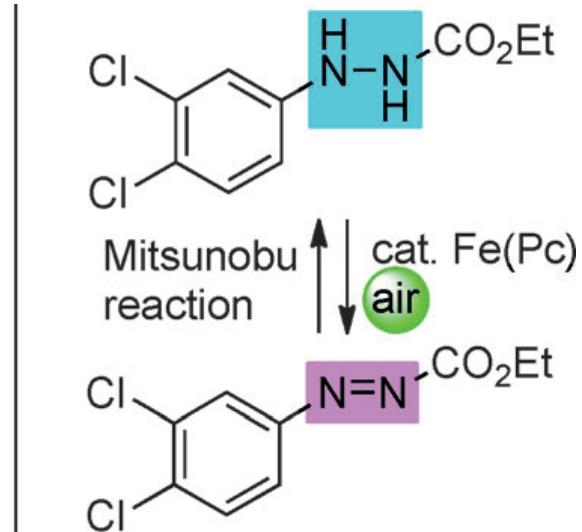
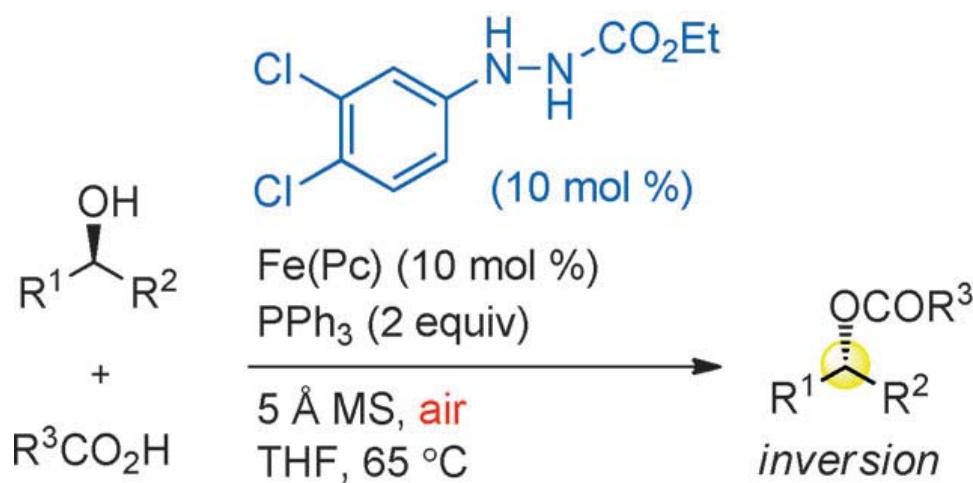


Recyclable Mitsunobu Reagents: Catalytic Mitsunobu Reactions with an Iron Catalyst and Atmospheric Oxygen

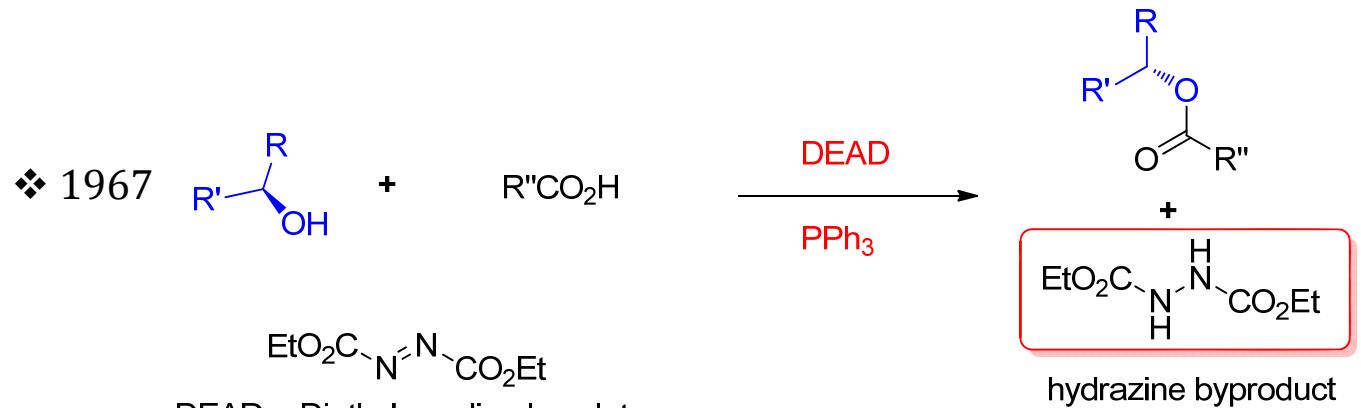
T.Taniguchi et. al., *Angew. Chem. Int. Ed.* 2013, 52, 1 – 6

Recyclable Mitsunobu Reagents: Catalytic Mitsunobu Reactions with an Iron Catalyst and Atmospheric Oxygen



T.Taniguchi et. al., *Angew. Chem. Int. Ed.* 2013, 52, 1 – 6

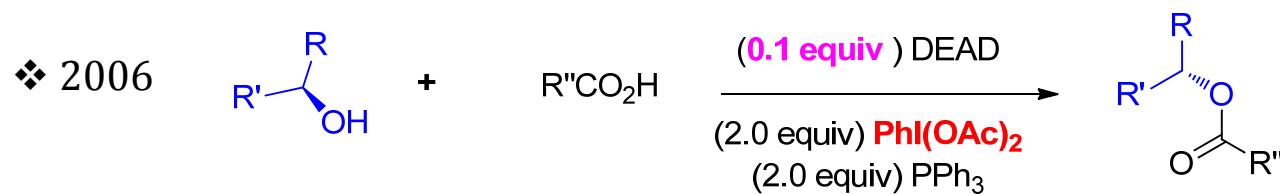
Literature precedent and background



Oyo Mitsunobu

O. Mitsunobu, Y. Yamada, *Bull. Chem. Soc. Jap.* **1967**, 40, 2380–2382.

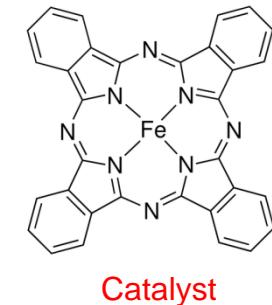
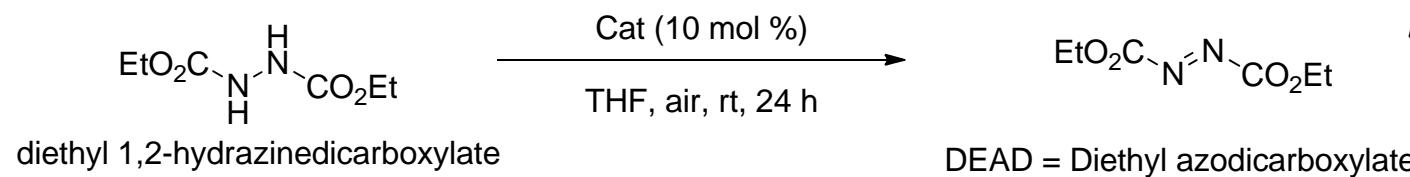
✓ Natural products synthesis : Tedanolide, spongidepsin



T. Y. S. But, P. H. Toy, *J. Am. Chem. Soc.*, **2006**, 128, 9636-9637.

Optimization of the new catalytic approach

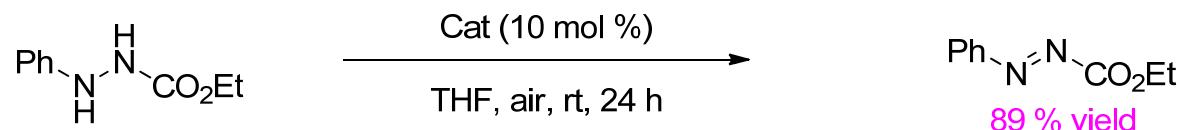
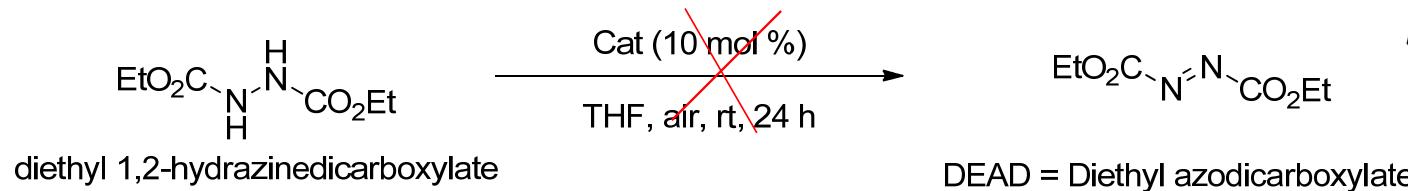
❖ Aerobic oxidation of hydrazines



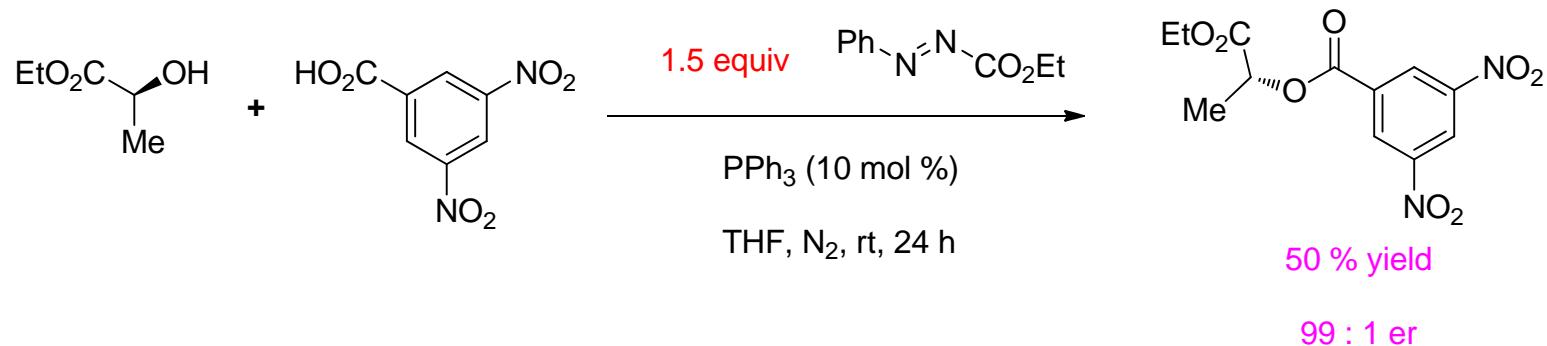
T.Taniguchi et. al., *Angew. Chem. Int. Ed.* 2013, 52, 1 – 6

Optimization of the new catalytic approach

❖ Aerobic oxidation of hydrazines

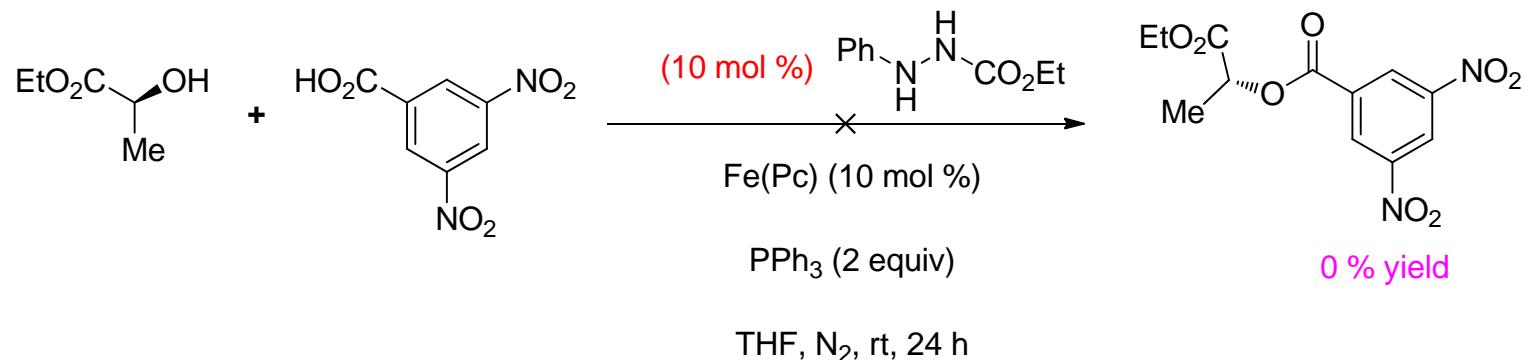
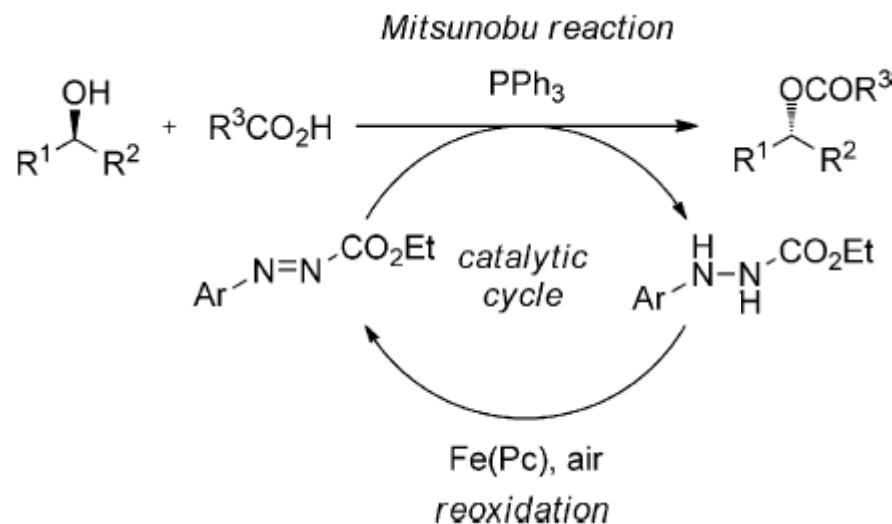


❖ Evaluation in the Mitsunobu reaction



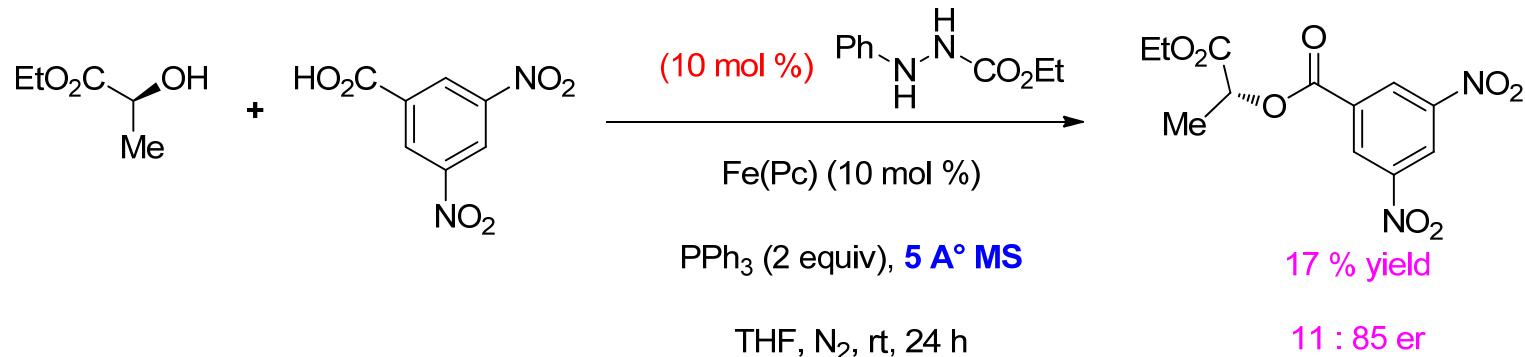
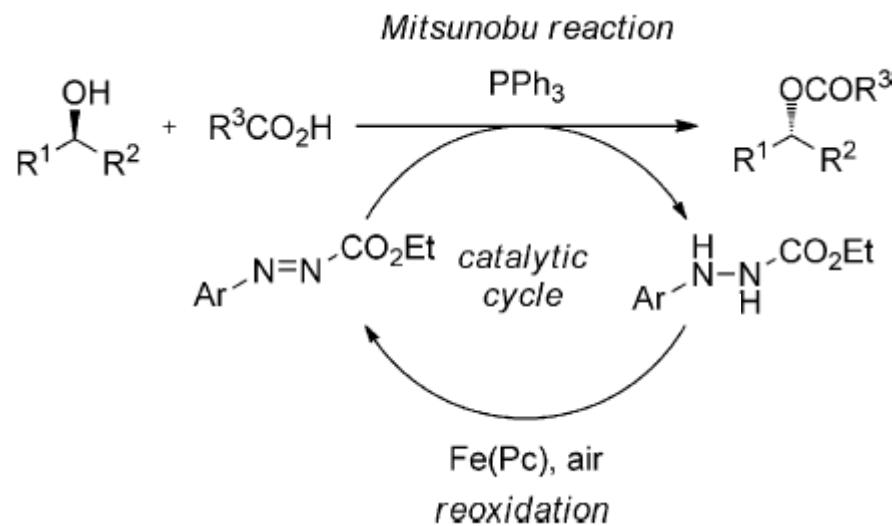
Optimization of the new catalytic approach

❖ Concept for a catalytic Mitsunobu reaction



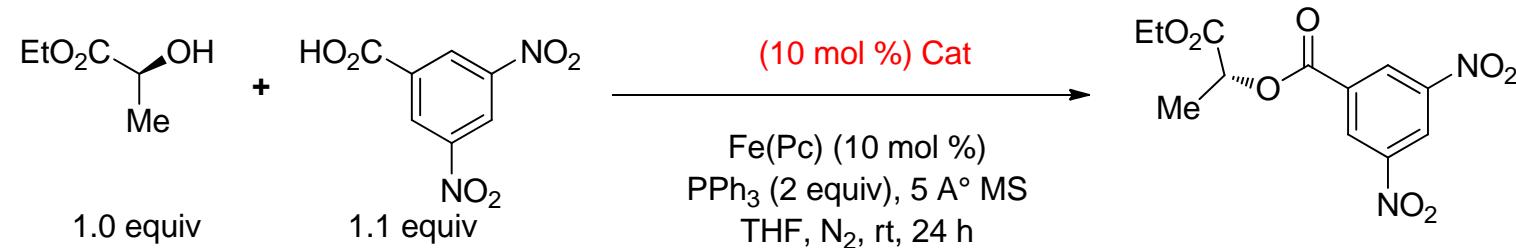
Optimization of the new catalytic approach

❖ Concept for a catalytic Mitsunobu reaction



Optimization of the new catalytic approach

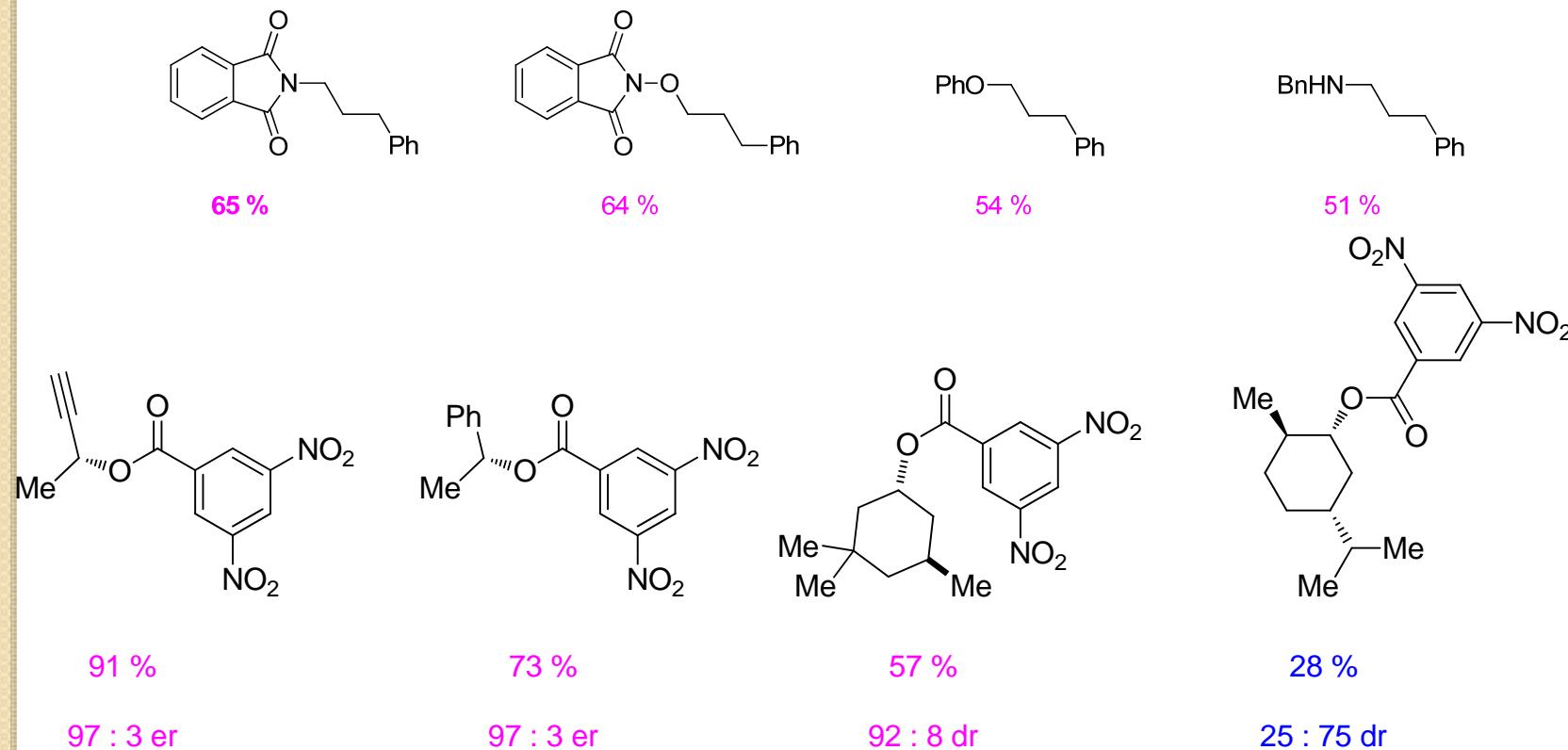
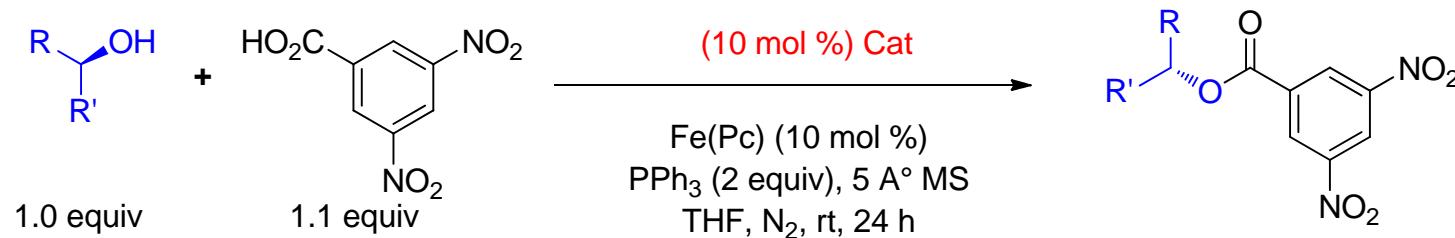
❖ Optimization of the hydrazines catalyst



	20 %	11 : 89 er
	30 %	87 : 13 er
	19 %	8 : 92 er
	37 %	80 : 20 er
	25 %	94 : 6 er
	48 %	92 : 8 er
	53 %	94 : 6 er
	23 %	85 : 15 er
	79 %	98 : 2 er

Good yield and good enantiomeric ratio

Scope of the new catalytic Mitsunobu reaction



✓ With inversion of configuration

Selected examples



Conclusion

- Innovative catalytic Mitsunobu reaction
- Inexpensive and nontoxic iron phthalocyanine
- Hydrazine catalyst easily prepared and fully recovered
- Limitation: Moderate reactivity and low yields

Further improvement need to be done