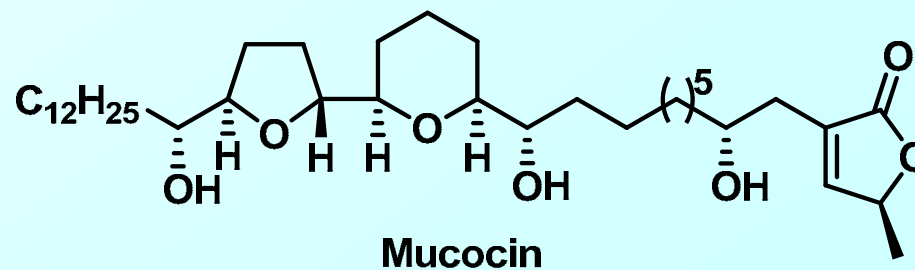
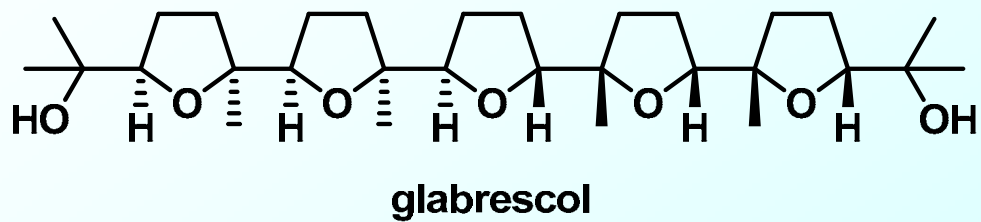
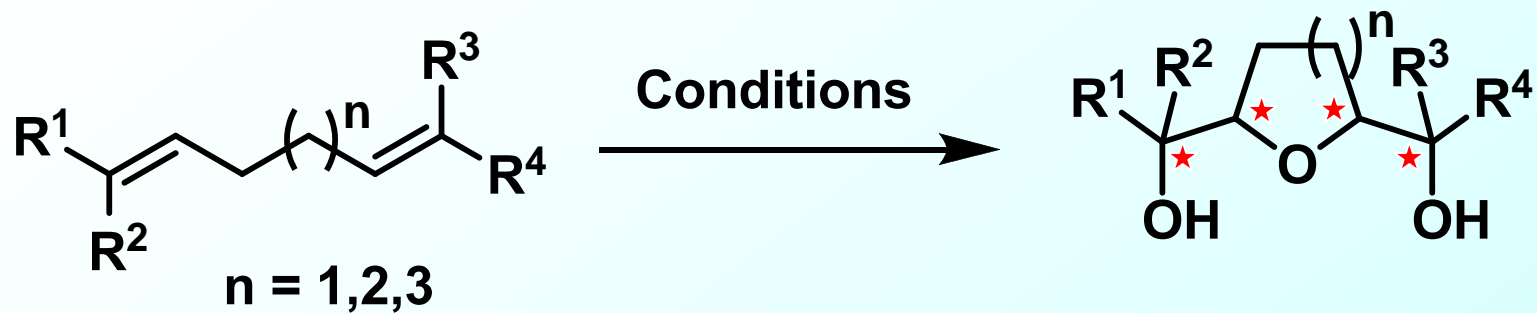


# Oxidative Cyclization of Dienes mediated by Transition-Metal-Oxo Species.

Mathieu Candy

Séminaire bibliographie STeRéO

# Introduction



# Contents

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## Introduction

### 1. Formation of Tetrahydrofurans

1.1. Permanganate mediated

1.2. Ruthenium mediated

1.3. Osmium mediated

1.4. Chromium mediated

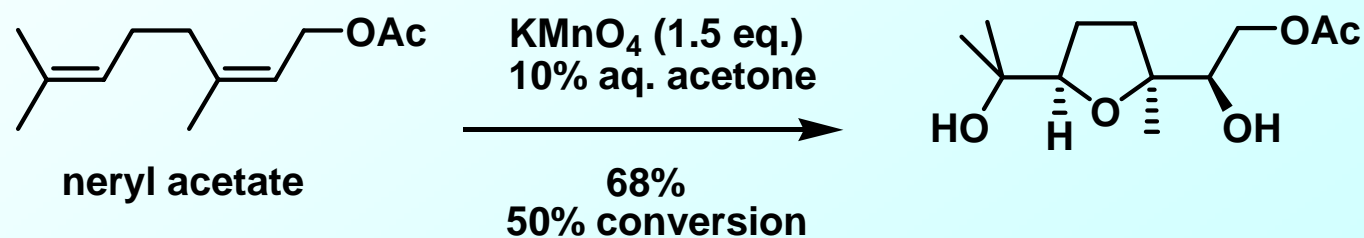
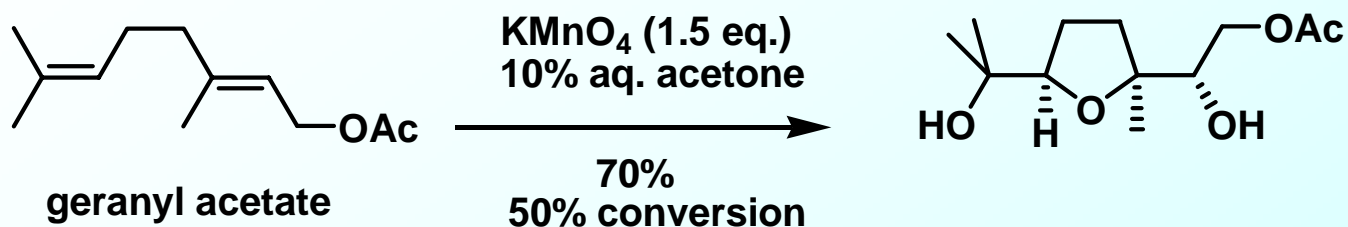
### 2. Formation of Tetrahydropyrans

2.1. Permanganate mediated

2.2. Ruthenium mediated

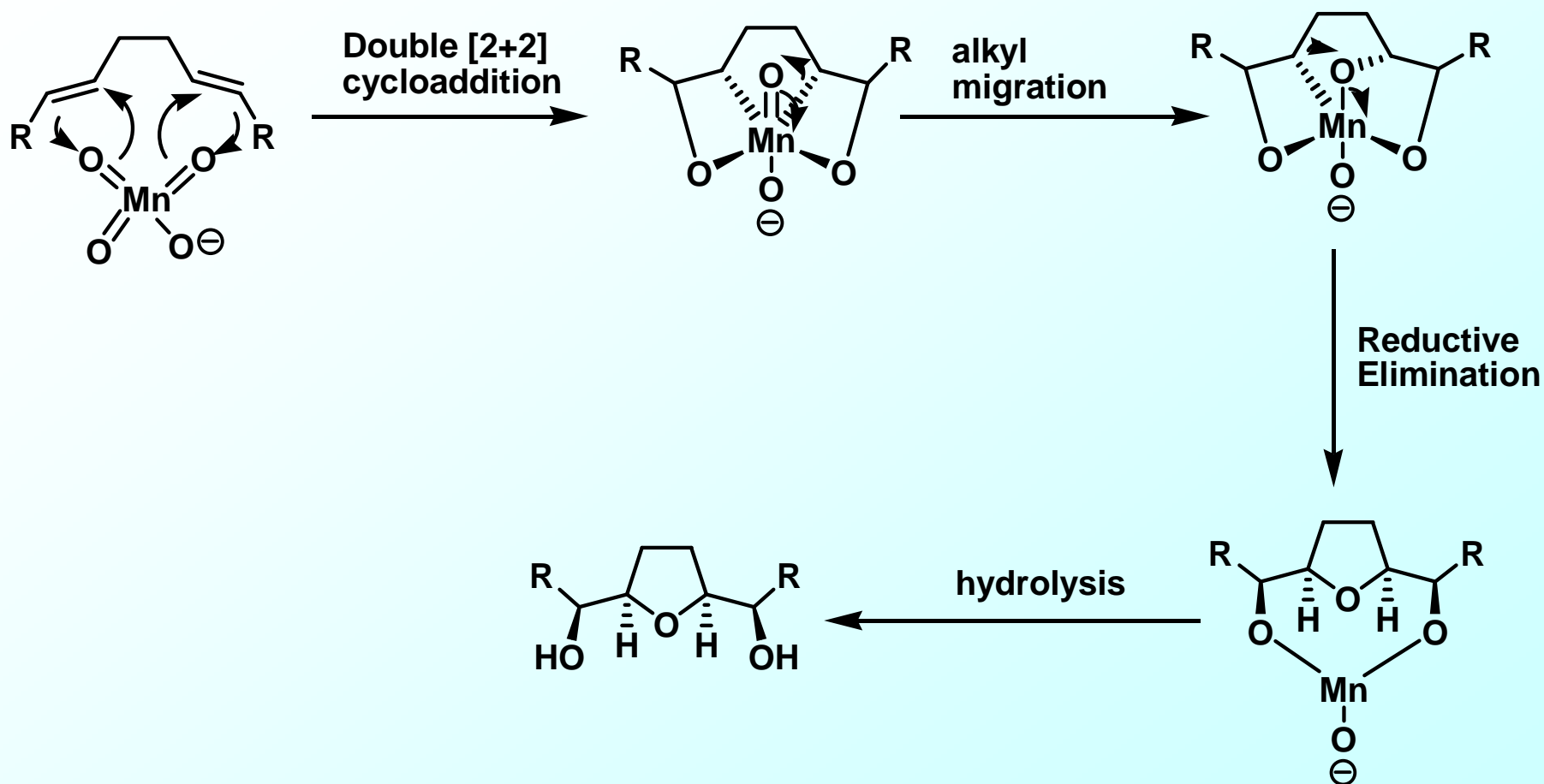
### 3. Formation of Oxepanes

# 1.1 Formation of Tetrahydrofurans using $\text{KMnO}_4$

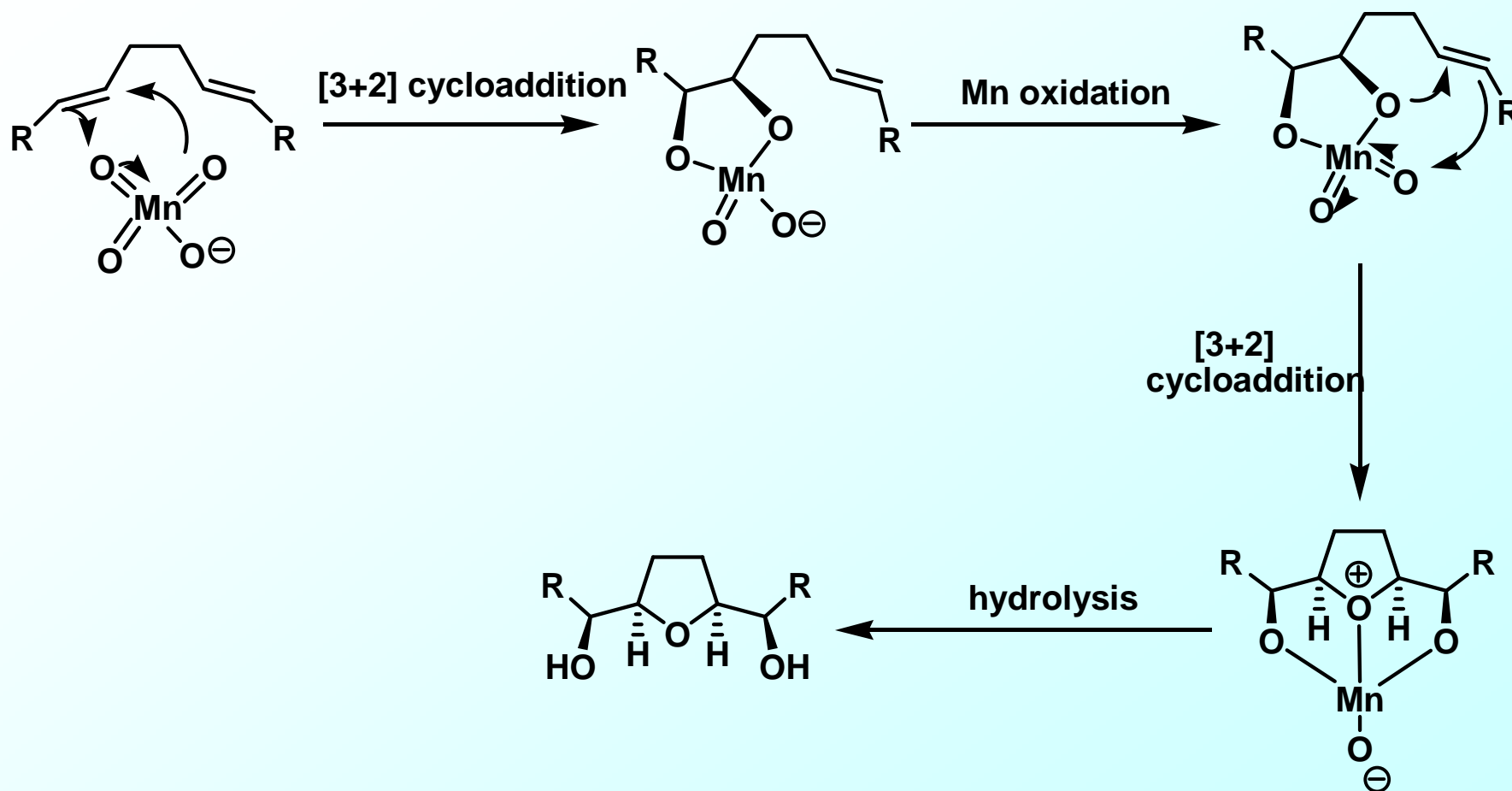


Klein, E.; Rojahn, W. *Tetrahedron* **1965**, 21, 2353.  
Kötz, A.; Steche, T. *J. Prakt. Chem.* **1924**, 107, 3353.

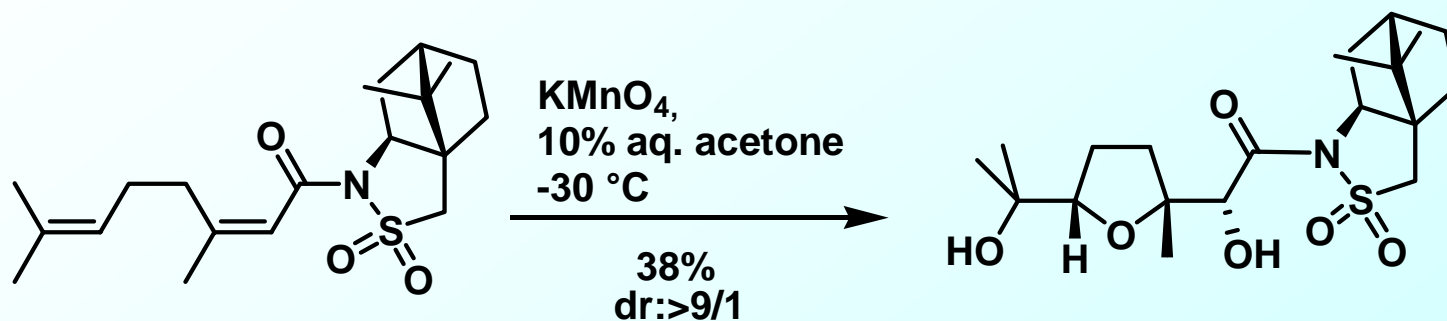
# 1.1 Formation of Tetrahydrofurans using $\text{KMnO}_4$



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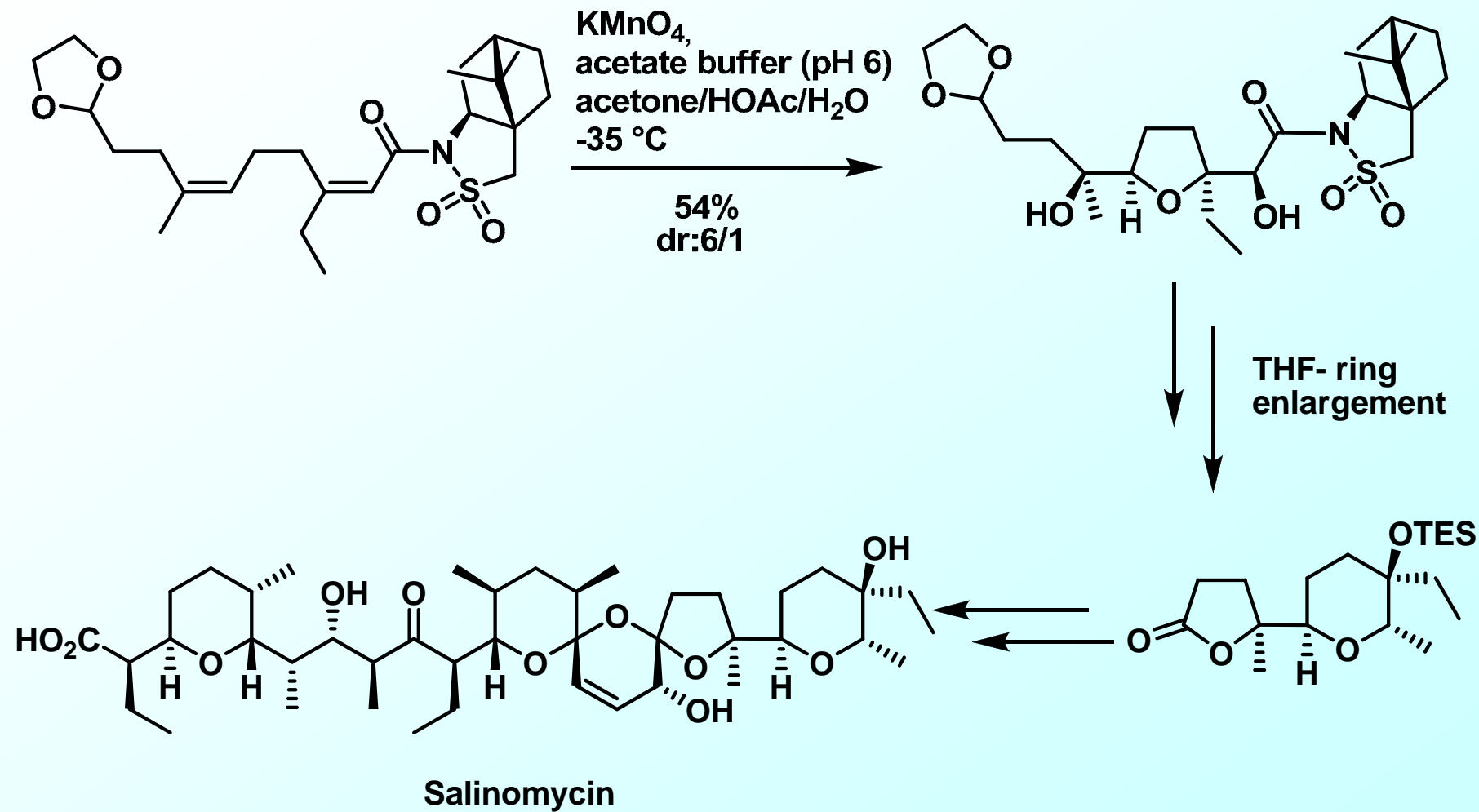


## 1.1 Formation of Tetrahydrofurans using $\text{KMnO}_4$



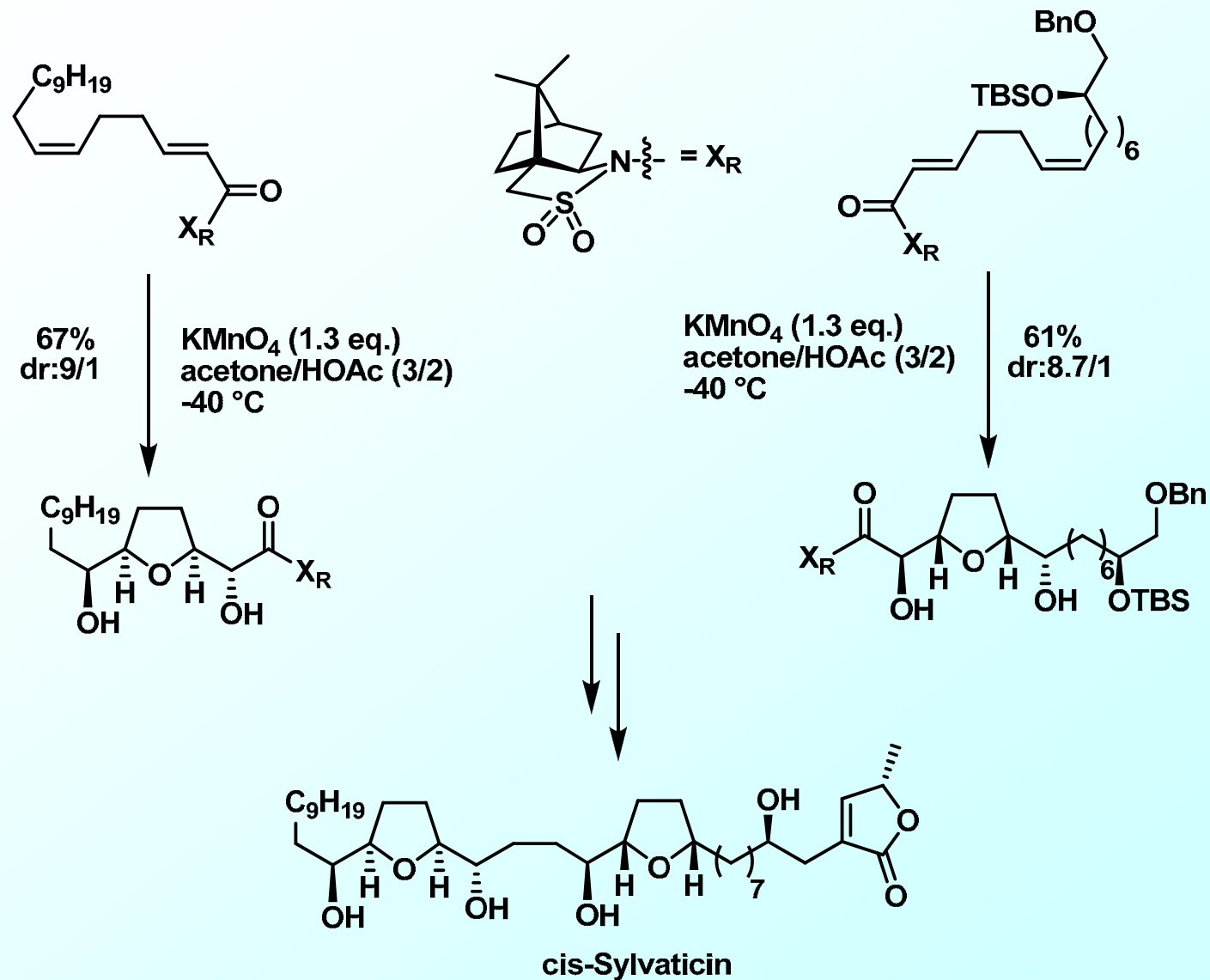
$\text{KMnO}_4$  : Attack on the electron-deficient double bond

# 1.1 Formation of Tetrahydrofurans using $\text{KMnO}_4$

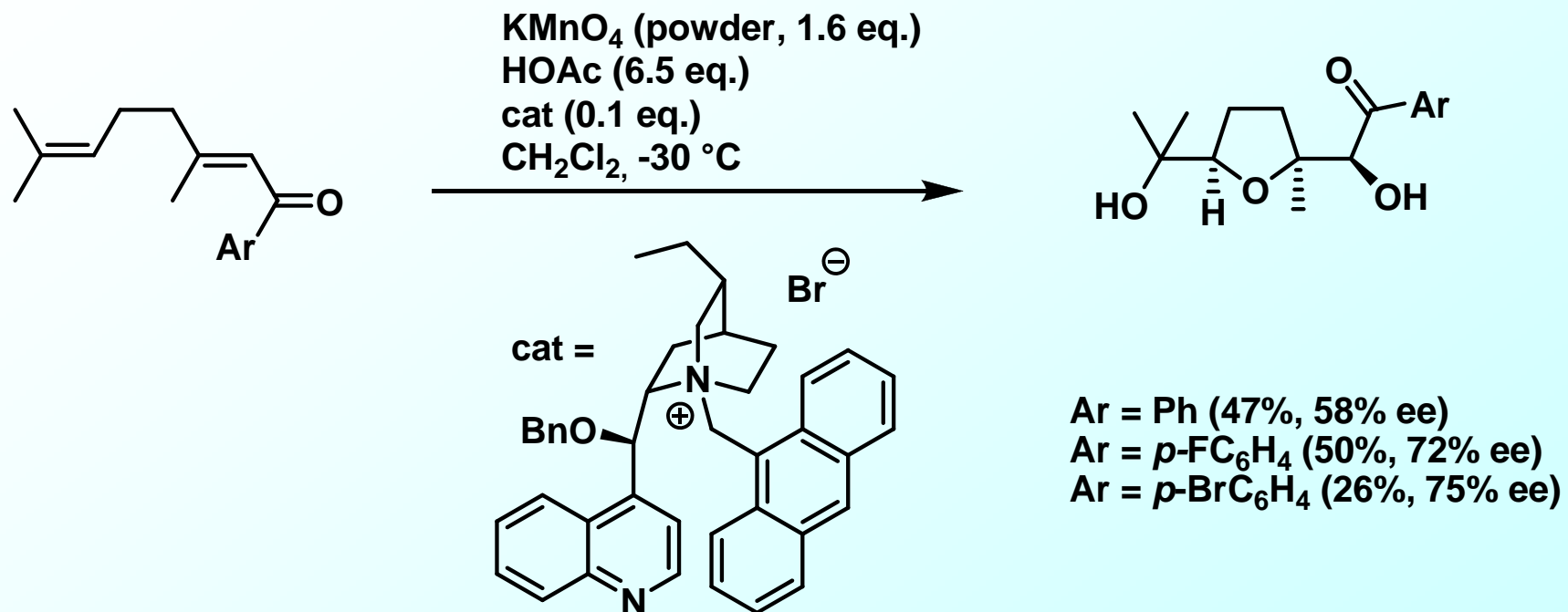




# 1.1 Formation of Tetrahydrofurans using $\text{KMnO}_4$

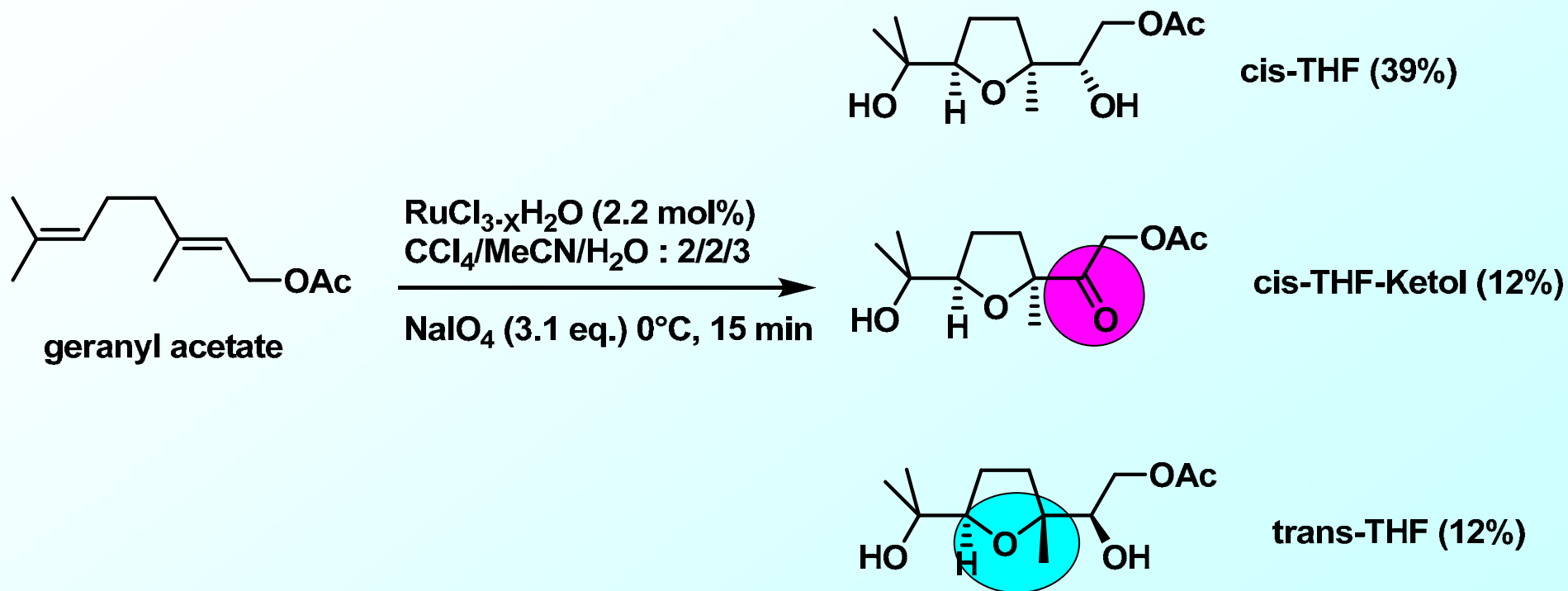


# 1.1 Formation of Tetrahydrofurans using $\text{KMnO}_4$

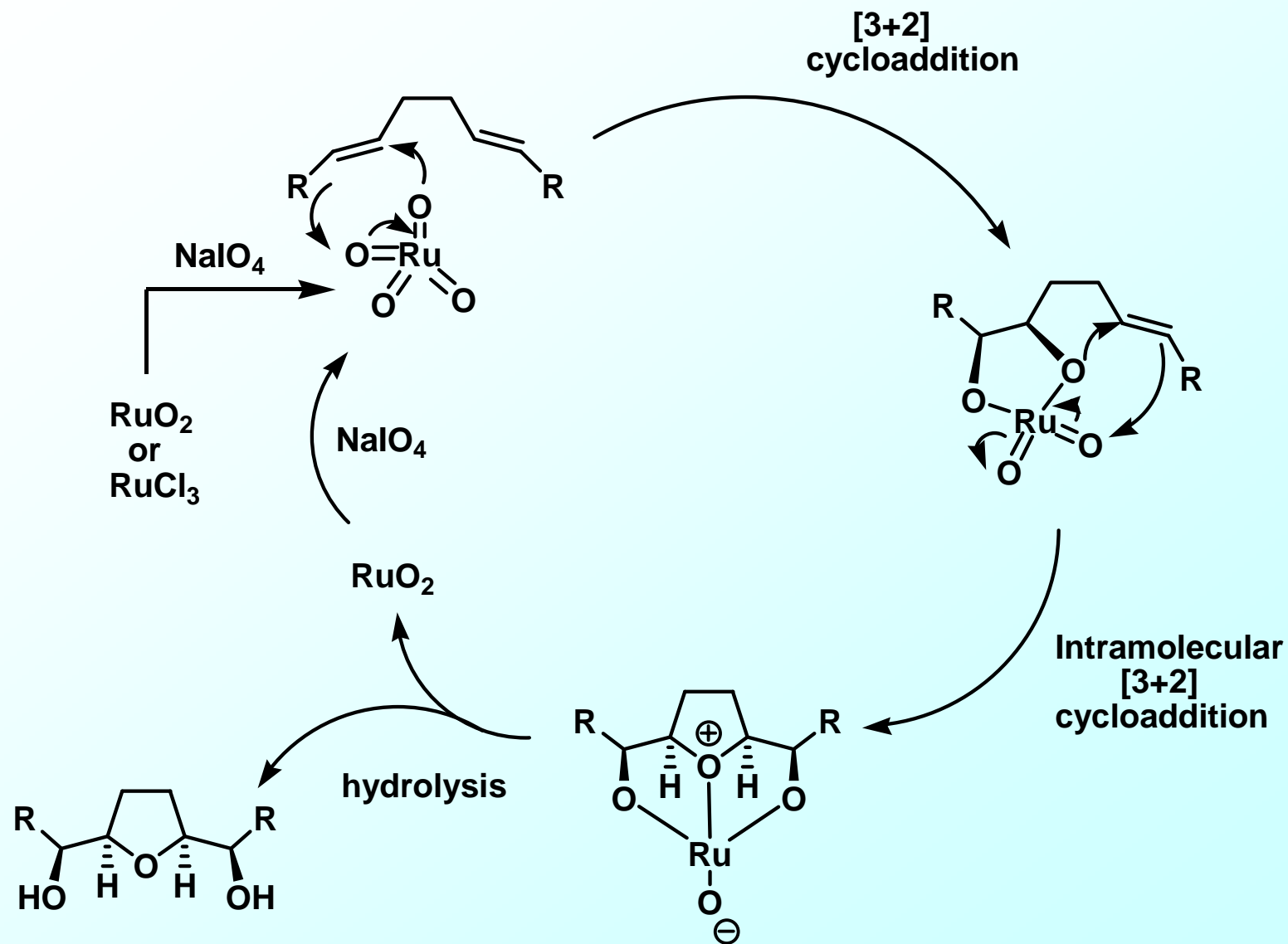


The unique example of enantioselective oxidative cyclization !!!

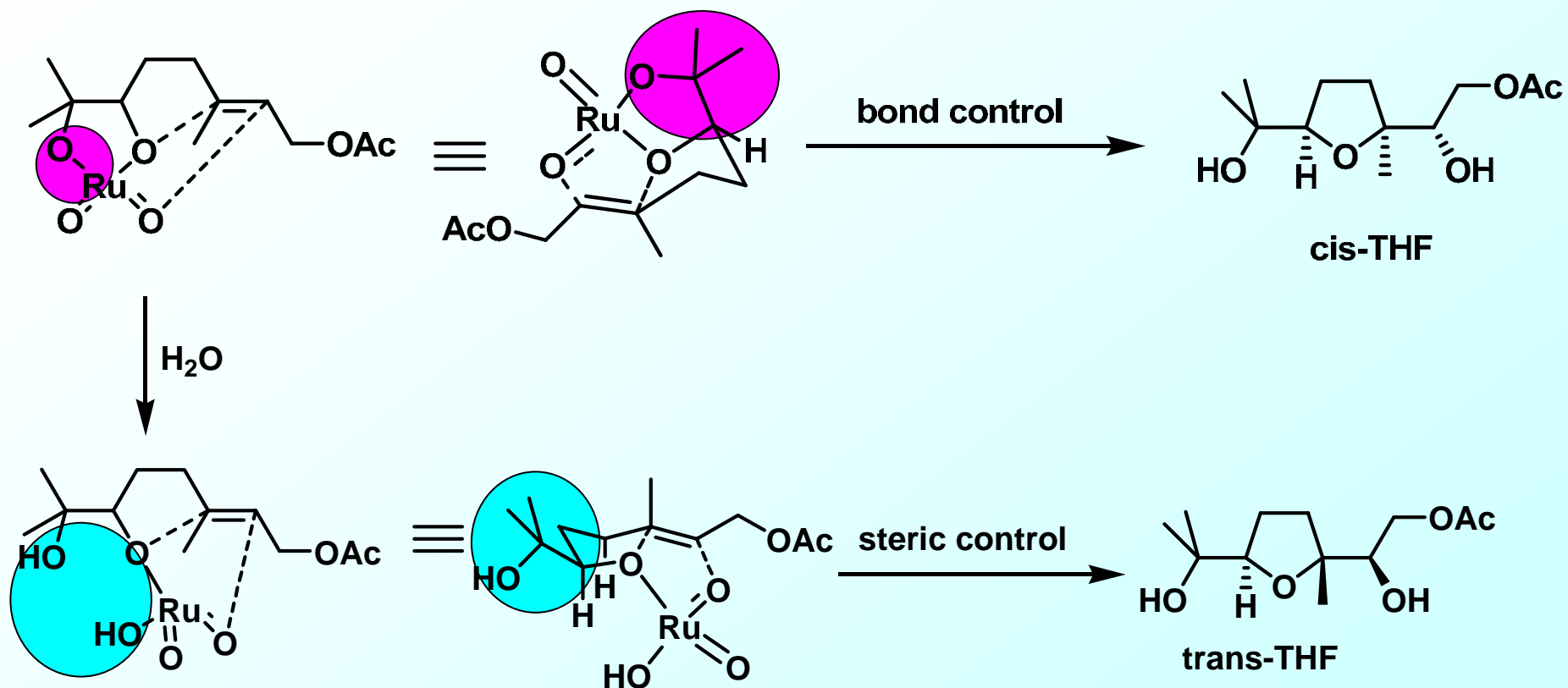
## 1.2 Formation of Tetrahydrofurans using Ru



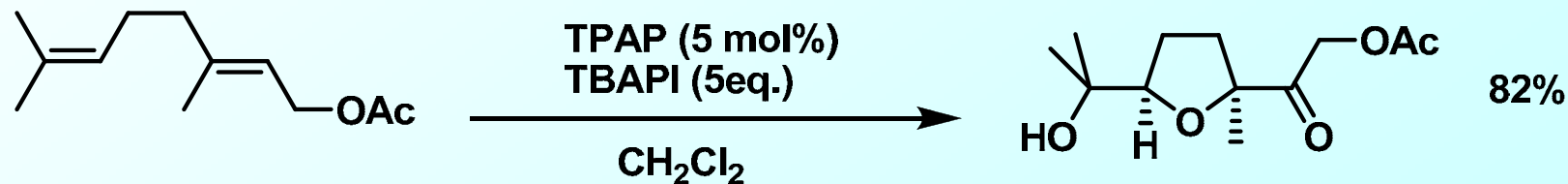
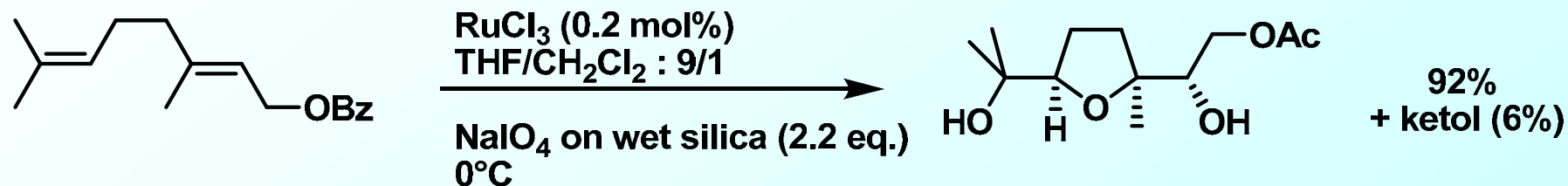
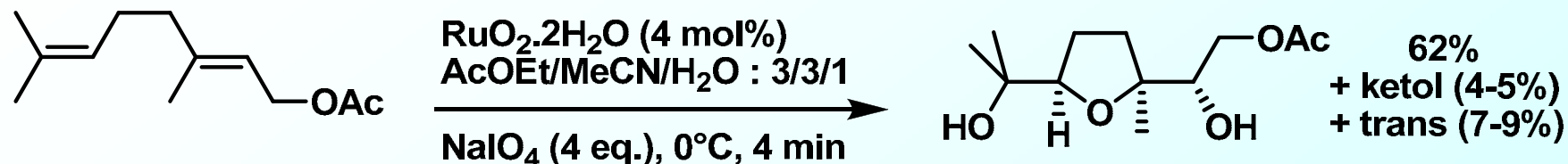
## 1.2 Formation of Tetrahydrofurans using Ru



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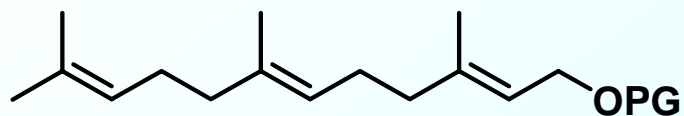


Piccialli, V. *et al. Tetrahedron Lett.* **2001**, 42, 4695.

Stark, C. B. W. *et al. Synthesis* **2007**, 2751.

Piccialli, V. *et al. Tetrahedron Lett.* **2004**, 45, 303.

## 1.2 Formation of Tetrahydrofurans using Ru

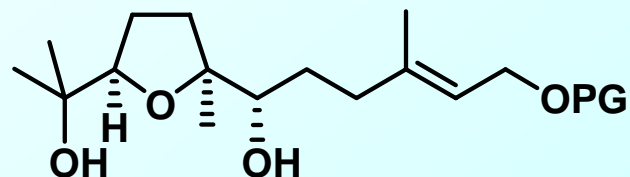


PG = TBDPS, Bz, P-NO<sub>2</sub>-Bz, Ac

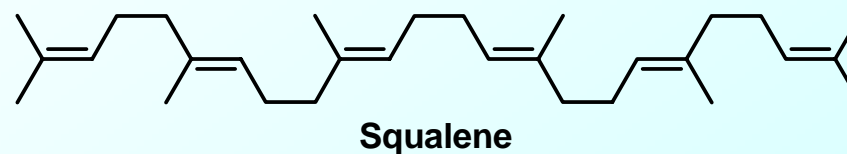
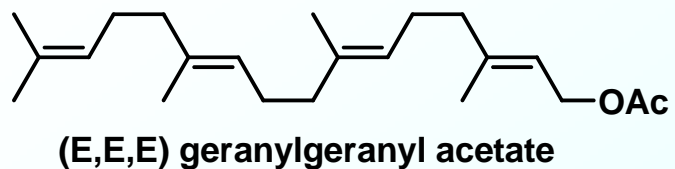
RuCl<sub>3</sub> (1 mol%)  
NaIO<sub>4</sub> on wet silica (3 eq.)  
THF, RT

47-67%  
+ 12-17% ketol

Ru : Attack on the  
electron-rich double bond

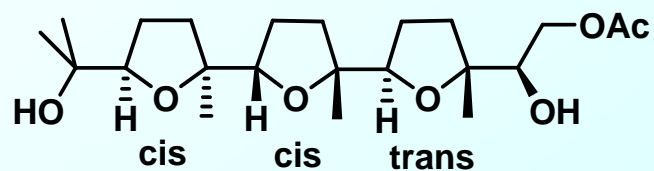


## 1.2 Formation of Tetrahydrofurans using Ru



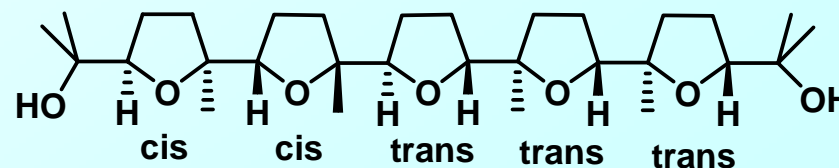
RuO<sub>2</sub>·2H<sub>2</sub>O (20 mol%)  
AcOEt/MeCN/H<sub>2</sub>O : 3/3/1  
NaIO<sub>4</sub> (6 eq.), 0°C, 30 min

30%



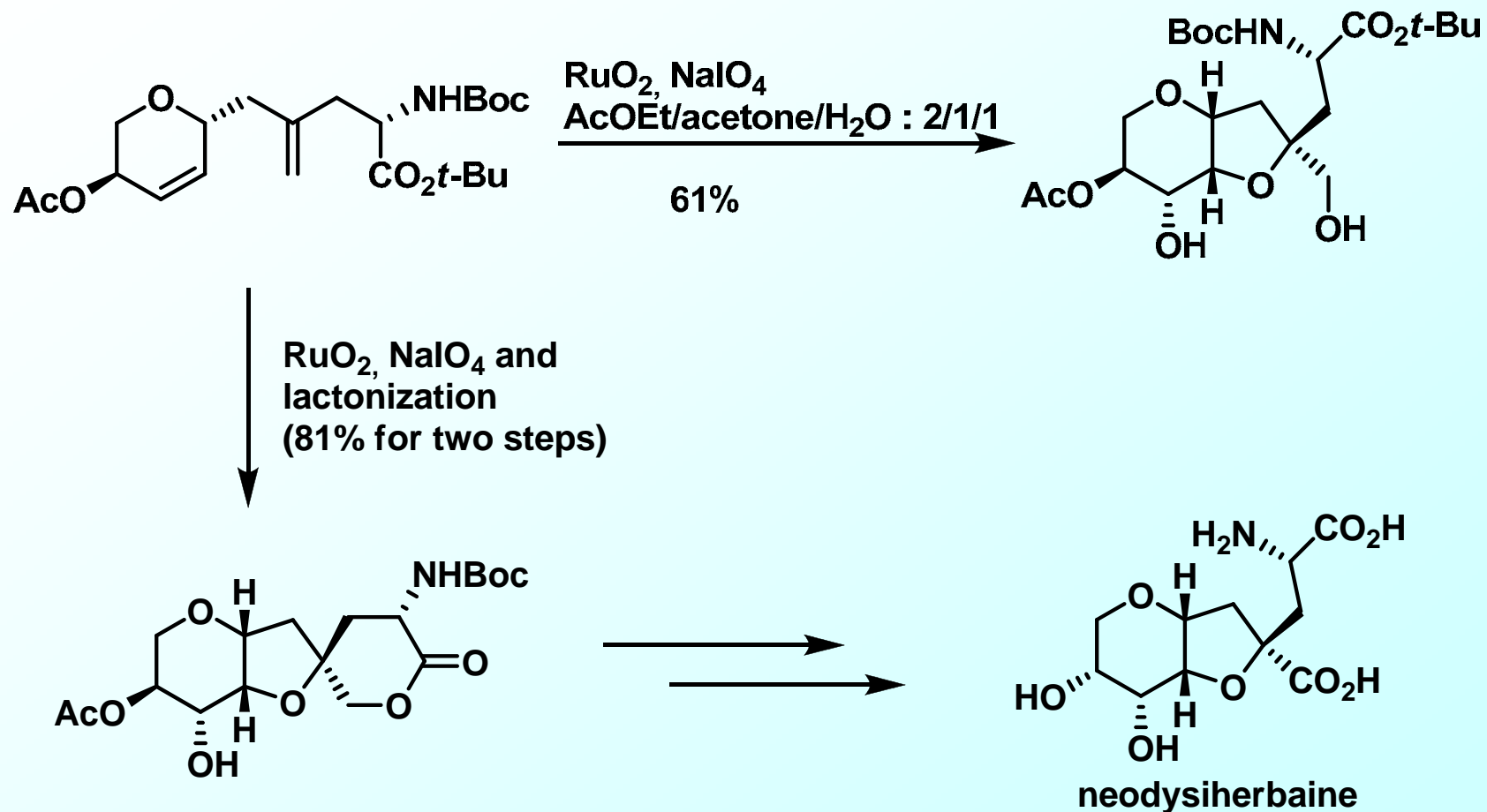
RuO<sub>2</sub>·2H<sub>2</sub>O (20 mol%)  
AcOEt/MeCN/H<sub>2</sub>O : 3/3/1  
NaIO<sub>4</sub> (8 eq.), 0°C, 30 min

45%

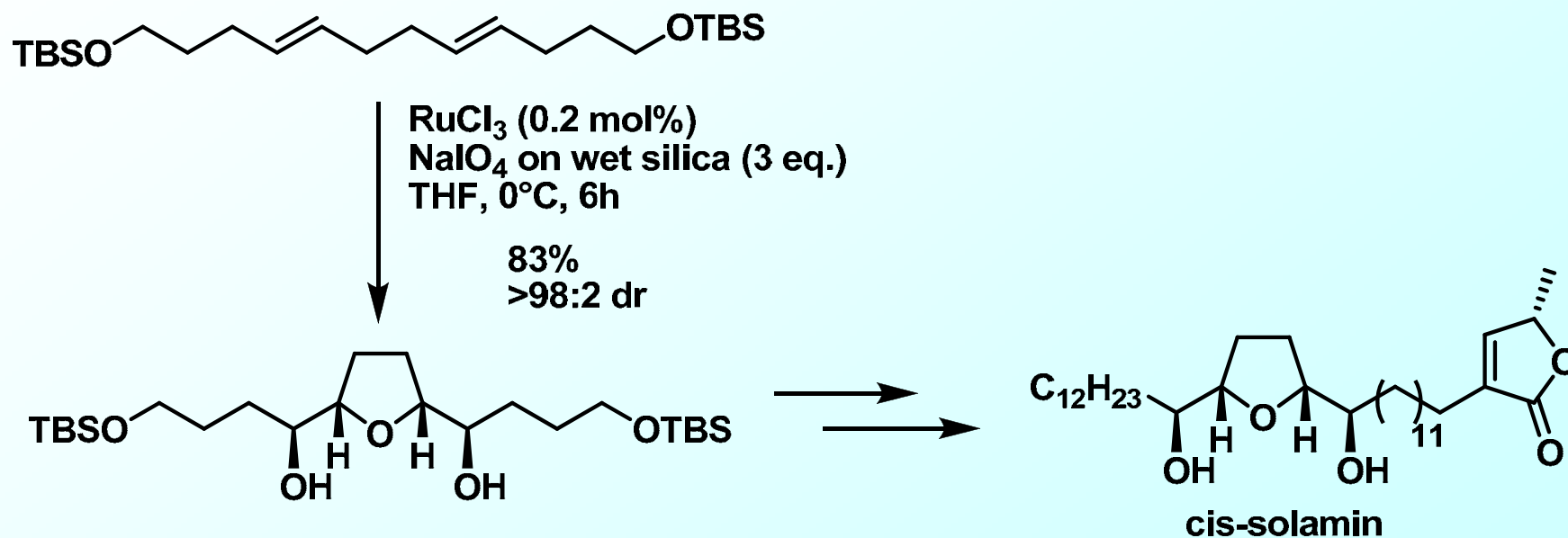




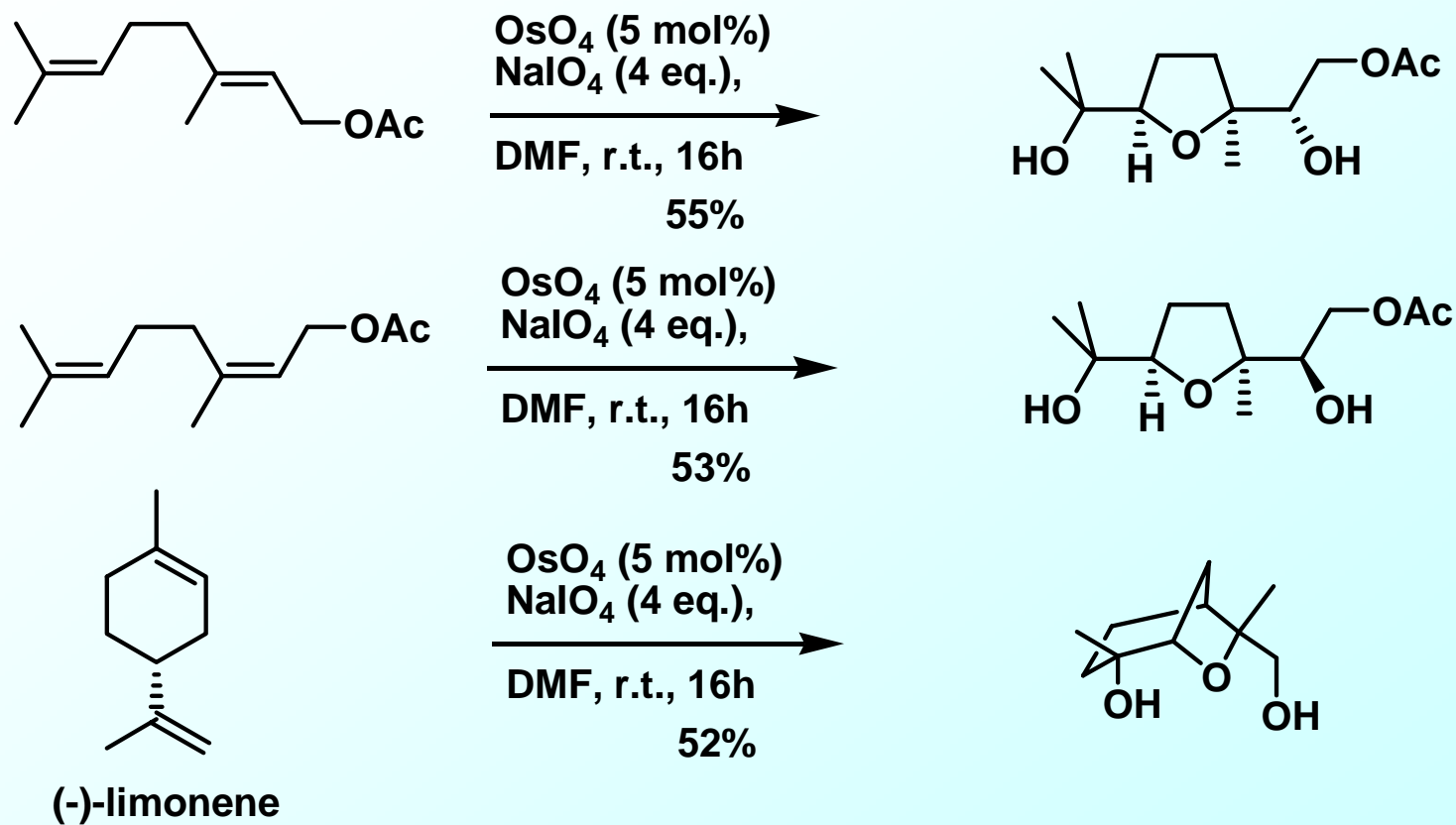
## 1.2 Formation of Tetrahydrofurans using Ru



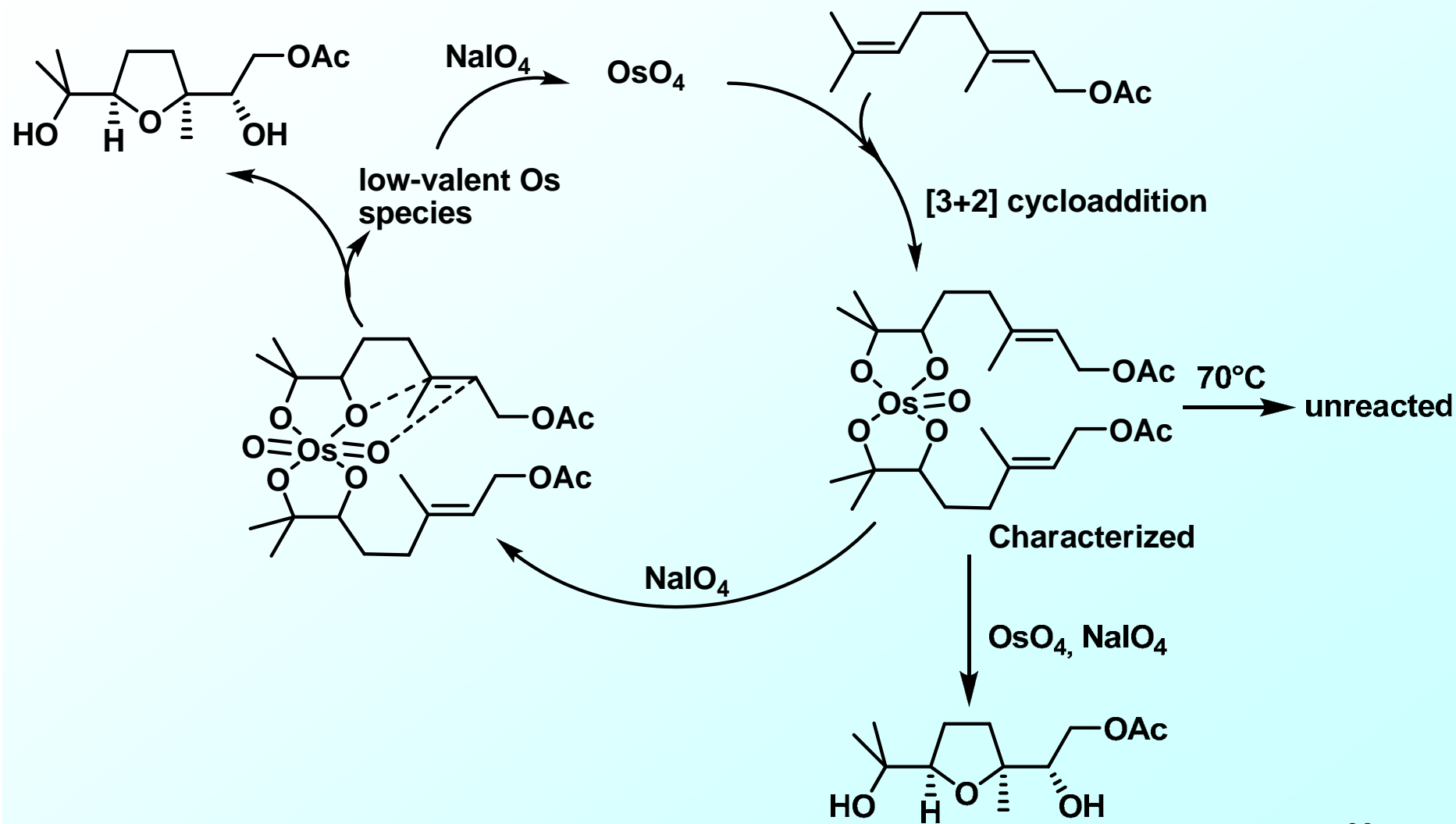
## 1.2 Formation of Tetrahydrofurans using Ru



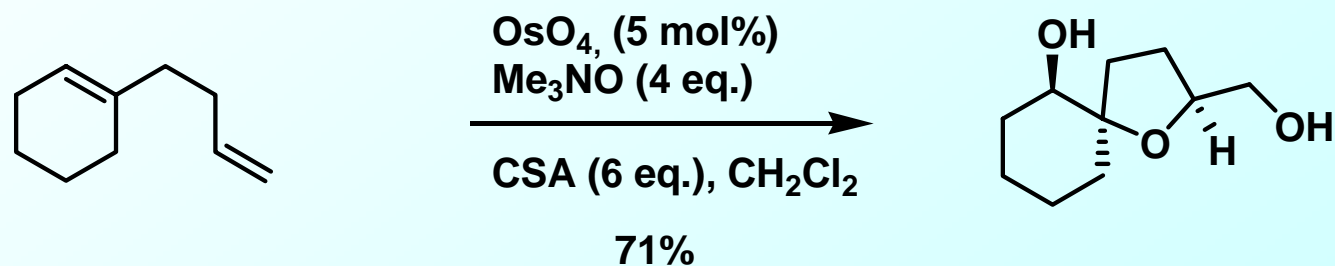
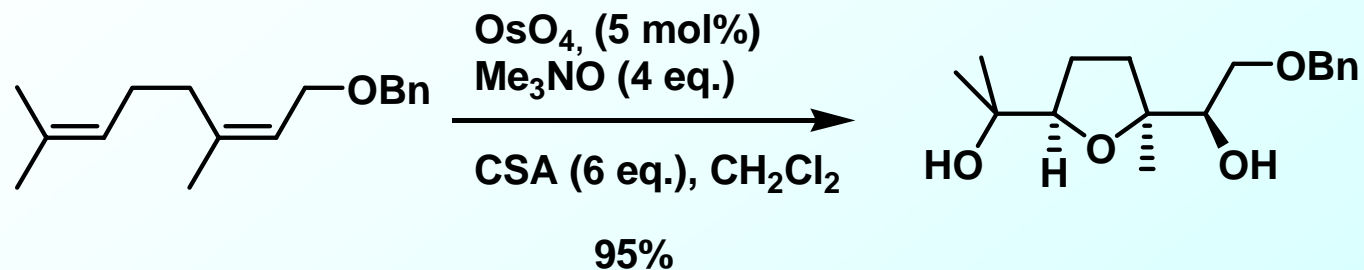
## 1.2 Formation of Tetrahydrofurans using Os



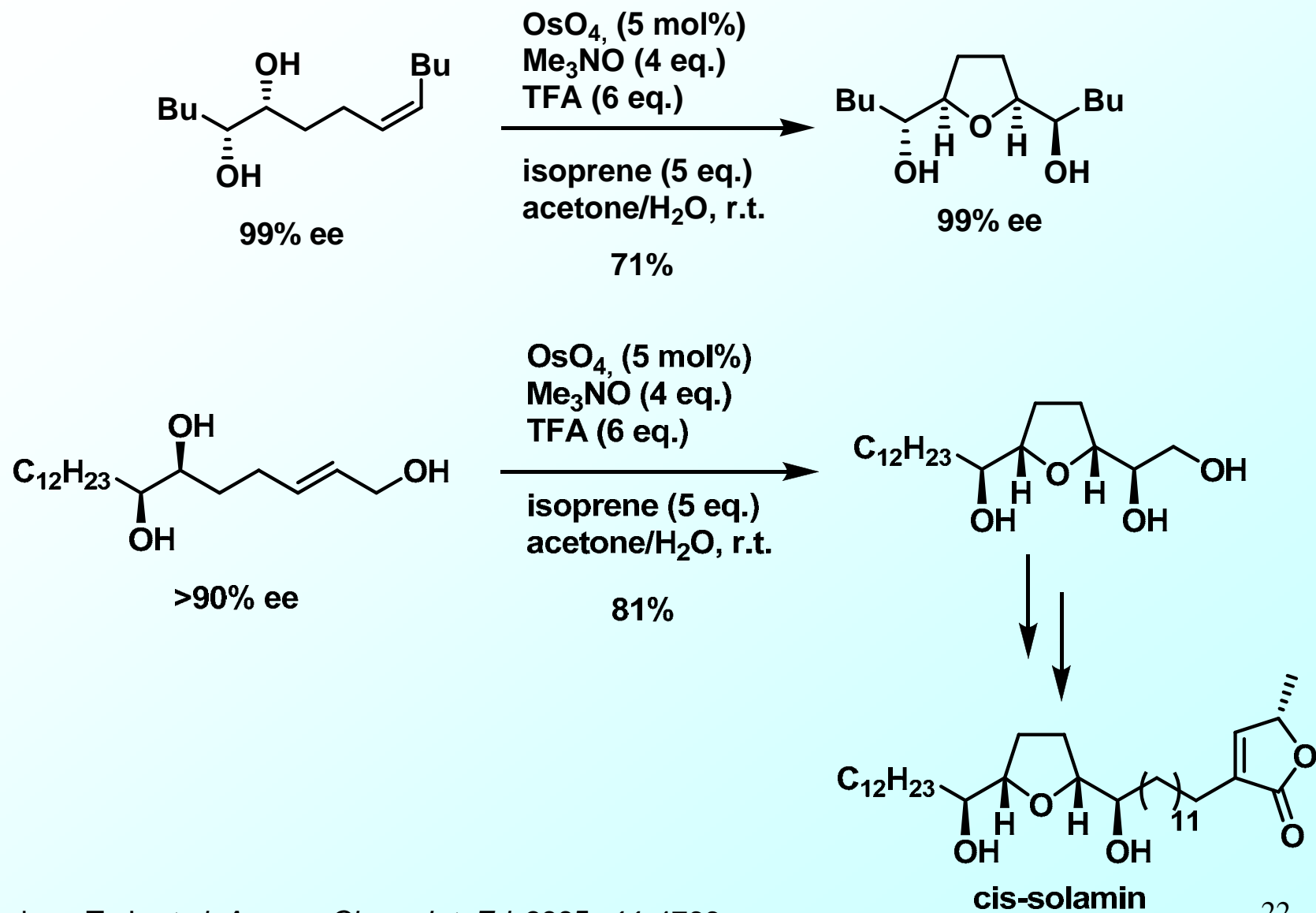
## 1.2 Formation of Tetrahydrofurans using Os



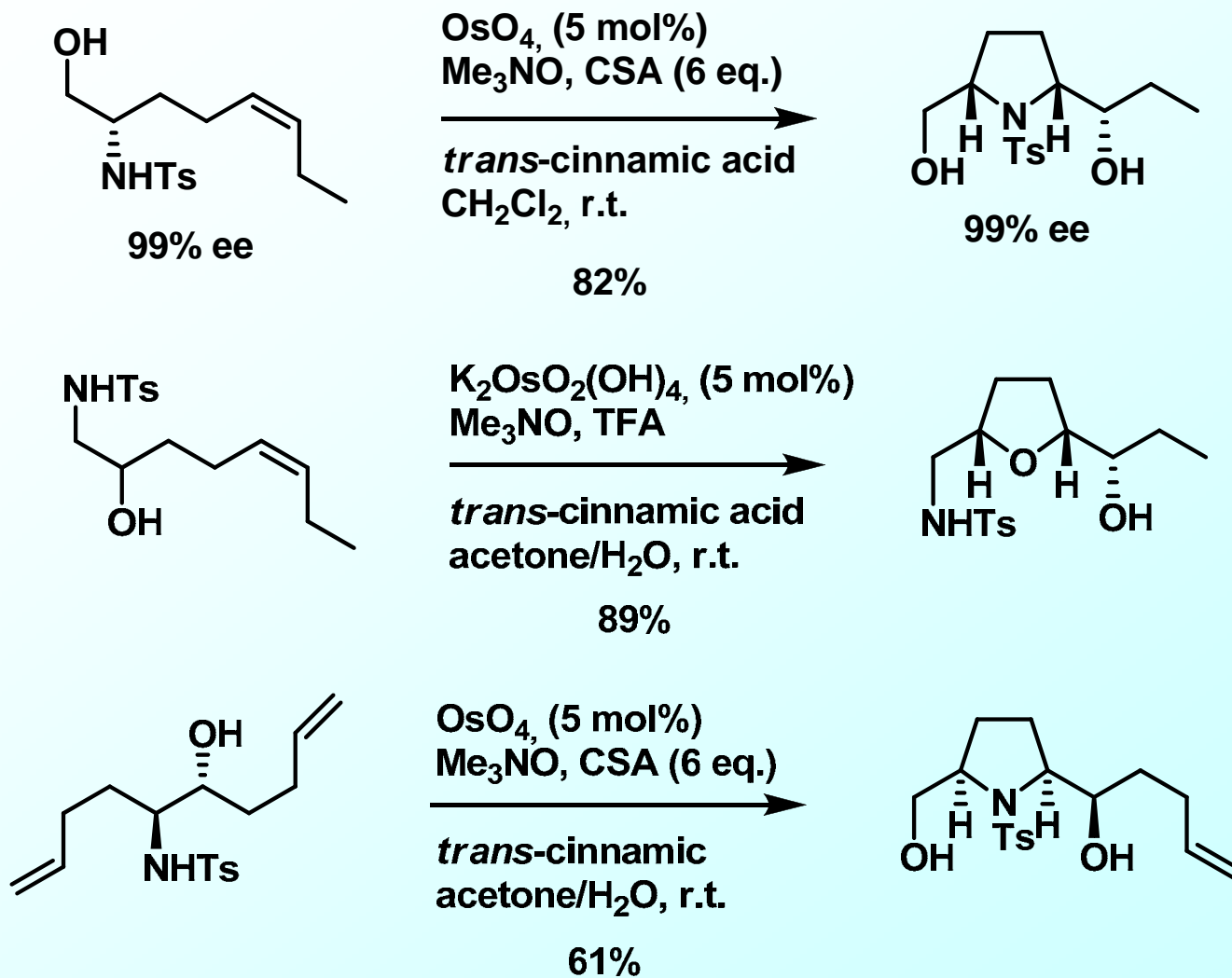
## 1.2 Formation of Tetrahydrofurans using Os



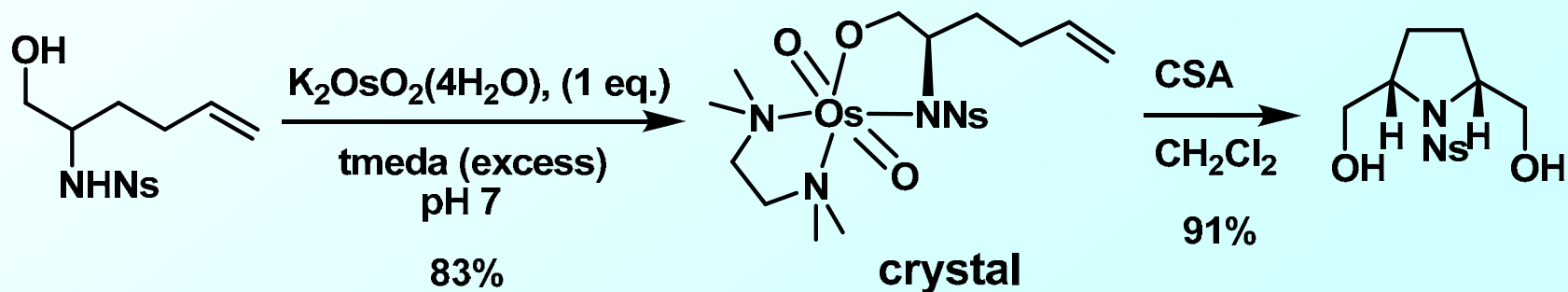
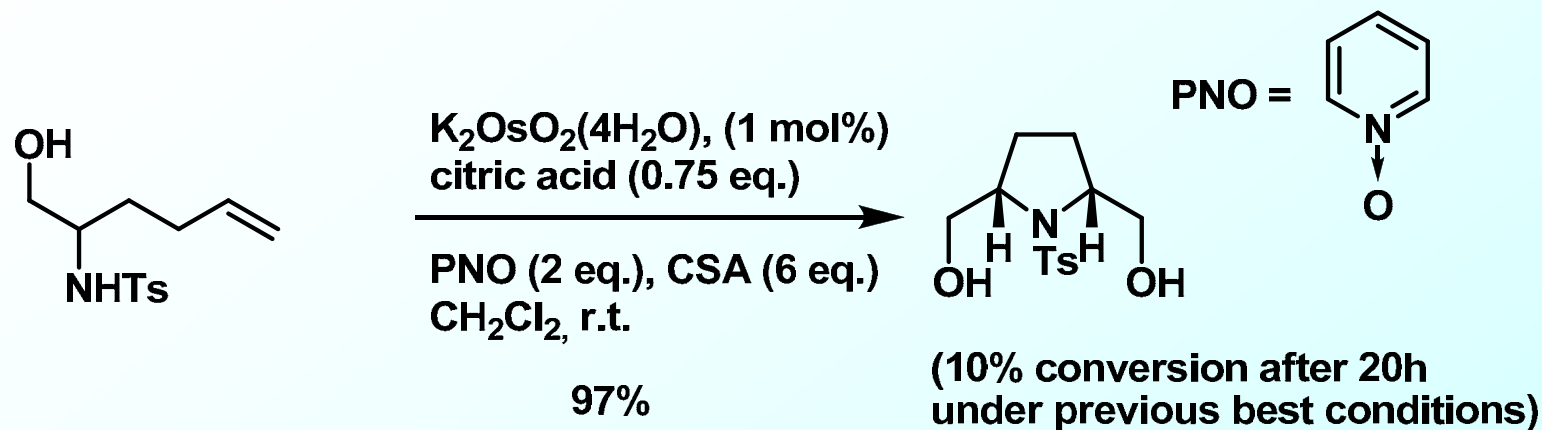
## 1.2 Formation of Tetrahydrofurans using Os



## 1.2 Formation of Pyrrolidines using Os

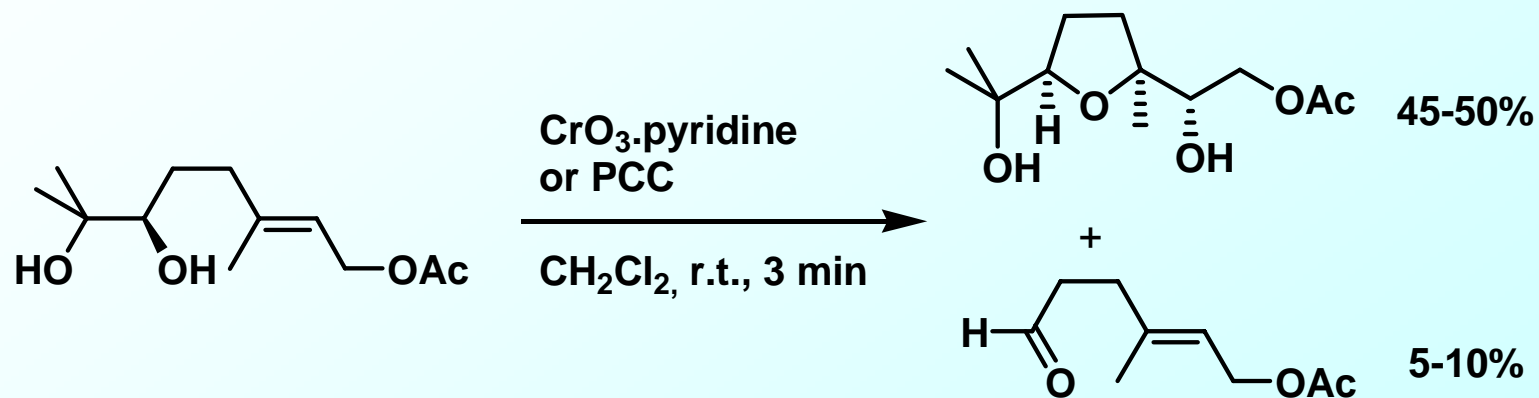


## 1.2 Formation of Pyrrolidines using Os



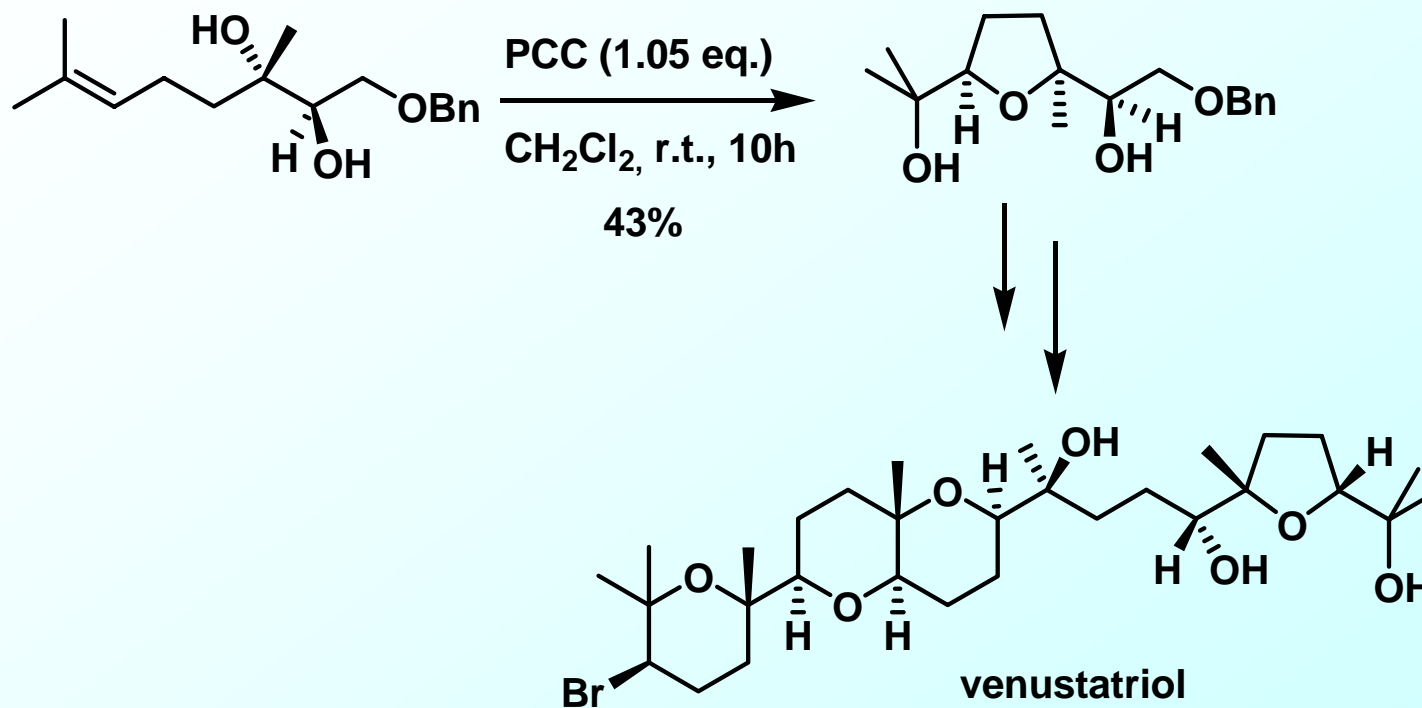


## 1.2 Formation of Tetrahydrofurans using Cr

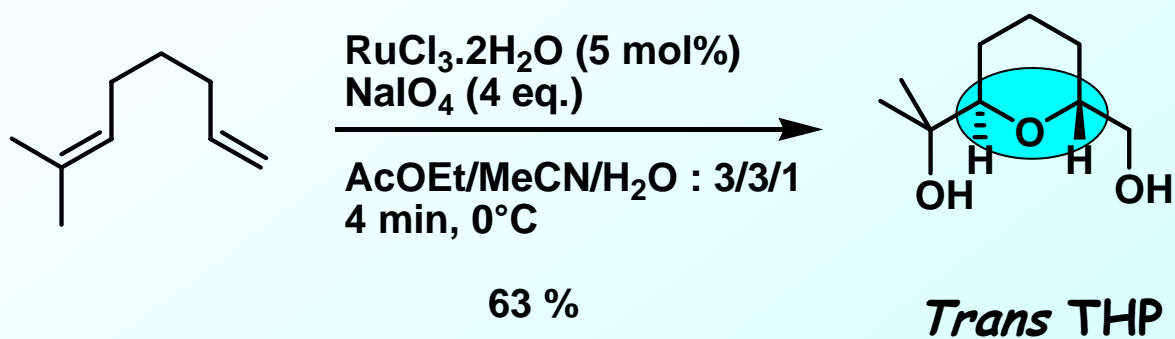


Casida, J. E. *et al. J. Agric. Food Chem.* **1974**, *22*, 379.  
Walba, D. M. *et al. Tetrahedron Lett.* **1982**, *23*, 727.

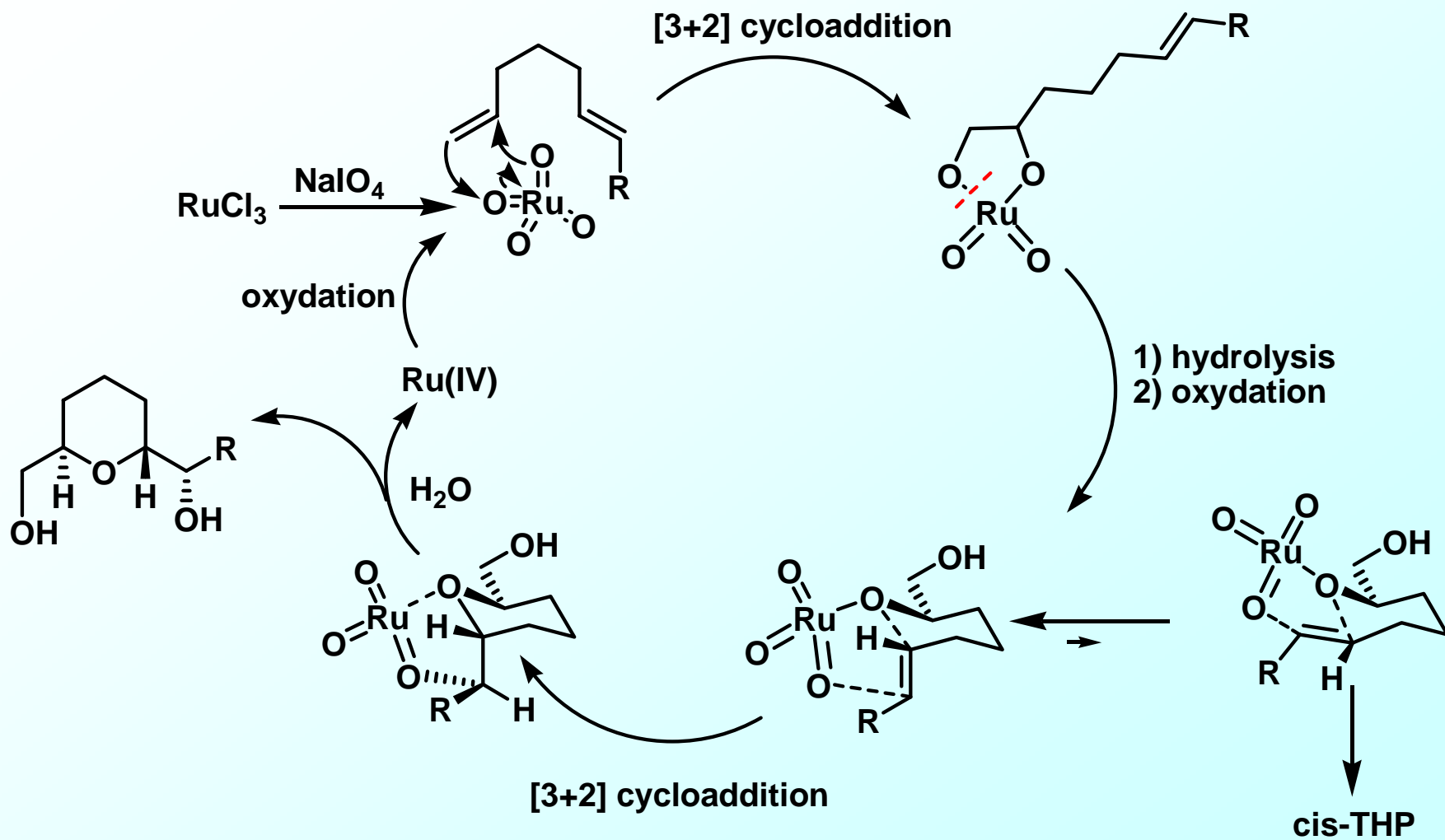
## 1.2 Formation of Tetrahydrofurans using Cr



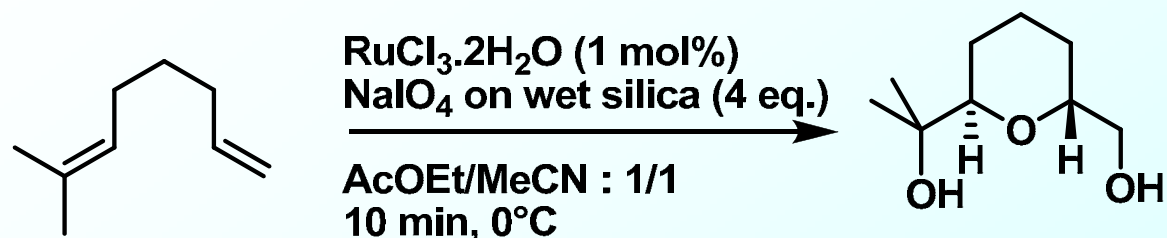
## 2.1 Formation of Tetrahydropyrans using Ru



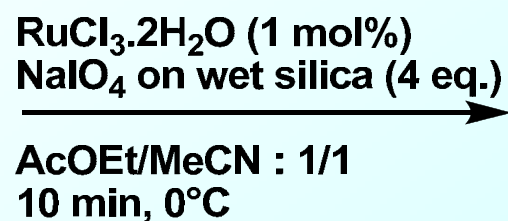
## 2.1 Formation of Tetrahydropyrans using Ru



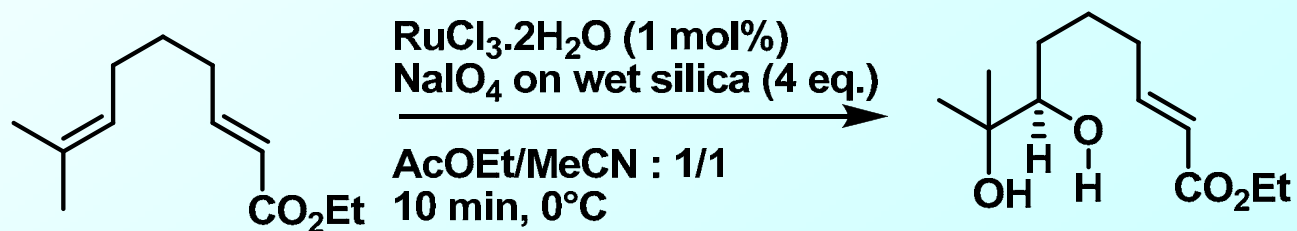
## 2.1 Formation of Tetrahydropyrans using Ru



85 %

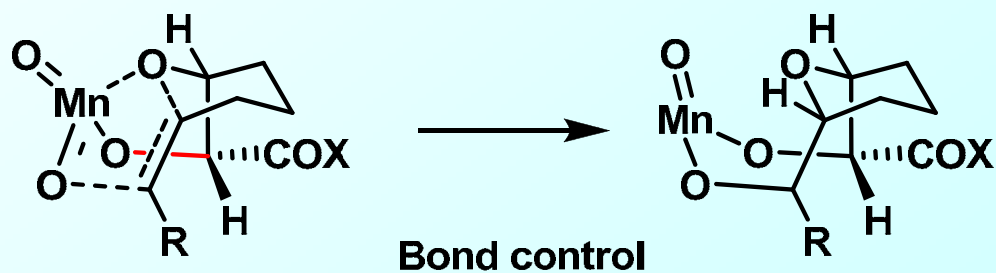
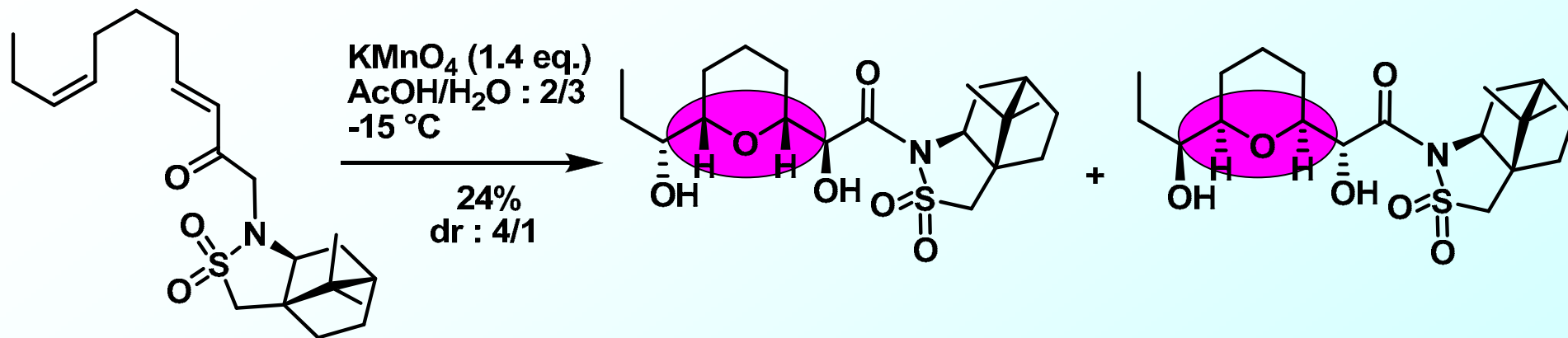


60 %

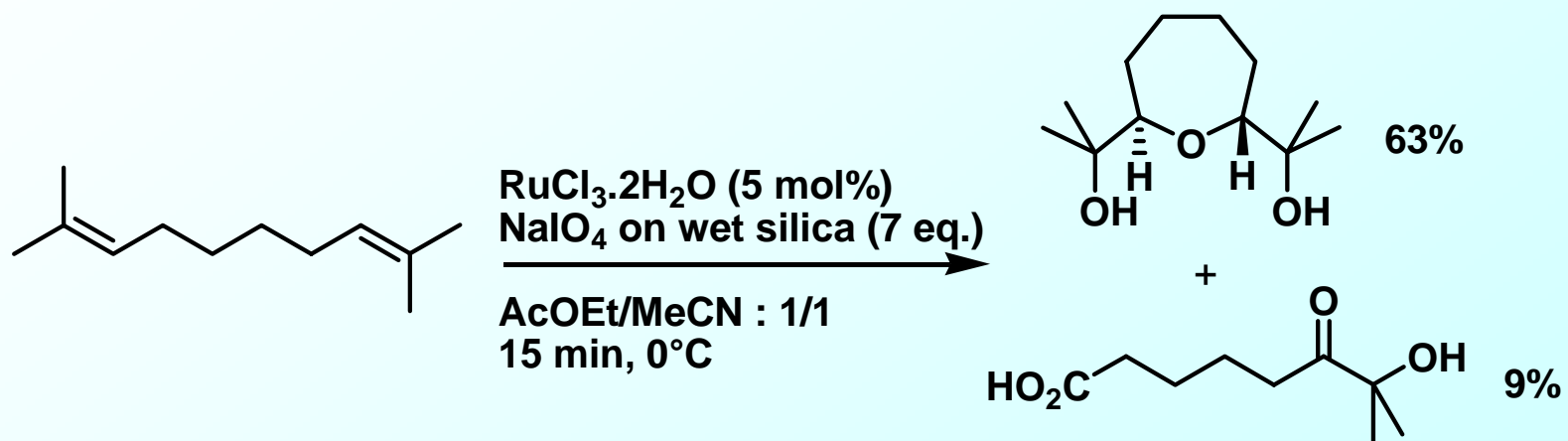


60 %

## 2.2 Formation of Tetrahydropyrans using $\text{KMnO}_4$



### 3 Formation of Oxepanes



## Conclusion

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- Stereospecific and highly stereoselective process
- *Cis* THF
- *Cis* (KMnO<sub>4</sub>) + *trans* (Ru) THP
- *Trans* oxepanes
- Only one example of enantioselective cyclization