>p</i> -EtC <sub>6</sub> H <sub>4</sub>	c	DMF	24	-40	50	96
7	<i>p</i> -ClC <sub>6</sub> H <sub>4</sub>	e	THF	24	0	71	42
8	<i>p</i> -ClC <sub>6</sub> H <sub>4</sub>	e	THF	24	-20	65	63
9	<i>p</i> -ClC <sub>6</sub> H <sub>4</sub>	e	MeCN	24	-30	80	81
10	<i>p</i> -ClC <sub>6</sub> H <sub>4</sub>	e	DMF	24	-30	51	95

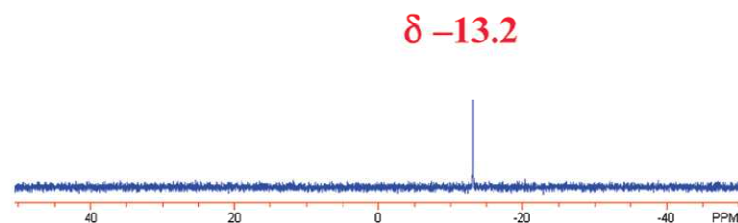
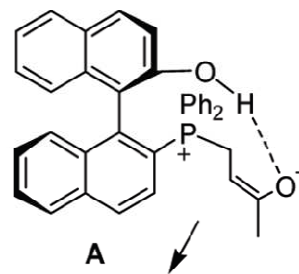
[a] Yields of isolated products.

# Multifunctional Chiral Phosphine Catalysts in Aza-Morita-Baylis-Hillman Reaction:

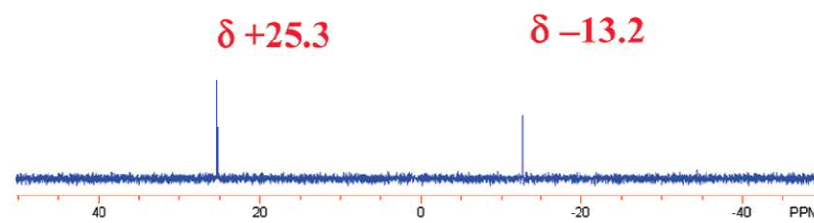
The presence of a phenolic hydroxyl group in catalyst **CP1** seems crucial for good yield and high ee.



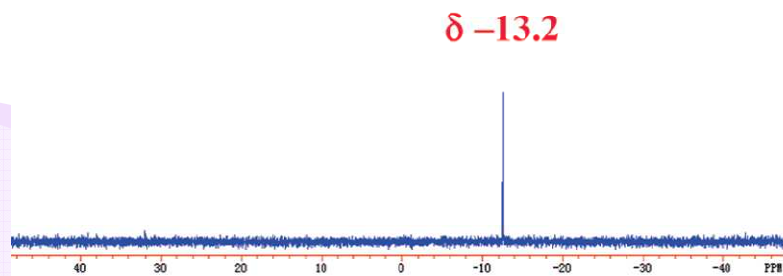
# Multifunctional Chiral Phosphine Catalysts in Aza-Morita-Baylis-Hillman Reaction:



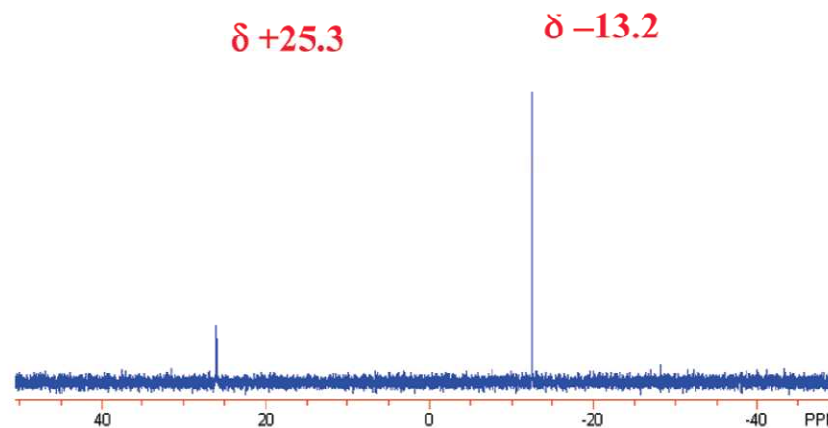
$^{31}\text{P}$  NMR of CP1



$^{31}\text{P}$  NMR of the mixture of CP1 and methyl vinyl ketone



$^{31}\text{P}$  NMR of the mixture of CP1 and imine

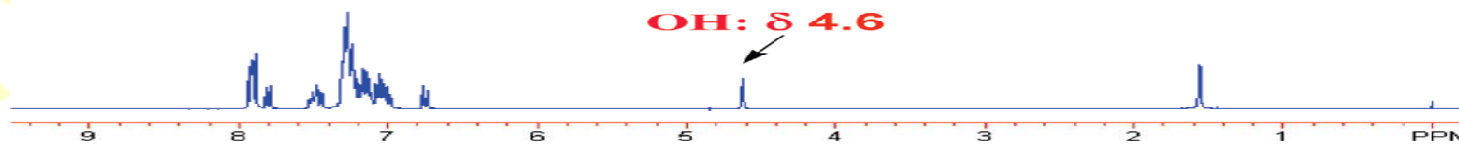


$^{31}\text{P}$  NMR of the mixture of CP1, methyl vinyl ketone and imine

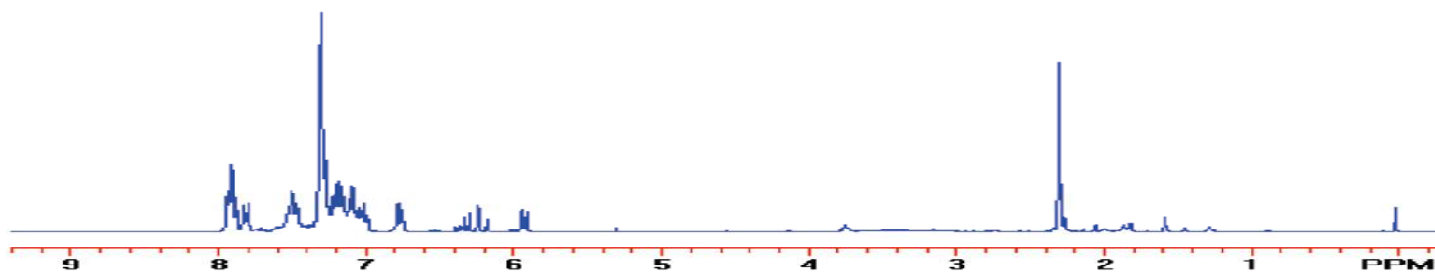
# Multifunctional Chiral Phosphine Catalysts in Aza-Morita-Baylis-Hillman Reaction:



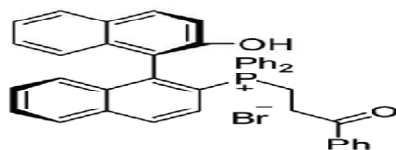
**OH:  $\delta$  4.6**



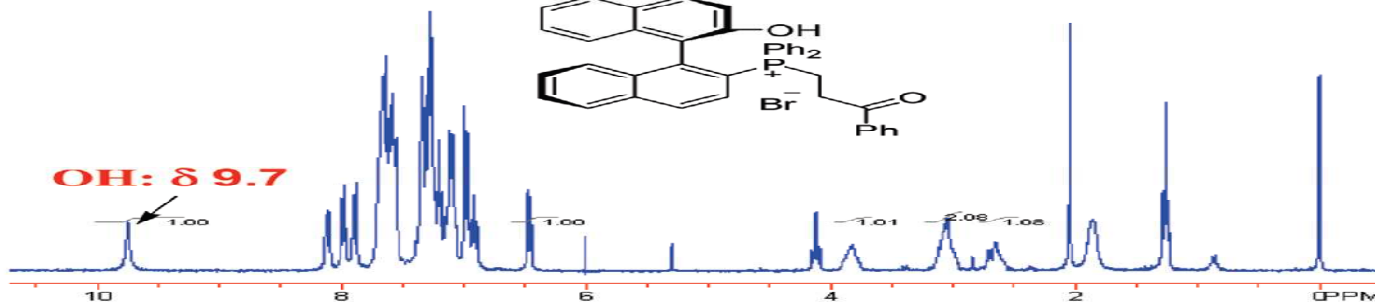
$^1\text{H}$  NMR of CP1



$^1\text{H}$  NMR of CP1 with MVK



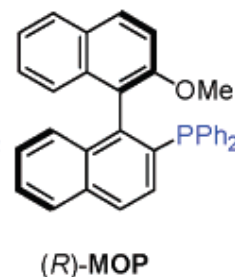
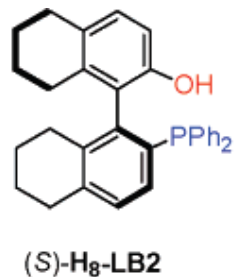
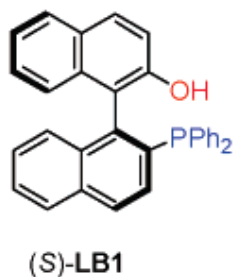
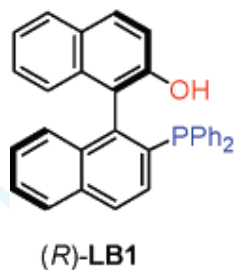
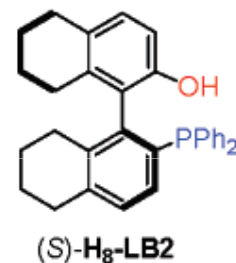
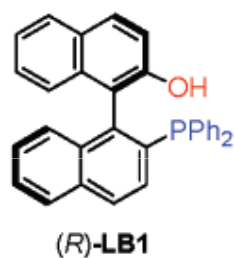
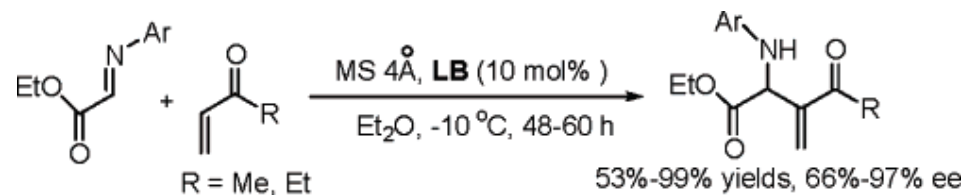
**OH:  $\delta$  9.7**



$^1\text{H}$  NMR of phosphonium bromide

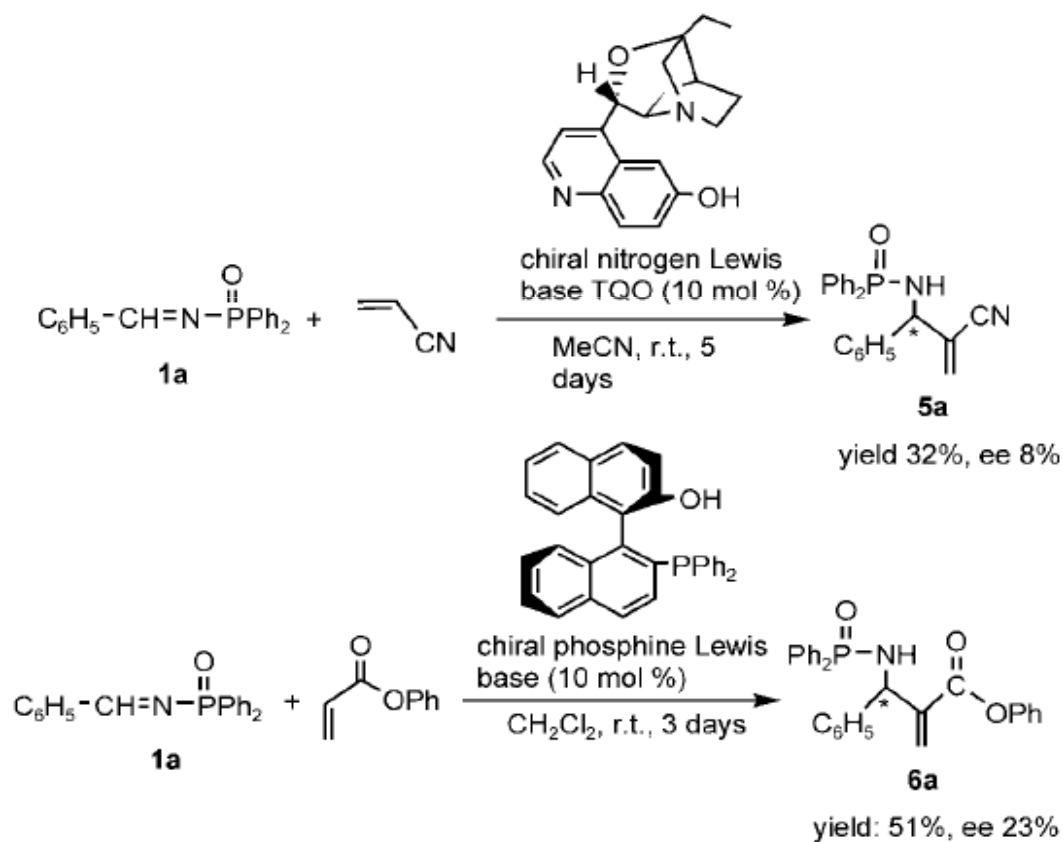
## Multifunctional Chiral Phosphine Catalysts in Aza-Morita-Baylis-Hillman Reaction:

The chiral catalyst **CP1** did not induce any racemization on a similar time scale.

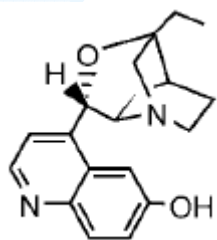
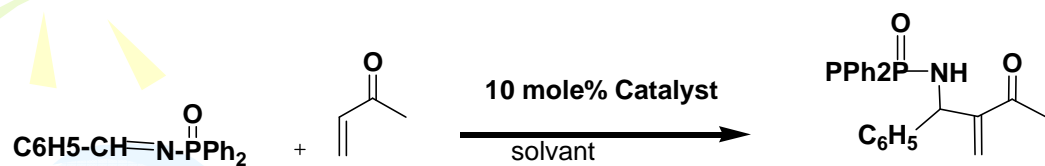


## Multifunctional Chiral Phosphine Catalysts in Aza-Morita-Baylis-Hillman Reaction:

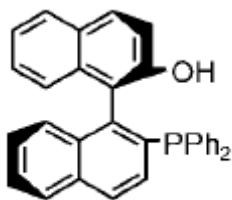
However, catalyst **CP1** could not give good enantiomeric excess in the reaction of *N*-arylmethylidenediphenylphosphinamides with activated alkenes such as MVK, acrylonitrile, or phenyl acrylate



# Multifunctional Chiral Phosphine Catalysts in Aza-Morita-Baylis-Hillman Reaction:



chiral nitrogen Lewis base TQO



chiral phosphine Lewis base

**Table 18.** Catalytic, asymmetric aza-Baylis–Hillman reactions of *N*-benzylidenediphenylphosphinamide (**1a**; 1.0 equiv.) with MVK (1.2 equiv.) in the presence of chiral phosphine and nitrogen catalysts (10 mol %).

Entry	Catalyst	Solvent	Temp. [°C]	Time [h]	Yield [%] <sup>[a]</sup> 2a	ee [%] <sup>[b]</sup>	[α] <sub>D</sub> <sup>[c]</sup>
1	TQO	DMF	rt.	48	43	23	+7.7
2		CH <sub>2</sub> Cl <sub>2</sub>	rt.	48	73	28	+8.0
3		MeCN	rt.	48	13	12 <sup>[d]</sup>	+3.0
4		DMF	-20	72	32	23	+8.4
5		CH <sub>2</sub> Cl <sub>2</sub>	-20	120	<10	ND <sup>[e]</sup>	ND
6		DMF	rt.	48	27	38	+0.7
7		CH <sub>2</sub> Cl <sub>2</sub>	rt.	48	32	47	+14.6
8		CH <sub>2</sub> Cl <sub>2</sub>	-20	120	<10	ND	ND
9		DMF	-20	72	16	43 <sup>[d]</sup>	+13.2
10		THF	rt.	72	26	31 <sup>[d]</sup>	+8.9

<sup>[a]</sup> Yields of isolated products.

<sup>[b]</sup> Determined by chiral HPLC

<sup>[c]</sup> Measured in chloroform at 20 °C.

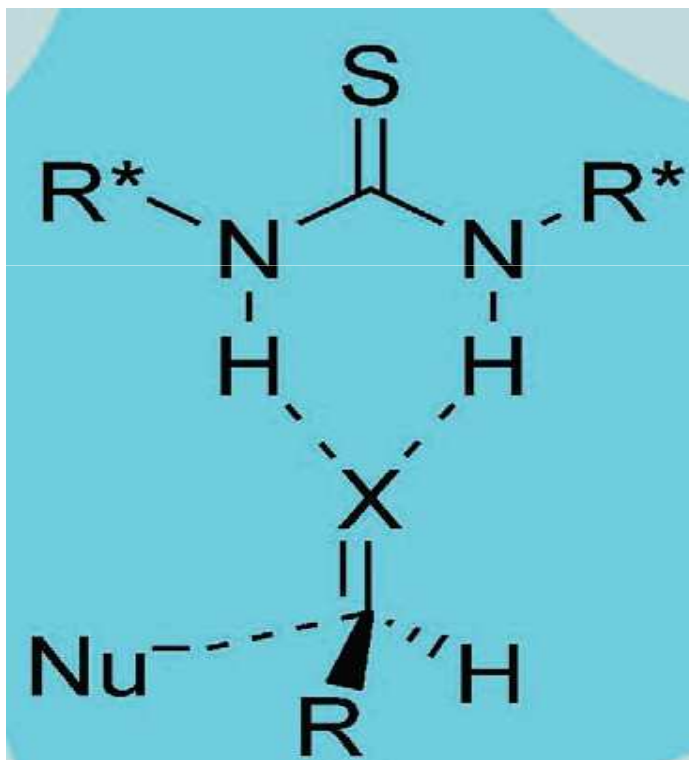
<sup>[d]</sup> Calculated by comparing the optical rotation.

<sup>[e]</sup> Not determined



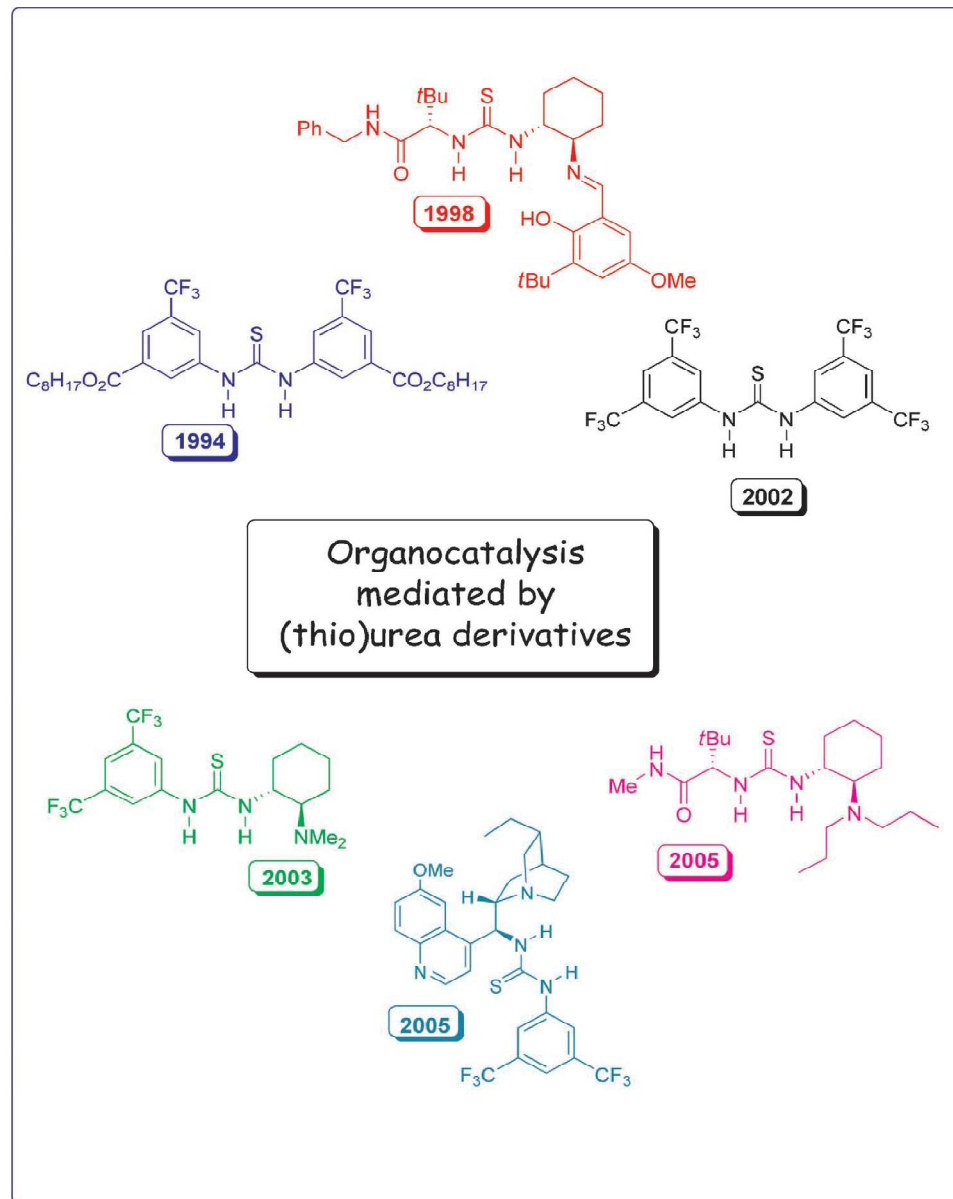
## Multifunctional Chiral Phosphine Catalysts in Aza-Morita-Baylis-Hillman Reaction:

(thio-)urea group might also give high catalytic activity and good asymmetric induction, because the acidic NH protons provide good opportunity to form a hydrogen bond, which may stabilize certain intermediates



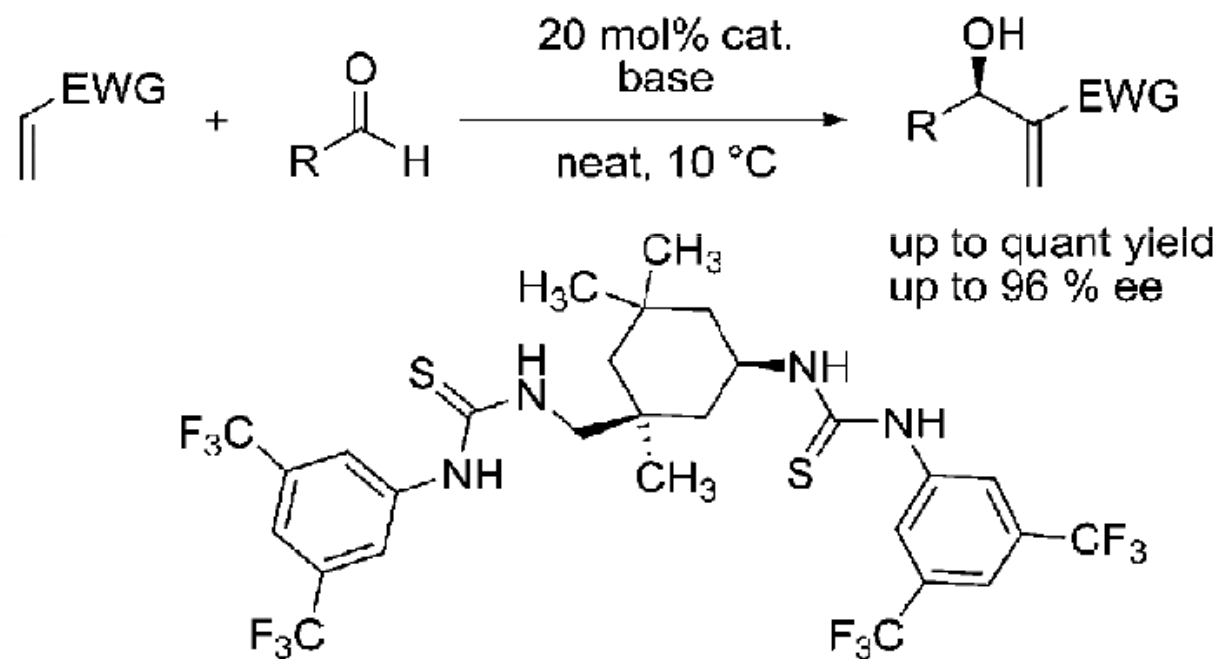
# Multifunctional Chiral Phosphine Catalysts in Aza-Morita-Baylis-Hillman Reaction:

Organocatalysis mediated by (thio) urea derivatives



## Multifunctional Chiral Phosphine Catalysts in Aza-Morita-Baylis-Hillman Reaction:

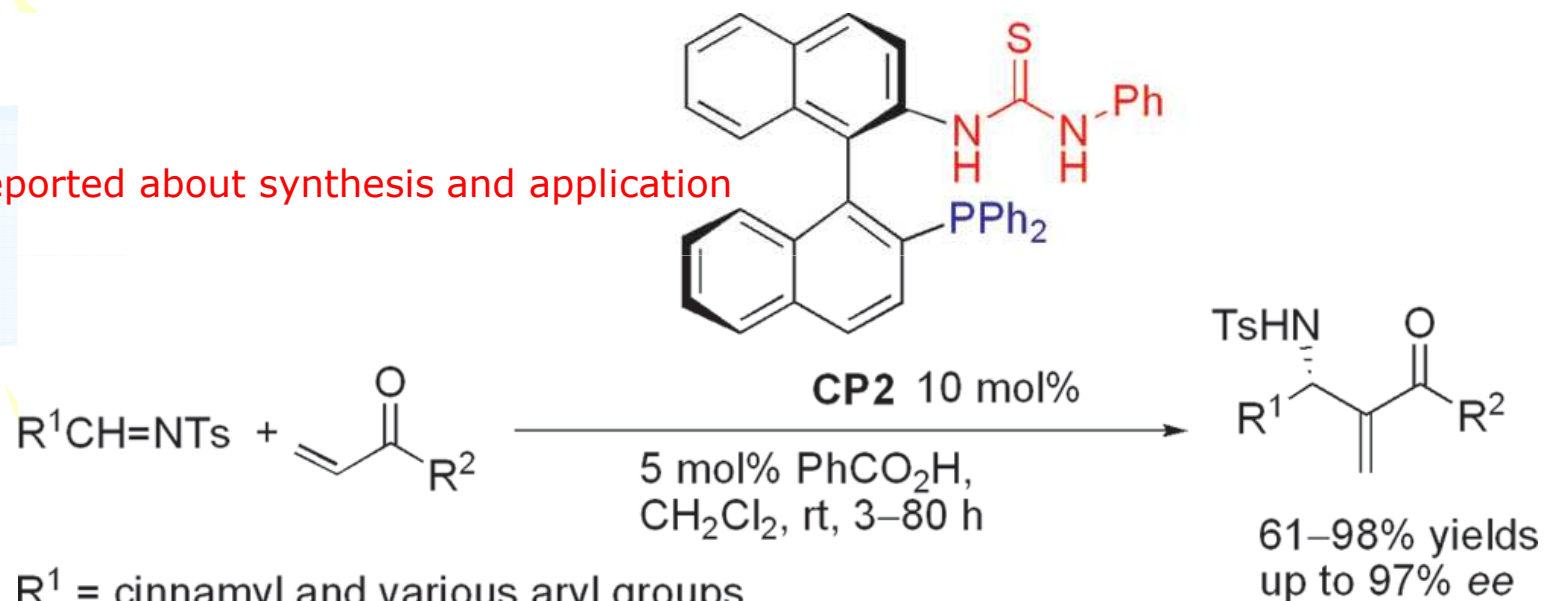
In Baylis Hillman reaction previously used organocatalyst which has thio urea functional group and give quantitative yield and high ee.



## Multifunctional Chiral Phosphine Catalysts in Aza-Morita-Baylis-Hillman Reaction:

The chiral thiourea-phosphine **CP2** in combination with benzoic acid was a very successful catalytic system for the aza-MBH reaction of *N*-tosyl imines with MVK, PVK, EVK.

First reported about synthesis and application



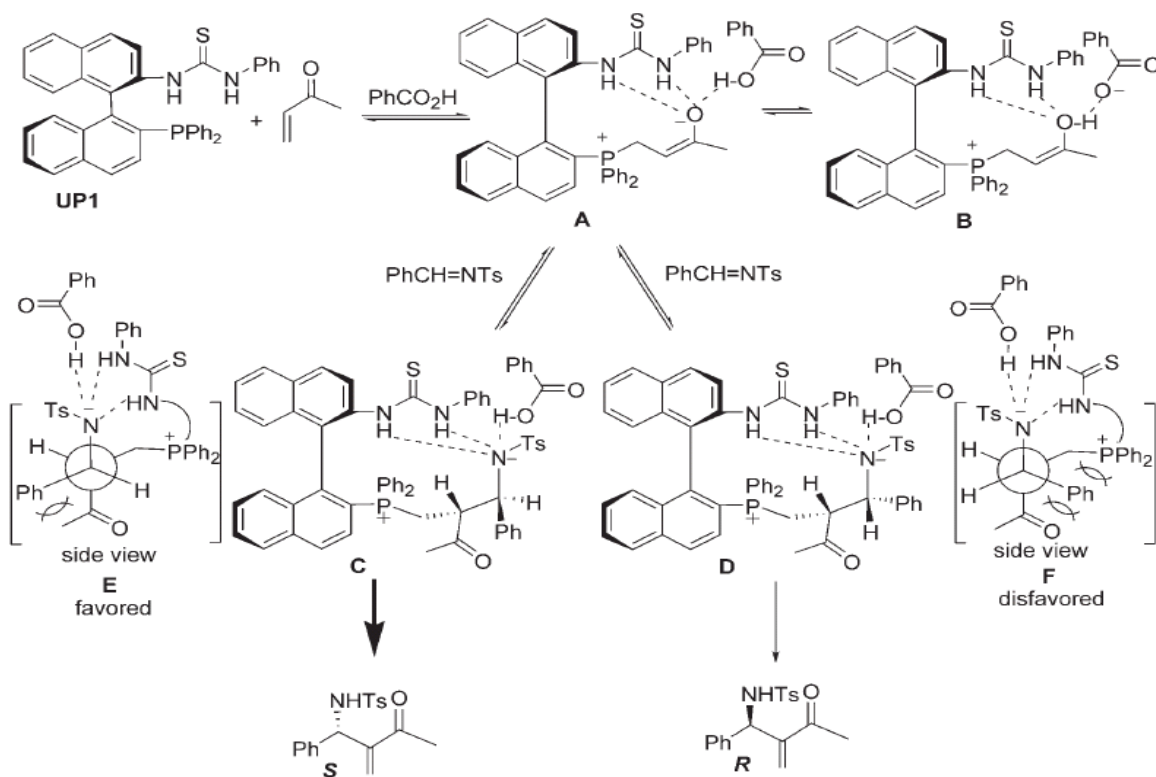
R<sup>1</sup> = cinnamyl and various aryl groups.

R<sup>2</sup> = H, Me, Et, Ph.

# Multifunctional Chiral Phosphine Catalysts in Aza-Morita-Baylis-Hillman Reaction:

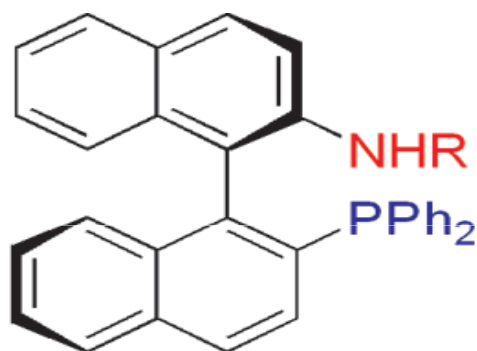
A new kind of bifunctional (thio)ureaphosphine catalyst.

A plausible reaction mechanism.



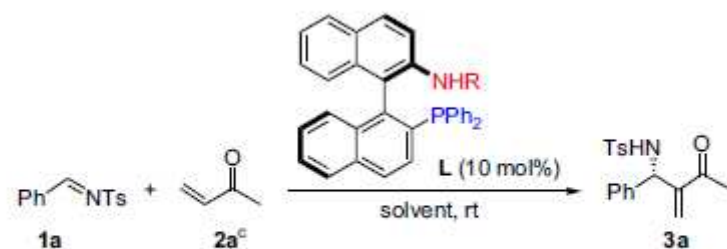
## Multifunctional Chiral Phosphine Catalysts in Aza-Morita-Baylis-Hillman Reaction:

In order to further improve the catalytic activity and enantioselectivity, designed and synthesized a series of bifunctional chiral phosphine amides

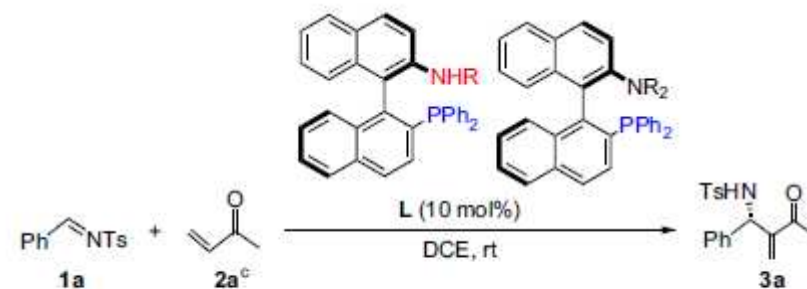


**CP3:** R = SO<sub>2</sub>CH<sub>3</sub>; **CP4:** R = SO<sub>2</sub>CF<sub>3</sub>,  
**CP5:** R = SO<sub>2</sub>C<sub>6</sub>H<sub>4</sub>CH<sub>3</sub>-*p*; **CP6:** R = COC<sub>6</sub>H<sub>5</sub>;  
**CP7:** R = COCH<sub>3</sub>; **CP8:** R = CO<sub>2</sub>CH<sub>3</sub>; **CP9:** R = PO(C<sub>6</sub>H<sub>5</sub>)<sub>2</sub>

# Multifunctional Chiral Phosphine Catalysts in Aza-Morita-Baylis-Hillman Reaction:

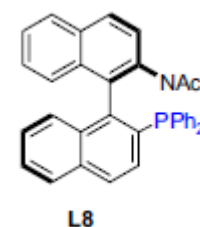


Entry	Catalyst	Solvent	Time (h)	Yield <sup>a</sup> (%)	ee <sup>b</sup> (%)
1	L1	DCM	72	99	75
2	L1	PhMe	72	99	51
3	L1	THF	120	49	45
4	L1	CH <sub>3</sub> CN	74	99	75
5	L1	CHCl <sub>3</sub>	72	99	82
6 <sup>d</sup>	L1	DCM	48	99	61
7 <sup>e</sup>	L1	DCM	90	96	90
8	L5	DCM	24	99	88
9 <sup>e</sup>	L5	DCM	24	99	95
10	L5	PhMe	160	90	27
11	L5	THF	160	91	39
12	L5	DMSO	36	89	52
13	L5	CH <sub>3</sub> CN	48	94	79
14 <sup>d</sup>	L5	DCM	36	99	86
15 <sup>g</sup>	L5	DCM	20	95	91
16 <sup>f</sup>	L5	DCM	23	89	96

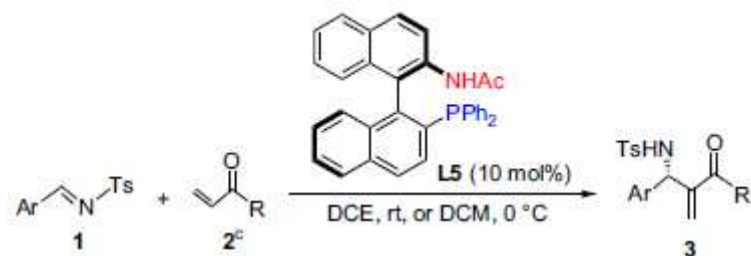


Entry	Catalyst	Time (h)	Yield <sup>a</sup> (%)	ee <sup>b</sup> (%)
1	L1	48	95	89
2	L2	48	0	—
3	L3	96	94	86
4	L4	48	89	78
5	L5	48	99	93
6	L6	60	91	59
7	L7	96	85	72
8	L8	96	0	—

- L1: R = SO<sub>2</sub>CH<sub>3</sub>
- L2: R = SO<sub>2</sub>CF<sub>3</sub>
- L3: R = SO<sub>2</sub>C<sub>6</sub>H<sub>4</sub>CH<sub>3</sub>-*p*
- L4: R = COC<sub>6</sub>H<sub>5</sub>
- L5: R = COCH<sub>3</sub>
- L6: R = CO<sub>2</sub>CH<sub>3</sub>
- L7: R = PO(C<sub>6</sub>H<sub>5</sub>)<sub>2</sub>



# Multifunctional Chiral Phosphine Catalysts in Aza-Morita-Baylis-Hillman Reaction:

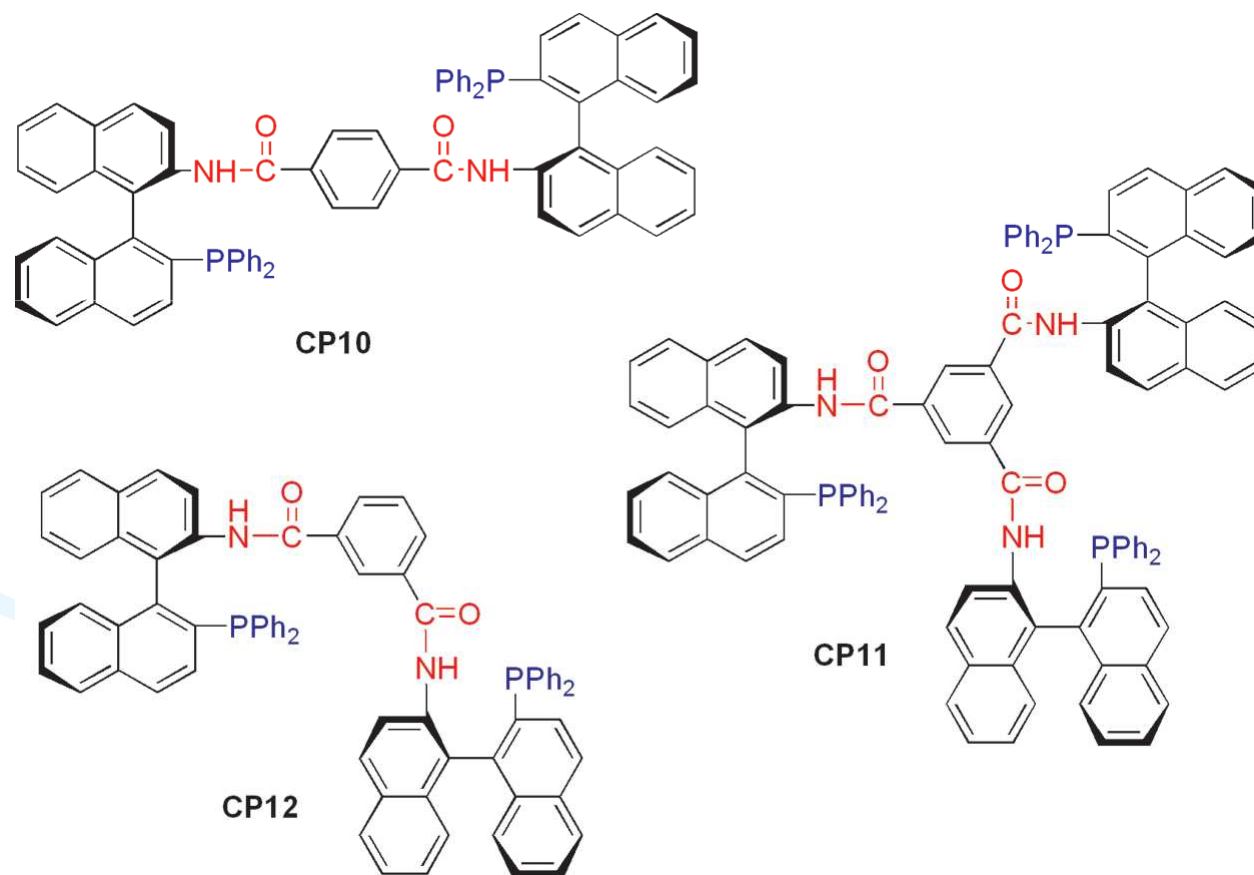


Entry	Ar	R	DCE at rt			DCM at 0 °C		
			Time (h)	Yield <sup>a</sup> (%)	ee <sup>b</sup> (%)	Time (h)	Yield <sup>a</sup> (%)	ee <sup>b</sup> (%)
1	<i>p</i> -BrC <sub>6</sub> H <sub>4</sub> <b>1b</b>	Me	48	<b>3b</b> , 98	82	24	<b>3b</b> , 86	90
2	<i>p</i> -NO <sub>2</sub> C <sub>6</sub> H <sub>4</sub> <b>1c</b>	Me	24	<b>3c</b> , 75	80	7	<b>3c</b> , 99	90
3	<i>o</i> -NO <sub>2</sub> C <sub>6</sub> H <sub>4</sub> <b>1d</b>	Me	12	<b>3d</b> , 78	46	28	<b>3d</b> , 85	61
4	<i>m</i> -NO <sub>2</sub> C <sub>6</sub> H <sub>4</sub> <b>1e</b>	Me	12	<b>3e</b> , 99	80	8	<b>3e</b> , 99	82
5	<i>o</i> -ClC <sub>6</sub> H <sub>4</sub> <b>1f</b>	Me	12	<b>3f</b> , 76	65	48	<b>3f</b> , 80	65
6	<i>p</i> -ClC <sub>6</sub> H <sub>4</sub> <b>1g</b>	Me	36	<b>3g</b> , 99	82	24	<b>3g</b> , 99	90
7	<i>m</i> -ClC <sub>6</sub> H <sub>4</sub> <b>1h</b>	Me	12	<b>3h</b> , 89	84	24	<b>3h</b> , 90	89
8	<i>p</i> -FC <sub>6</sub> H <sub>4</sub> <b>1i</b>	Me	36	<b>3i</b> , 99	85	24	<b>3i</b> , 90	91
9	<i>m</i> -FC <sub>6</sub> H <sub>4</sub> <b>1j</b>	Me	20	<b>3j</b> , 77	75	24	<b>3j</b> , 91	80
10	<i>p</i> -MeOC <sub>6</sub> H <sub>4</sub> <b>1k</b>	Me	48	<b>3k</b> , 99	46	48	<b>3k</b> , 98	82
11	<i>p</i> -MeC <sub>6</sub> H <sub>4</sub> <b>1l</b>	Me	60	<b>3l</b> , 99	84	48	<b>3l</b> , 99	90
12	C <sub>6</sub> H <sub>5</sub> <b>1a</b>	Et				60	<b>3m</b> , 95	74
13	<i>p</i> -BrC <sub>6</sub> H <sub>4</sub> <b>1b</b>	Et				60	<b>3n</b> , 80	65
14	<i>p</i> -ClC <sub>6</sub> H <sub>4</sub> <b>1g</b>	Et				60	<b>3o</b> , 86	51



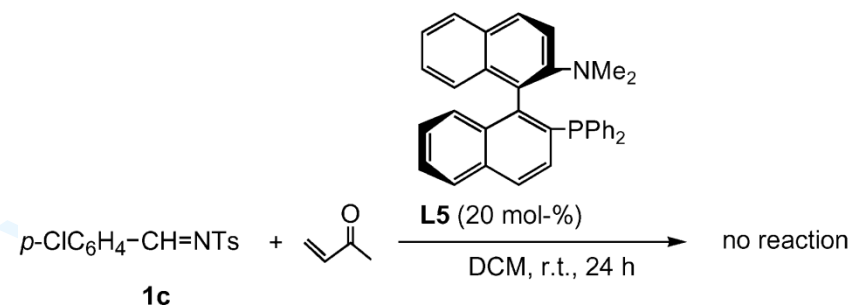
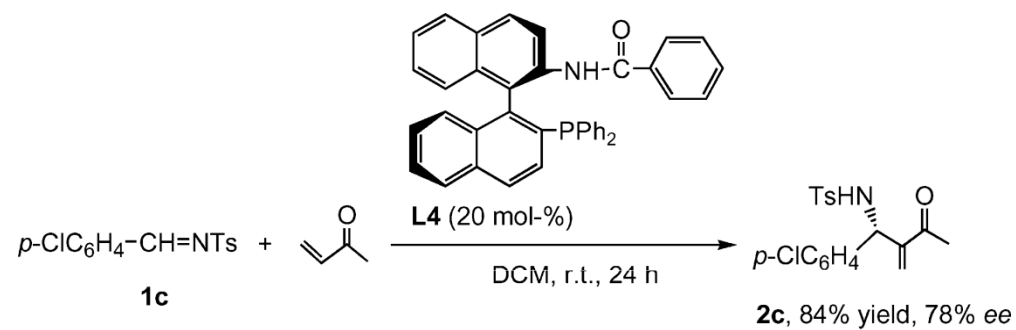
## Multifunctional Chiral Phosphine Catalysts in Aza-Morita-Baylis-Hillman Reaction:

Three sterically congested bifunctional chiral phosphane-amides, **synthesized** in order to evaluate the steric effect for asymmetric induction.



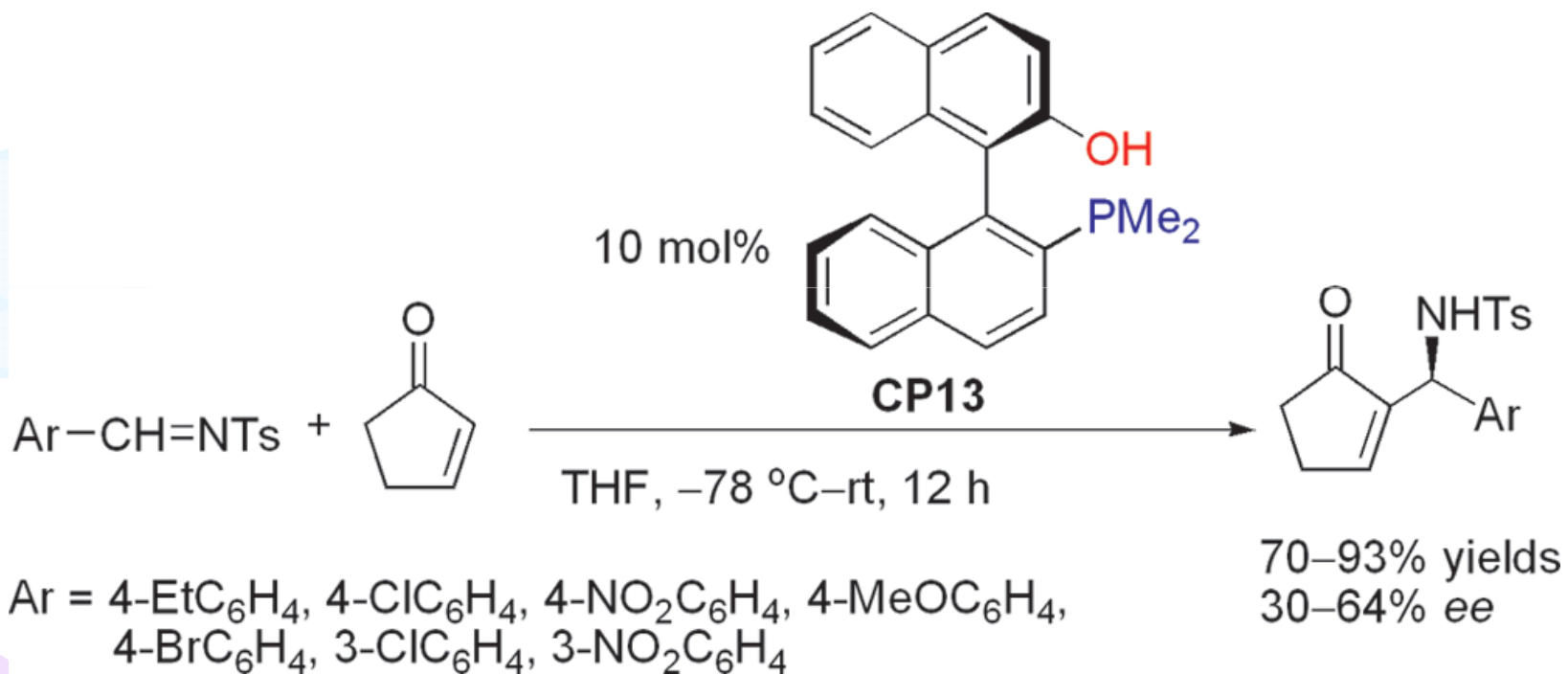
# Multifunctional Chiral Phosphine Catalysts in Aza-Morita-Baylis-Hillman Reaction:

The chiral sterically congested phosphane-amide bifunctional phosphanes



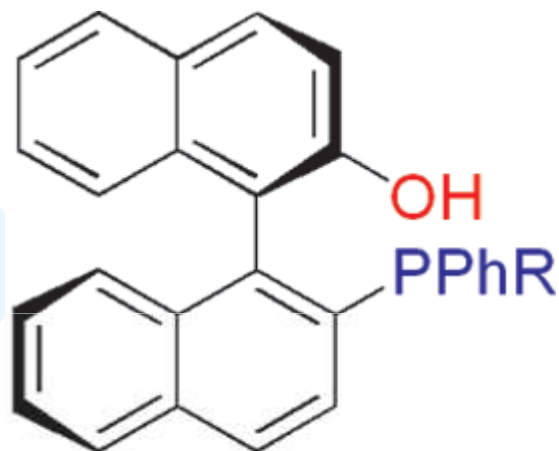
## Multifunctional Chiral Phosphine Catalysts in Aza-Morita-Baylis-Hillman Reaction:

The nucleophilicity of the phosphorus center in the catalyst may affect catalytic activity.



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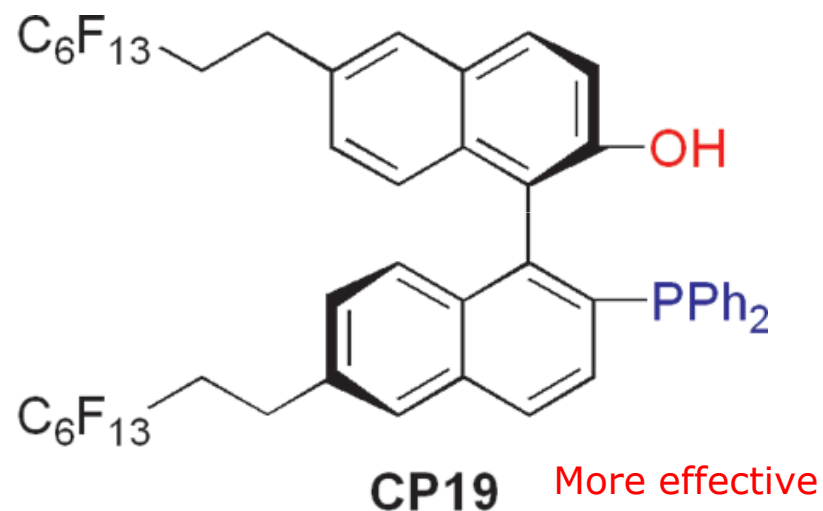
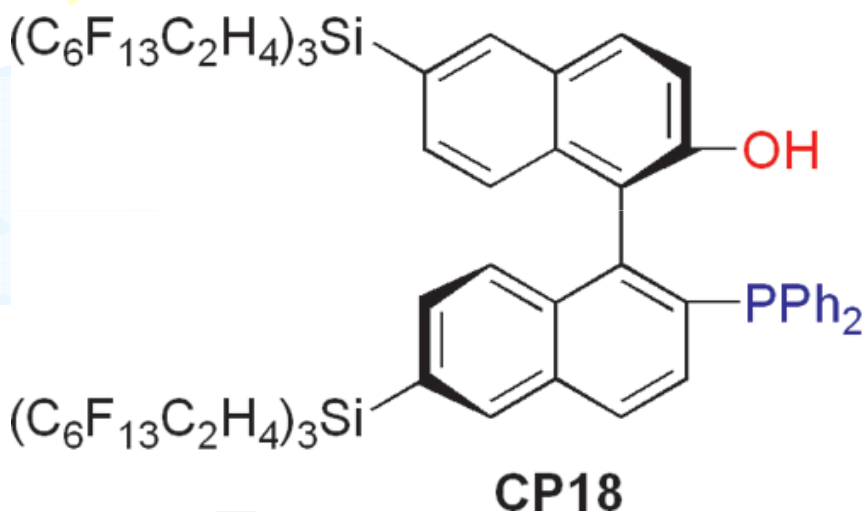


**CP14:** R = Et  
**CP15:** R = *i*Pr  
**CP16:** R = *n*Bu  
**CP17:** R = Cy

Structures of more nucleophilic phosphane-phenol type bifunctional chiral phosphines.

## Multifunctional Chiral Phosphine Catalysts in Aza-Morita-Baylis-Hillman Reaction:

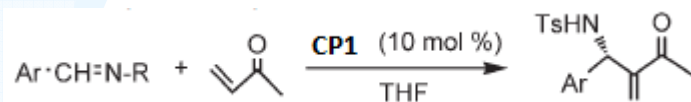
A few reports have demonstrated that introducing a longchainalkyl group in a variety of chiral ligands could improve the catalytic activity and enantioselectivity in homogeneous asymmetric.



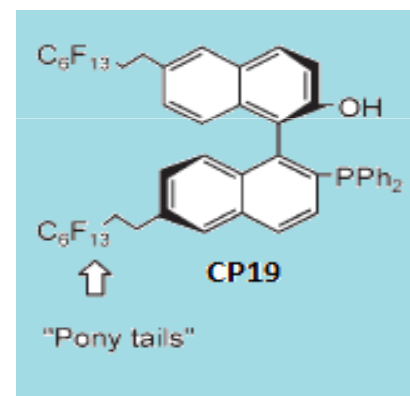
Phosphane-phenol-type bifunctional chiral phosphines bearing perfluoroalkane chains.

## Multifunctional Chiral Phosphine Catalysts in Aza-Morita-Baylis-Hillman Reaction:

catalyst **CP19** was more effective in the aza-MBH reaction of *N*-tosyl imines with MVK than the previously reported original chiral phosphine **CP1**.

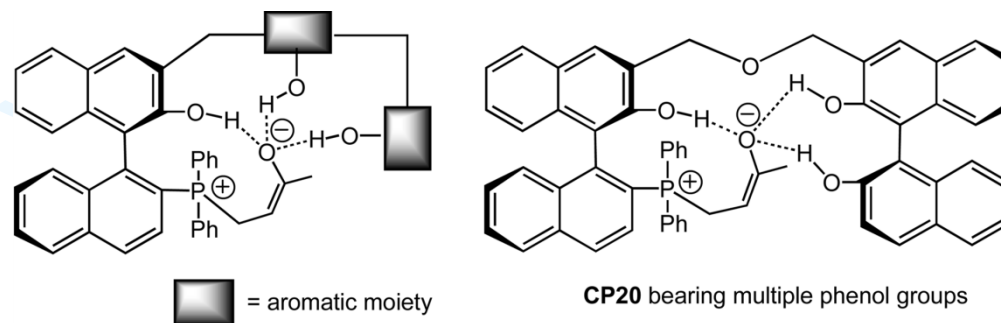
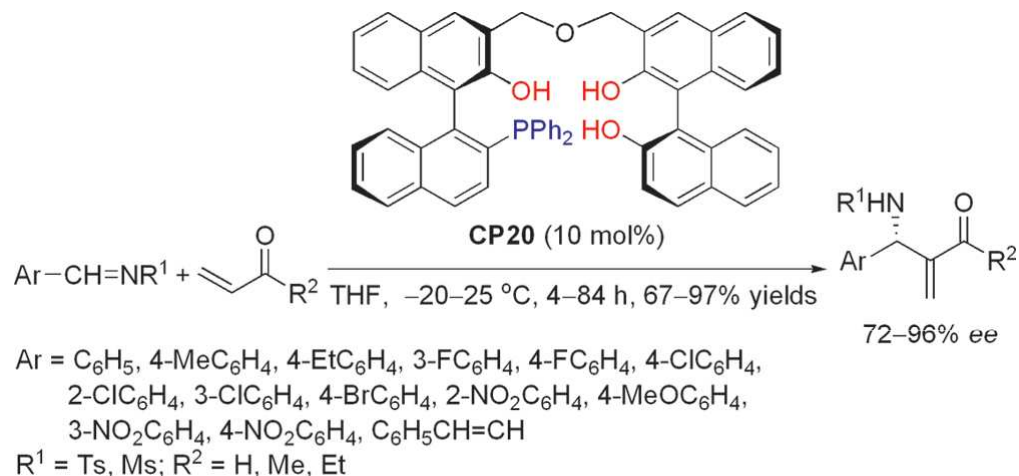


Up to 88% yield, and 82% ee.

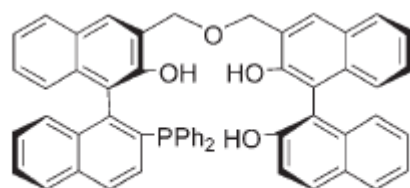


## Multifunctional Chiral Phosphine Catalysts in Aza-Morita-Baylis-Hillman Reaction:

Another approach to improve the catalytic activity and enantioselectivity is to increase the number of hydrogen bond donors in the bifunctional chiral phosphines.

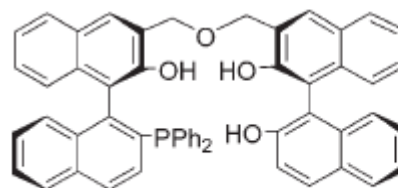


# Multifunctional Chiral Phosphine Catalysts in Aza-Morita-Baylis-Hillman Reaction:



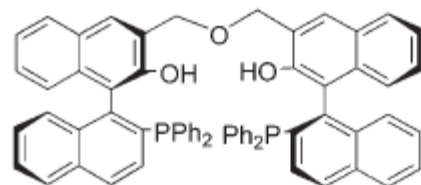
**(*R,R*)-CPLB1**

12 h, yield: 98%, ee: 92%



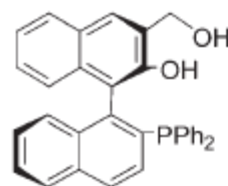
**(*R,S*)-CPLB2**

12 h, yield: 90%, ee: 90%



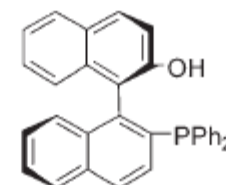
**(*R,R*)-CPLB3**

12 h, yield: 94%, ee: 92%



**(*R*)-CPLB4**

12 h, yield: 96%, ee: 91%



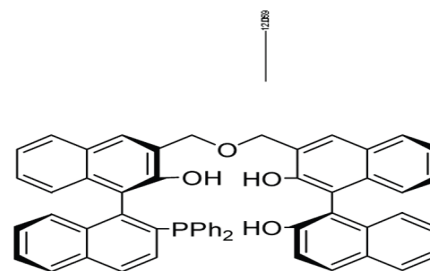
**(*R*)-CPLB5**

12 h, yield: 86%, ee: 84%

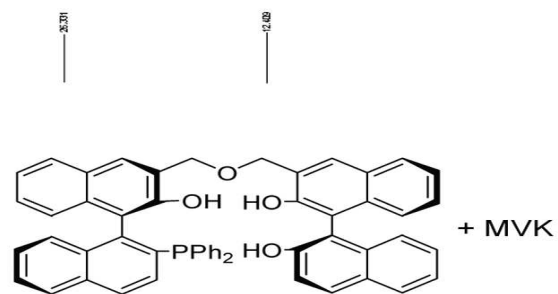
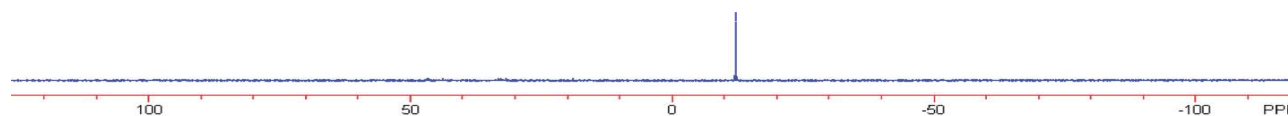


# Multifunctional Chiral Phosphine Catalysts in Aza-Morita-Baylis-Hillman Reaction:

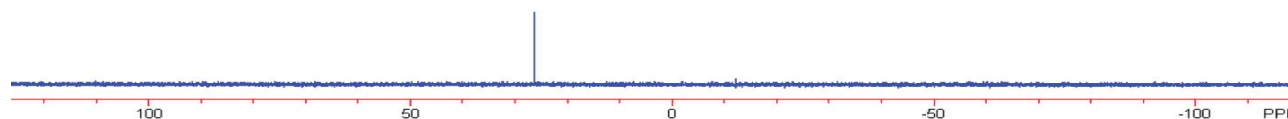
The  $^{31}\text{P}$  NMR measurements for CP20 and the mixture of CP20 and MVK



$\delta -12.1$

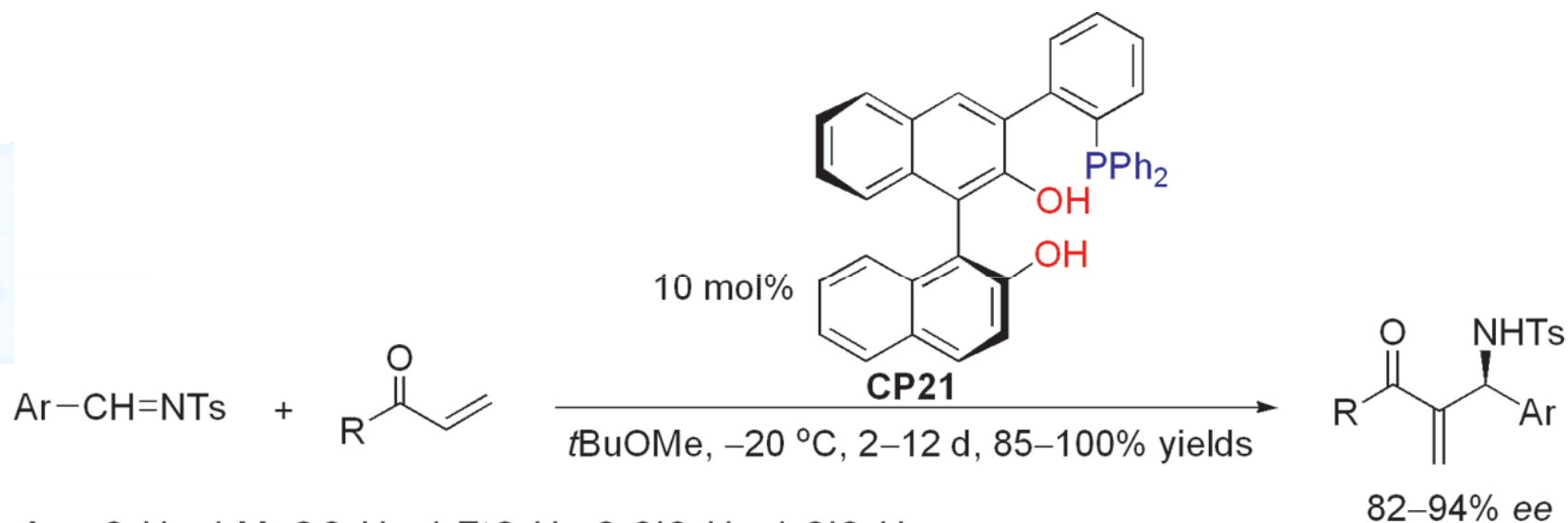


$\delta +26.4$



## Multifunctional Chiral Phosphine Catalysts in Aza-Morita-Baylis-Hillman Reaction:

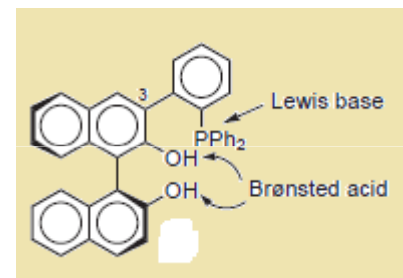
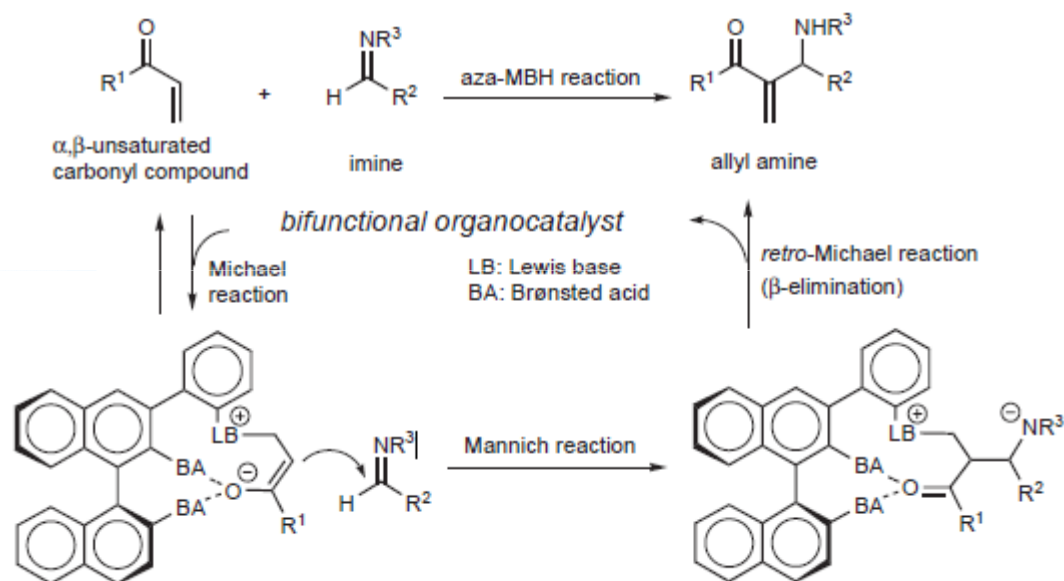
Catalyst **CP21** could effectively catalyze asymmetric aza-MBH reaction of *N*-tosyl imines with vinyl ketones



Ar = C<sub>6</sub>H<sub>5</sub>, 4-MeOC<sub>6</sub>H<sub>4</sub>, 4-EtC<sub>6</sub>H<sub>4</sub>, 2-ClC<sub>6</sub>H<sub>4</sub>, 4-ClC<sub>6</sub>H<sub>4</sub>,  
4-NO<sub>2</sub>C<sub>6</sub>H<sub>4</sub>, 4-BrC<sub>6</sub>H<sub>4</sub>, 2-furyl, 1-naphthyl, 2-naphthyl  
R = Me, Et, Ph

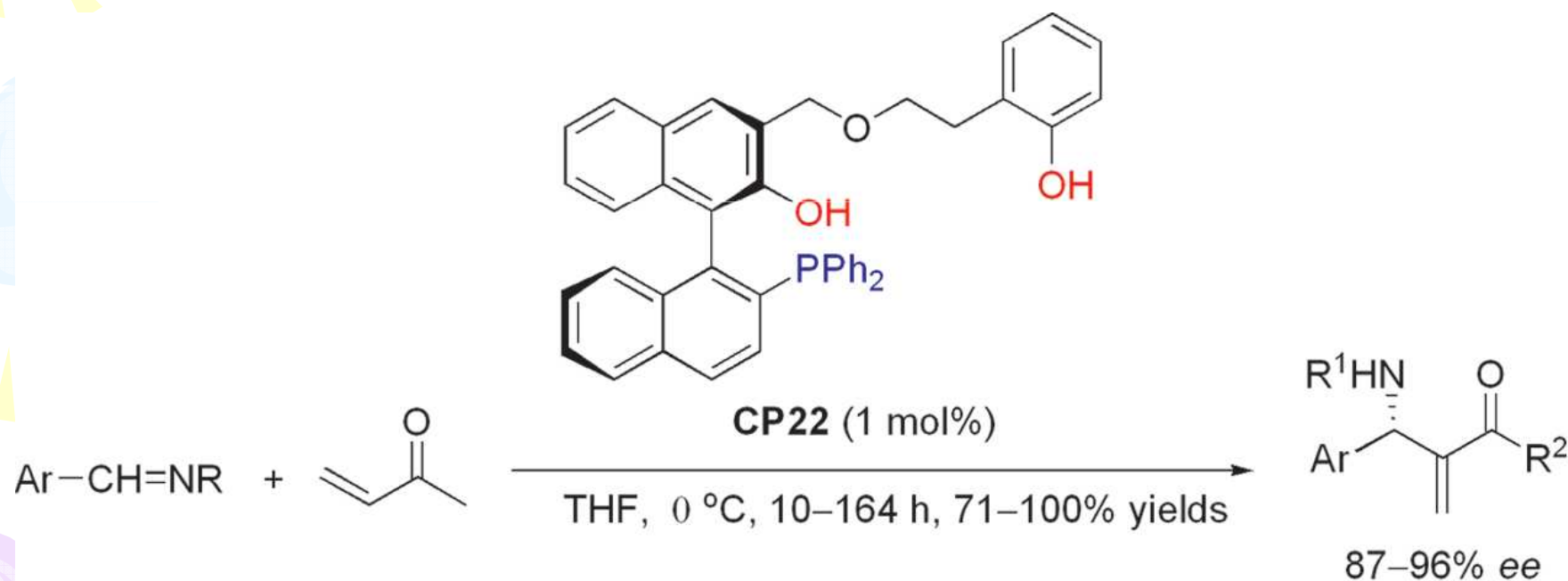
# Multifunctional Chiral Phosphine Catalysts in Aza-Morita-Baylis-Hillman Reaction:

The aza-MBH reaction promoted by asymmetric bifunctional catalysts



## Multifunctional Chiral Phosphine Catalysts in Aza-Morita-Baylis-Hillman Reaction:

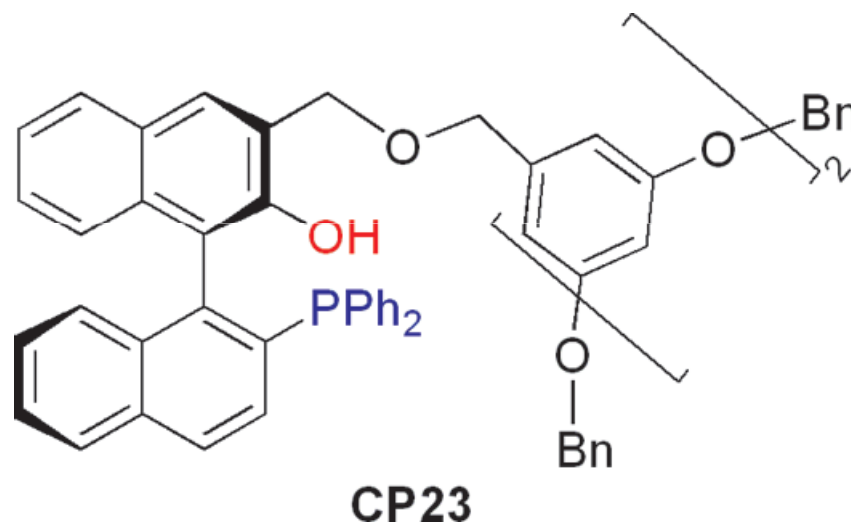
biphenol-based bifunctional catalyst **CP22** for aza- MBH reaction of *N*-tosyl imines with MVK High enantioselectivity up to 96% ee was achieved by **CP22** with catalyst loading of 1 mol %.



Ar = C<sub>6</sub>H<sub>5</sub>, 4-MeC<sub>6</sub>H<sub>4</sub>, 4-ClC<sub>6</sub>H<sub>4</sub>, 4-MeOC<sub>6</sub>H<sub>4</sub>,  
4-NO<sub>2</sub>C<sub>6</sub>H<sub>4</sub>, 2-Naphthyl, (*E*)-Cinnamyl  
R = Ts

## Multifunctional Chiral Phosphine Catalysts in Aza-Morita-Baylis-Hillman Reaction:

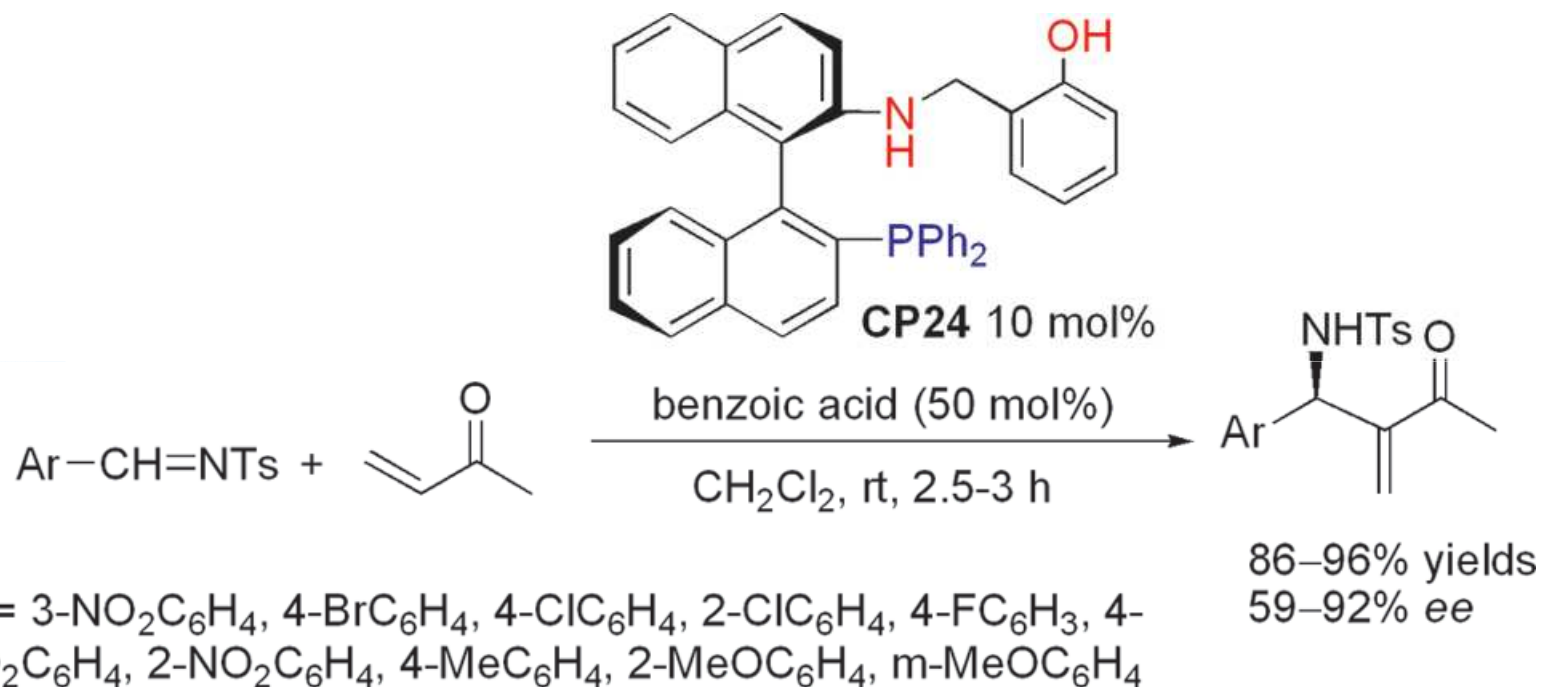
In order to recycle the catalyst, immobilized **CP1** on a series of dendrimers. It was found that the dendrimer-immobilized catalyst **CP23** was more effective than catalyst **CP1** for the aza-MBH reaction of *N*-sulfonylimines with MVK, EVK, or acrolein.



Dendrimer immobilized phosphine-phenol type of multifunctional chiral phosphines

# Multifunctional Chiral Phosphine Catalysts in Aza-Morita-Baylis-Hillman Reaction:

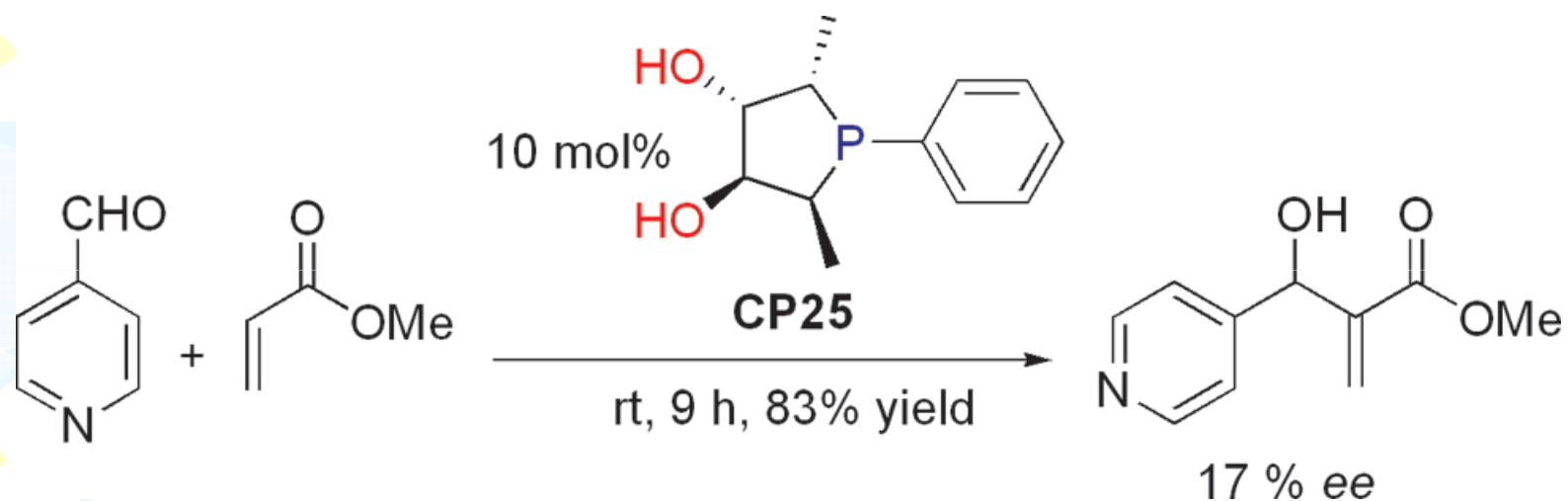
## Trifunctional Phosphine Organocatalyst-Promoted Aza- MBH Reaction



It was used for the first time to catalyze aza-Morita-Baylis-Hillman reactions between N-tosylimines and methyl vinyl ketone with fast reaction rates and good enantioselectivity at room temperature

## Multifunctional Chiral Phosphine Catalysts in Morita-Baylis-Hillman Reaction

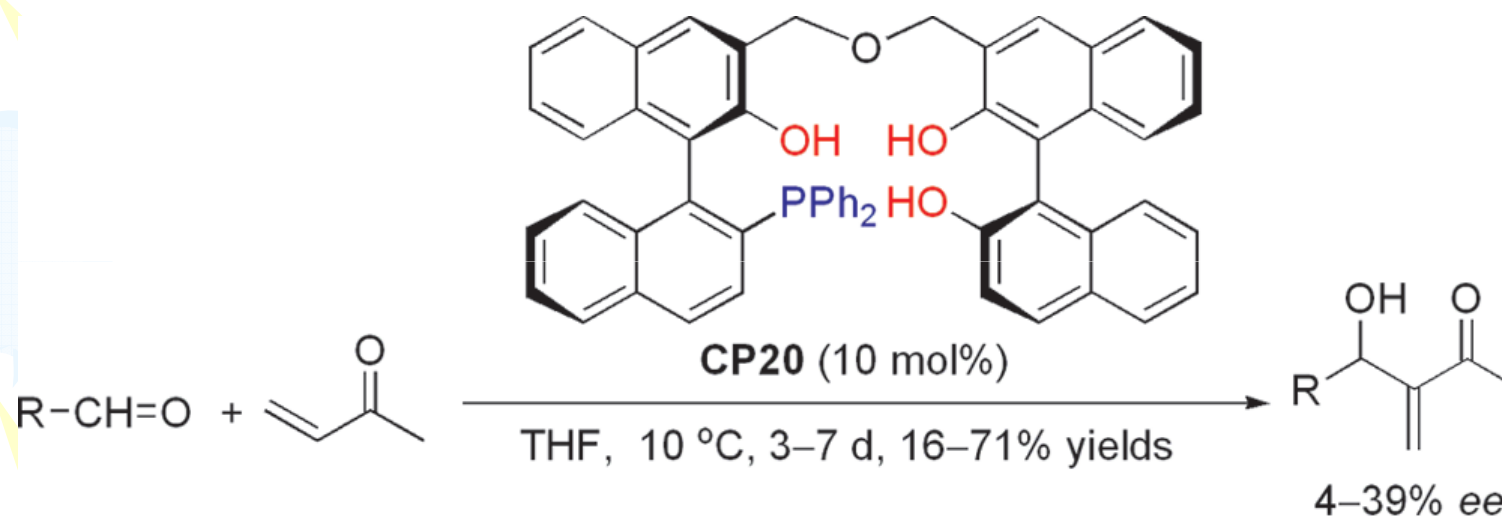
CP25 catalyzed MBH reaction between 4-pyridinecarbaldehyde and methyl acrylate



Phosphane-Hydroxy-Type Multifunctional Chiral Phosphine-Catalyzed Asymmetric MBH Reaction

## Multifunctional Chiral Phosphine Catalysts in Morita-Baylis-Hillman Reaction

Phosphane-Multiphenol Groups of Chiral Phosphines in Asymmetric MBH Reaction of Aldehydes with MVK

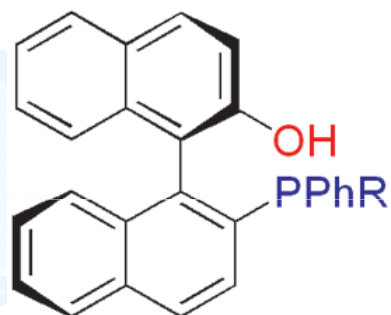


$R = \text{C}_6\text{H}_5, 4\text{-ClC}_6\text{H}_4, 2\text{-ClC}_6\text{H}_4,$   
 $2\text{-NO}_2\text{C}_6\text{H}_4, 3\text{-NO}_2\text{C}_6\text{H}_4, \text{C}_6\text{H}_5\text{CH}_2\text{CH}_2$

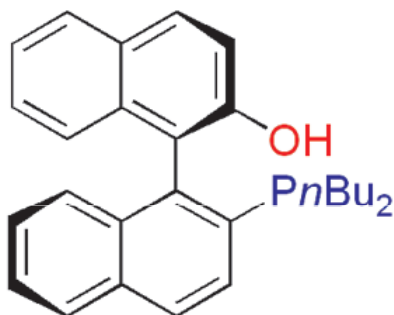


## Multifunctional Chiral Phosphine Catalysts in Morita-Baylis-Hillman Reaction

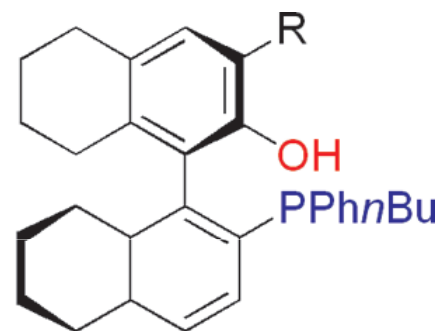
Unfortunately, the effective catalyst **CP1** for aza-MBH reaction did not show catalytic activity for the reaction of 3-phenylpropanal and MVK. **CP16** was still the most effective catalyst with respect to a wide range of substrates, affording the corresponding products in good yields with moderate ee's.



CP14: R = Et  
CP15: R = *i*Pr  
CP16: R = *n*Bu  
CP17: R = Cy  
CP26: R = *i*Bu



CP27

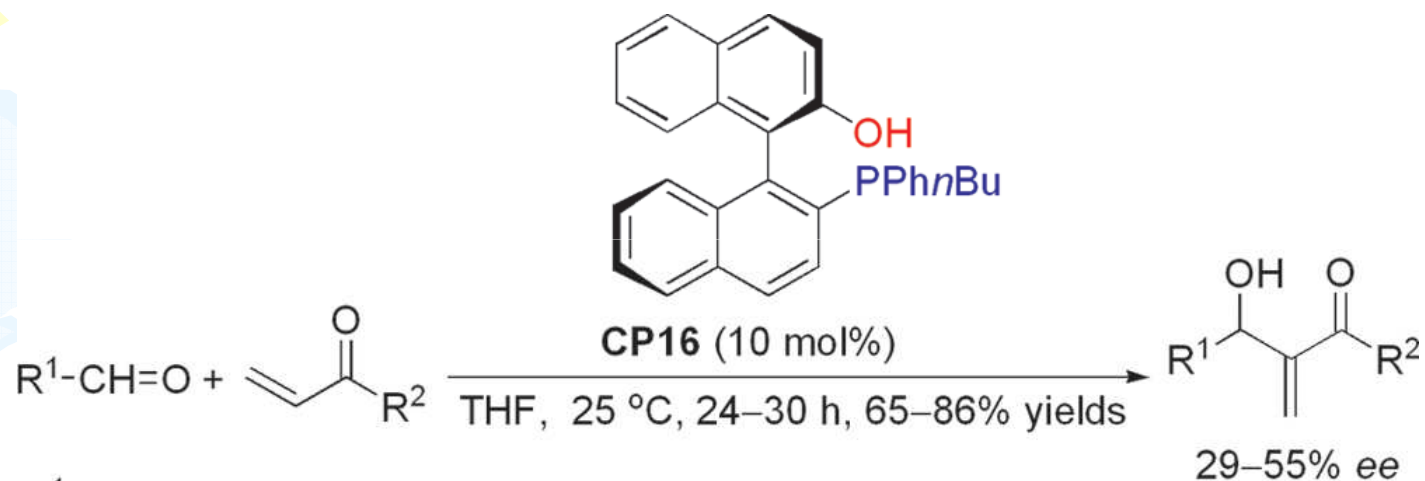


CP28: R = H  
CP29: R = Ph  
CP30: R = Br

Structures of more nucleophilic phosphine-phenol-type bifunctional chiral phosphines.

## Multifunctional Chiral Phosphine Catalysts in Morita-Baylis-Hillman Reaction

Phosphane-Phenol-Type Multifunctional Chiral Phosphine-Catalyzed Asymmetric MBH Reaction of Aldehydes with  $\alpha,\beta$ -Unsaturated Ketones

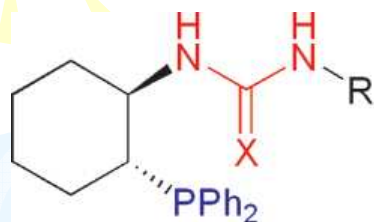


$R^1$  = various aliphatic or aromatic groups

$R^2$  = Me, Et

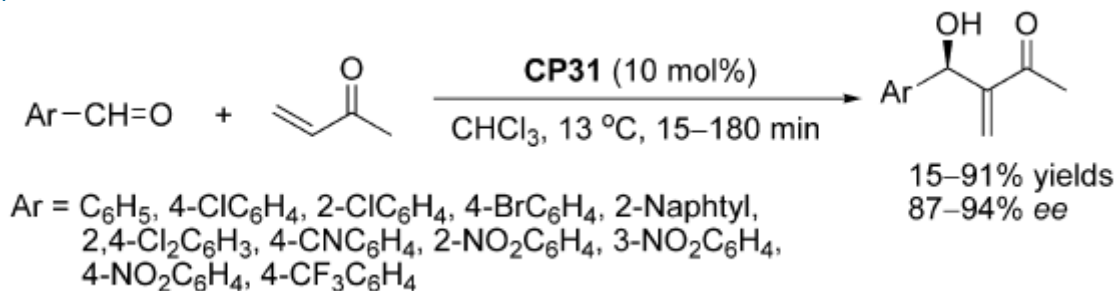
## Multifunctional Chiral Phosphine Catalysts in Morita-Baylis-Hillman Reaction

Recently, Wu's group reported a series of chiral phosphino( thio)ureas **CP31-CP36** derived from *trans*-2-amino-1-(diphenylphosphino)cyclohexane. **CP31 was the best** catalyst for the MBH reaction of various aromatic aldehydes with MVK giving the products with excellent enantiomeric excesses under mild conditions in relative short reaction time.



Structures of phosphine-(thio)urea multifunctional chiral phosphines.

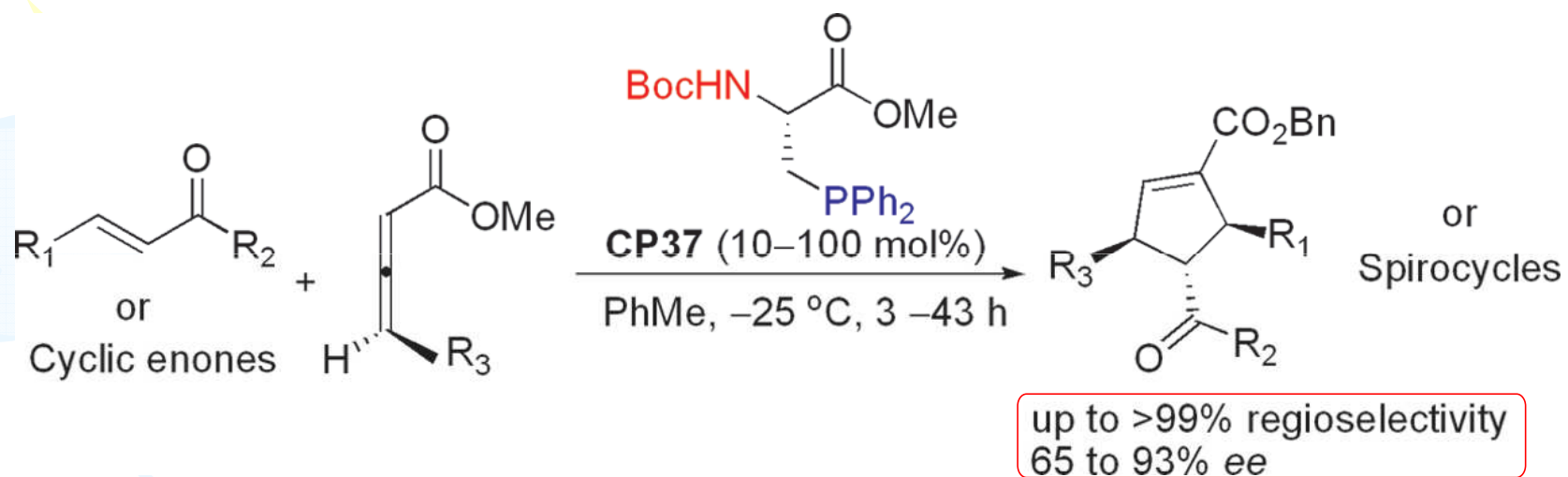
- CP31:** X = S, R = C<sub>6</sub>H<sub>5</sub>
- CP32:** X = O, R = C<sub>6</sub>H<sub>5</sub>
- CP33:** X = S, R = 3,5-(CF<sub>3</sub>)<sub>2</sub>C<sub>6</sub>H<sub>3</sub>
- CP34:** X = S, R = 4-ClC<sub>6</sub>H<sub>4</sub>
- CP35:** X = S, R = 4-MeOC<sub>6</sub>H<sub>4</sub>
- CP36:** X = S, R = *c*-Hexyl



**CP31 is the best** catalyst for the MBH reaction.  
*Products with excellent enantiomeric excesses*

## Multifunctional Chiral Phosphine Catalysts in Other Reactions

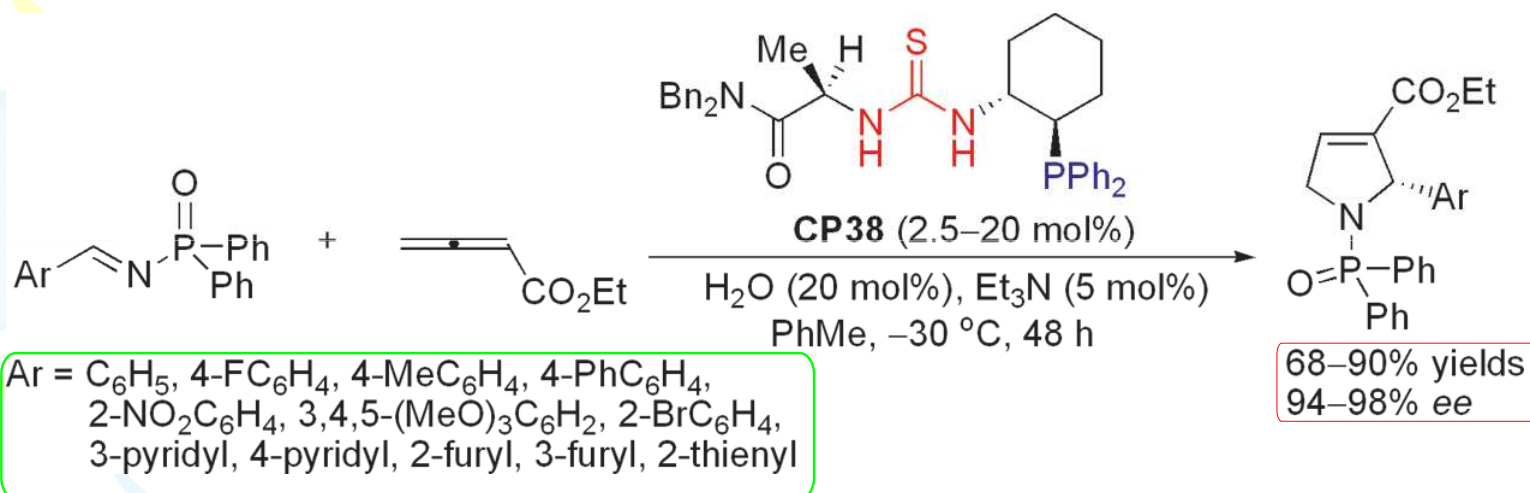
Since 2007, bifunctional chiral phosphines can be applied in reactions beyond MBH/aza-MBH reaction, such as enantioselective [3 + 2] cycloaddition reactions.



*Phosphine-amide-Type Multifunctional Chiral Phosphine Catalyzed Asymmetric [3 + 2]Cycloaddition Reaction*

## Multifunctional Chiral Phosphine Catalysts in Other Reactions

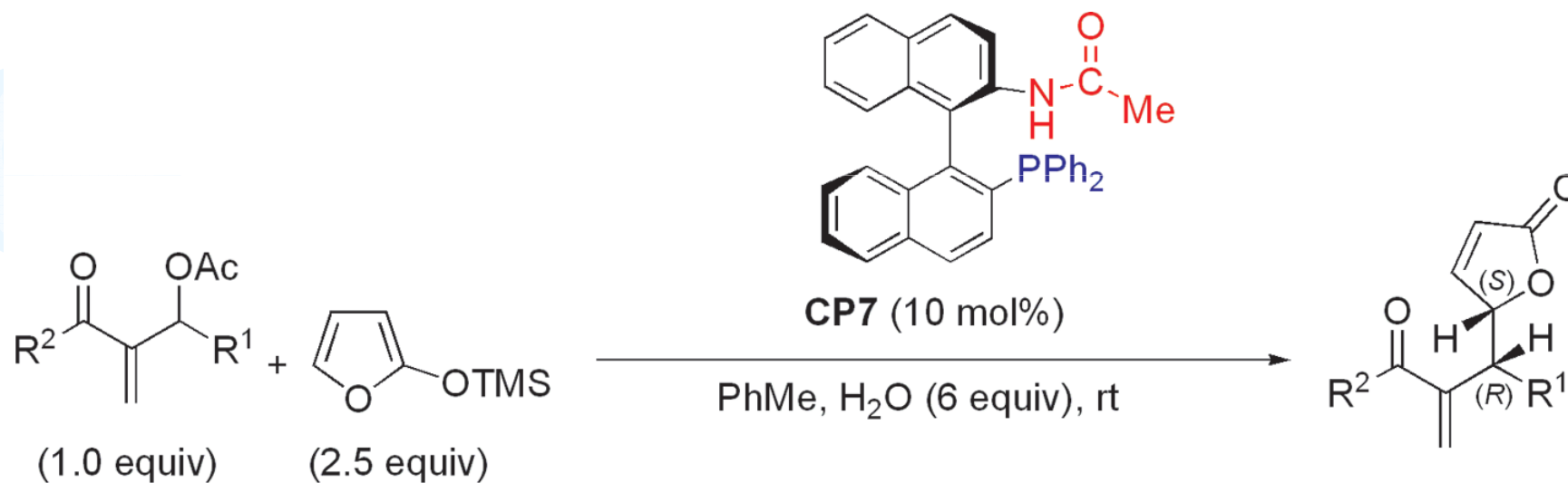
More recently, **Jacobsen** developed a series of bifunctional phosphorus thiourea derivatives for highly enantioselective synthesis of chiral dihydropyrroles via imine-allene [3 + 2] cycloaddition.



**CP38 as the best catalyst**

## Multifunctional Chiral Phosphine Catalysts in Other Reactions

Interestingly, previously developed catalyst **CP7** for aza-MBH reaction has a new application. Catalyst **CP7** achieved high yield and excellent ee for the reaction of MBH acetates with 2-trimethylsilyloxy furan, which is an effective approach for the asymmetric synthesis of  $\gamma$ -butenolides.

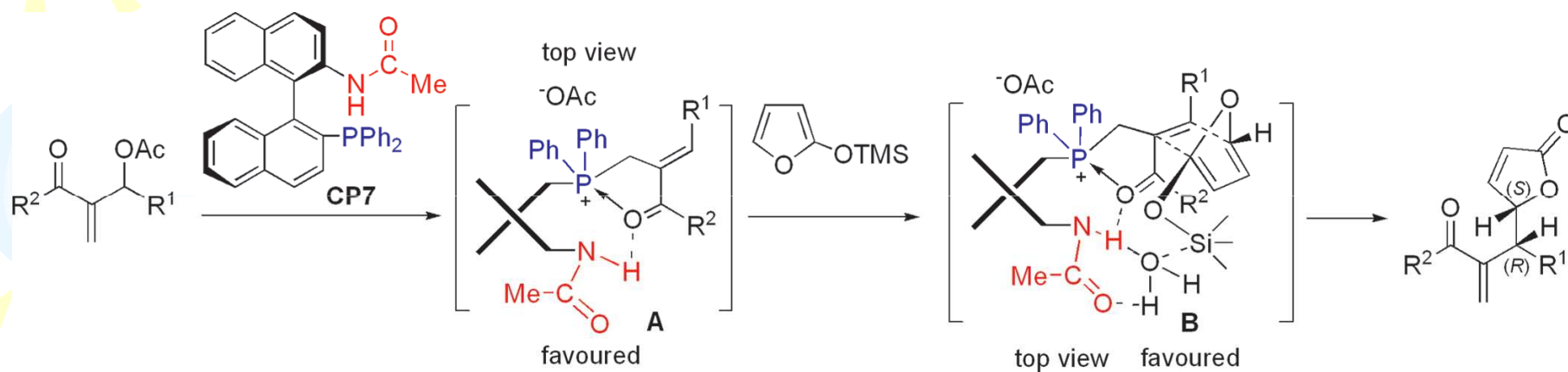


$R^1$  = aromatic group or C<sub>3</sub>H<sub>7</sub>,  $R^2$  = Me, Et, OMe

45-98% yields  
71-96% ee

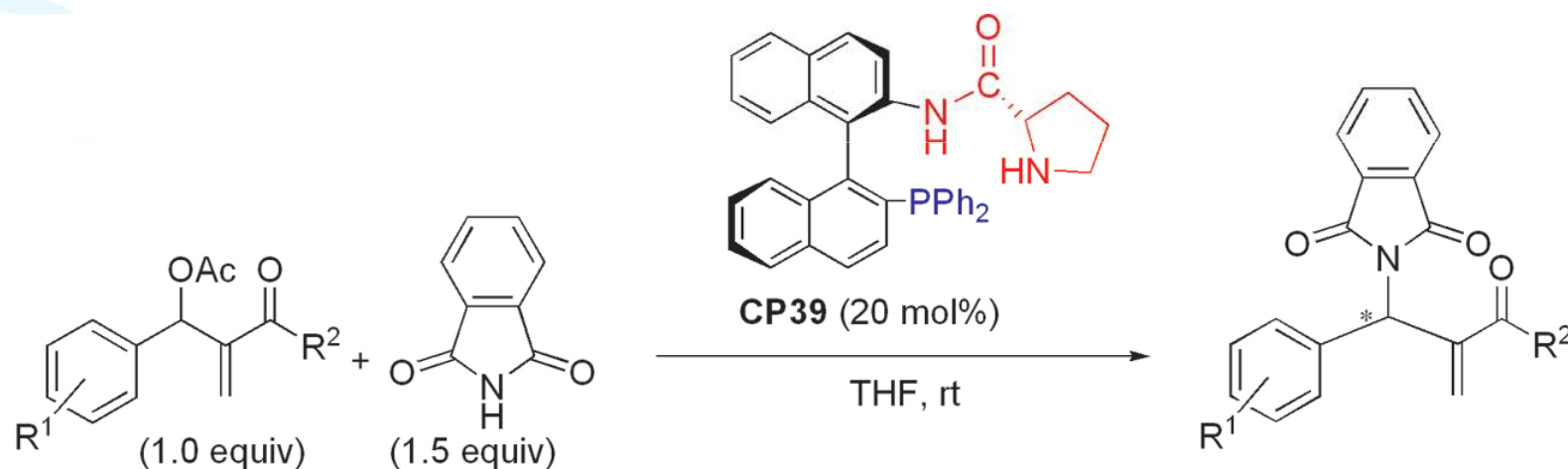
## Multifunctional Chiral Phosphine Catalysts in Other Reactions

### The proposed Mechanism



## Multifunctional Chiral Phosphine Catalysts in Other Reactions

### Chiral Phosphine-Catalyzed Regio- and Enantioselective Allylic Amination of Morita-Baylis-Hillman Acetates



$R^1 = \text{H}, m\text{-NO}_2, p\text{-Cl}, m\text{-Cl}, o\text{-Cl}, p\text{-Br}, p\text{-F}, p\text{-CF}_3,$   
 $p\text{-CN}, p\text{-Me}, m\text{-Me}, p\text{-NO}_2, R^2 = \text{Me}, \text{Et}$

70-91% yields  
34-58% ee  
the product in good yields  
with moderate ee

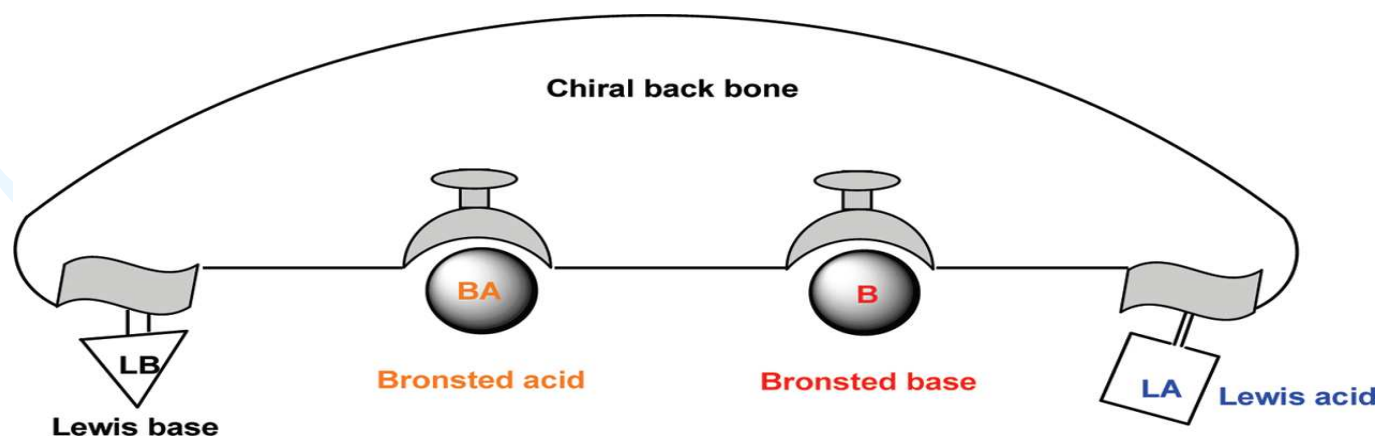


## Conclusion

Multifunctional/bifunctional chiral phosphine organocatalysts have established themselves as efficient enantioselective catalysts in catalytic asymmetric MBH and related reactions due to the combination of a hydrogen-bonding motif with a highly nucleophilic phosphorus center within one molecule.

The reactivities and enantioselectivities of these multifunctional/bifunctional chiral phosphine organocatalysts can be finely tuned through enhancing the reactive center's nucleophilicity and varying and increasing hydrogen bond donors.

Multifunctional catalysis will remain as a powerful strategy to inspire the design of new efficient and selective catalysts.





*Thank you*

Mahodand Lake: Pakistan