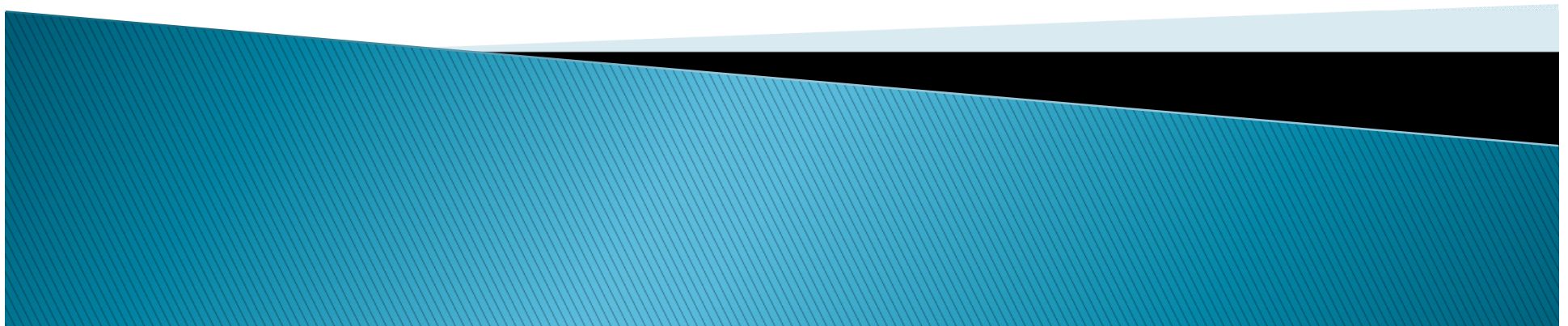
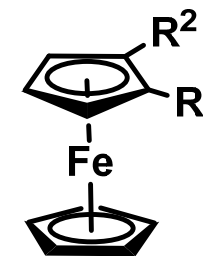
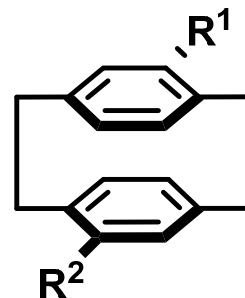
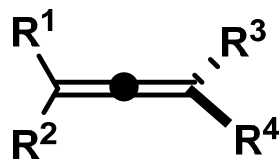
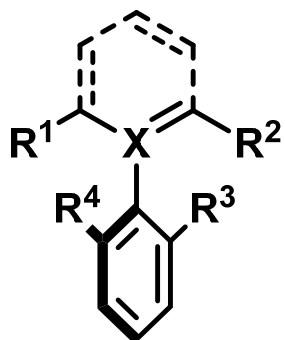


Helicenes, more than a century of interest

Fabien PEREZ – STeRéO – November 17th, 2015



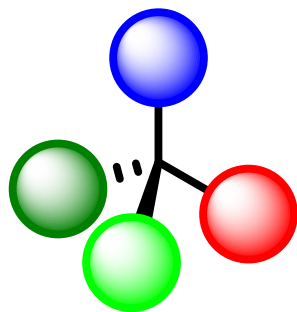
Chirality modes



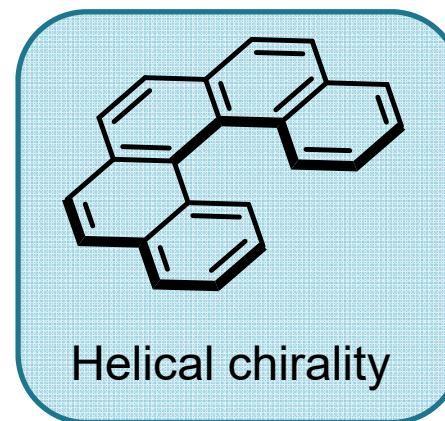
paracyclophane

Axial chirality

Planar chirality

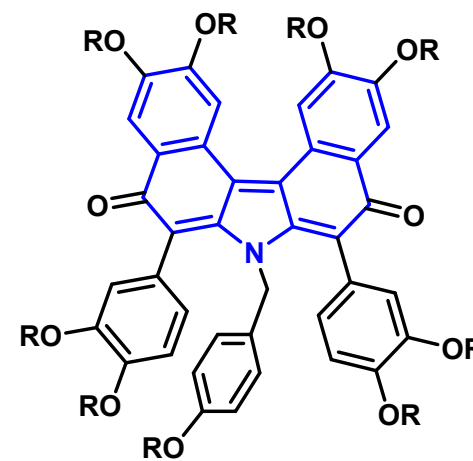
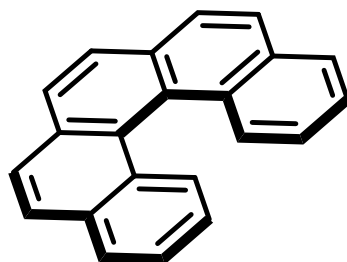
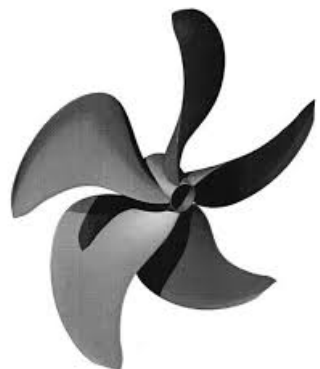


Stereogenic center



Helical chirality

Helix in our world

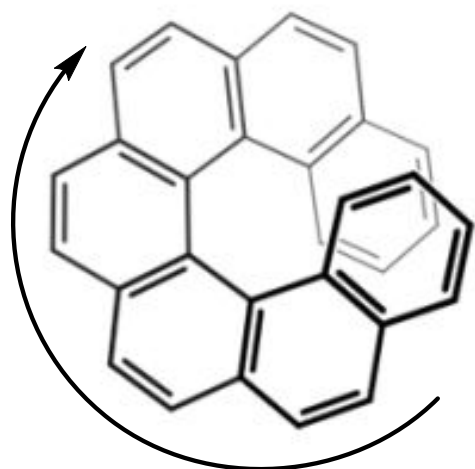


Outline

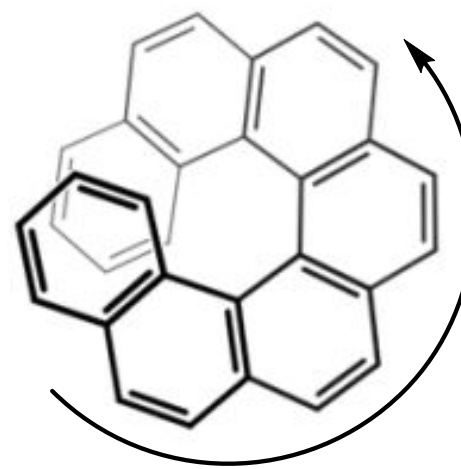
- ▶ Introduction on helicenes
- ▶ Racemic synthesis
- ▶ To enantiopure helicene
- ▶ Applications

Introduction on helicenes

- [n]helicene with n = number of *orthofused* aromatic rings



P-(+)
Clockwise helicity
Dextrorotatory rotation

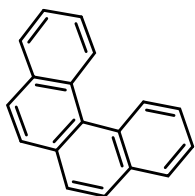


M-(-)
Anticlockwise helicity
Levorotatory rotation

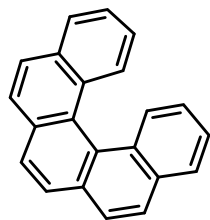
Helicene timeline



1913
[4]helicene R. Weitzenböck



Helicene timeline

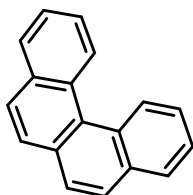


1918
[5]helicene

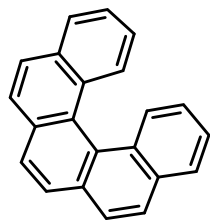
T. Oppenheimer



1913
[4]helicene



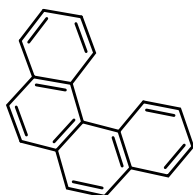
Helicene timeline



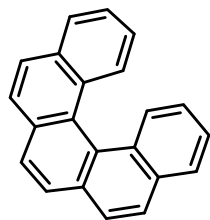
1918
[5]helicene



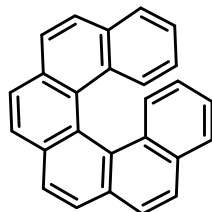
1913
[4]helicene



Helicene timeline



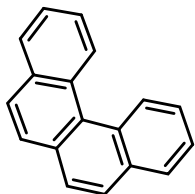
1918
[5]helicene



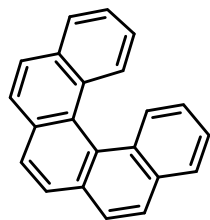
1955
[6]helicene



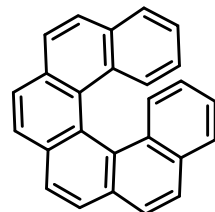
1913
[4]helicene



Helicene timeline

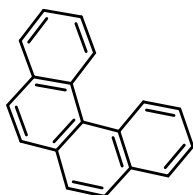


1918
[5]helicene

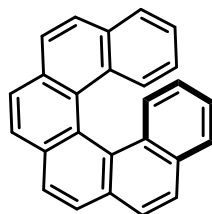


1955
[6]helicene

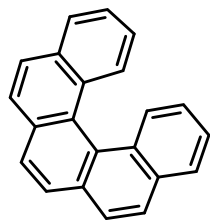
1913
[4]helicene



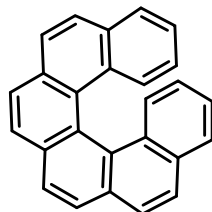
1956
Resolution of
[6]helicene



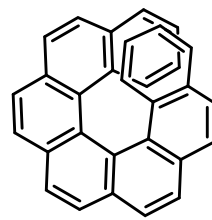
Helicene timeline



1918
[5]helicene

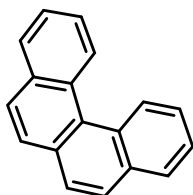


1955
[6]helicene

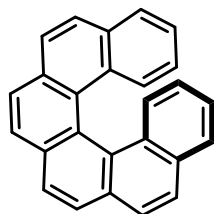


1967
[7]helicene

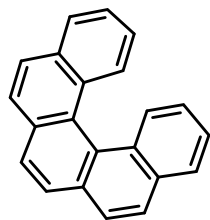
1913
[4]helicene



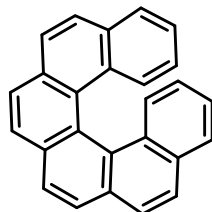
1956
Resolution of
[6]helicene



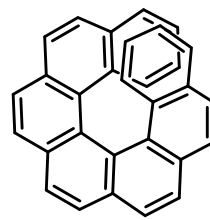
Helicene timeline



1918
[5]helicene

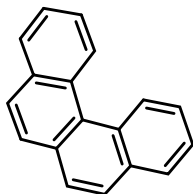


1955
[6]helicene

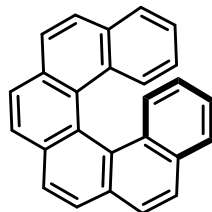


1967
[7]helicene

1913
[4]helicene



1956
Resolution of
[6]helicene

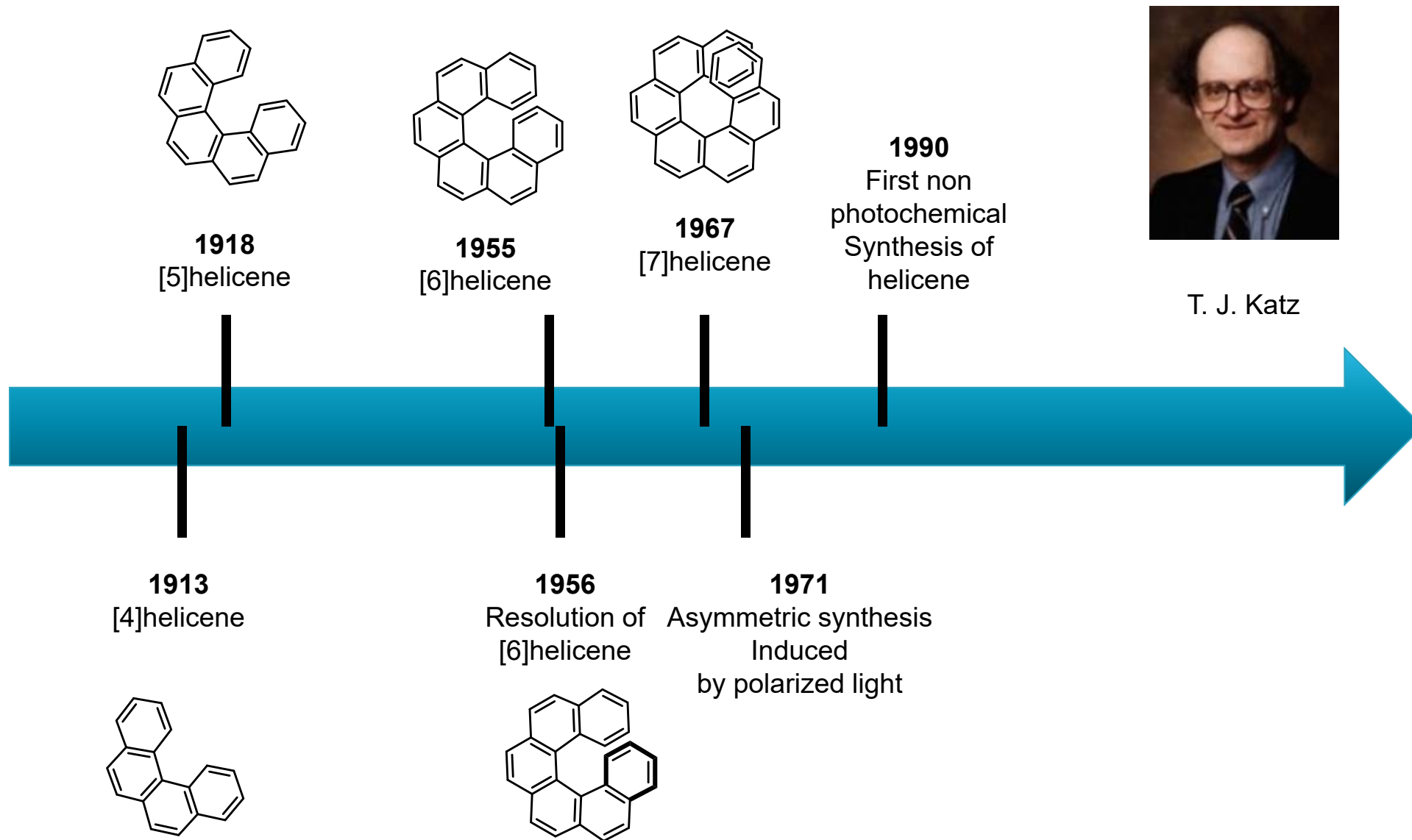


1971
Asymmetric synthesis
Induced
by polarized light



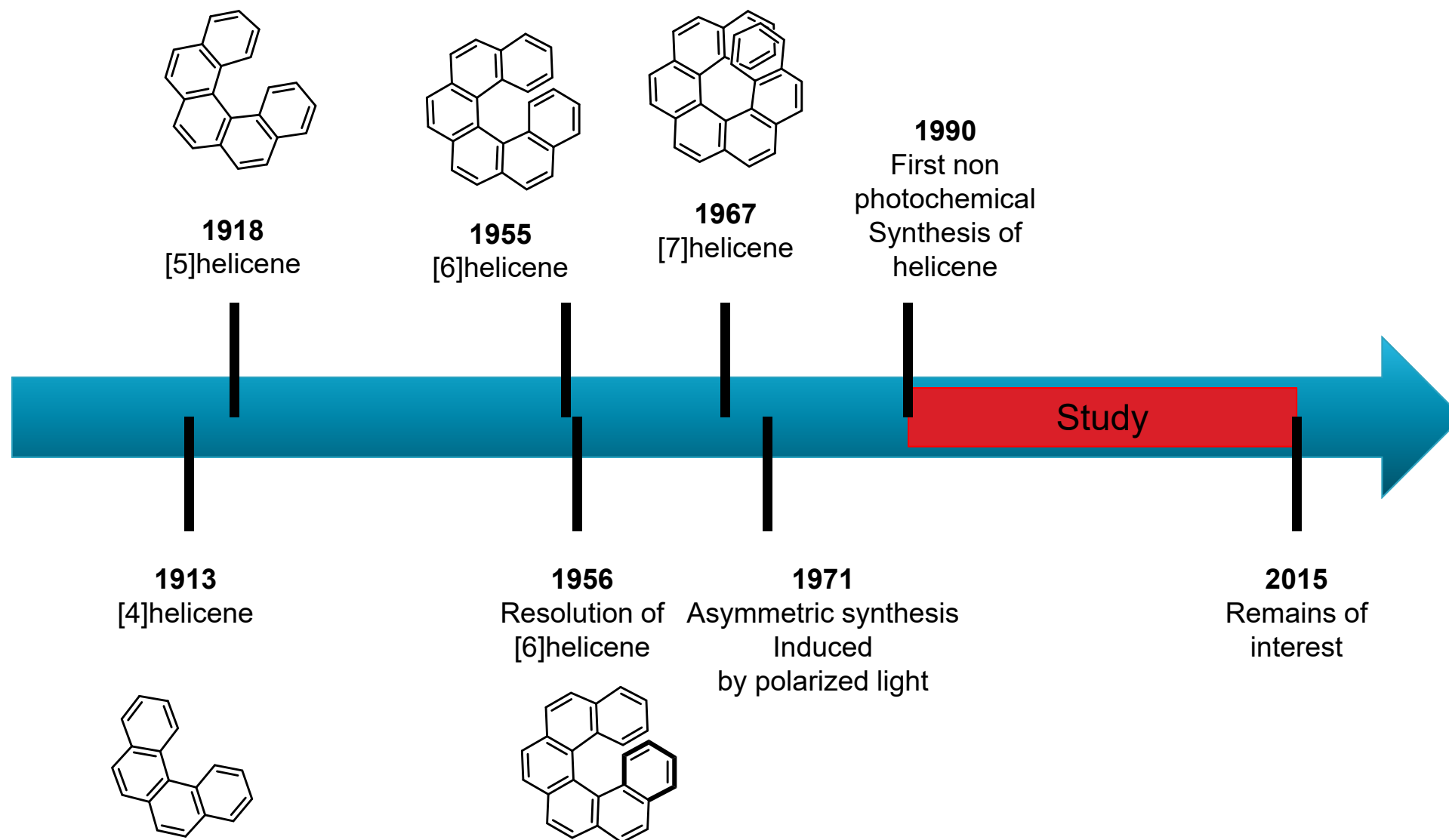
H. Kagan

Helicene timeline

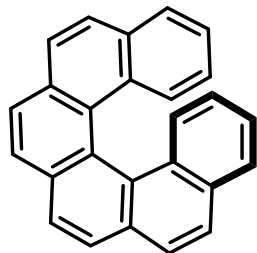


T. J. Katz

Helicene timeline

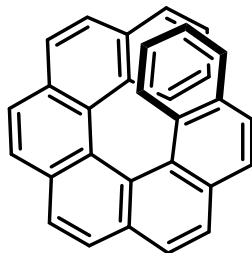


General properties



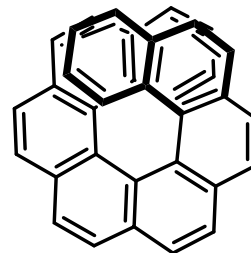
[6]helicene
 $t_{1/2} = 13$ min
 at 221,4 °C

$E_{a(27\text{ °C})} = 35,6$ kcal/mol



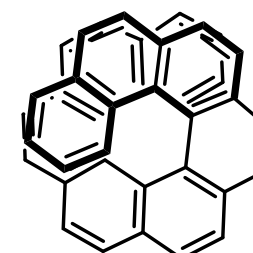
[7]helicene
 $t_{1/2} = 26$ min
 at 286,2 °C

$E_{a(27\text{ °C})} = 41,1$ kcal/mol



[8]helicene
 $t_{1/2} = 54$ min
 at 286,4 °C

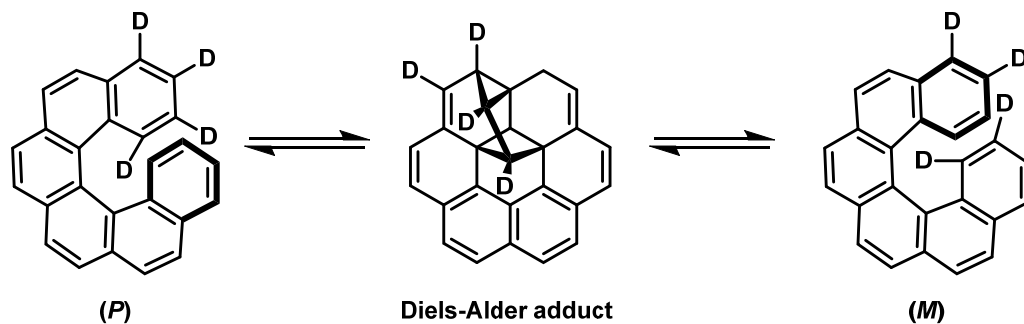
$E_{a(27\text{ °C})} = 41,6$ kcal/mol



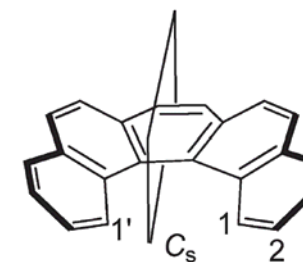
[9]helicene
 $t_{1/2} = 123$ min
 at 293,5 °C

$E_{a(27\text{ °C})} = 42,3$ kcal/mol

R. H. Martin. *Tetrahedron. Lett.* **1972**, 35, 3707

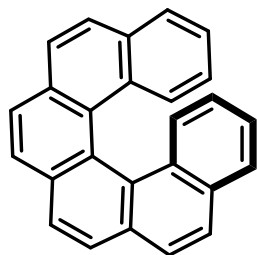


R. H. Martin. *Tetrahedron.* **1974**, 30, 347



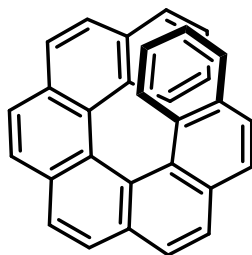
M. Patzschke. *Chem. Eur. J.* **2009**, 30, 347

General properties



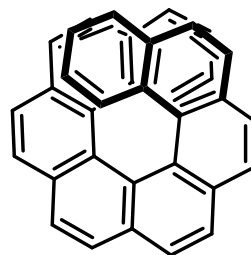
[6]helicene
 $t_{1/2} = 13$ min
 at 221,4 °C

$E_{a(27\text{ °C})} = 35,6$ kcal/mol



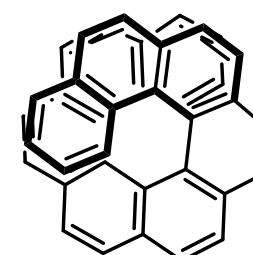
[7]helicene
 $t_{1/2} = 26$ min
 at 286,2 °C

$E_{a(27\text{ °C})} = 41,1$ kcal/mol



[8]helicene
 $t_{1/2} = 54$ min
 at 286,4 °C

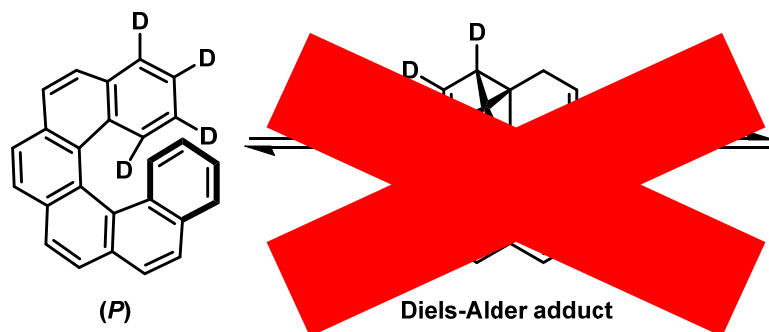
$E_{a(27\text{ °C})} = 41,6$ kcal/mol



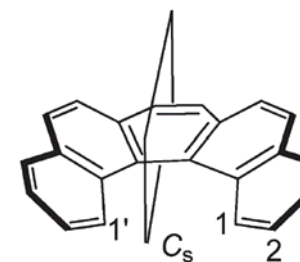
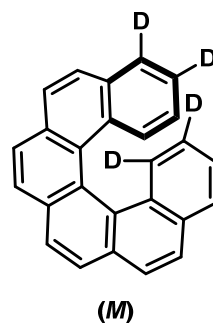
[9]helicene
 $t_{1/2} = 123$ min
 at 293,5 °C

$E_{a(27\text{ °C})} = 42,3$ kcal/mol

R. H. Martin. *Tetrahedron. Lett.* **1972**, 35, 3707



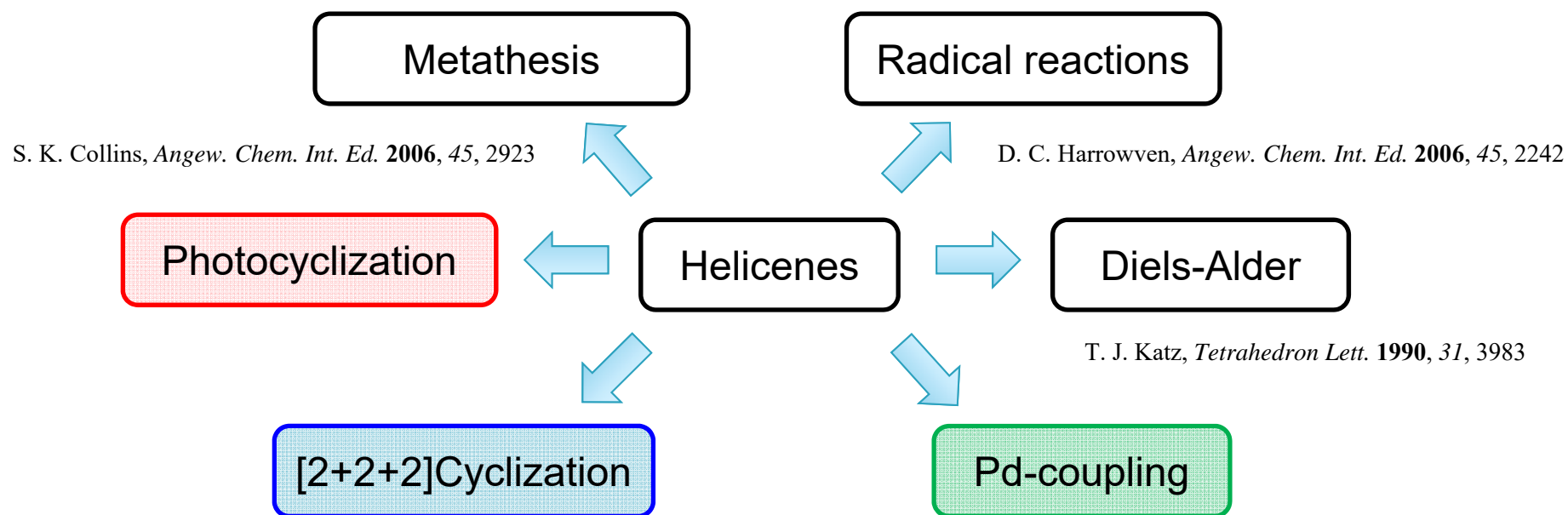
R. H. Martin. *Tetrahedron.* **1974**, 30, 347



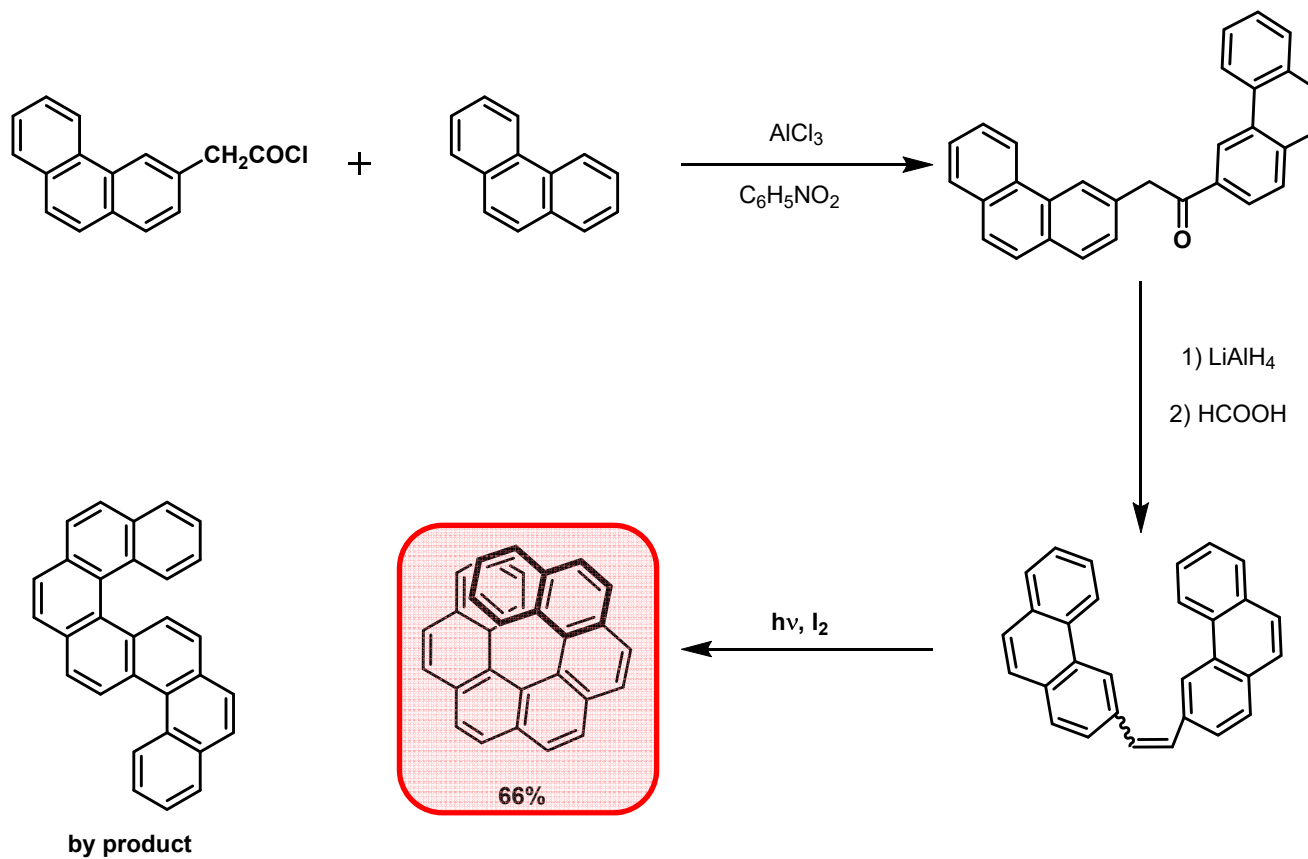
M. Patzschke. *Chem. Eur. J.* **2009**, 30, 347

Synthesis of helicenes, racemic series

Racemic synthesis overview

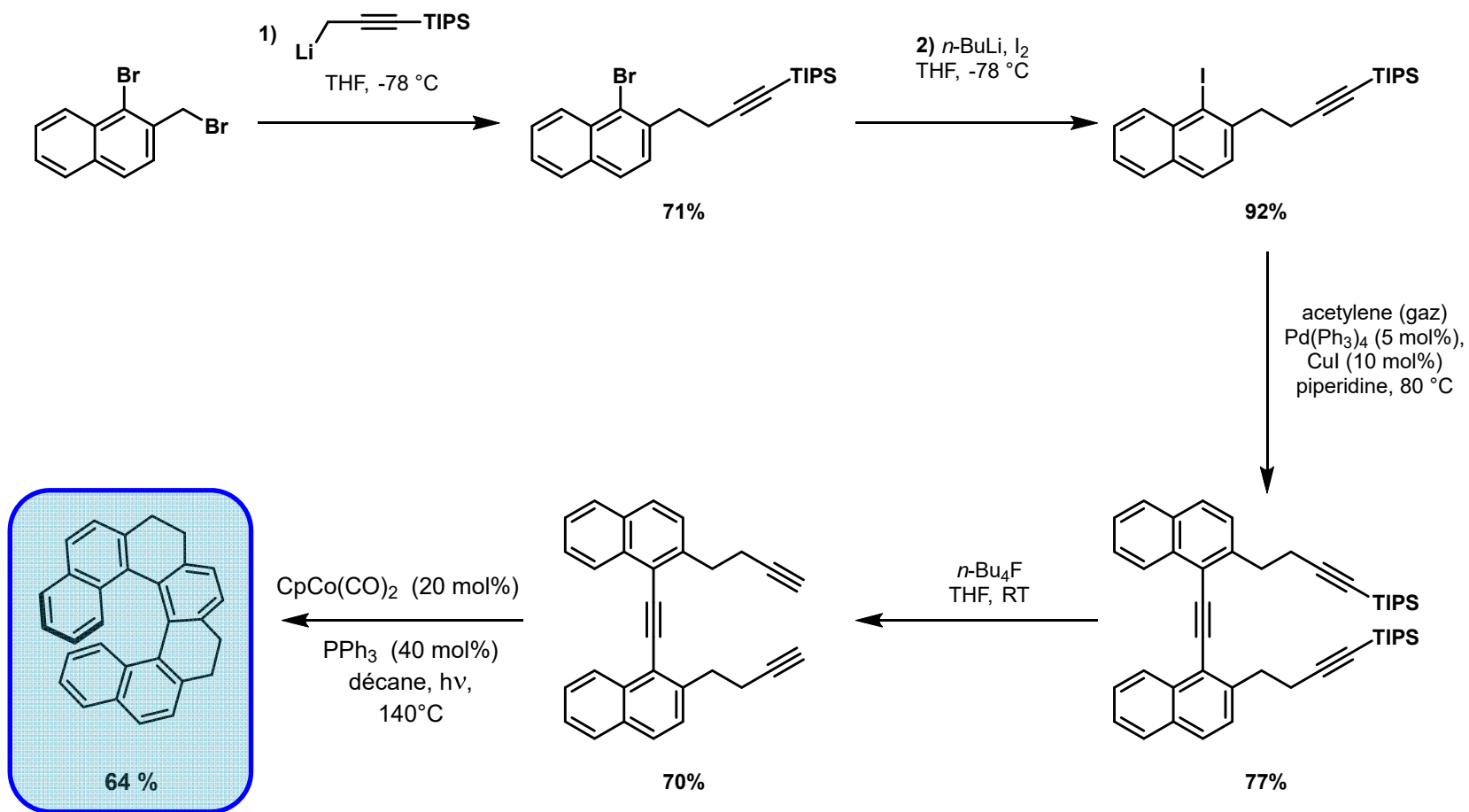


Photocyclisation



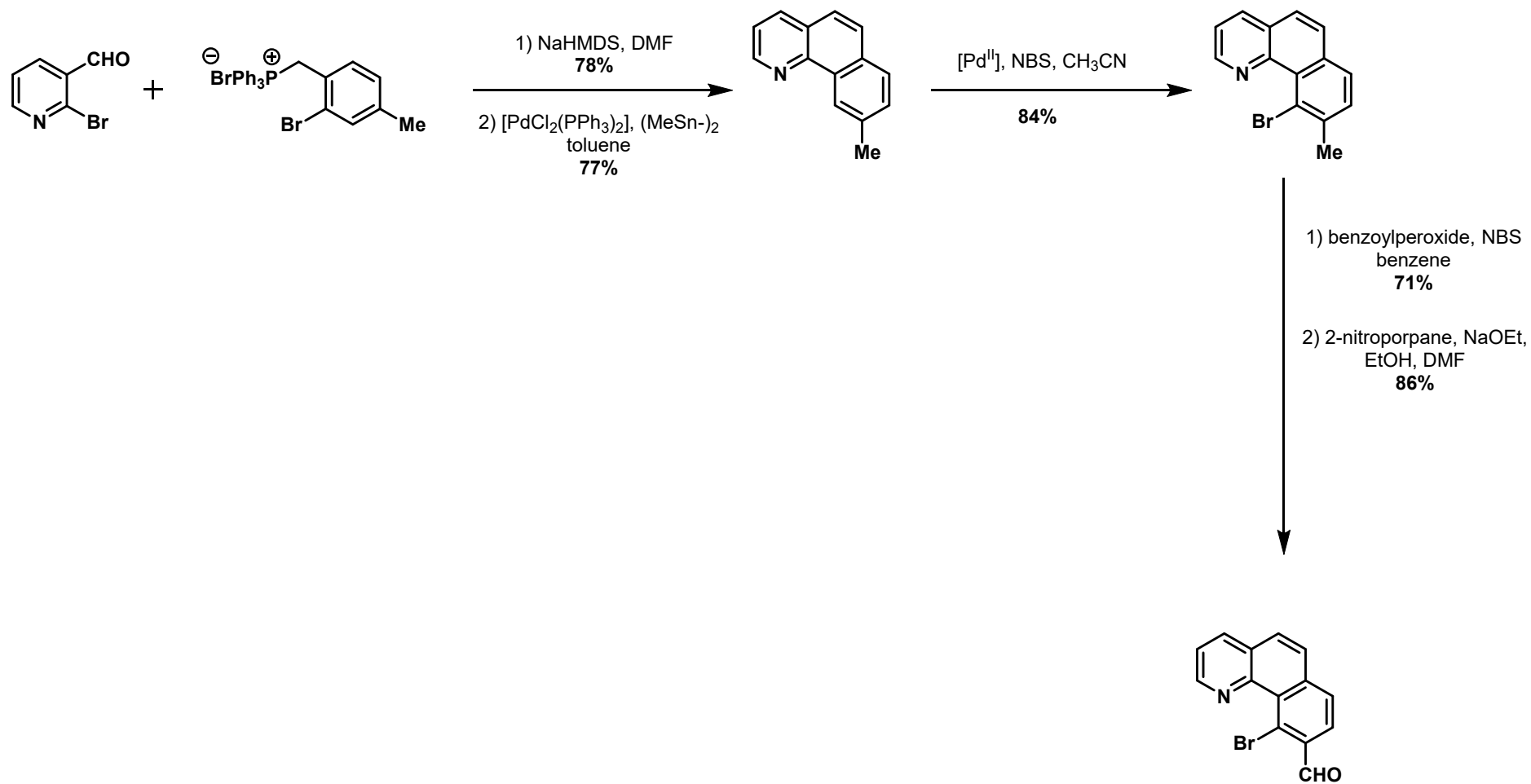
R. H. Martin *Tetrahedron. Lett.* **1967**, 8, 743

[2+2+2] cyclization



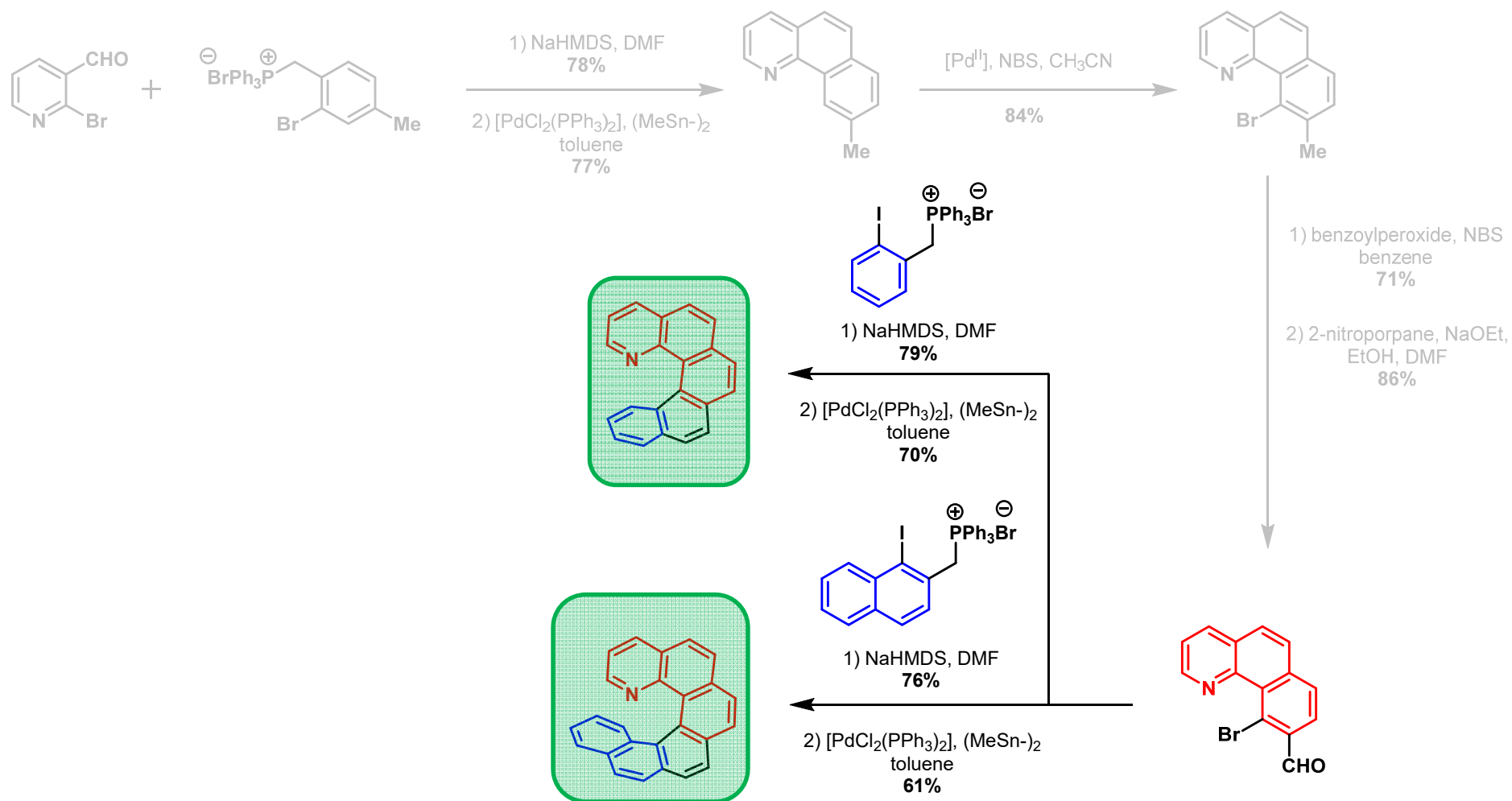
I. G. Stará, I. Starý *Tetrahedron. Lett.* **1999**, *40*, 1993

azahelicene



N. Takenaka, *Angew. Chem. Int. Ed.* **2008**, 9708

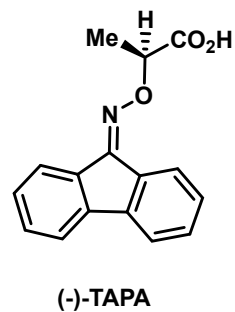
azahelicene



N. Takenaka, *Angew. Chem. Int. Ed.* **2008**, 9708

En route for enantiopure helicenes

Optical resolution



Cristal picking

chiral HPLC

M. S. Newmann, *J. Am. Chem. Soc.* **1956**, *78*, 4765

E. Gilav, *J. Chem. Soc. Chem. Commun.* **1976**, 99

Enzymatic resolution

enantiopure helicene

Asymmetric rearrangement

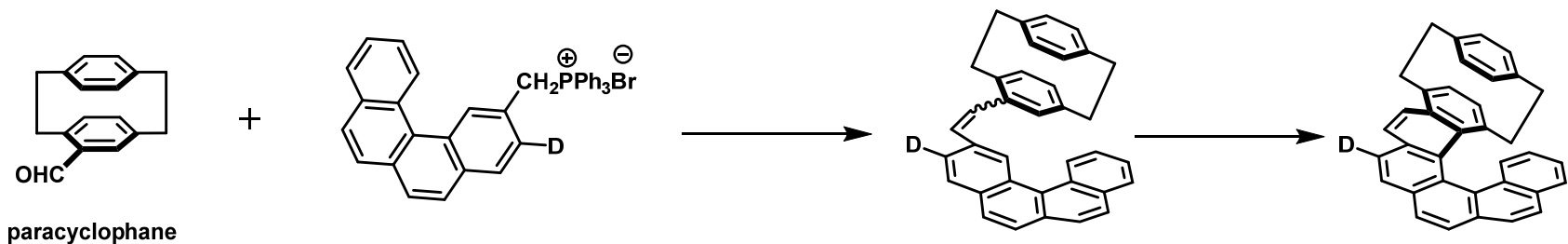
T. J. Katz, *Tetrahedron Lett.* **1990**, *31*, 3983

T. J. Katz, *Tetrahedron Lett.* **1990**, *31*, 3983

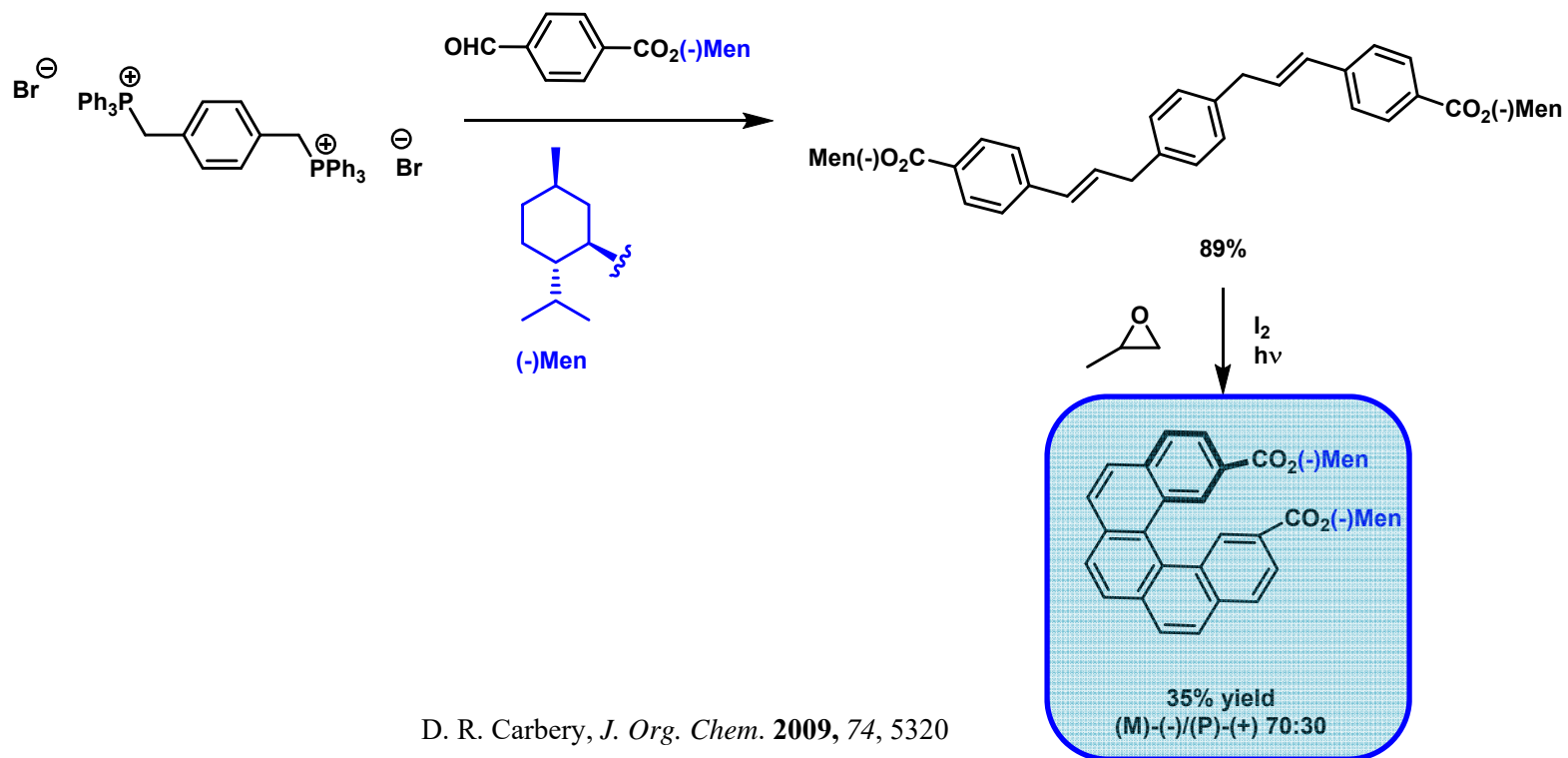
Enantioselective synthesis

Chiral auxiliary

Use of chiral auxiliary

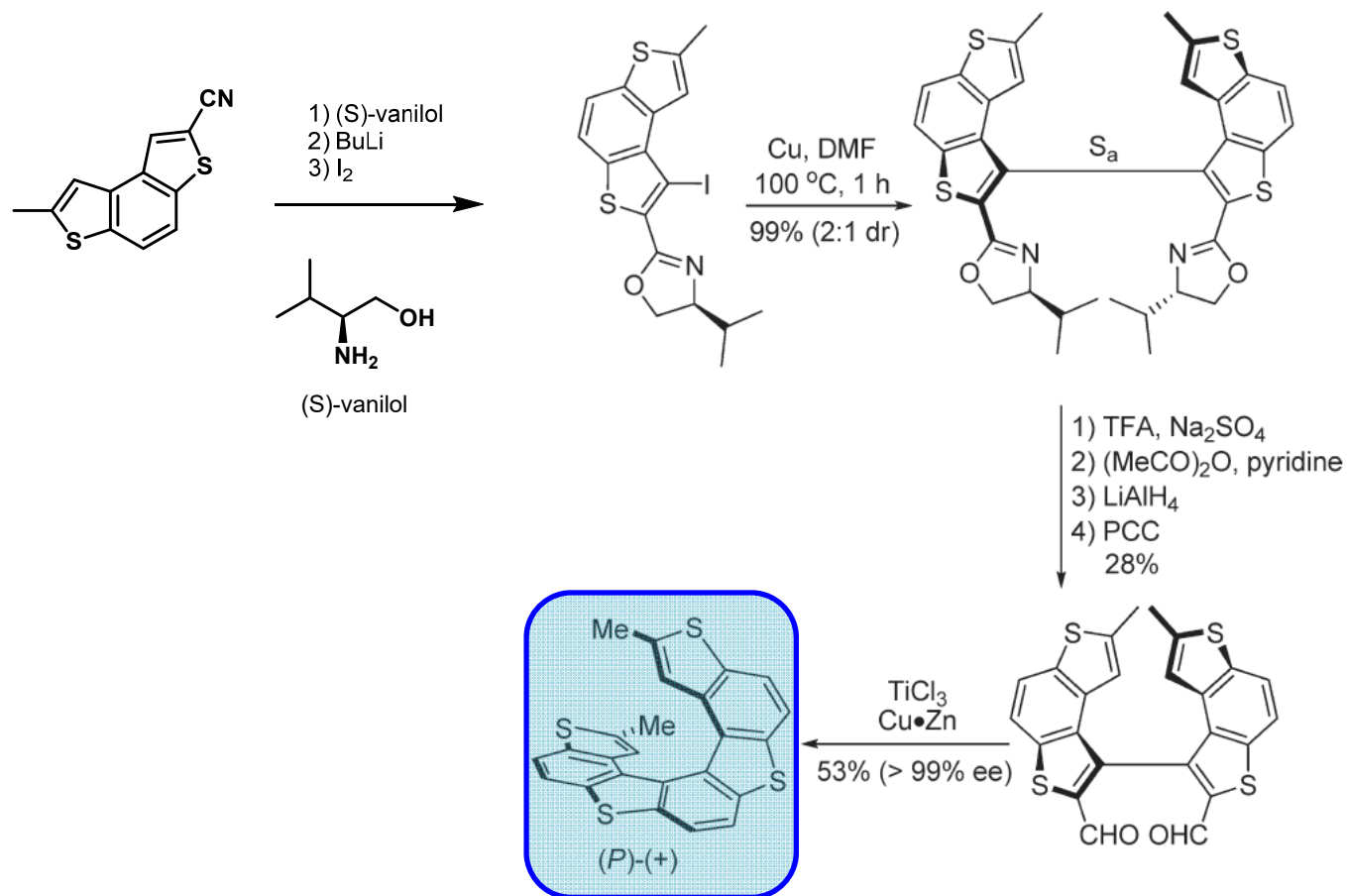


R. H. Martin, *Tetrahedron Lett.* **1972**, 28, 2842



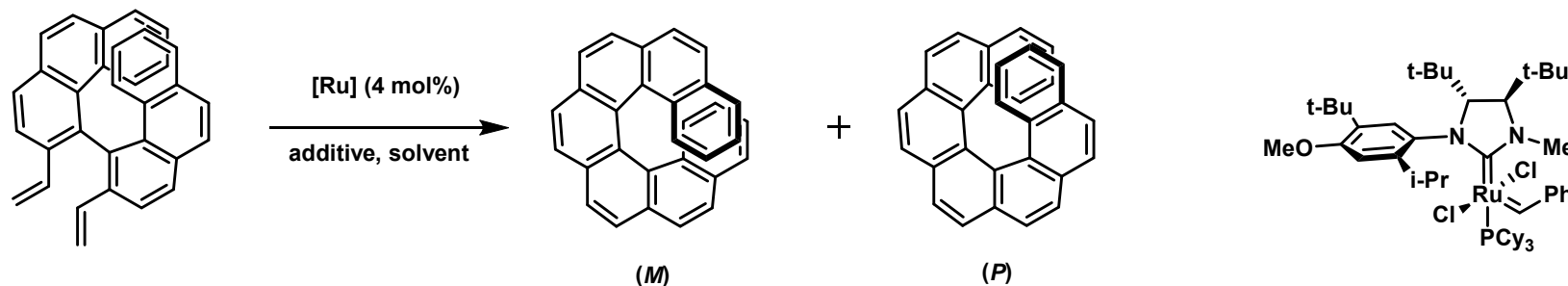
D. R. Carbery, *J. Org. Chem.* **2009**, 74, 5320

Use of chiral auxiliary



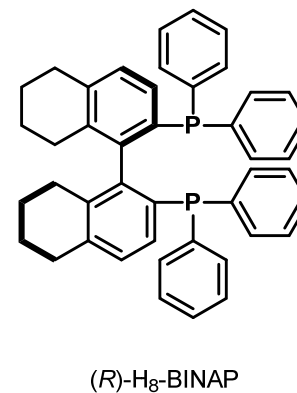
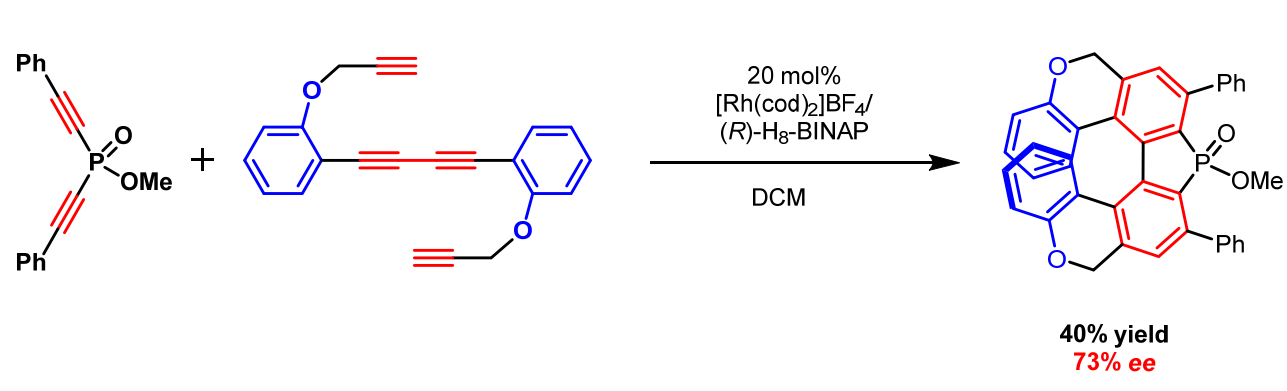
K. Tanaka, *J. Org. Chem.* **1997**, *62*, 4465

Enantioselective synthesis

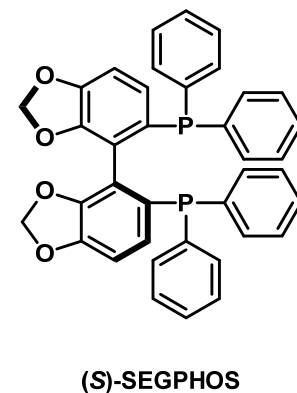
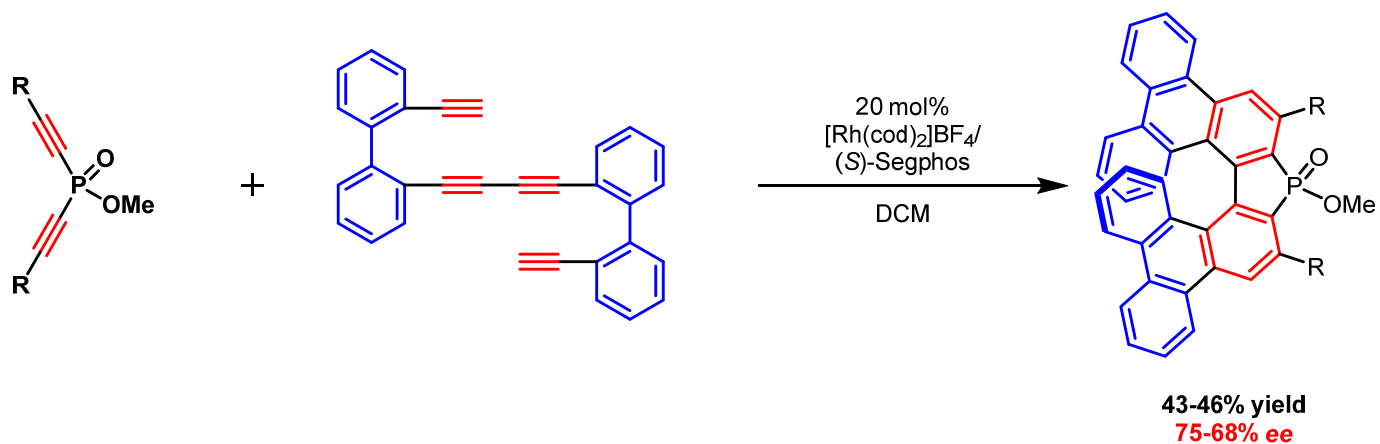


Solvent	Additive	Conversion (%)	ee (%) (conf)
DCM	1-hexene	40	48 (<i>M</i>)
DCM	vinycyclohexane	43	53 (<i>M</i>)
Benzene	vinycyclohexane	32	65 (<i>M</i>)
C ₆ F ₆	vinycyclohexane	38	80 (<i>M</i>)

Enantioselective synthesis



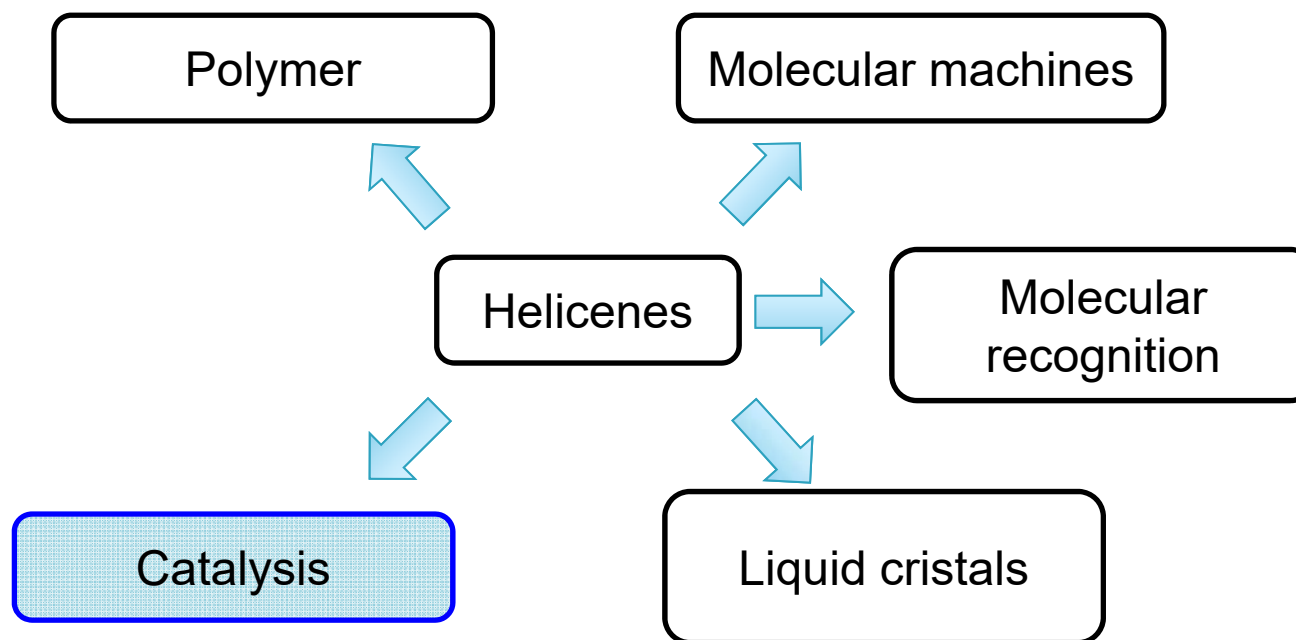
Tanaka, *Org. Lett.* **2010**, *12*, 1324



K. Tanaka, *J. Am. Chem. Soc.* **2012**, *134*, 4080

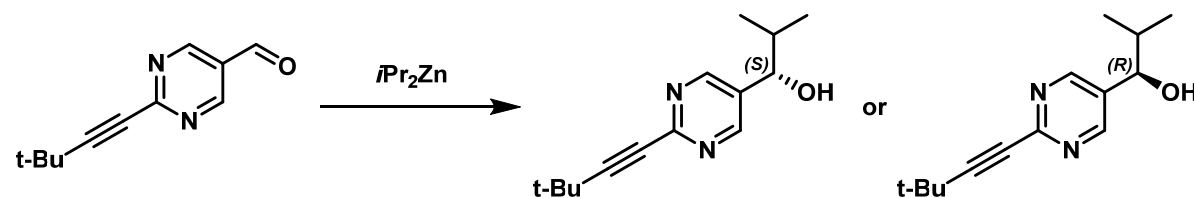
Applications

Applications



Helicene in catalysis

▶ Autocatalyzed reaction

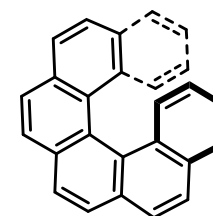


(P)-(+)-[5]helicene
(P)-(+)-[6]helicene

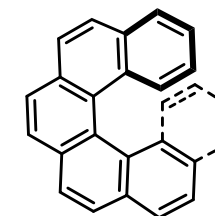
88-90% yield
95% ee

(M)-(-)-[5]helicene
(M)-(-)-[6]helicene

89-93% yield
93-94% ee



(P)-(+)-[5]helicene
(P)-(+)-[6]helicene



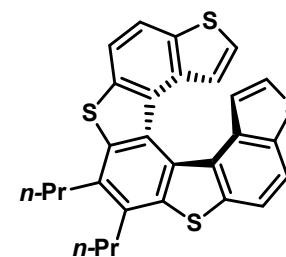
(M)-(-)-[5]helicene
(M)-(-)-[6]helicene

(P)-(+)-tetrathia[7]helicene

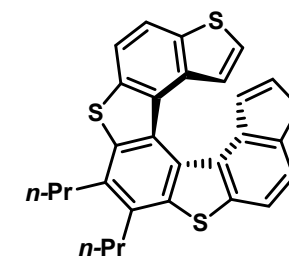
88% yield
99% ee

(M)-(-)-tetrathia[7]helicene

88% yield
95% ee



(P)-(+)-tetrathiahelicene



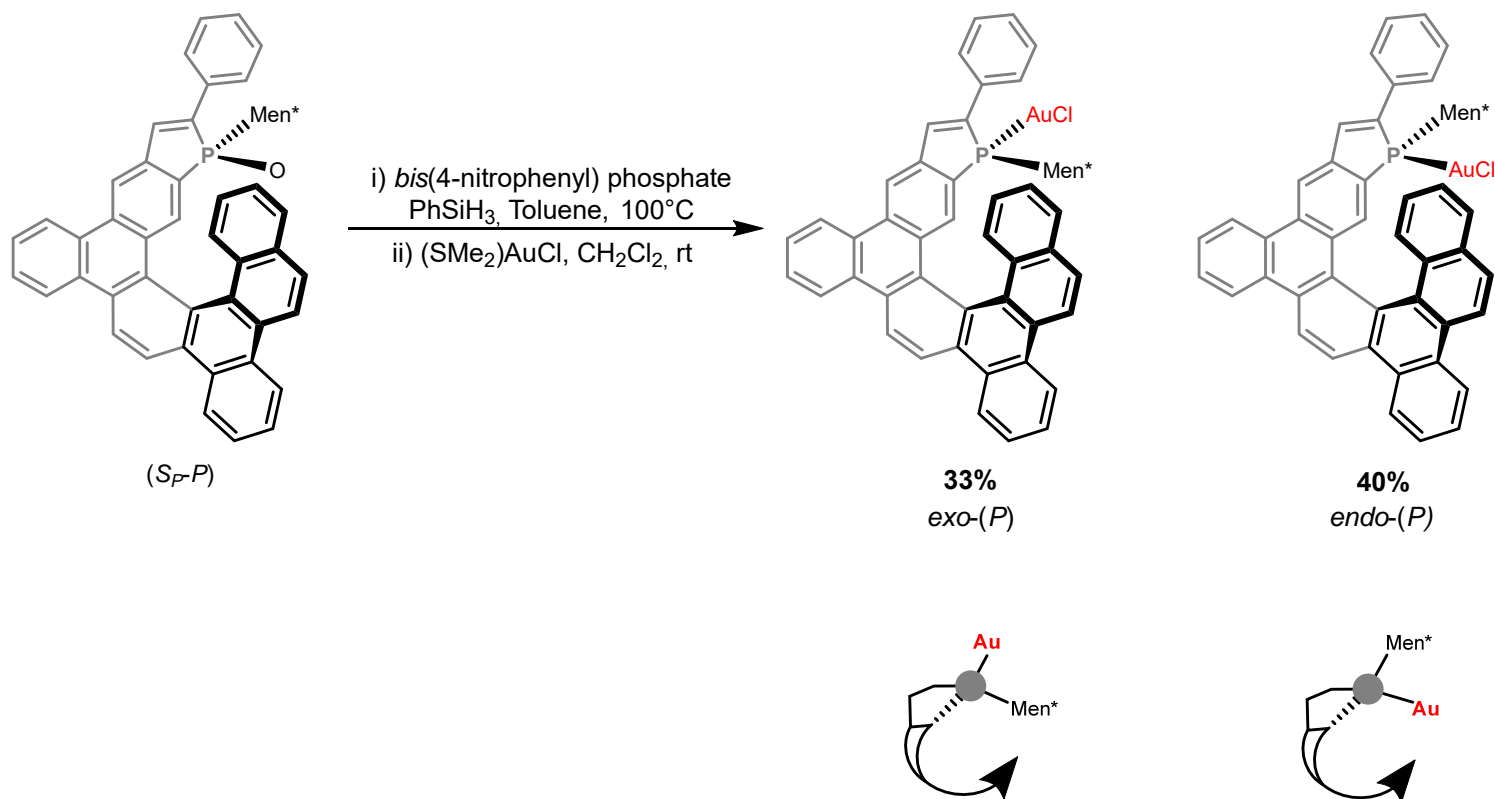
(M)-(-)-tetrathiahelicene

K. Soai, *Angew. Chem. Int. Ed.* **2001**, *40*, 1098

K. Soai, *Tetrahedron : Asymmetry.* **2006**, *17*, 2050

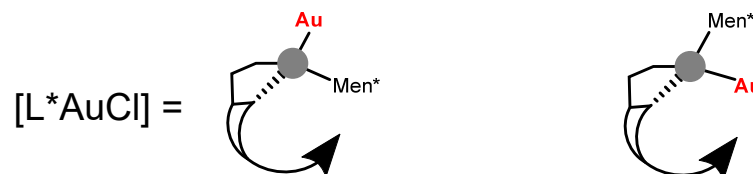
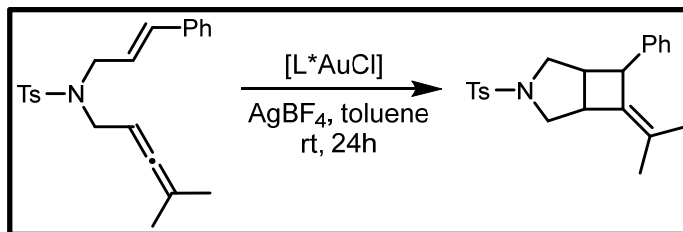
Helicene in catalysis

▶ Chiral ligand

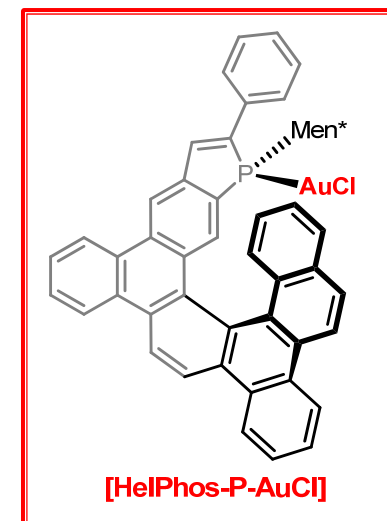


Helicene in catalysis

▶ Chiral ligand

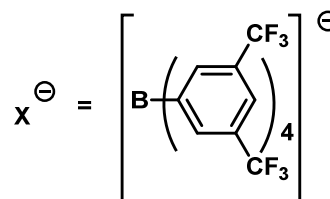
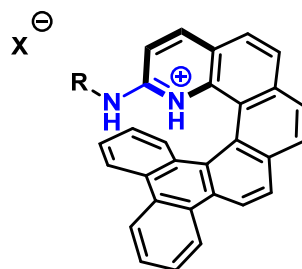
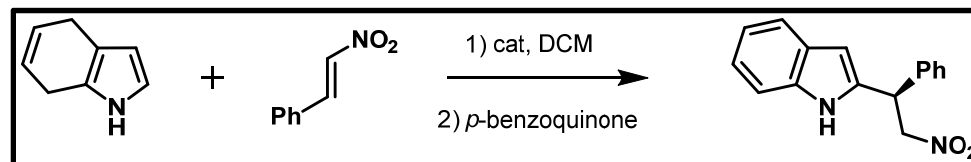


	Exo-(P)	Endo-(P)
Conv (%)	30	>95
ee (%)	41	88



Helicene in catalysis

► Organocatalyst



R	mol%	Temp (°C)	time (h)	Yield (%)	er
H	10	-40	20	85	69:31
Bn	2	-40	20	72	69:31
<i>t</i> -Bu	2	-40	20	79	92:8
ad	2	-40	20	88	93:7
ad	0,5	-40	48	80	92:8

Takenaka, *J. Am. Chem. Soc.* **2010**, *132*, 4536

Conclusion

- ▶ Numerous ways to generate helicene
- ▶ Special properties
- ▶ Wide range of application
- ▶ Still studied after more a century

Thank you for you
attention

