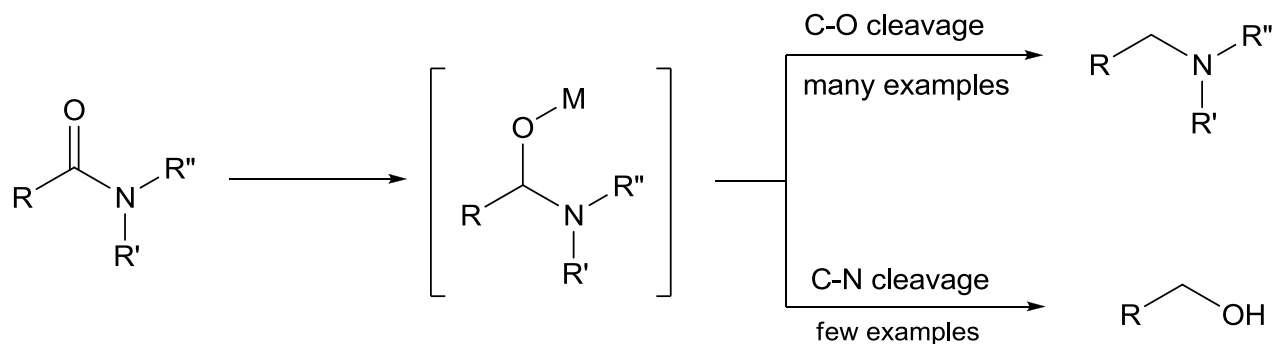


Highly Chemoselective Reduction of Amides (Primary, Secondary, Tertiary) to Alcohols using SmI_2 /Amine/ H_2O under mild conditions

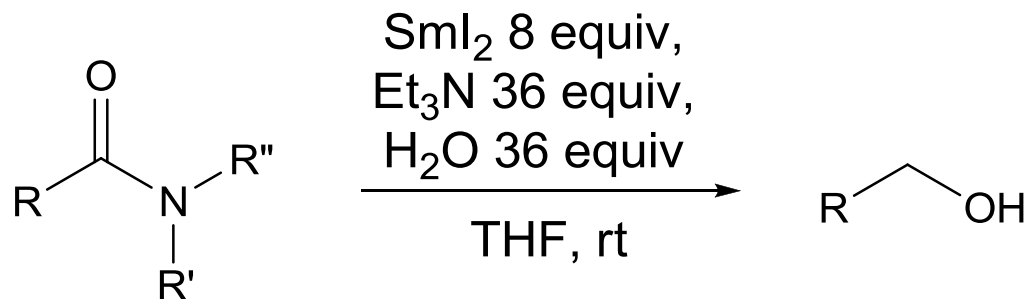
M. Szostak, M. Spain, A. J. Eberhart and D. Procter, *J. Am. Chem. Soc.*, **2014**, asap
Doi : 10.1021/ja412578t

State of the art



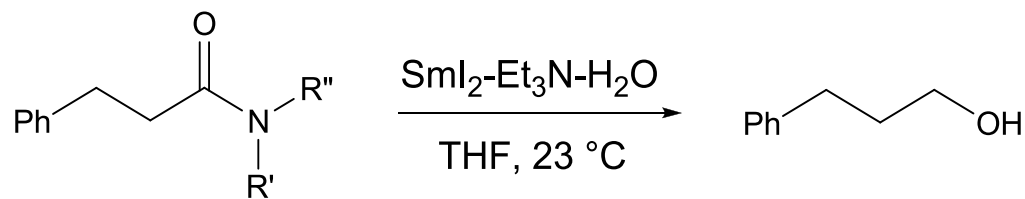
- Lots of examples involving C-O bond cleavage
- Few involving C-N bond cleavage
- C-N bond cleavage involves specific substrates

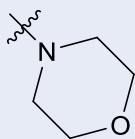
Experiment design



- Mild, operationnally conditions
- General method for primary, secondary, tertiary amides
- Excellent C-N bond cleavage selectivity

Selected examples



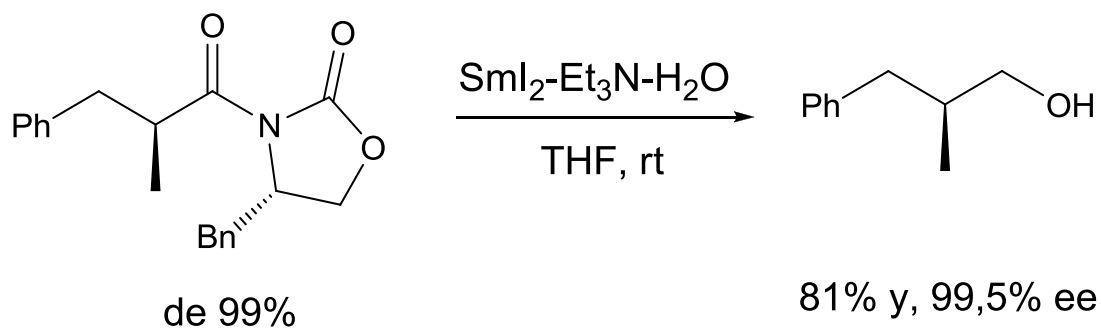
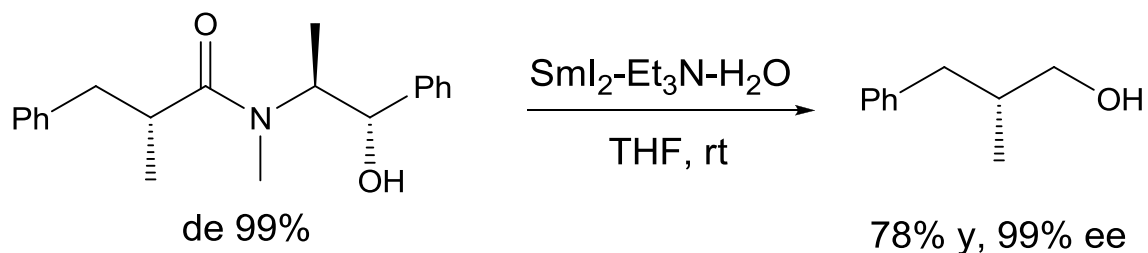
Entry	NR'R''	Yield (%)	Selectivity
1	NH ₂	82	>95:5
2	NH <i>n</i> -Bu	89	>95:5
3	NHPh	84	>95:5
4	NEt ₂	82	>95:5
5	N(OMe)Me	97	>95:5
6		83	>95:5
7	N(<i>i</i> -Pr) ₂	<5	>95:5

Selected examples

entry	3	amide	NR'R''	yield (%)
1	3a		NH ₂	91
2	3b		NH <i>n</i> -Bu	76
3	3c		NH ₂	95
4	3d		NH ₂	86
5	3e		NEt ₂	75
6	3f		NH ₂	93
7	3g		NH <i>n</i> -Bu	82
8	3h		NEt ₂	84
9	3i		NH ₂	94

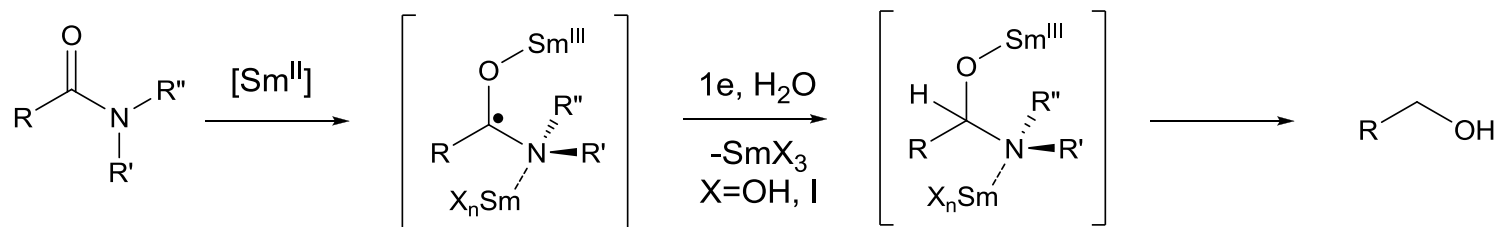
entry	3	amide	NR'R''	yield (%)
10	3j		X = H, NH <i>n</i> -Bu	96
11	3k		X = F, NH ₂	94
12	3l		X = Cl, NH ₂	85
13	3m		X = Br, NH ₂	63
14	3n		X = CF ₃ , NH ₂	73
15	3o		X = MeO, NH <i>n</i> -Bu	91
16	3p		NH ₂	82
17	3q		R = CH ₂ CO ₂ Na Glycochenodeoxycholic acid sodium salt	79
18	3r			61

Enantioselective pathway

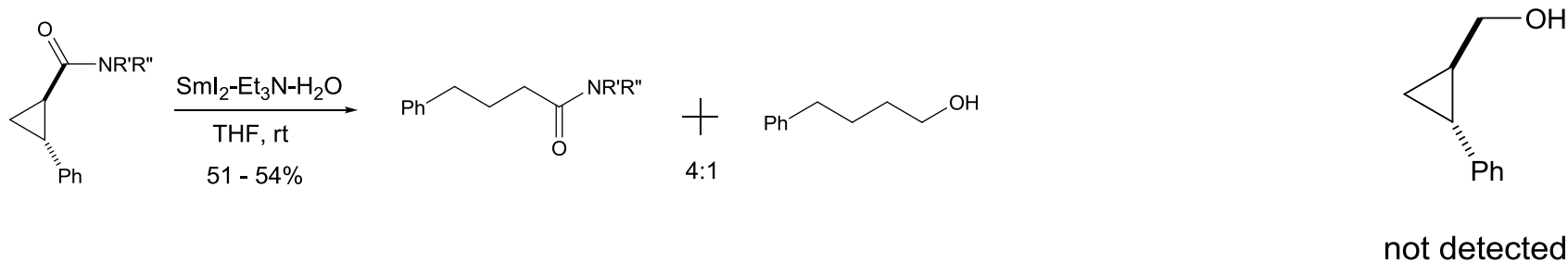


Mechanistic studies

- Proposed mechanism

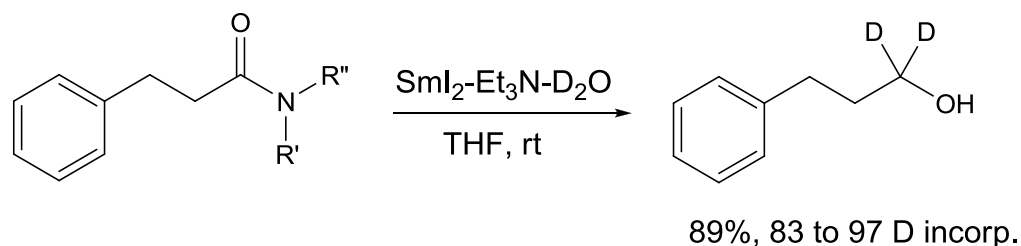


- Radical clock fragmentation

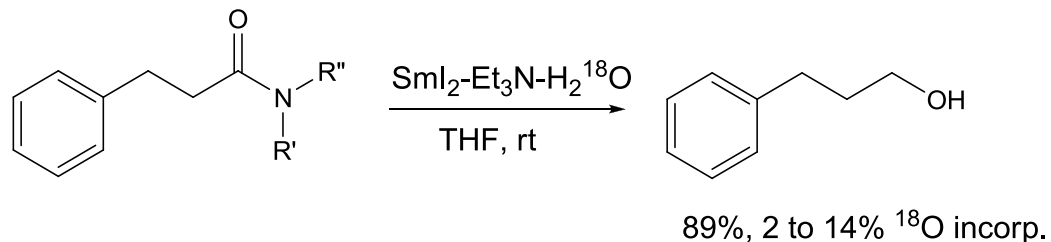


Mechanistic studies

- D₂O incorporation study



- H₂¹⁸O incorporation study



Conclusion

- General method to obtain alcohol from amide
- High yield and selectivity
- Large scope
- Mechanism to clarify



Thank you for your attention