



Catalytic Intramolecular Ketone Alkylation with Olefins by Dual Activation

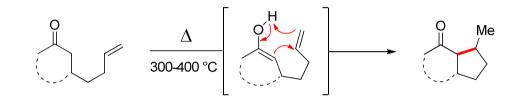
Hee Nam Lim and Guangbin Dong*

Angew. Chem. Int. Ed. 2015, 54, 15294-15298

Ophélie Quinonero 08/12/2015

Conia-ene type reactions

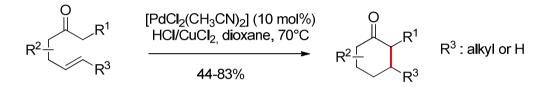
• Thermal Conia-Ene reaction



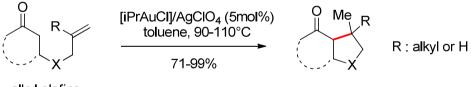
J.M. Conia, P. Le Perchec, Synthesis 1975, 1

Catalytic Conia-ene type reactions

• *π*-Acid-catalyzed Ketone-Ene Cyclization



S.Wang, T.Pei, X. Han, R.A. Widenhoefer, Org. Lett. 2003, 5, 2699

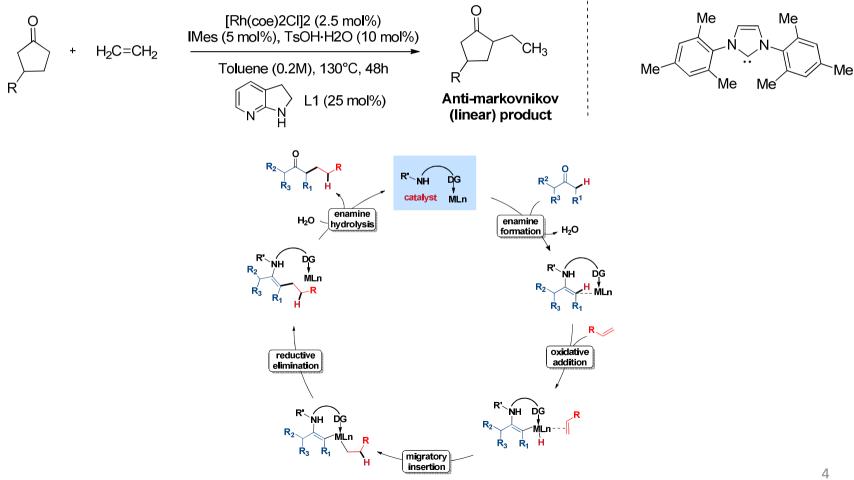


alkyl olefins

Y.-P. Xiao, X.-Y. Liu, C.-M. Che, Angew. Chem. Int. Ed. 2011, 50, 4937

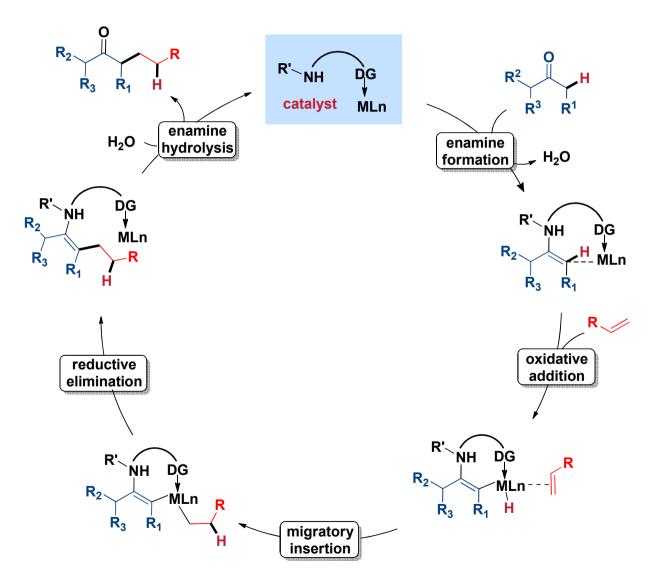
Intermolecular ketone α -alkylation reaction with simple olefins

• Bifunctionnal catalysis

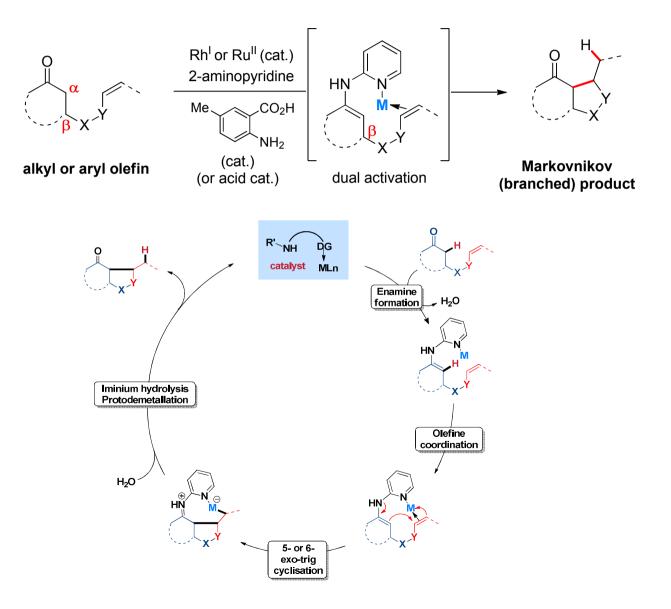


F. Mo, G. Dong, Science, 2014, 345, 68

Intermolecular ketone α -alkylation reaction with simple olefins

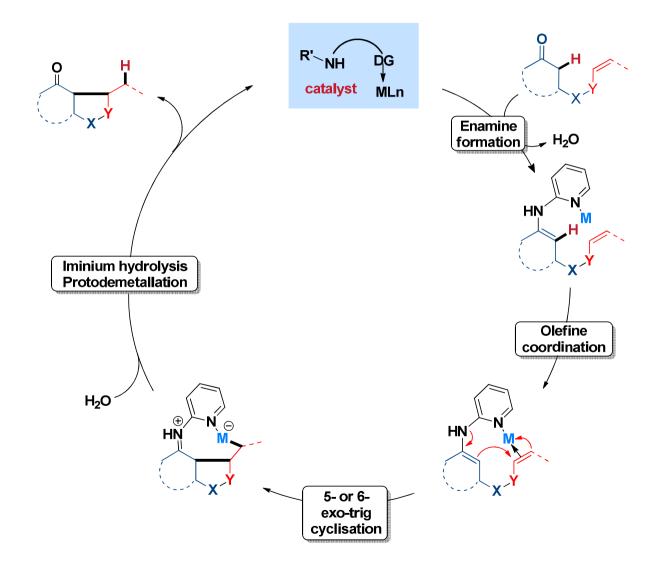


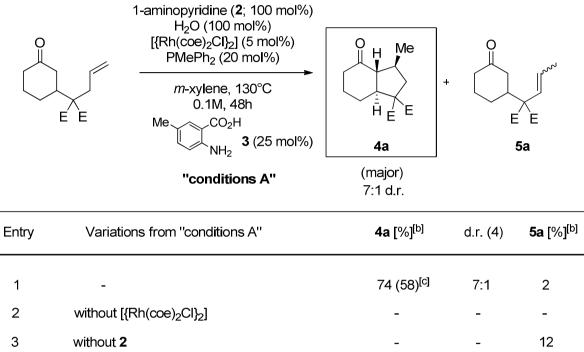
This work Conia-ene type reaction by dual activation



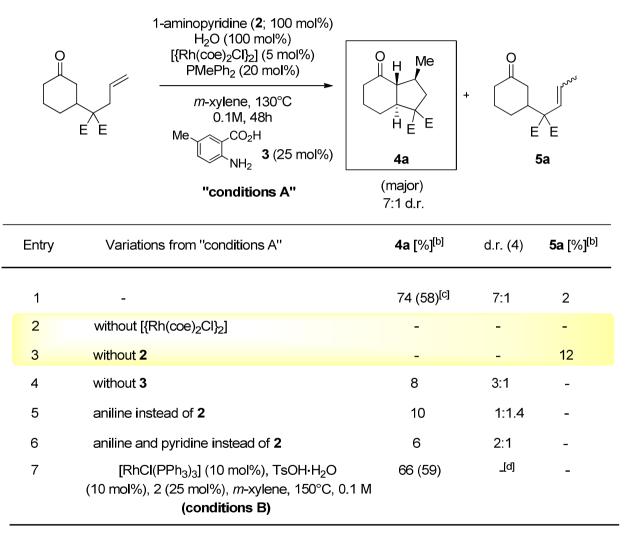
This work

Conia-ene type reaction by dual activation



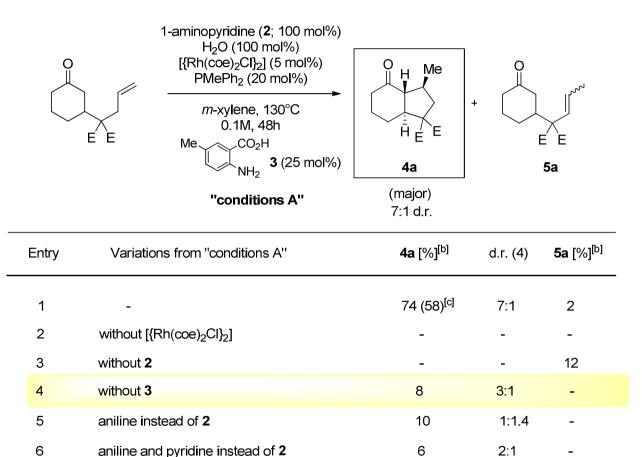


I	-	74 (50) ¹³	7.1	2	
2	without [{Rh(coe) ₂ Cl} ₂]	-	-	-	
3	without 2	-	-	12	
4	without 3	8	3:1	-	
5	aniline instead of 2	10	1:1.4	-	
6	aniline and pyridine instead of 2	6	2:1	-	
7	[RhCl(PPh ₃) ₃] (10 mol%), TsOH⋅H ₂ O (10 mol%), 2 (25 mol%), <i>m</i> -xylene, 150°C, 0.1 M (conditions B)	66 (59)	_[d]	-	



6

7



[a] All reactions were run on 0.1 mmol scale with 1.0 mL of the indicated solvent. [b] Determined by 1H NMR spectroscopy using 1,2-tetrachloroethane as the internal standard. [c] Yield of the isolated major diastereomer. [d] Single diastereomer. coe = cyclooctene.

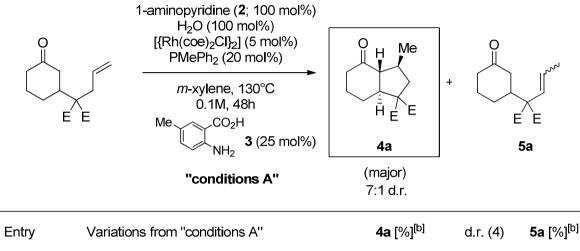
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(10 mol%), 2 (25 mol%), *m*-xylene, 150°C, 0.1 M (conditions B)

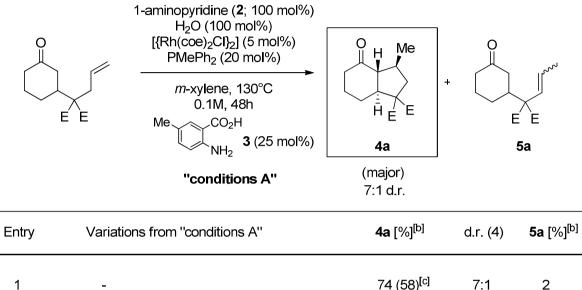
6

66 (59)

2:1 [d]

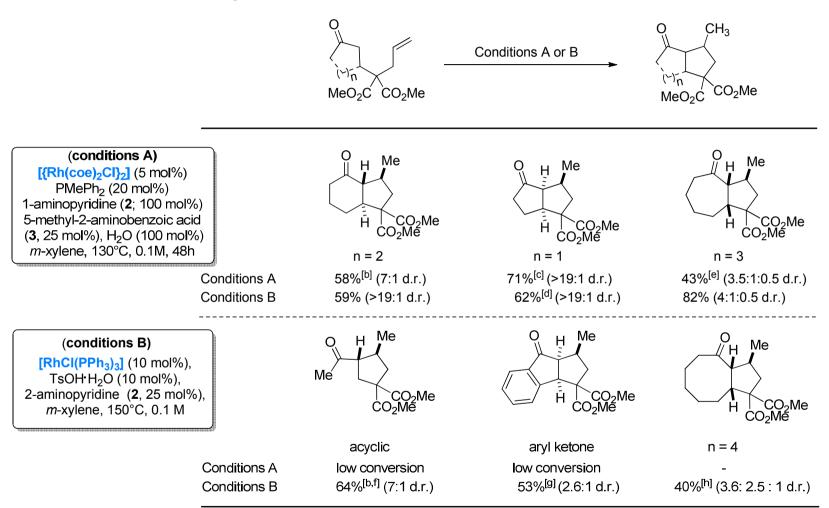


1	-	74 (58) ^[c]	7:1	2
2	without [{Rh(coe) ₂ Cl} ₂]	-	-	-
3	without 2	-	-	12
4	without 3	8	3:1	-
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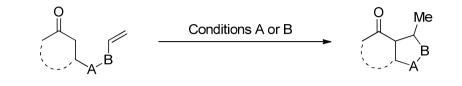
Ketone scope



[a] Yields of isolated products are given. [b] Yield of the major isomer [c] 2-Amino-3-methylpyridine (100 mol%) was used instead of **2**. [d] 2-Amino-3-methyl-pyridine (25 mol%) was used instead of **2**. [e] 150°C

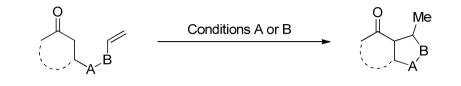
[f] 2-Amino-3-methylpyridine (100 mol%) and $AgPF_6$ (10 mol%) were used. [g] **2** (100 mol%) was used. [h][{Rh(coe)₂Cl}₂] (5 mol%), tris (3,5-di(trifluoromethyl)phenyl)phosphine (30 mol%), and $AgPF_6$ (10 mol%) were used.

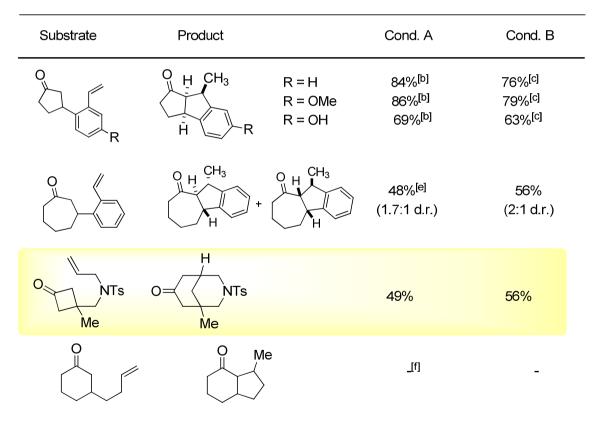
Further substrate scope



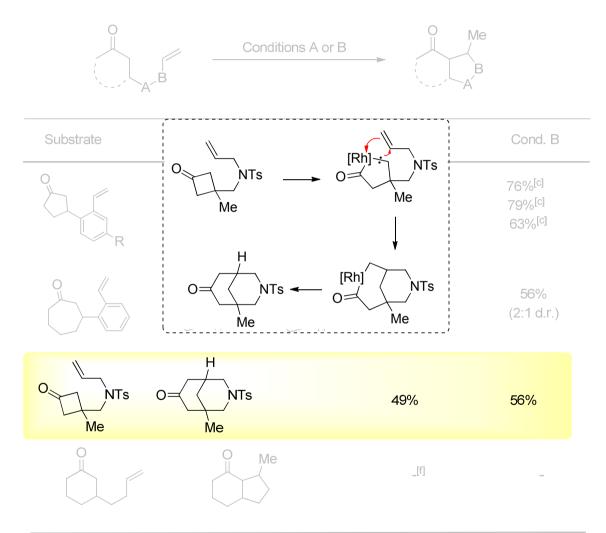
Substrate	Product		Cond. A	Cond. B
O C R	O H CH ₃ H R	R = H R = OMe R = OH	84% ^[b] 86% ^[b] 69% ^[b]	76% ^[c] 79% ^[c] 63% ^[c]
	OH H H +	O H CH ₃	48% ^[e] (1.7:1 d.r.)	56% (2:1 d.r.)
O Me	O= Me		49%	56%
0 U		e	្បា	-

Further substrate scope

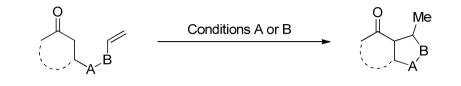


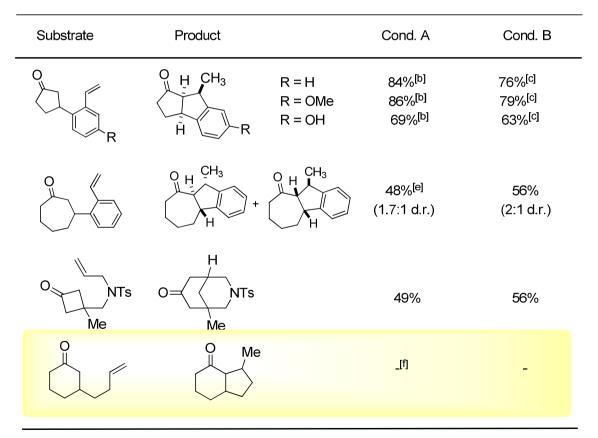


Further substrate scope

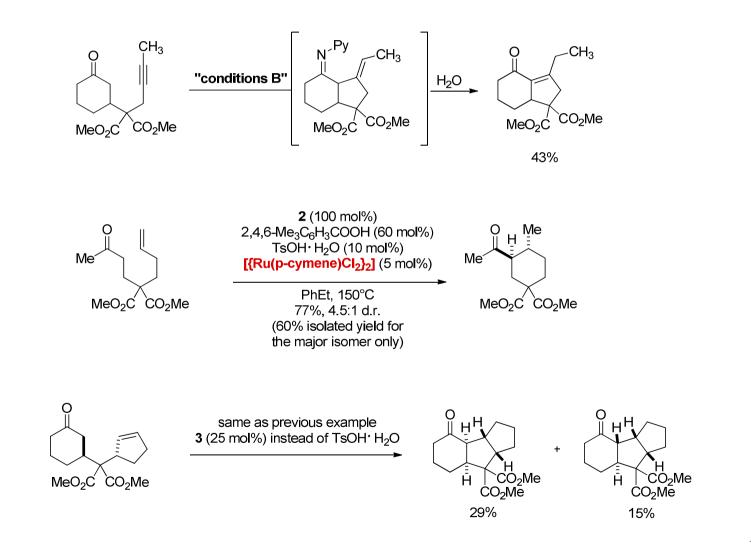


Further substrate scope





Further substrate scope



Conclusion

- Catalytic intramolecular ketone-olefin coupling
- 2 atom economical complementary methods
- Unique dual activation mode
- [Rh]-based system : formation of 5-membered ring
- [Ru]-based system : formation of 6-membered ring

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Thank you for your attention