



Organocatalytic Carbonyl-Olefin Metathesis

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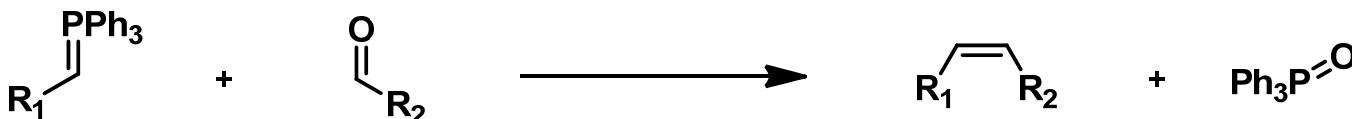
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Christèle Roux

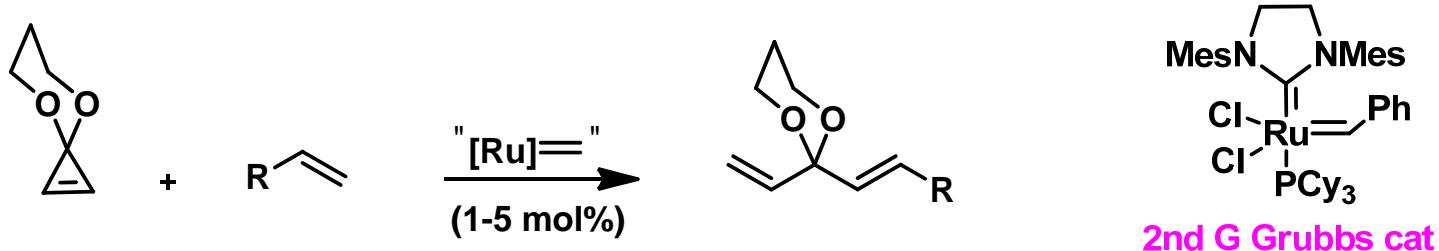
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The most widely used olefination

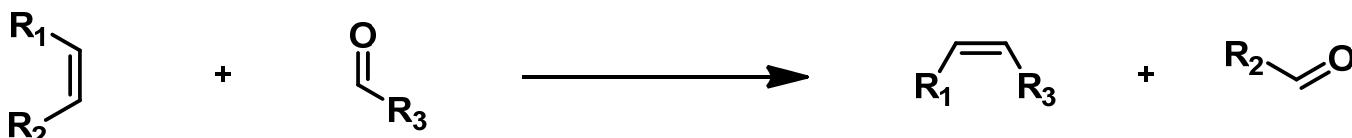
✓ Wittig olefination



✓ Metathesis olefination



✓ Carbonyl-olefin metathesis



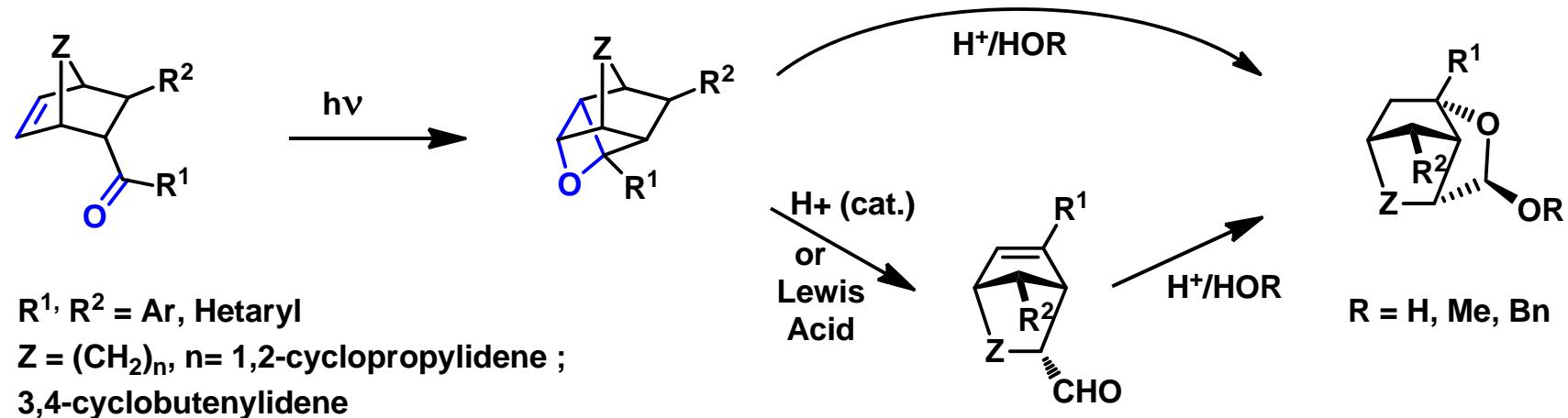
Maryanoff, B. E.; Reitz, A. B. *Chem. Rev.* **1989**, *89*, 863–927.

Michaut, M.; Parrain, J.-L.; Santelli, M. *Chem. Commun.* **1998**, 2567–2568.

Hoveyda, A. H.; Zhugralin, A. R. *Nature* **2007**, *450*, 243–251.

Carbonyl-olefin metathesis precedents

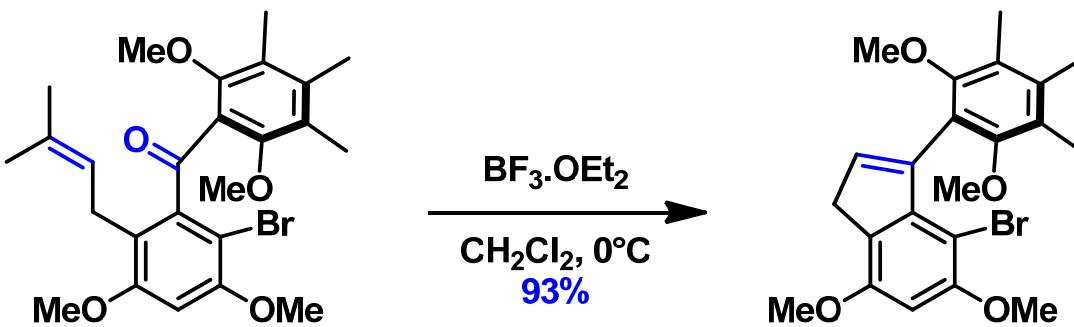
✓ Photochemical carbonyl-olefin metathesis reaction :



- high-yielding two-step sequence

Carbonyl-olefin metathesis precedents

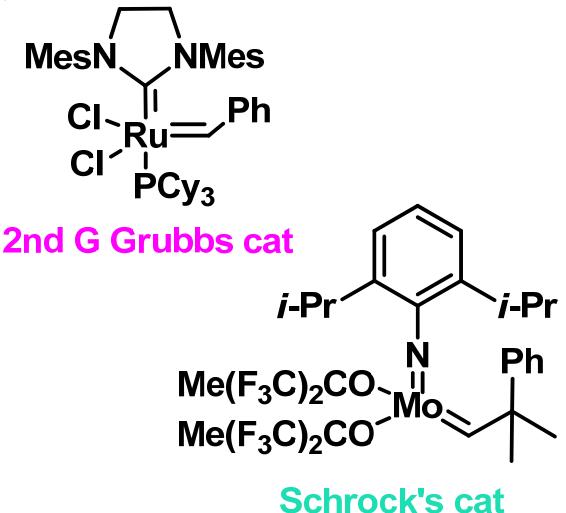
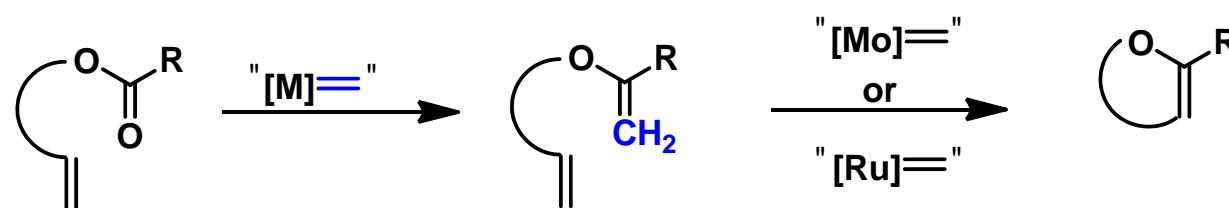
✓ Lewis acid promoted carbonyl-olefin metathesis reaction :



- Cationic cyclization process => formation of an oxetane intermediate followed by a fragmentation reaction
- Several substrates were shown to afford the metathesis products with up to 93% yield.

Carbonyl-olefin metathesis precedents

✓ Metal mediated carbonyl-olefin metathesis reaction :



- **Tebbe reagent :** Cp₂TiCH₂(Cl)AlMe₂

- *disadvantages* : sensitive to moisture and air / Lewis acidic character / limited to methylenation

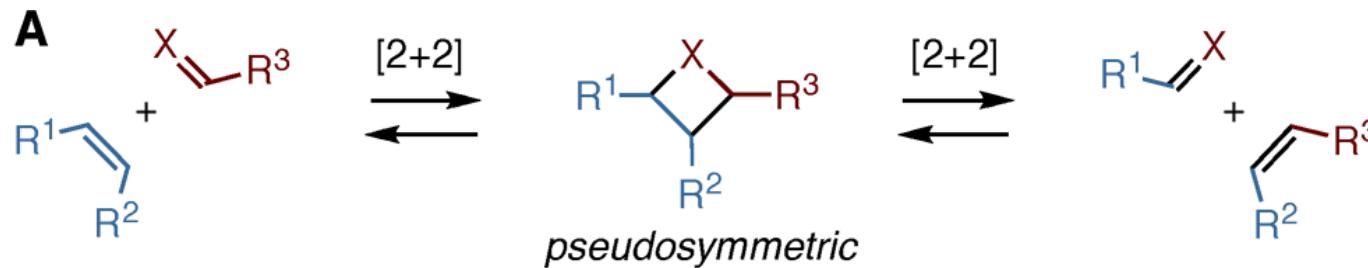
- **Petasis reagent :** Cp₂Ti(CH₃)₂

- *disadvantages* : high temperature needed to induce α -elimination

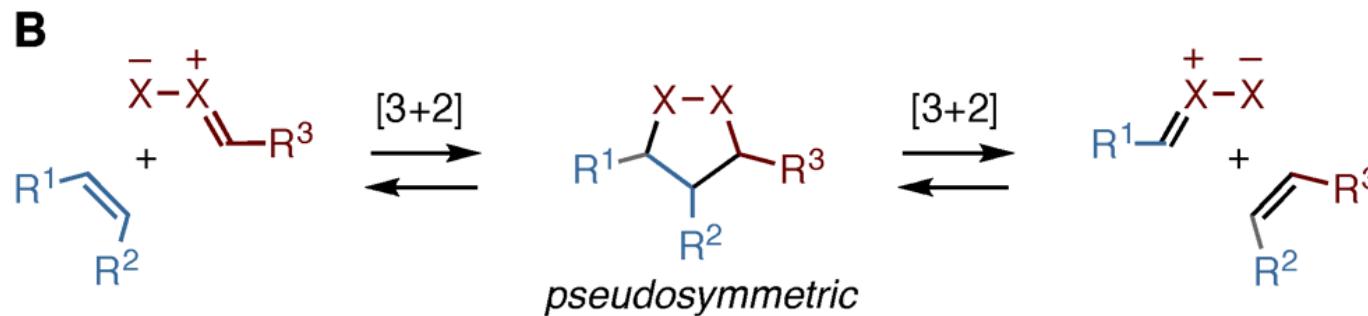
- **Takai reagent :** [TiCl₄, Zn, PbCl₂, AlkX₂, TMEDA]

Metathesis paradigm

A - Traditional [2 + 2] metathesis paradigm :

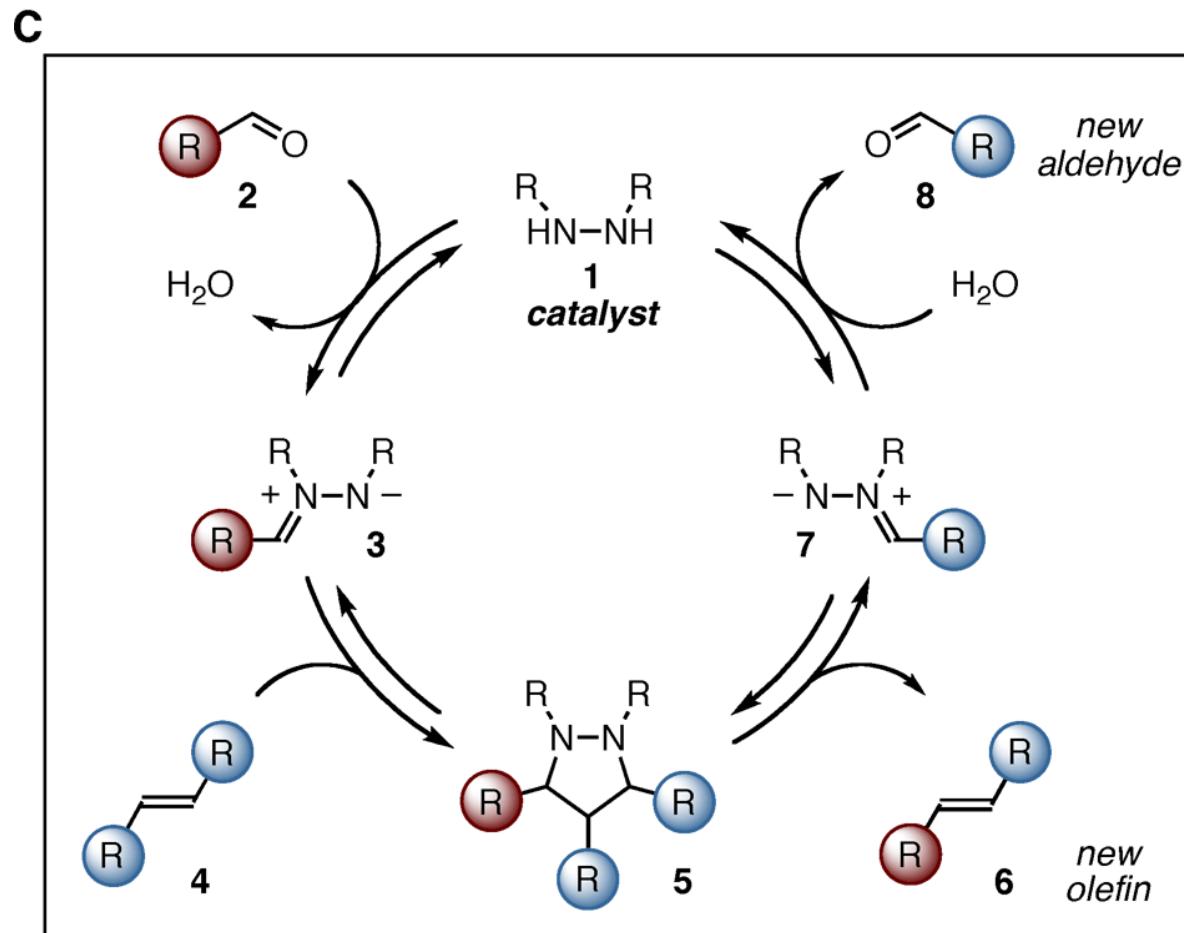


B - [3 + 2] metathesis paradigm.

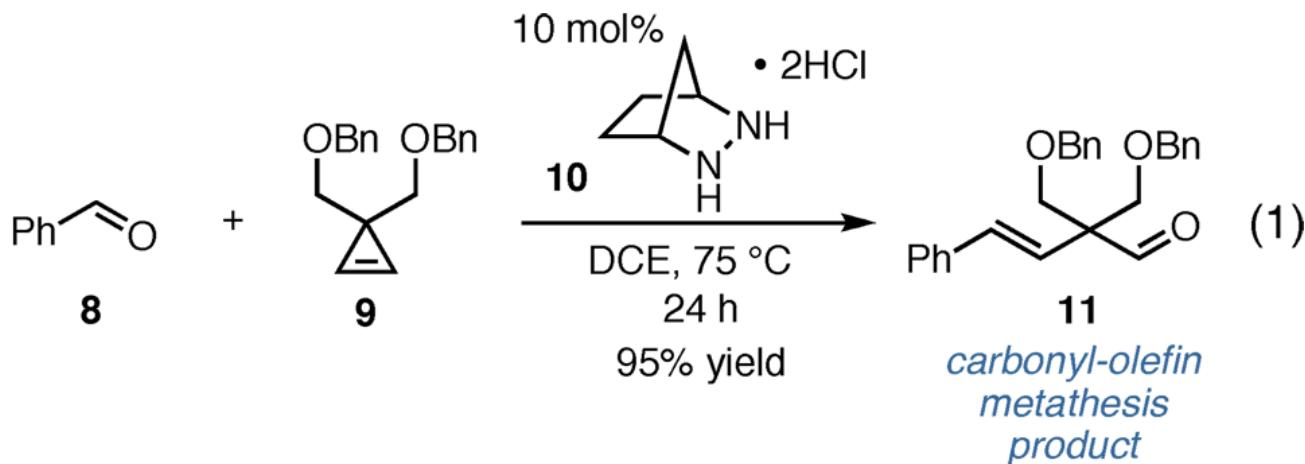


Metathesis paradigm

C- Carbonyl-olefin metathesis reaction based on azomethine imine 1,3-dipolar cycloadditions.

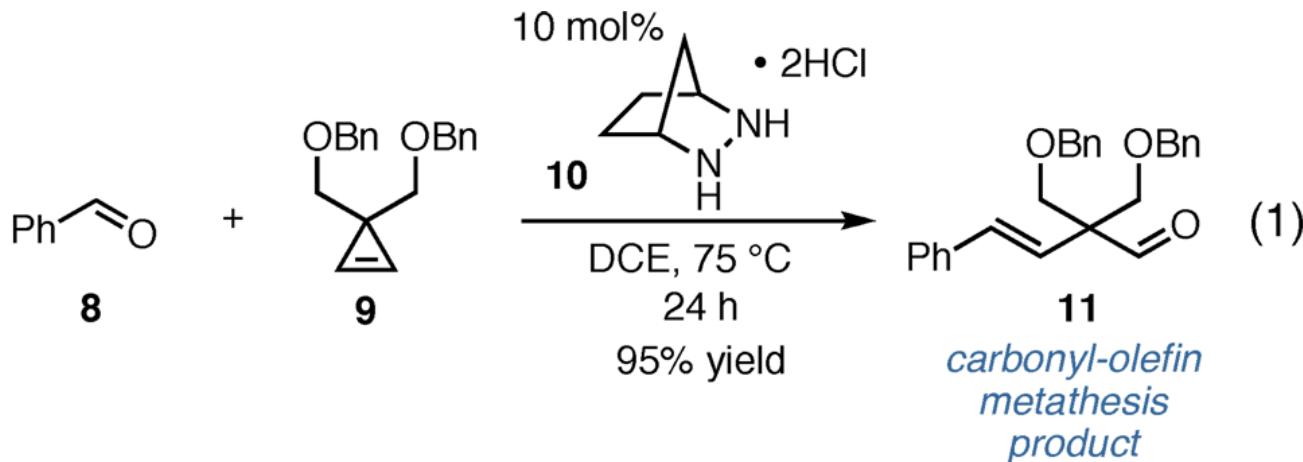


Hydrazine catalyzed carbonyl-olefin metathesis reaction

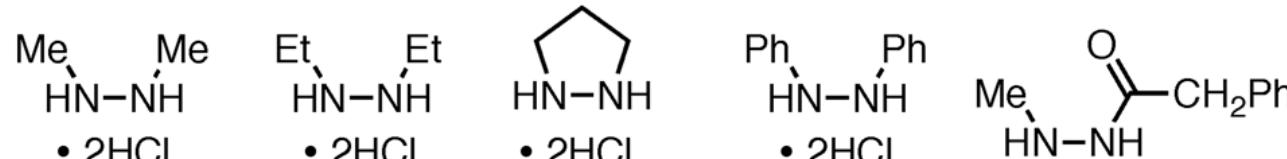


- No reaction was observed between **8** and **9** without catalyst **10**
 - With 50 mol % of catalyst :
 - 10.2HCl gave the product **11** in 60% yield
 - 10.HCl gave the product **11** in 35% yield
 - free base gave the product **11** in 15% yield
-

Hydrazine catalyzed carbonyl-olefin metathesis reaction



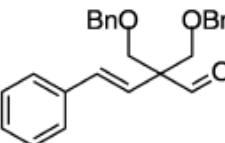
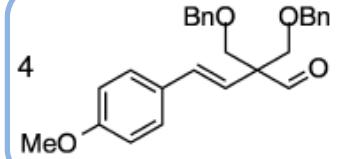
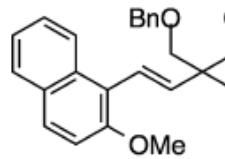
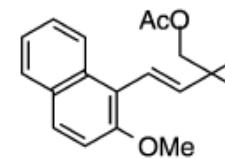
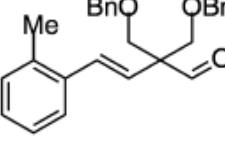
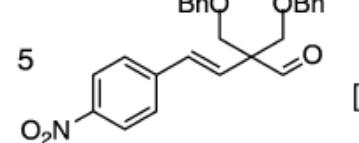
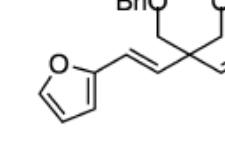
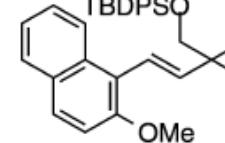
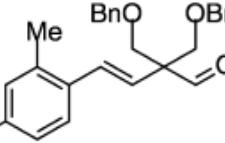
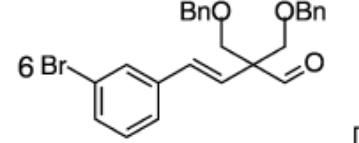
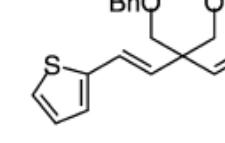
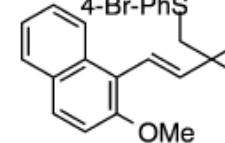
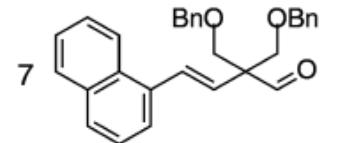
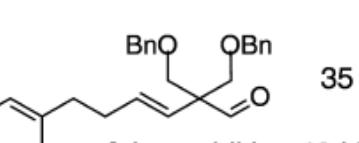
other catalysts (50 mol%)



— <10% product — | — no product —

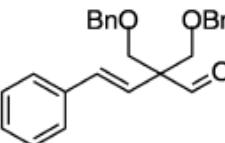
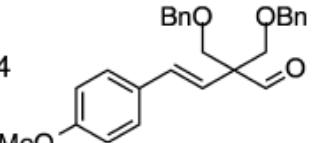
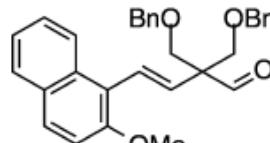
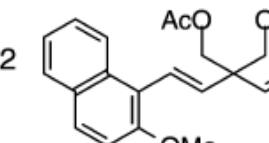
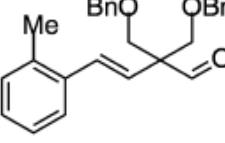
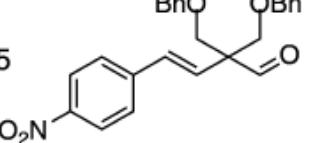
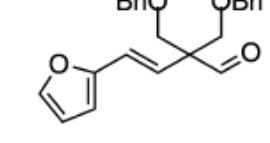
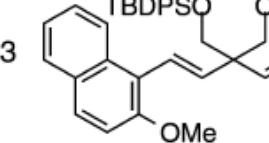
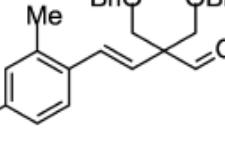
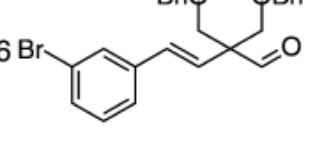
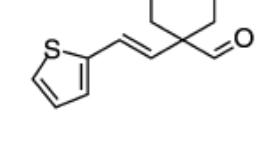
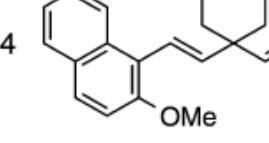
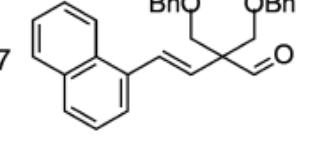
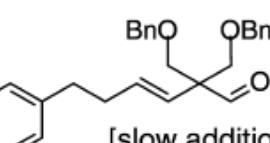
- Bicyclic structure => plays a key role in the high performance of this hydrazine catalyst.

Scope hydrazine catalyzed carbonyl-olefin metathesis reaction

entry	product	% yield	entry	product	% yield	entry	product	% yield	entry	product	% yield
1		95 (80)	4		50	8		68 [6 h]	12		68
2		85	5		60 [48 h]	9		80	13		79
3		66	6		75 (60) [48 h]	10		35	14		50
7		67	11		[slow addition 48 h]						

Inverse correlation between the electron-rich character of the aldehyde and the yield of isolated metathesis product.

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With substrates bearing EWG the conversion rate is slower => HOMO of the dipole => typical demand 1,3-dipolar cycloadditions.

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		67	11		35 [slow addition 48 h]						

Entry 10 : modest yield due to the sensitive nature of the vinylthiophene functionality

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Aliphatic aldehydes participate in a variety of amine-catalyzed transformations.

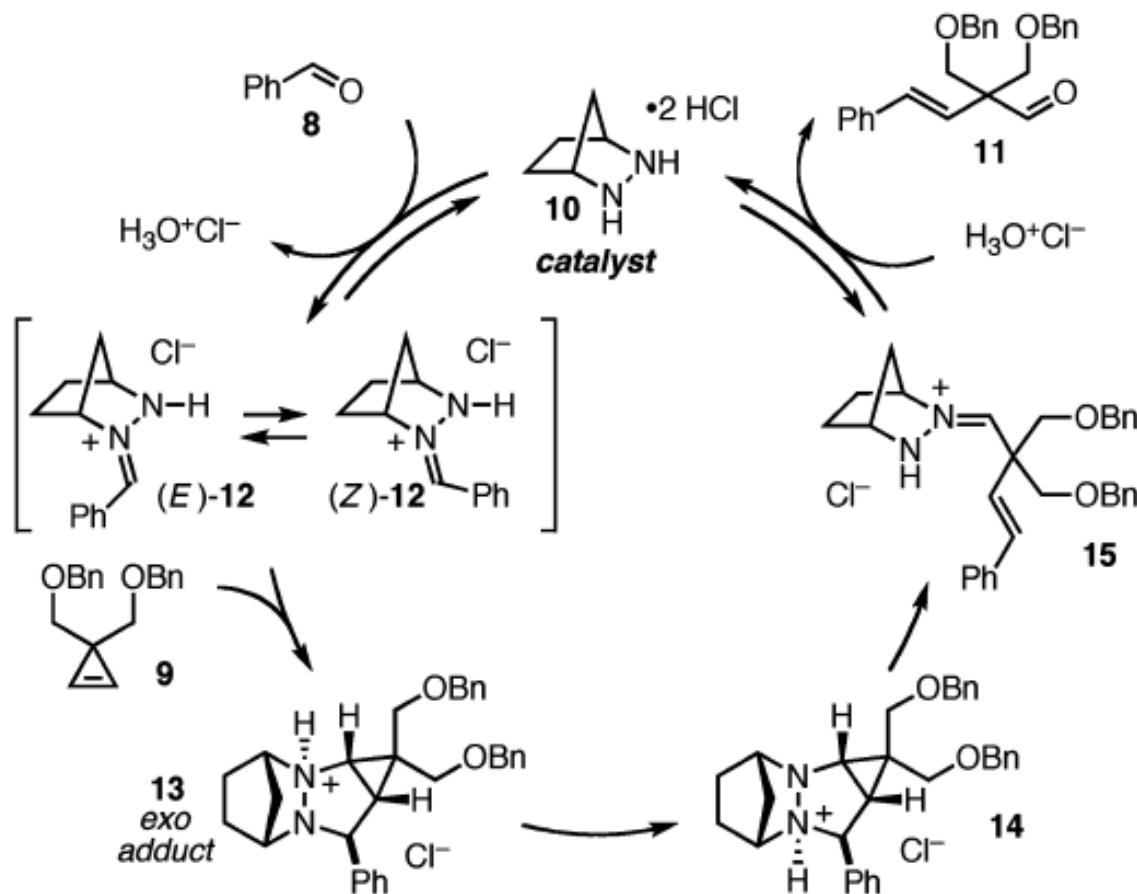
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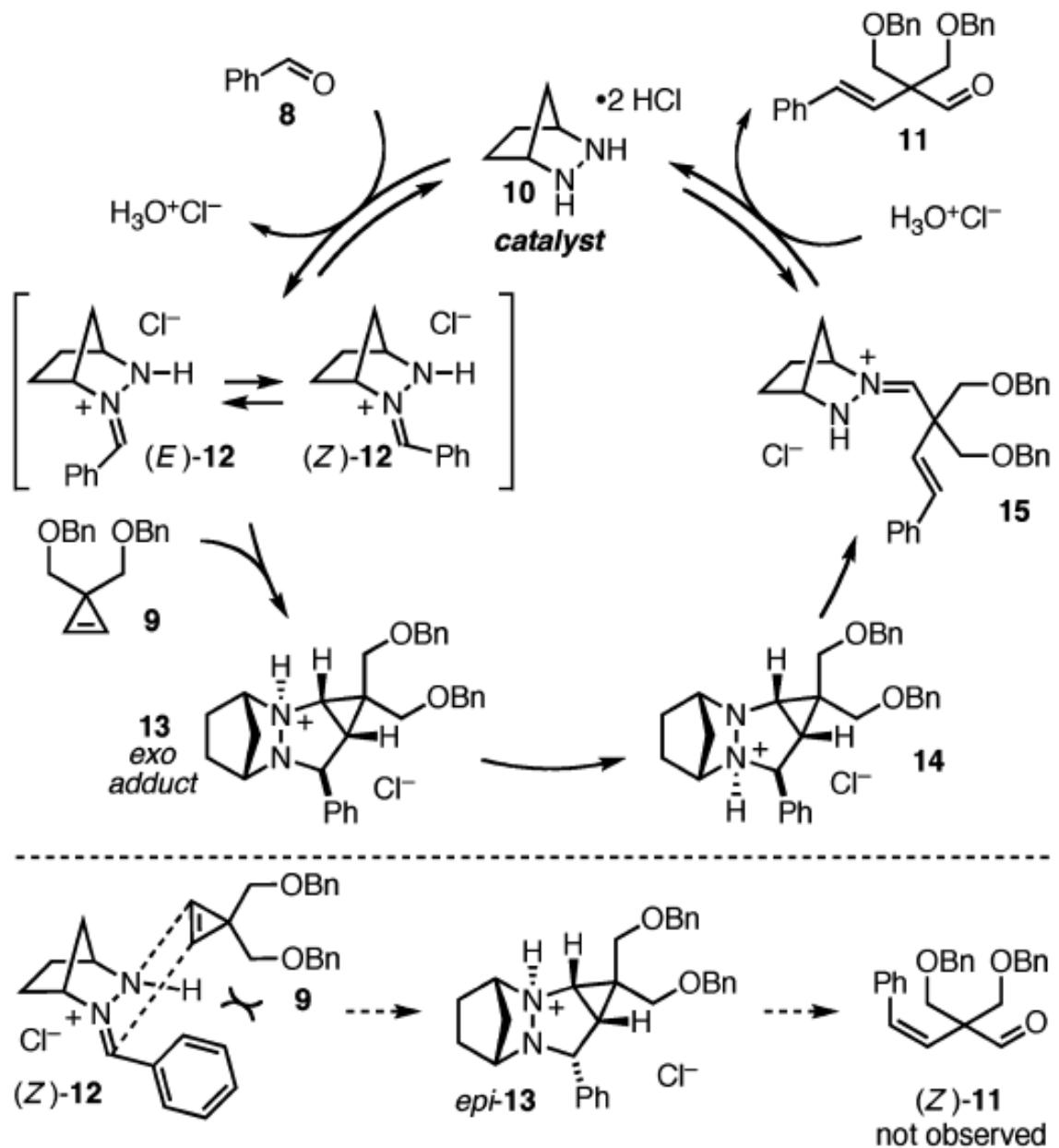
O-linkages are well tolerated.

Thioether were also found to be compatible.

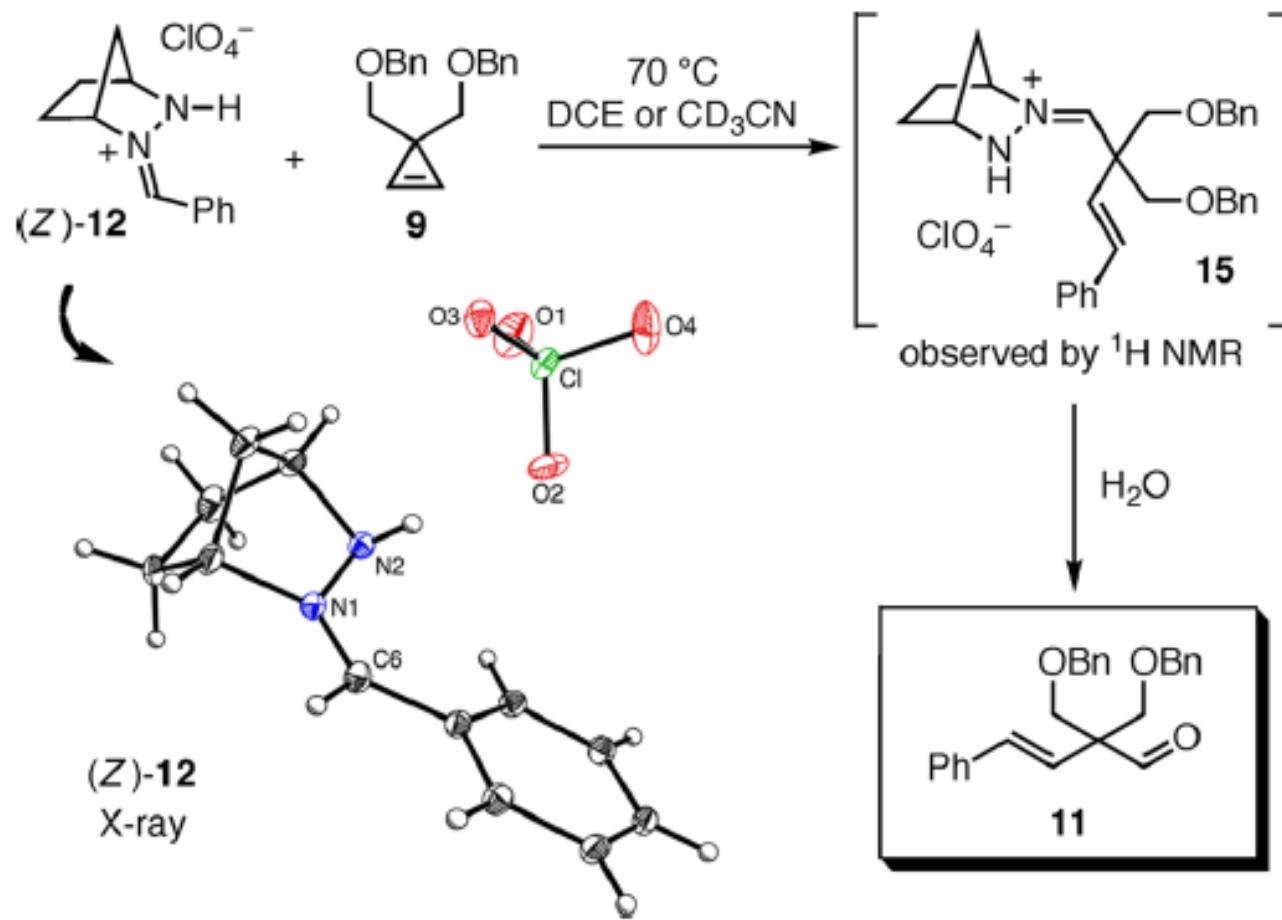
Mechanistic studies



Mechanistic studies



Mechanistic studies



Conclusion

- Carbonyl-olefin metathesis using simple organic catalysts and pericyclic reactions
- Development of catalysts able to perform the cycloreversion step
- Methodology applicable on a variety of substrates
- No asymmetric version

Thank you for your attention
