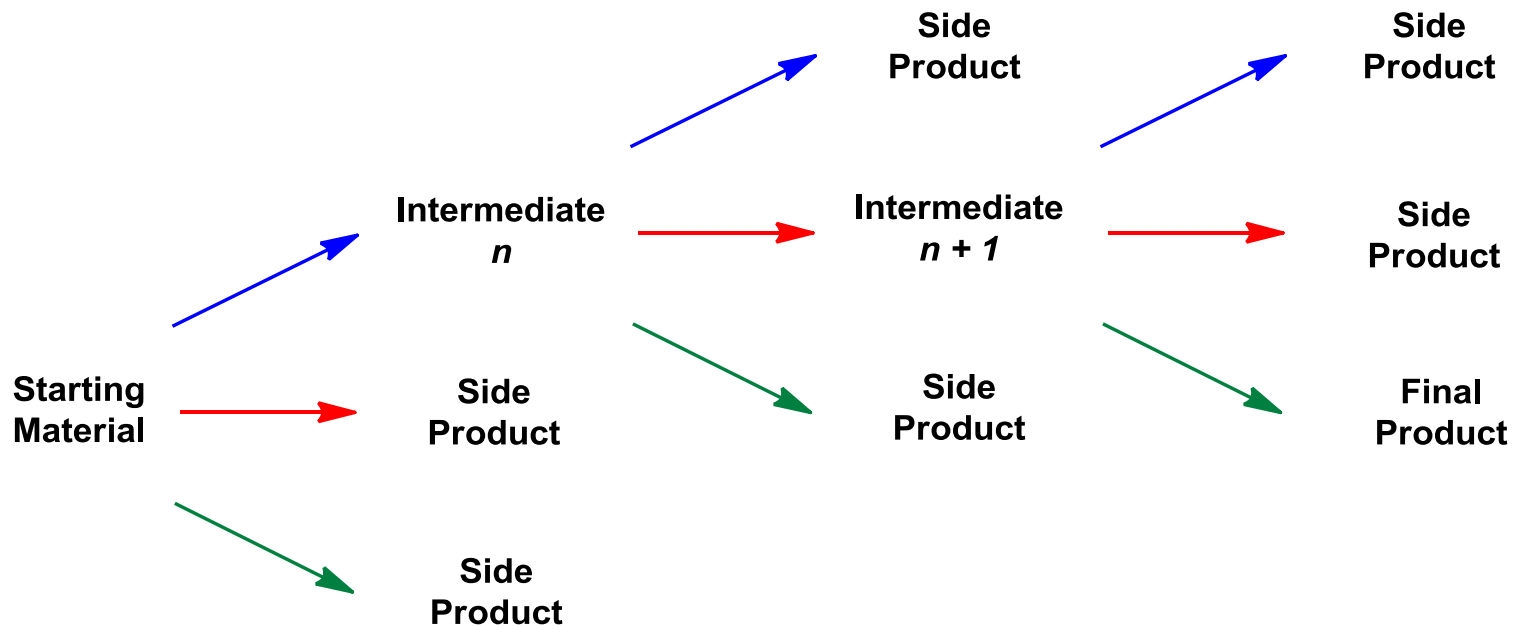
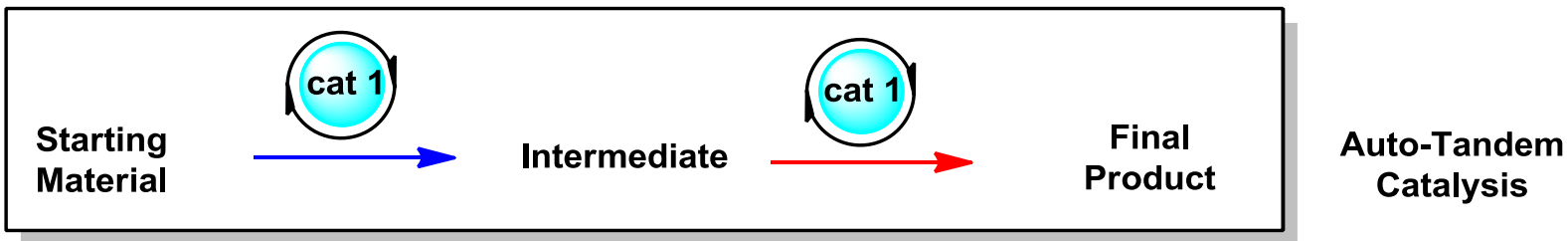


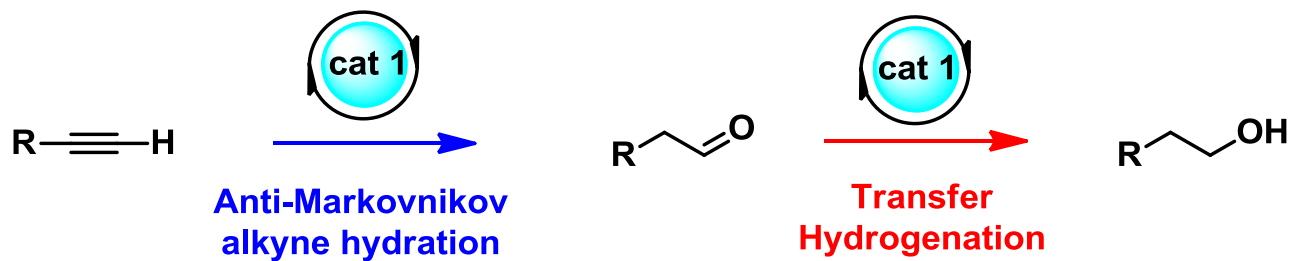
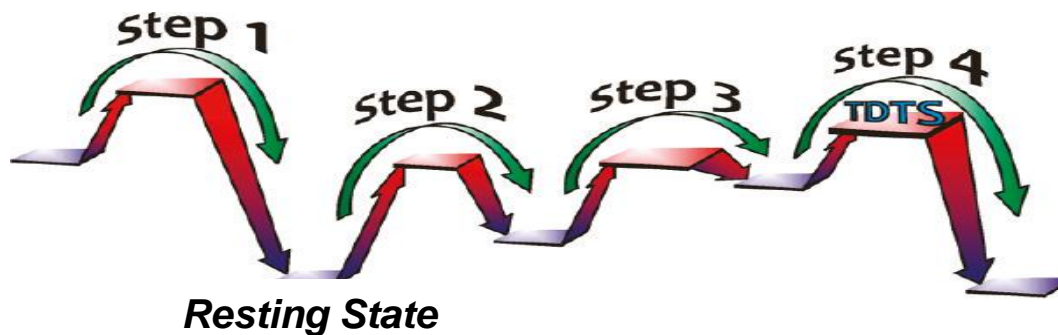
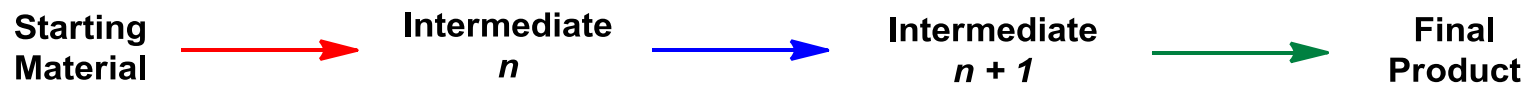
***Temporal separation of catalytic activities  
allows anti-Markovnikov reductive  
functionalization of terminal alkynes***

**L. Li, S. B. Herzon, *Nature Chemistry*, 2014, 6, 22**

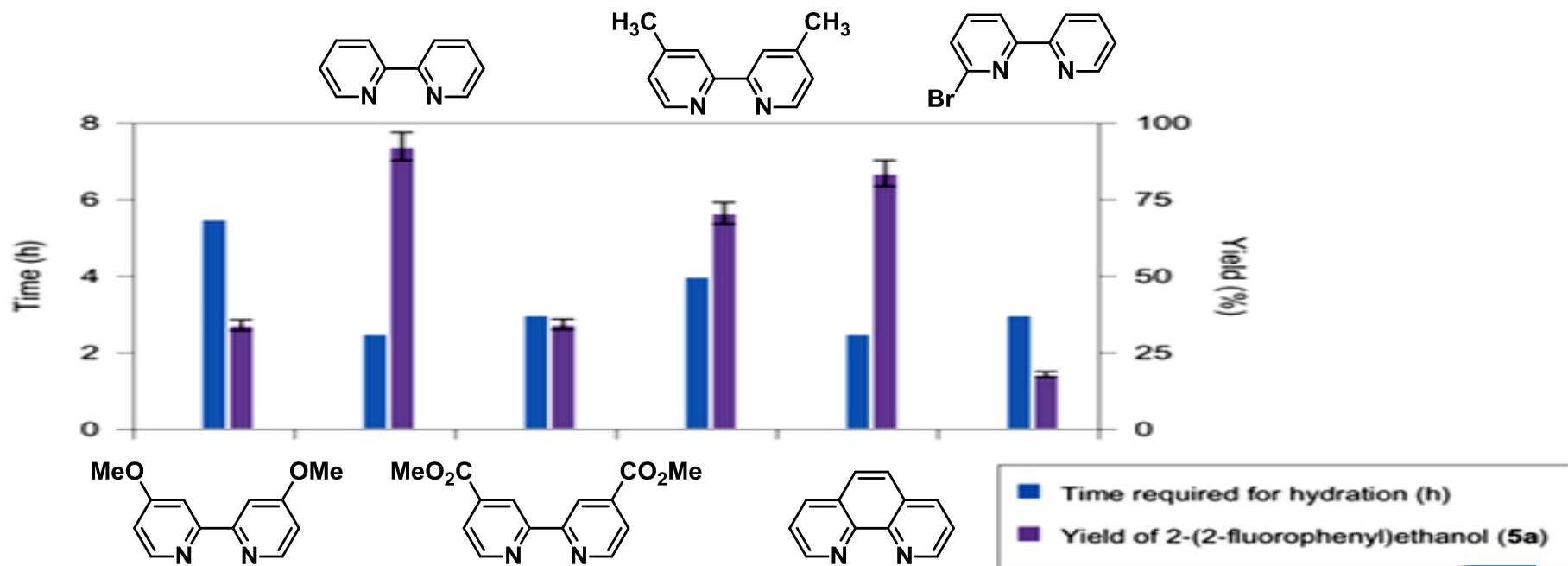
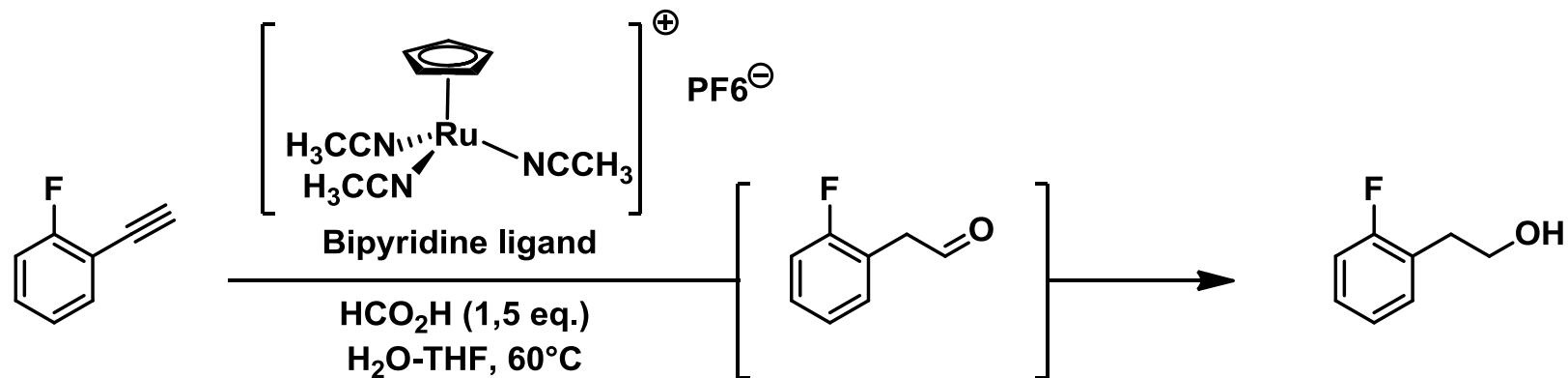
# Auto-Tandem Catalysis : Efficiency Increases with Complexity



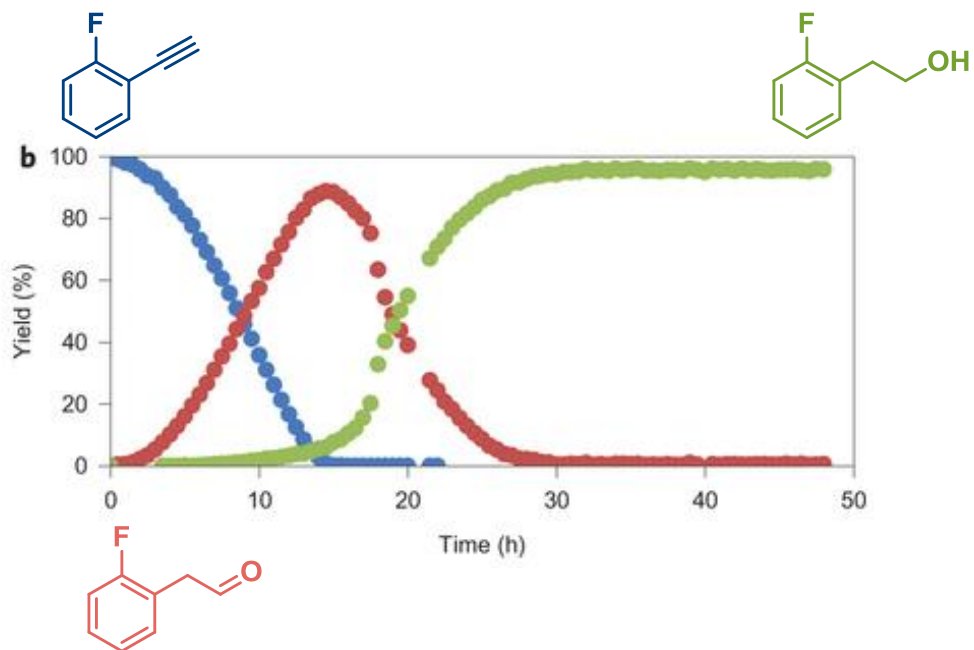
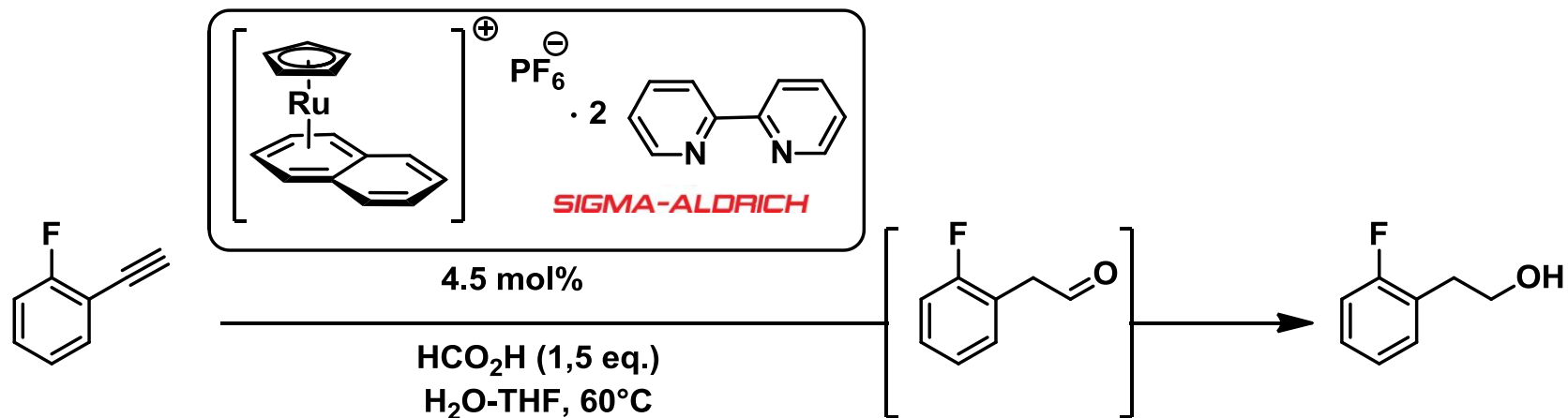
# An Alternative Strategy : Temporal Separation



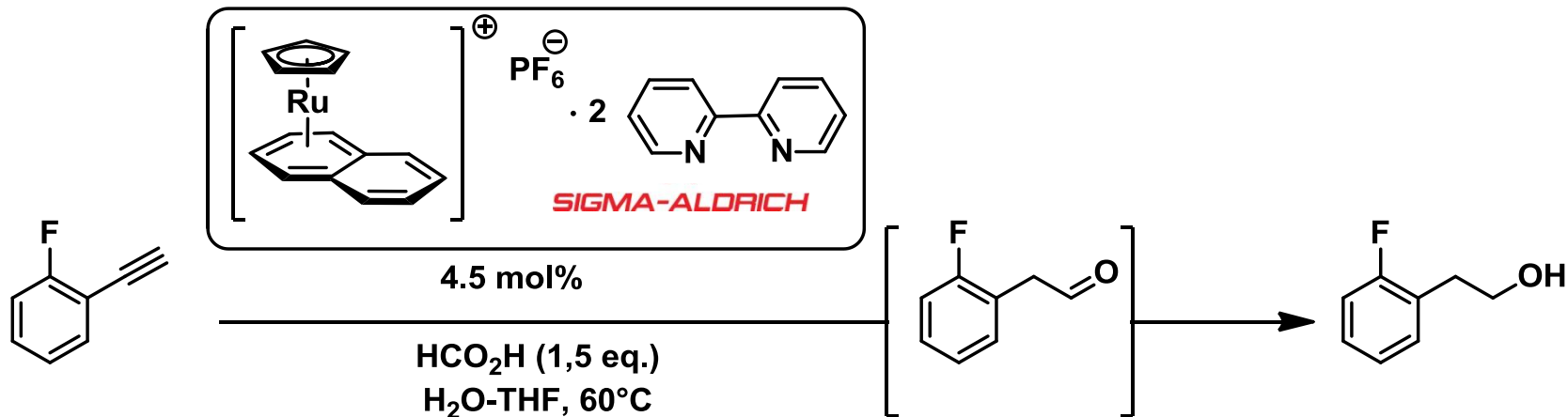
# Preliminary Studies



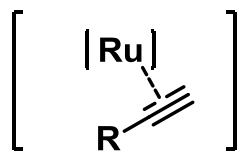
# Temporal Separation Probe



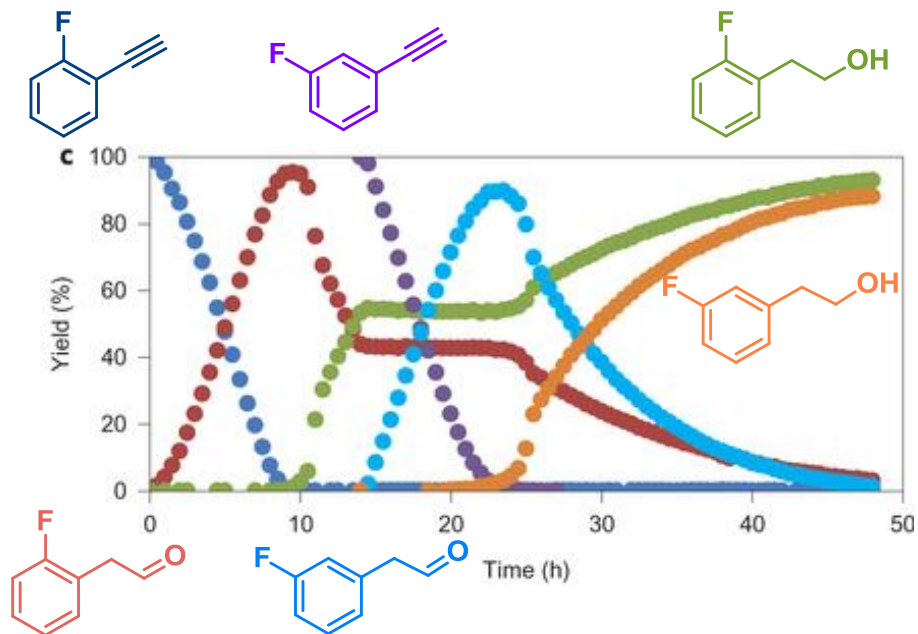
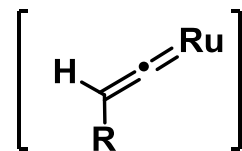
# Temporal Separation Probe



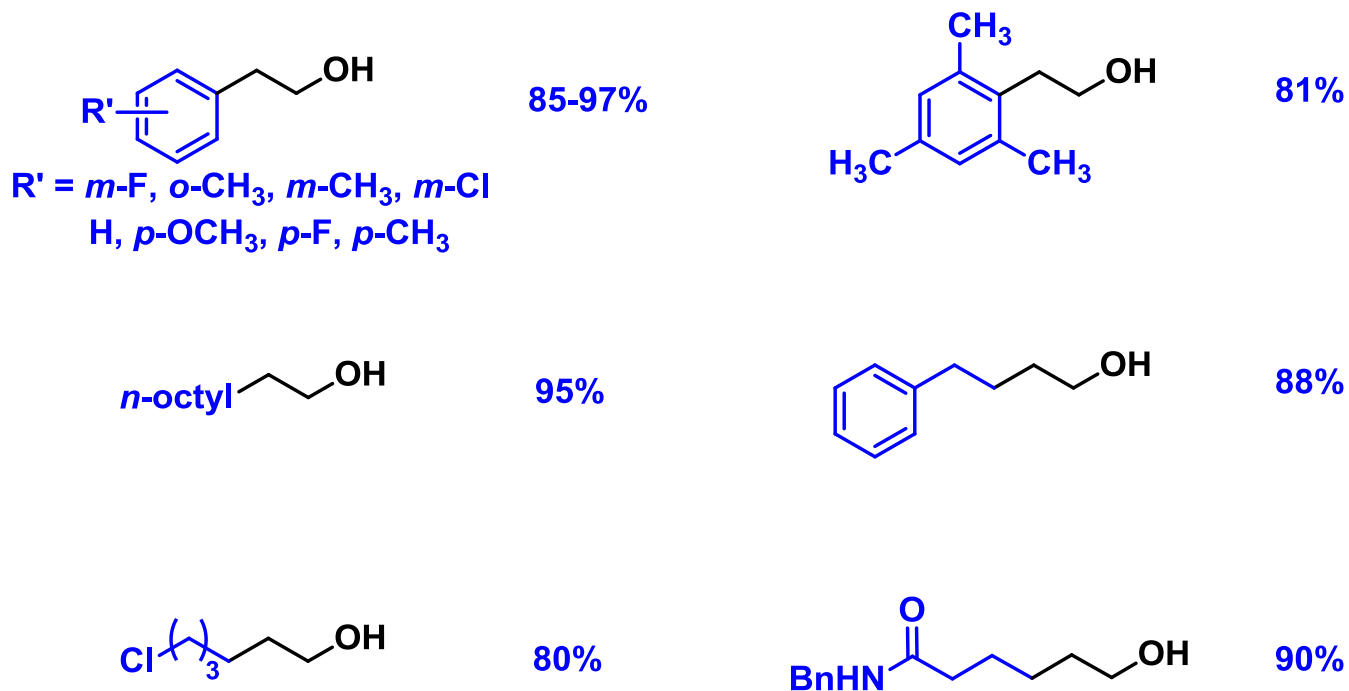
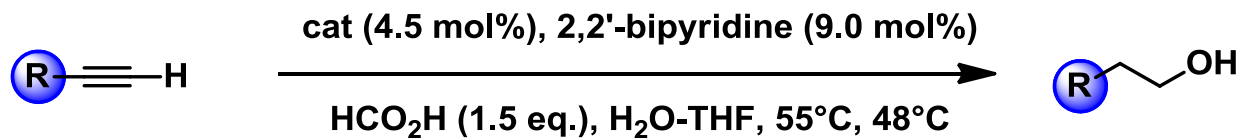
## Possible Resting-States



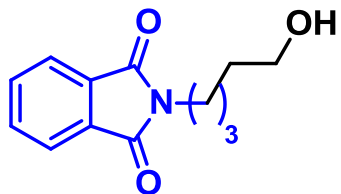
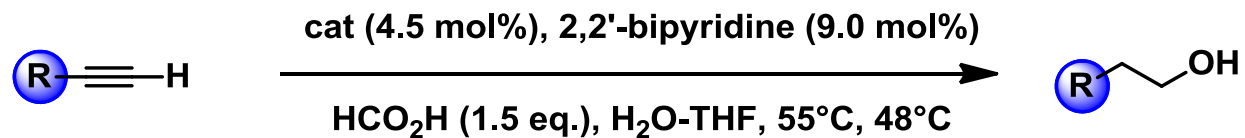
or



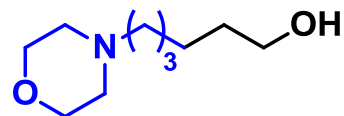
# Scope of the Reaction



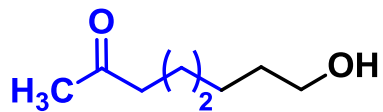
# Scope of the Reaction



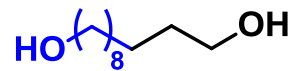
93%



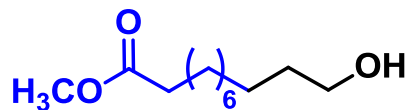
81%



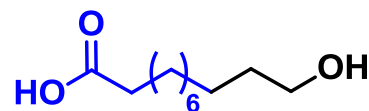
87%



80%



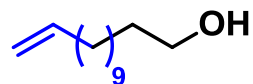
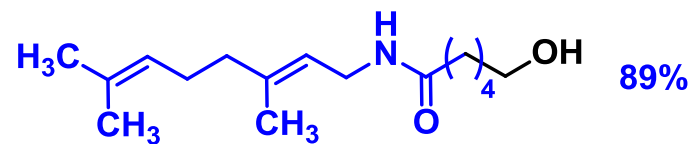
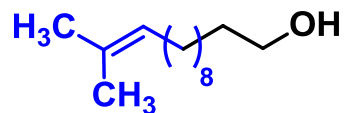
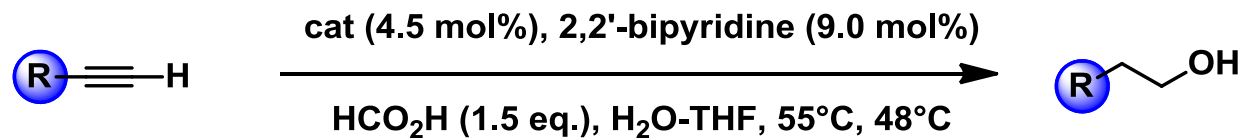
90%



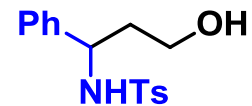
80%



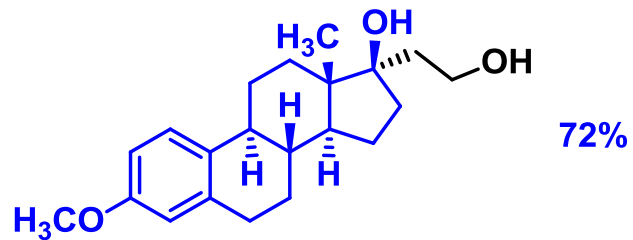
# Scope of the Reaction



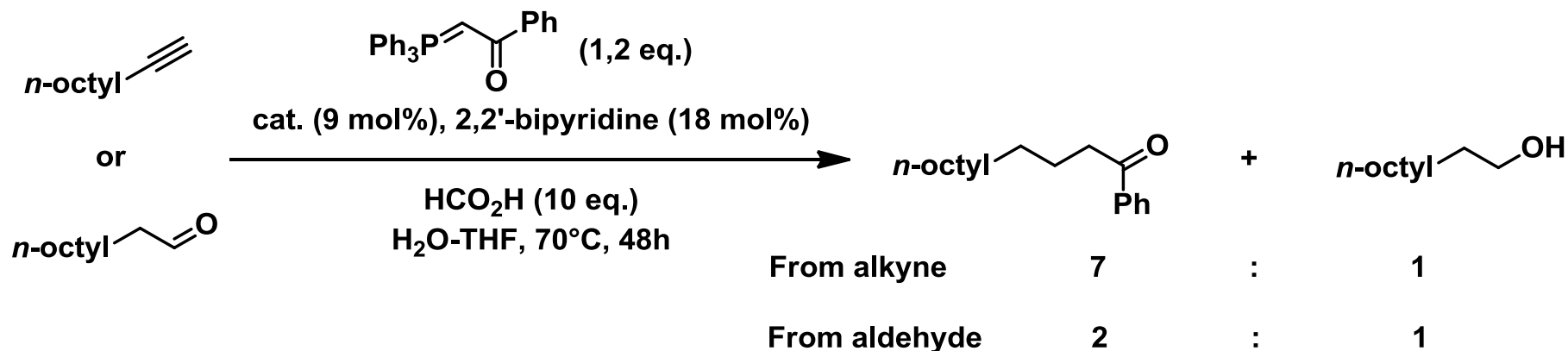
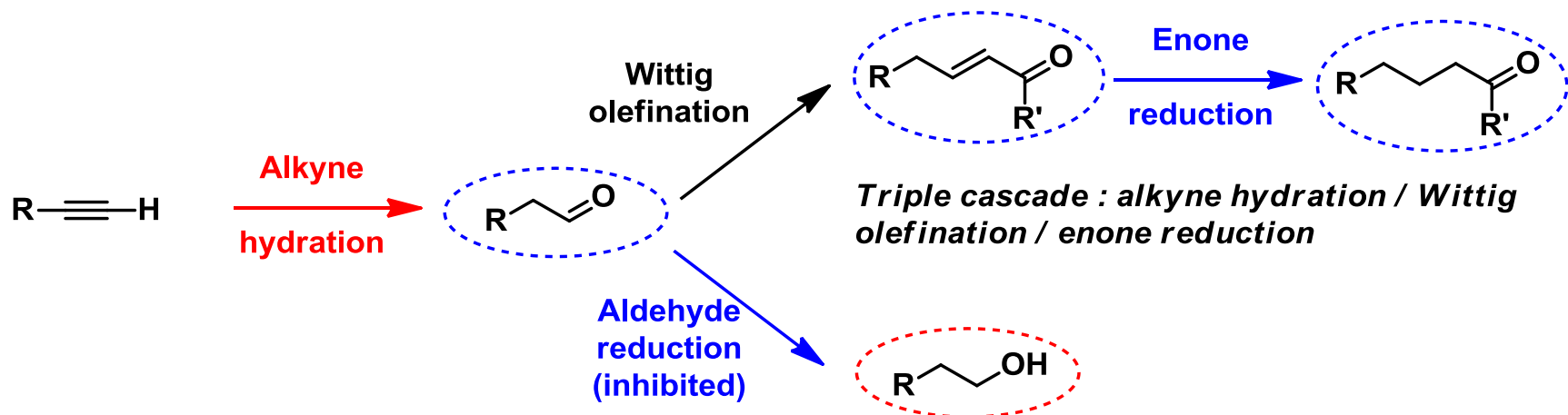
90%



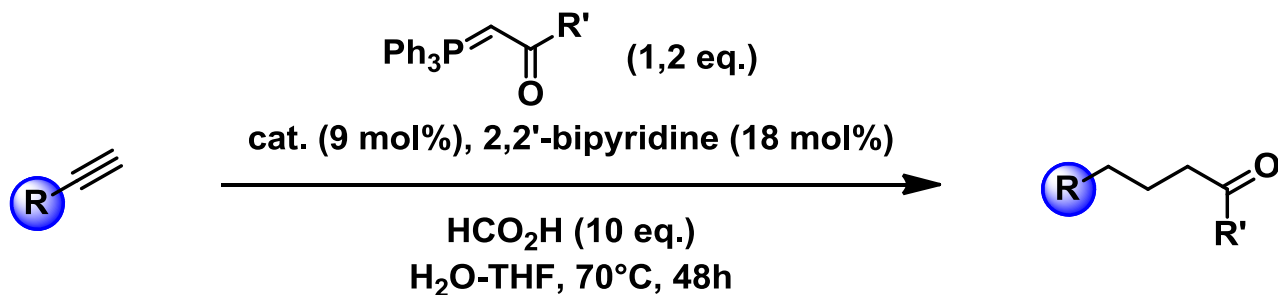
83%



# Toward a Triple Cascade



# Scope of the Reaction



	R' = Ph	R' = CH <sub>3</sub>		R' = Ph	R' = CH <sub>3</sub>
	75%	71%		81%	76%
<i>n</i> -octyl	72%	70%		72%	76%
	75%	70%			

# Conclusion

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- ***The first recognition and investigation of temporal separation in tandem catalysis***
- ***A strategy which cannot be used in all catalytic systems***
- ***Novel candidate reactions need to be developed***
- ***A further understanding of the reaction mechanism must be provided notably with the highlighting of the resting state structure***