Palladium catalyzed intermolecular aminocarbonylation of alkenes

J. Cheng, X. Qi, M. Li, P. Chen, G. Liu, JACS, **2015**, 137, 2480.

RCC Mylène ROUDIER 19,03,2015 β-aminoacids identified as essential component in natural products

HO₂C, N CO₂H

pyrrolidine-2.4-dicarboxylic acid

Me Me

 β -prolinebetaine = found in citrus foods

HO₂C OH NHAC

siastatine B (natural iminoglycoside)

β-aminoacids identified as essential component in natural products

pyrrolidine-2.4-dicarboxylic acid

 β -prolinebetaine = found in citrus foods

siastatine B (natural iminoglycoside)

Br
$$\frac{1}{1}$$
 $\frac{1}{1}$ \frac

Jasplakinolide
= from marine organism
insectividal, antifungal properties

= phenylisoserine

Taxol = antitumour agent

I. Braschi, G. Cardillo, C. Tomasini, R. Venezia, J. Org. Chem., 1994, 59, 7292

Chem. Soc. Rev., 1996, 117-128

For a recent review of asymmetric synthesis of α - and β -amino acids : *ACIE*, **2003**, *42*, 4290-4299.

I. Braschi, G. Cardillo, C. Tomasini, R. Venezia, J. Org. Chem., 1994, 59, 7292

\Leftrightarrow Amination of α , β -insaturated esters

S. G. Davies, O. Ichihara, Tetrahedron asymmetry, 1991, 2, 183

Chem. Soc. Rev., 1996, 117-128

For a recent review of asymmetric synthesis of α - and β -amino acids : *ACIE*, **2003**, *42*, 4290-4299.

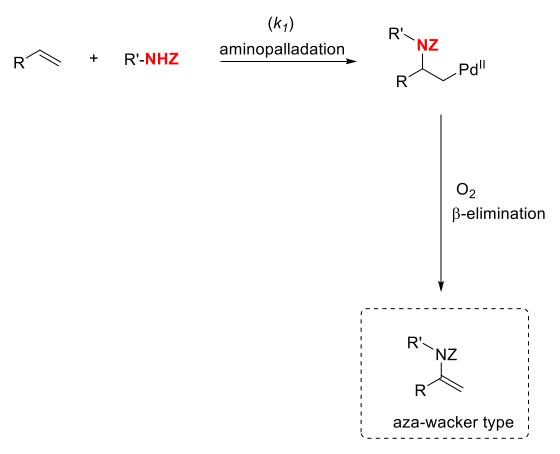
 \diamond Hydrogenation of β - aminoacrylic acids

Achn O Ru or Rh complex
$$H_2$$
 R^1 OR^2

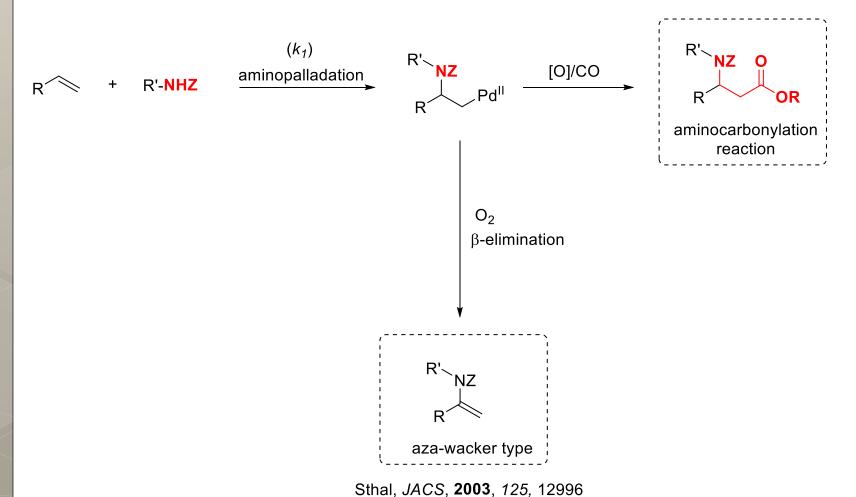
Y.-G. Zhou, W. Tang, W.-B. Wang, W. Li, X. Zhang, JACS, 2002, 124, 4952.

Chem. Soc. Rev., 1996, 117-128

For a recent review of asymmetric synthesis of α - and β -amino acids : *ACIE*, **2003**, *42*, 4290-4299.



Sthal, JACS, 2003, 125, 12996

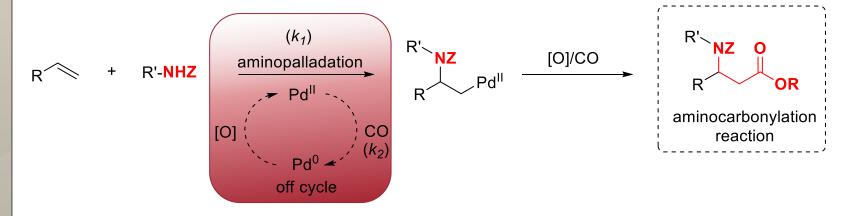


-> Pd-benzoquinone or Pd-Cu(II) salt systems as oxidants for aminocarbonylation

$$R' + R'-NHZ \xrightarrow{aminopalladation} R' NZ \qquad [O]/CO$$

$$Pd^{0} \leftarrow (k_{2})$$

-> Pd-benzoquinone or Pd-Cu(II) salt systems as oxidants for aminocarbonylation



$$X k_2 > k_1$$

How to favor the aminopalladation?

$X k_2 > k_1$

How to favor the aminopalladation?

Togni's reagent could react with lodide Nevado and al., ACIE, 2013, 52, 13086

$$R + R'-NHZ$$

$$(k_1)$$

$$aminopalladation$$

$$Pd^{\parallel}$$

$$(k_2)$$

$$off cycle$$

$$R' NZ O$$

$$R' NZ$$

$$X k_2 > k_1$$

How to favor the aminopalladation?

$$R + R'-NHZ$$

$$(k_1)$$

$$aminopalladation$$

$$Pd^{\parallel}$$

$$OR$$

$$aminocarbonylation$$

$$reaction$$

$$80\%$$

$$X \quad k_2 > k_1$$

$$k_1 = k_1$$

$$k_2 < k_1$$

How to favor the aminopalladation?

Table 1. Optimization of Reaction Conditions^a

		yield (%) ^b	
Pd catalyst	oxidant	3a	3a' (R)
Pd(OAc) ₂	$PhI(OAc)_2$	45	4 (Me)
$Pd(CH_3CN)_2Cl_2$	$PhI(OAc)_2$	65	5 (Me)
Pd(acac) ₂	$PhI(OAc)_2$	36	3 (Me)
Pd(OTFA) ₂	$PhI(OAc)_2$	80	5 (Me)
$Pd(OTFA)_2$	PhI(OPiv) ₂	55	5 (*Bu)
Pd(OTFA) ₂	$PhI(O_2CAd)_2$	90 (83) ^f	5 (Ad)
$Pd(OTFA)_2$	$PhI(O_2CAd)_2$	80	13 (Ad)
$Pd(OTFA)_2$	$PhI(O_2CAd)_2$	85	10 (Ad)
$Pd(OTFA)_2$	$(NH_4)_2S_2O_8$	31	0
Pd(OTFA) ₂	oxone	27	0
Pd(OTFA) ₂	35% aq H_2O_2	21	0
none	$PhI(O_2CAd)_2$	0	0
	Pd(OAc) ₂ Pd(CH ₃ CN) ₂ Cl ₂ Pd(acac) ₂ Pd(OTFA) ₂	Pd(OAc) ₂ PhI(OAc) ₂ Pd(CH ₃ CN) ₂ Cl ₂ PhI(OAc) ₂ Pd(acac) ₂ PhI(OAc) ₂ Pd(OTFA) ₂ PhI(OAc) ₂ Pd(OTFA) ₂ PhI(OPiv) ₂ Pd(OTFA) ₂ PhI(O ₂ CAd) ₂ Pd(OTFA) ₂ ONH ₄) ₂ S ₂ O ₈ Pd(OTFA) ₂ oxone Pd(OTFA) ₂ 35% aq H ₂ O ₂	Pd catalyst oxidant 3a Pd(OAc) ₂ PhI(OAc) ₂ 45 Pd(CH ₃ CN) ₂ Cl ₂ PhI(OAc) ₂ 65 Pd(acac) ₂ PhI(OAc) ₂ 36 Pd(OTFA) ₂ PhI(OAc) ₂ 80 Pd(OTFA) ₂ PhI(OPiv) ₂ 55 Pd(OTFA) ₂ PhI(O ₂ CAd) ₂ 90 (83) ^f Pd(OTFA) ₂ PhI(O ₂ CAd) ₂ 80 Pd(OTFA) ₂ PhI(O ₂ CAd) ₂ 85 Pd(OTFA) ₂ (NH ₄) ₂ S ₂ O ₈ 31 Pd(OTFA) ₂ oxone 27 Pd(OTFA) ₂ 35% aq H ₂ O ₂ 21

^aAll reactions were run at 0.2 mmol scale. ^bYield obtained by ¹H NMR with CF₃-DMA as internal standard. ^c2a (1.2 equiv). ^d2a (2.0 equiv). ^eOxidant (5.0 equiv). ^fIsolated yield of ester 4a.

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5	Pd(OTFA) ₂	PhI(OPiv) ₂	55	5 (*Bu)	
6	Pd(OTFA) ₂	$PhI(O_2CAd)_2$	90 (83) ^f	5 (Ad)	Best conditi
7^c	Pd(OTFA) ₂	$PhI(O_2CAd)_2$	80	13 (Ad)	
8^d	Pd(OTFA) ₂	$PhI(O_2CAd)_2$	85	10 (Ad)	
9	Pd(OTFA) ₂	$(NH_4)_2S_2O_8$	31	0	
10	Pd(OTFA) ₂	oxone	27	0	
11^e	Pd(OTFA) ₂	35% aq H_2O_2	21	0	
12	none	$PhI(O_2CAd)_2$	0	0	

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9	$Pd(OTFA)_2$	$(NH_4)_2S_2O_8$	31	0	
10	$Pd(OTFA)_2$	oxone	27	0	Strong oxidants
11^e	$Pd(OTFA)_2$	35% aq H_2O_2	21	0	
12	none	$PhI(O_2CAd)_2$	0	0	

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10	Pd(OTFA) ₂	oxone	27	0	
11^e	$Pd(OTFA)_2$	35% aq H_2O_2	21	0	
12	none	$PhI(O_2CAd)_2$	0	0	Without palladium

^aAll reactions were run at 0.2 mmol scale. ^bYield obtained by ¹H NMR with CF₃-DMA as internal standard. ^c2a (1.2 equiv). ^d2a (2.0 equiv). ^eOxidant (5.0 equiv). ^fIsolated yield of ester 4a.

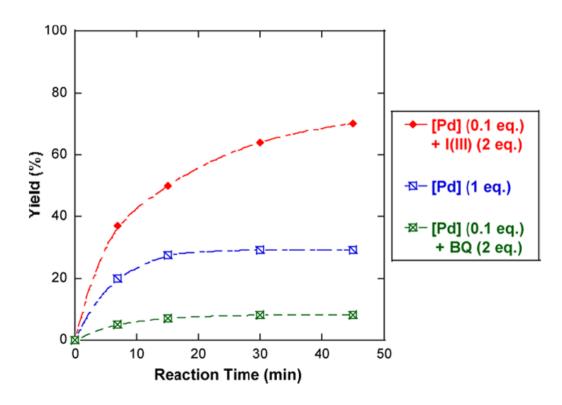


Figure 1. Time course of controlling experiments. $[Pd] = Pd-(O_2CCF_3)_2$, $I(III) = PhI(O_2CAd)_2$, BQ = benzoquinone.

Some examples..

2a = HN-oxa

Unreactive alkenes..

Deuterium labeling..

- First intermolecular aminocarbonylation pallado-catalyzed.
- Functionnalized β -aminoacids from simple and available alkenes.
- PhI(O₂CAd)₂ not only used as oxidant but can improve the reactivity of Pd catalyst.
- The mechanism need to be more studied to understand exactly the role of the hypervalent iodine.

Thank you for your attention